

The effects of the initial gas conditions on the properties of the dense gas in GMCs

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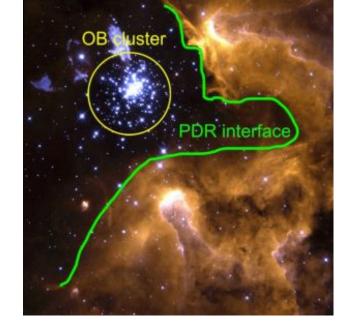


Key question(s)

- How do the initial physical and chemical conditions affect the chemistry of the gas that is/will be forming stars?
- How important *for the chemistry* is the mode of collapse of dense gas?

Many papers investigating the transition from atomic to molecular hydrogen $(H \rightarrow H_2)$ and from ionized to atomic to molecular carbon (C+

- →C→CO) [Glover & McLow (2011); Offner et al.
- (2013);, Bisbas et al. (2015); Bialy & Sternberg 2016 etc
- see Bolatto et al. 2013 review].
- → These transitions occurs at very low Avs (depending on several parameters (e.g. Bell et al. 2006)
- → let's look deeper into the gas at higher Av (>3 mags).



Some of the influencing factors:

Credit: Wolfgang Brandner (JPL/IPAC), Eva K. Grebel (Univ. Washington), You-Hua Chu (Univ. Illinois Urbana-Champaign), and NASA

- The H/H₂
- Initial elemental abundances
- "Dynamics"
- Time
- Flux(es)

Caveat: these are **not** independent parameters → we shall however look at them individually first but see later.



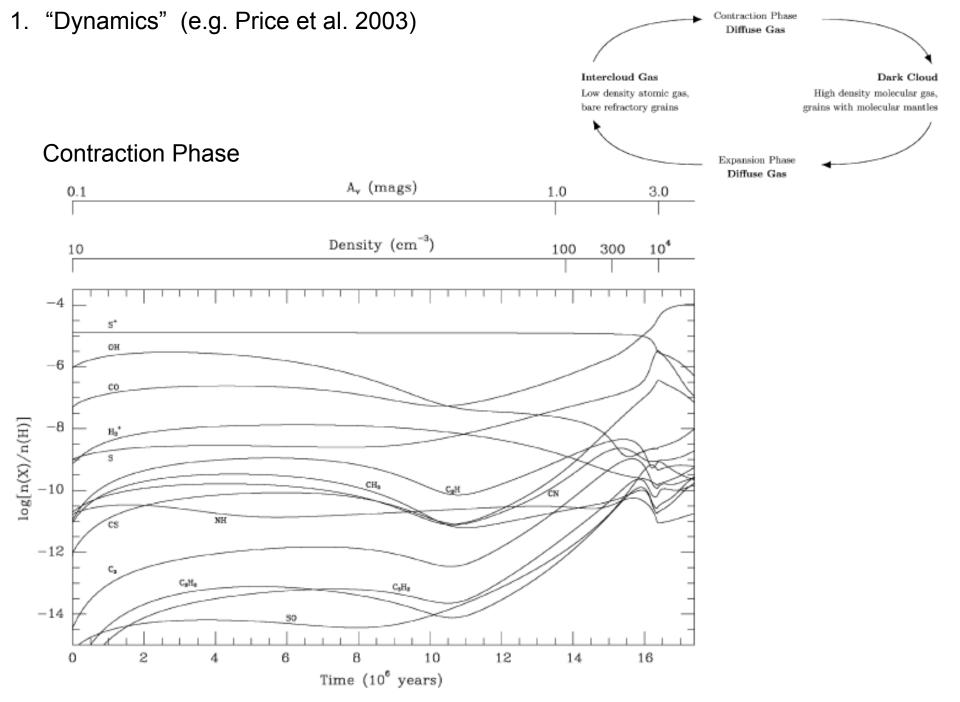


Figure 2. The fractional abundance of a selection of species relative to hydrogen over time, during the collapse phase.

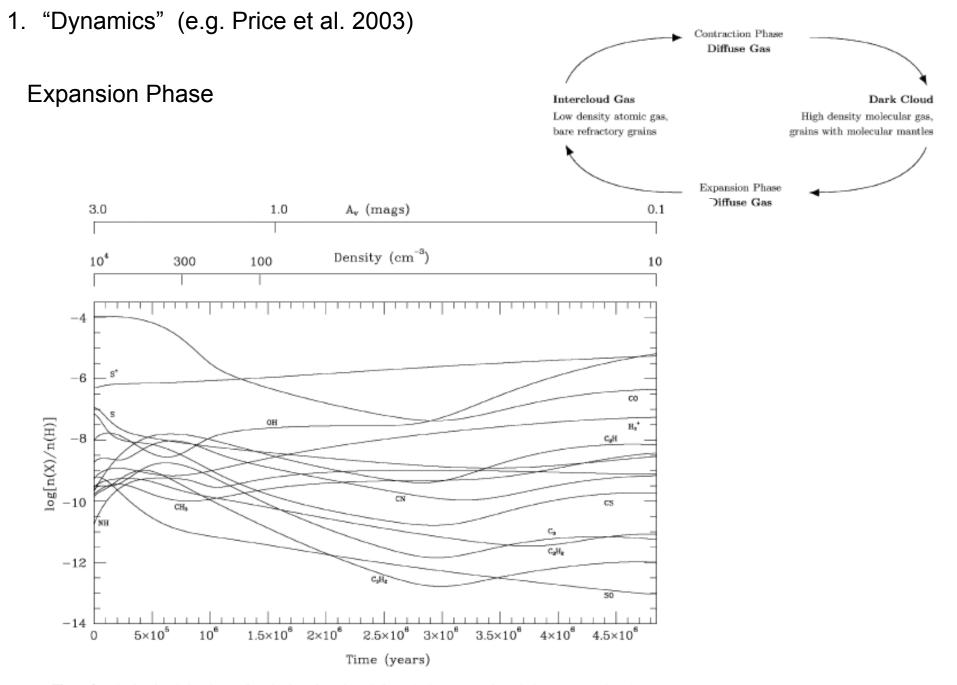


Figure 3. The fractional abundance of a selection of species relative to hydrogen over time, during the expansion phase.

Are there observational evidence of either 'direction' in the diffuse clouds?

Table 2. Column densities of species at a spatial density of 100 cm^{-3} . Observed column densities for ζ Per have been taken from the literature, where available. Observed column densities enclosed in brackets represent observations made in other lines of sight; some are discussed in the text. The notation a(b) represents $a \times 10^b$.

Species	Observed	Collapse	Expansion
H ₂	3.2(20)-7.1(20)a	8.2(+20)	8.5(+20)
S	$1.8(13)^b$	8.1(+13)	7.1(+12)
O	_	7.4(+17)	5.3(+17)
C	_	1.7(+16)	8.2(+15)
S^+	$1.6(16)^{c}$	2.1(+16)	1.8(+15)
CS	$[0.5(12)-2(12)]^d$	8.2(+11)	3.5(+11)
SO	$[1.0(12)-1.0(13)]^d$	2.0(+09)	7.0(+09)
CO	$6.1(14)^b$	8.7(+14)	1.0(+15)
C_2H	$[7.0(12)]^{e}$	1.9(+12)	6.7(+12)
C_3H_2	$[4.8(12)]^e$	5.5(+09)	3.1(+10)
NH	$1.0(12)^f$	5.0(+10)	8.6(+11)
CN	$3.0(12)^g$	4.0(+11)	2.5(+12)
OH	$4.0(13)^g$	3.6(+13)	4.0(+13)
H^{+}_{3}	$8.0(14)^h$	2.2(+12)	3.8(+12)
C ₃	_	6.4(+10)	1.8(+11)
HCO^{+}	_	1.5(+11)	2.2(+11)
HCN	_	3.7(+09)	6.0(+10)

^aSavage et al. (1977), ^bSnow (1977), ^cSnow, Lamers & Joseph (1987), ^dLucas & Liszt (2002), ^eLucas & Liszt (2000), ^fMeyer & Roth (1991), ^gFelenbok & Roueff (1996), ^hMcCall et al. (2003).

Table 3. Column densities of species at a spatial density of 300 cm⁻³. Observed column densities for ζ Oph have been taken from the literature, where available. Observed column densities enclosed in brackets represent observations made in other lines of sight; some are discussed in the text. The notation a(b) represents a \times 10^b.

Species	Observed	Collapse	Expansion
H ₂	4.6(20) ^a	1.2(+21)	1.2(+21)
S	$8.5(13)^b$	2.7(+14)	2.0(+13)
S^+	$1.2(16)^b$	2.9(+16)	1.9(+15)
O	5.0(17) ^c	1.0(+18)	7.0(+17)
C	$2.2(17)^{c}$	1.9(+17)	9.4(+16)
SO	$[1.0(12)-1.0(13)]^e$	4.7(+10)	2.7(+10)
CO	$2.2(15)^a$	4.8(+15)	4.9(+16)
C_2H	$[7.0(12)]^f$	9.8(+12)	3.7(+13)
C_3H_2	$[4.8(12)]^f$	1.2(+11)	8.3(+11)
NH	$8.8(11)^g$	8.0(+10)	1.3(+12)
CN	$2.5(12)^a$	5.0(+12)	2.2(+13)
OH	$4.6(13)^a$	1.0(+13)	8.9(+12)
H^{+}_{3}	$[3.8(14)]^h$	9.8(+11)	1.7(+12)
C_3	$1.6(12)^i$	1.3(+12)	3.7(+12)
HCO+	$9.1(11)^a$	2.2(+10)	4.0(+10)
HCN	$3.6(11))^a$	1.2(+10)	1.5(+11)
CH	$2.5(13)^a$	9.1(+13)	2.4(+14)

[&]quot;Liszt & Lucas (2001), ^bMorton (1975), ^cvan Dishoeck (1998), ^cLucas & Liszt (1997), ^fLucas & Liszt (2000), ^gCrawford & Williams (1997), ^hMcCall et al. (1999), ⁱMaier et al. (2001).

2. "Dynamics"

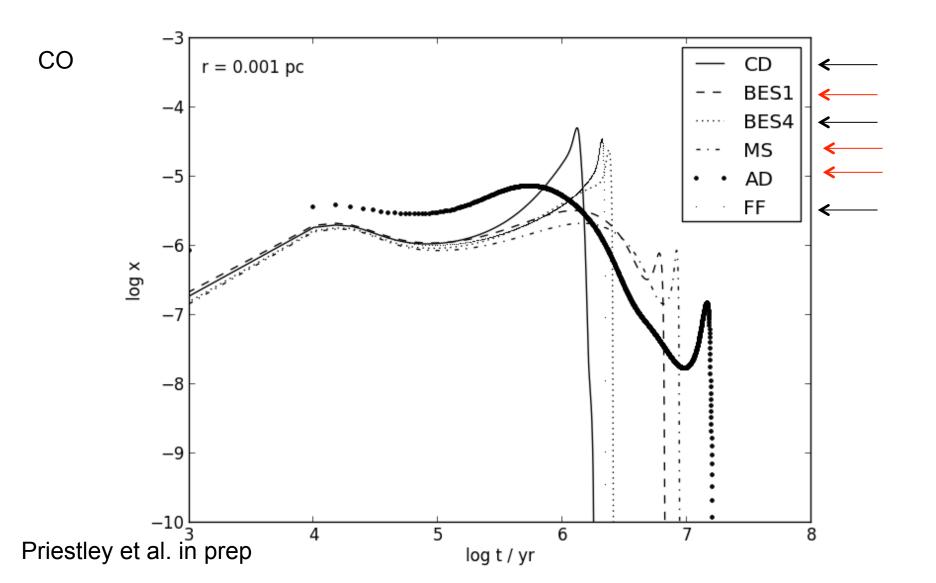


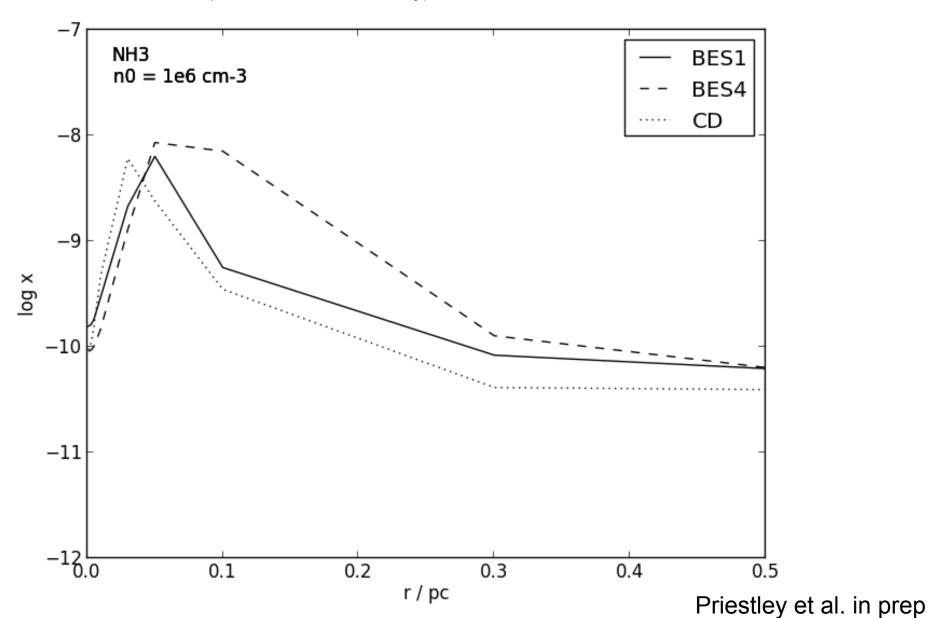
- 1. Collapse from constant density (CD, Larson 1969), 2. the collapse of an unstable (BES1, Foster & Chevalier 1993) or 3. highly unstable (BES4, Ogino et al. 1999) Bonnor-Ebert sphere, 4. collapse against magnetic support (MS, Nakamura et al. 1995) and 5. collapse resulting from ambipolar diffusion (AD, Fiedler & Mouschiavas 1993). The evolution under free-fall collapse (FF) was also included as a comparison.
 - → Then for each 'dynamical' model one can run a grid.....

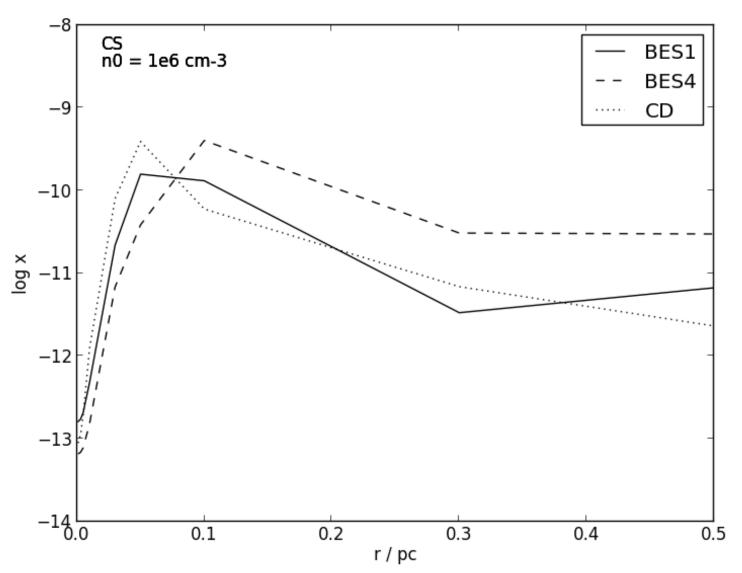
Model	n_f/cm^{-3}	$t_{eq}/{ m yr}$	$\zeta/1.3\times 10^{-17}{\rm s}^{-1}$	ε	ϕ	Y_UV
A	10 ⁸	0	1	0.01	10 ⁵	0.1
B1	10 ⁷	0	1	0.01	10 ⁵	0.1
B2	10 ⁹	0	1	0.01	10 ⁵	0.1
C1	108	10 ⁴	1	0.01	10 ⁵	0.1
C2	108	10 ⁶	1	0.01	10 ⁵	0.1
D1	108	0	5	0.01	10 ⁵	0.1
D2	108	0	10	0.01	10 ⁵	0.1
E1	108	0	1	0.01	10 ⁵	0.1
E2	108	0	1	0.01	10 ⁵	0.1
F1	108	0	1	0.1	10 ⁵	0.1
F2	108	0	1	1.0	10 ⁵	0.1
G1	108	0	1	0.01	10 ⁴	0.1
G2	108	0	1	0.01	10 ⁶	0.1
H1	108	0	1	0.01	10 ⁵	0.00
H2	108	0	1	0.01	10 ⁵	1.0

Priestley et al. in prep

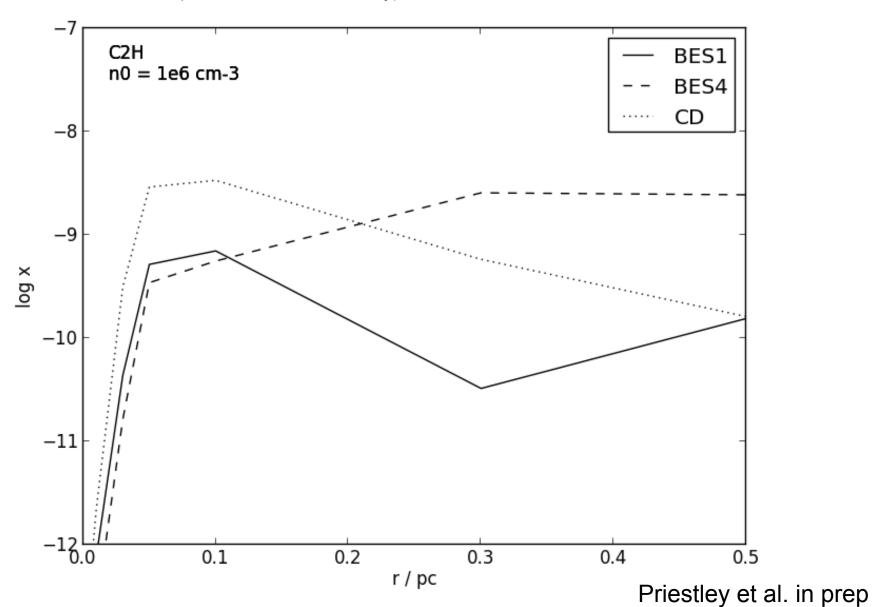
1. Collapse from constant density (CD) 2. Collapse of an unstable (BES1) or 3. highly unstable (BES4) Bonnor-Ebert sphere, 4. collapse against magnetic support (MS) and 5. collapse resulting from ambipolar diffusion (AD) cf free-fall collapse (FF)

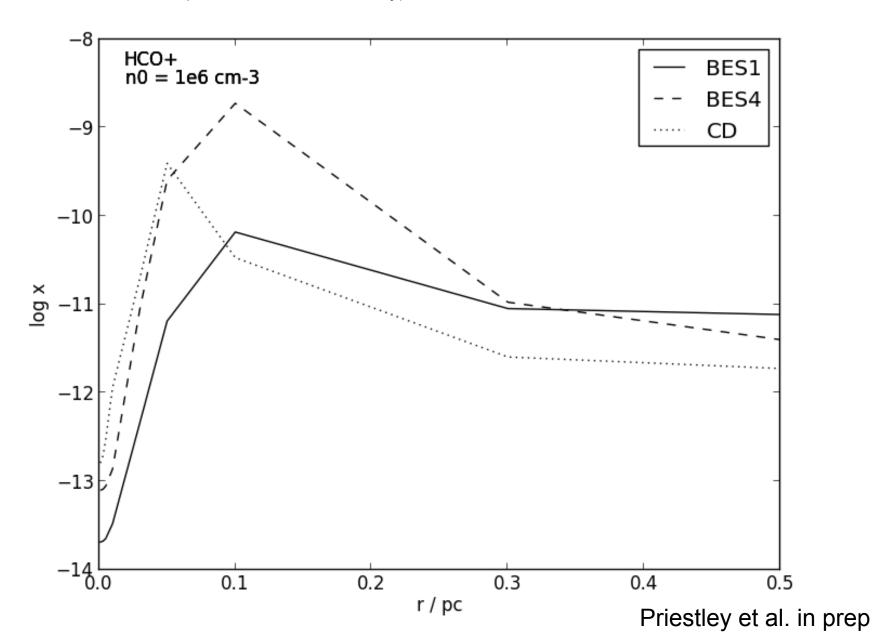


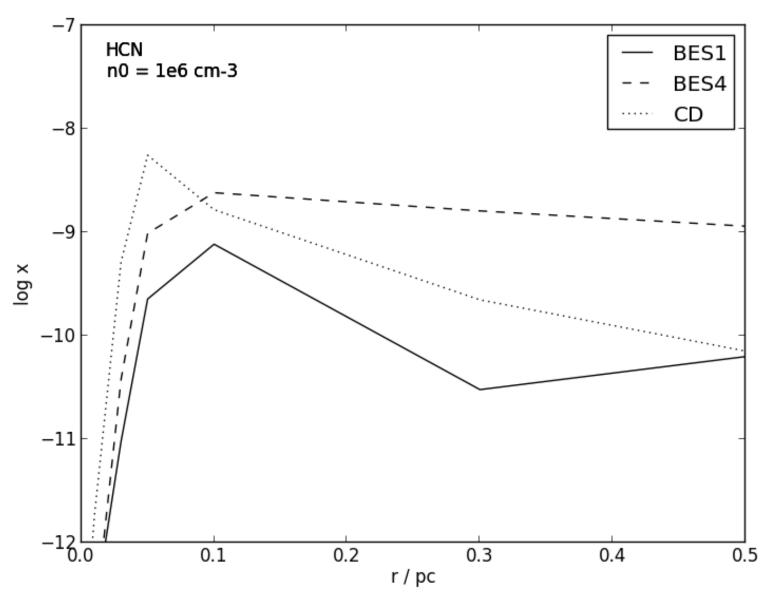




Priestley et al. in prep

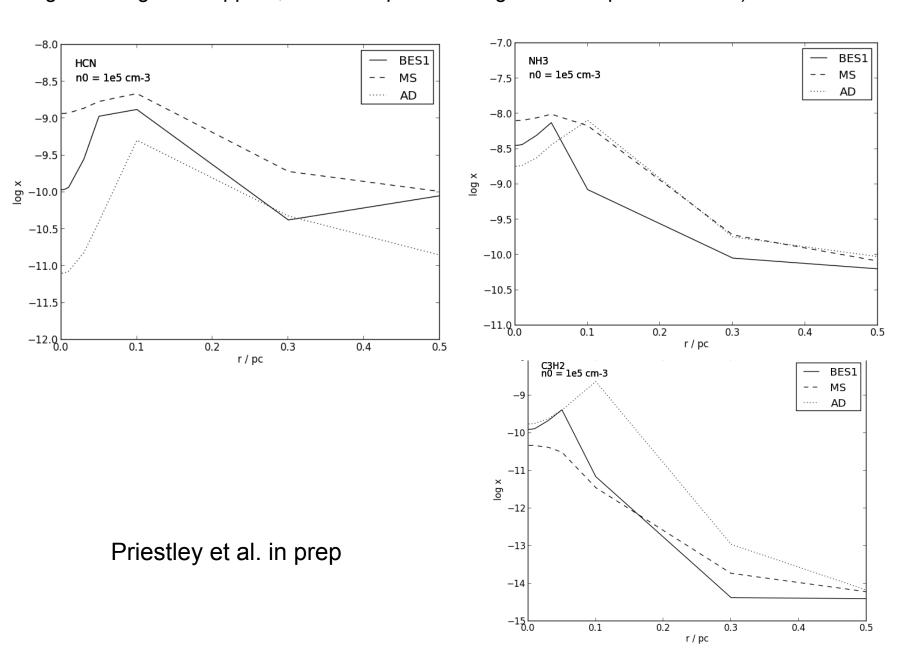




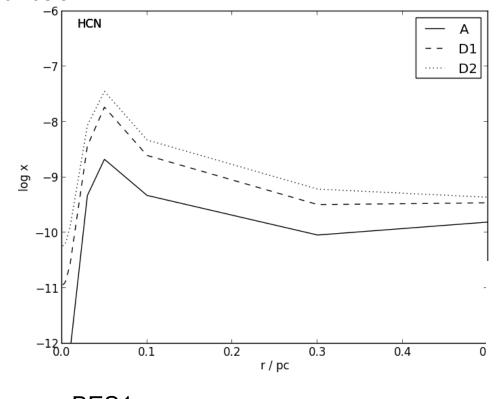


Priestley et al. in prep

The presence of magnetic fields: (BES1 = Collapse of an unstable BE; MS = collapse against magnetic support; AD = collapse resulting from ambipolar diffusion)



Dynamics + fluxes: the effects of cosmic rays → more important for models where collapse results from ambipolar diffusion

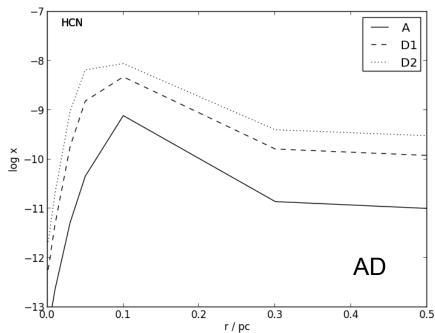


BES1

Priestley et al. in prep

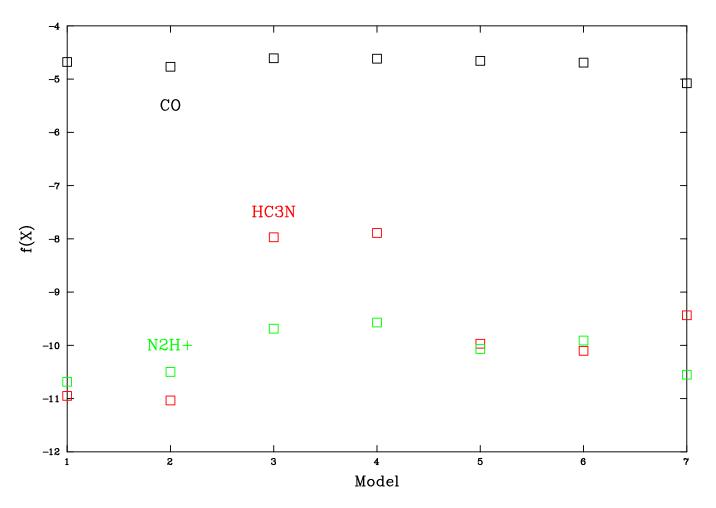
Table 1. Values of parameters selected for the models.

Model	n_f/cm^{-3}	$t_{eq}/{ m yr}$	$\zeta/1.3\times 10^{-17}{\rm s}^{-1}$	ε	φ	Y_UV
A	10 ⁸	0	1	0.01	10 ⁵	0.1
B1	10 ⁷	0	1	0.01	10 ⁵	0.1
B2	10 ⁹	0	1	0.01	10 ⁵	0.1
C1	108	104	1	0.01	10 ⁵	0.1
C2	108	10 ⁶	1	0.01	10 ⁵	0.1
D1	108	0	5	0.01	10 ⁵	0.1
D2	108	0	10	0.01	10 ⁵	0.1
E1	108	0	1	0.01	10 ⁵	0.1
E2	108	0	1	0.01	10 ⁵	0.1
F1	10 ⁸	0	1	0.1	10 ⁵	0.1
F2	10 ⁸	0	1	1.0	10 ⁵	0.1
G1	10 ⁸	0	1	0.01	10 ⁴	0.1
G2	108	0	1	0.01	10 ⁶	0.1
H1	108	0	1	0.01	10 ⁵	0.001
H2	108	0	1	0.01	10 ⁵	1.0



"Density-Time"





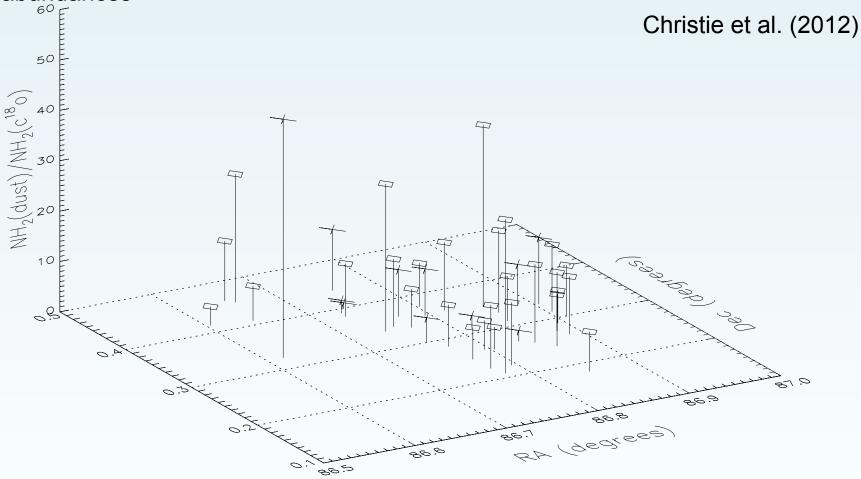
 $n f = 10^4 cm^{-3}$

Note that HC₃N/N₂H⁺ ratio is used as an age indicator for pre-stellar cores! (e.g. Hohashi et al. 2015)

Model 1: $n_{H_{_i}}$ = 10 cm⁻³, solar; Model 2: $n_{H_{_i}}$ = 100 cm⁻³; solar Model 3: $n_{H_{_i}}$ = 10 cm⁻³, TMC-1; Model 4: $n_{H_{_i}}$ = 100 cm⁻³; TMC-1 Model 5: $n_{H_{_i}}$ = 10 cm⁻³, L134N; Model 6: $n_{H_{_i}}$ = 100 cm⁻³; L134N Model 7: $n_{H_{_i}}$ = 100 cm⁻³; solar; long lived

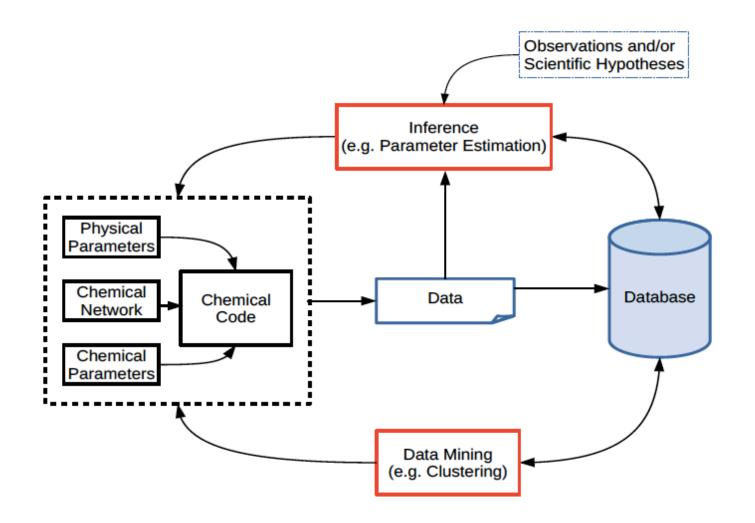


NGC 2071: squares are starless cores; crosses protostellar cores → small variations in CO abundances



Small variations in the abundance of **abundant** molecules → **large** variations in the abundance of **tracer** molecules

Probing effects of initial conditions on the chemistry: a statistical approach



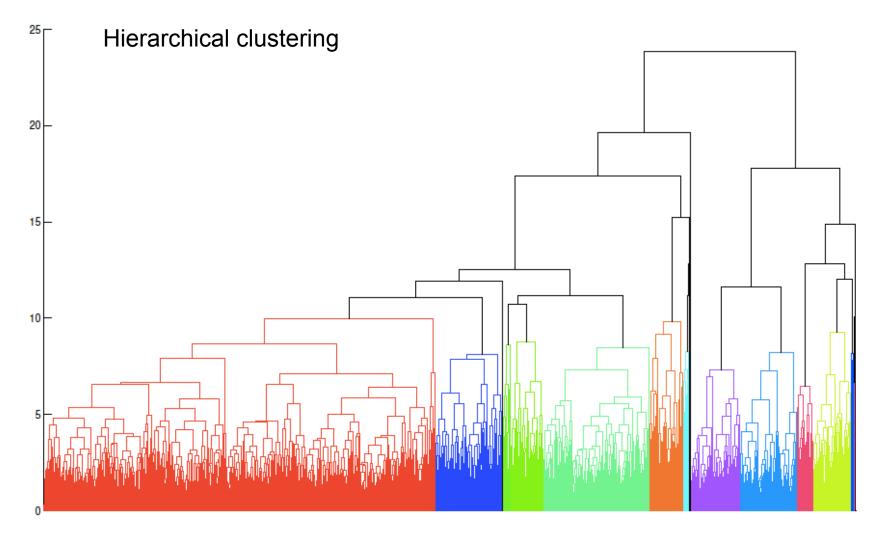


Figure 2.6: Full dendrogram for the bottom-up agglomerative hierarchical clustering on our data. Different clusters are indicated by different color according to a user defined level of pruning the tree. Each of the leaves in the dendrogram represents a model. The x-axis labeling though has been deactivated due to the large number of leaves.

Probabilistic hierarchical clustering

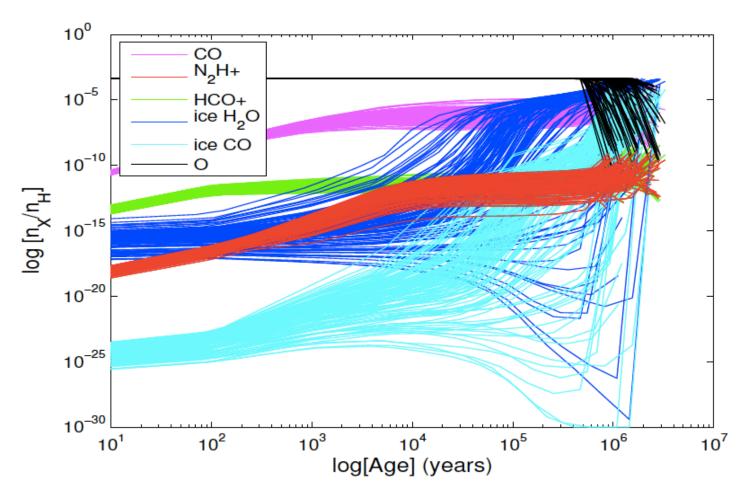


Figure 2.11: Probabilistic hierarchical clustering of our toy example application. Each color represents a distinct cluster and coincides with a distinct species as well.



Example: water icy mantle 'fits'

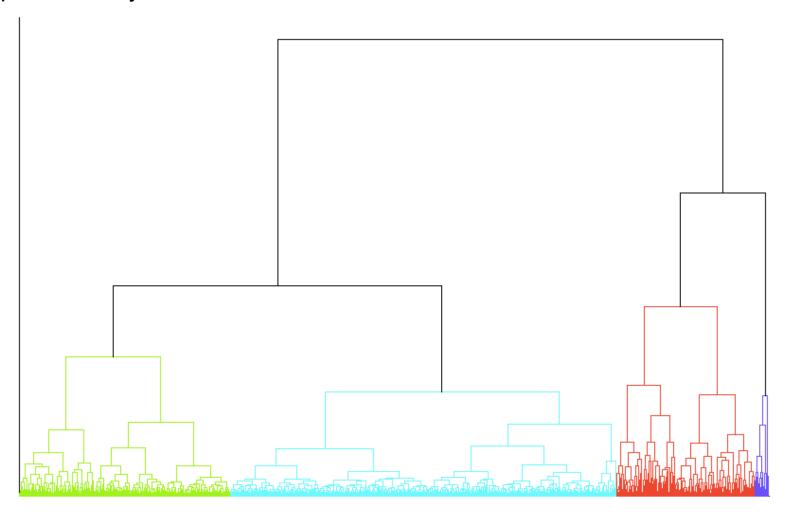


Figure 2.12: Dendrogram for the probabilistic hierarchical clustering on ice water data. Different clusters are indicated by different color according to the algorithm's suggested number of clusters. The tree is pruned for practical visualization reasons.

Accelerated
Bayesian
inference
applied to the
'problem' of ices

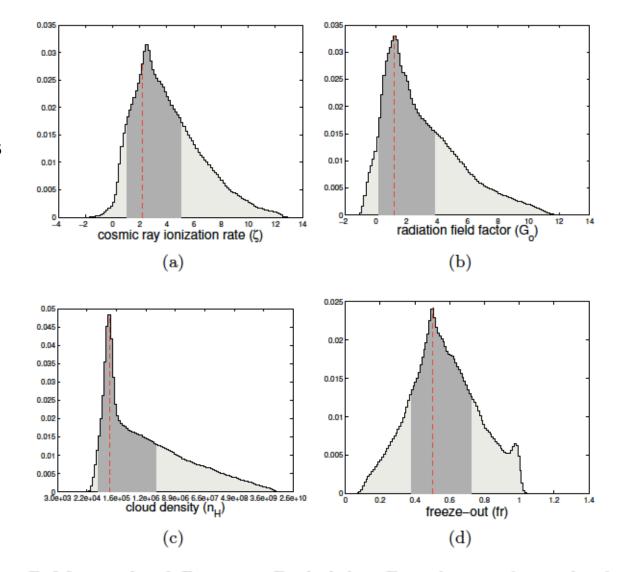


Figure 4.3: 1D Marginalized Posterior Probability Distribution for each of the four parameters using accelerated Bayesian inference with ANNs. The plots show the Gaussian kernel density estimator of each Probability Density Function. Dark gray area indicate 68% highest density region, while dashed red lines indicate the pre-defined parameter values.

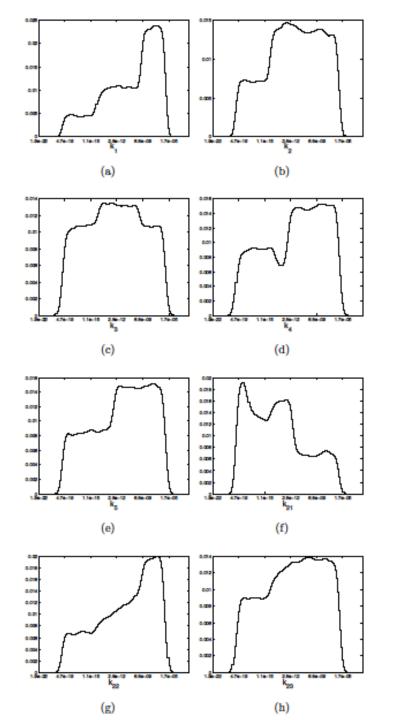
Bayesian Uncertainty Analysis of Surface Reactions in Icy Mantles (Makrymallis & Viti, 2016, submitted)

Can we prioritize specific experiments under specific conditions by the use of probabilistic inference?

Advantage of this method: we do not need to worry about how surface reaction rates are calculated

$$O_{int} = [10^{-8}, 10^{-4}] = \{x \in \mathbb{R} | 10^{-8} \le x \le 10^{-4}\}.$$

1. O + H \rightarrow OH 2. OH + H \rightarrow H ₂ O 3. CO + OH \rightarrow CO ₂ 4. S + H \rightarrow HS 5. HS + H \rightarrow H ₂ S 6. H ₂ S + S \rightarrow H ₂ S ₂ 7. CS + H \rightarrow HCS 8. HCS + H \rightarrow HCS 9. CO + S \rightarrow OCS 10. OCS + H \rightarrow HOCS 11. H ₂ S + CO \rightarrow OCS 12. H ₂ S + H ₂ S \rightarrow H ₂ S ₂ 13. H ₂ S ₂ + CO \rightarrow CS2 + O 14. H ₂ S + O \rightarrow SO ₂ 15. CS ₂ + O \rightarrow OCS 16. CO + HS \rightarrow OCS 17. S + O \rightarrow SO 18. SO + O \rightarrow SO 20. HSO + H \rightarrow HSO 21. CO + H \rightarrow HCO 22. HCO + H \rightarrow HCO	No.			React	ions	
3. $CO + OH \rightarrow CO_2$ 4. $S + H \rightarrow HS$ 5. $HS + H \rightarrow H_2S$ 6. $H_2S + S \rightarrow H_2S_2$ 7. $CS + H \rightarrow HCS$ 8. $HCS + H \rightarrow HCS$ 9. $CO + S \rightarrow CCS$ 10. $CCS + H \rightarrow HCS$ 11. $CCS + H \rightarrow HCS$ 12. $CCS + H \rightarrow HCS$ 13. $CCS + H \rightarrow HCS$ 14. $CCS + H \rightarrow HCS$ 15. $CCS + H \rightarrow HCS$ 16. $CCS + HCS \rightarrow CCS$ 17. $CCS + HCS \rightarrow CCS$ 18. $CCS + CCS$ 19.	1.	O	+	Н	\rightarrow	ОН
4. S + H → HS 5. HS + H → H ₂ S 6. H ₂ S + S → H ₂ S ₂ 7. CS + H → HCS 8. HCS + H → H ₂ CS 9. CO + S → OCS 10. OCS + H → HOCS 11. H ₂ S + CO → OCS 12. H ₂ S + H ₂ S → H ₂ S ₂ 13. H ₂ S ₂ + CO → CS2 + O 14. H ₂ S + O → SO ₂ 15. CS ₂ + O → OCS 16. CO + HS → OCS 17. S + O → SO 18. SO + O → SO 19. SO + H → HSO 20. HSO + H → HCO 21. CO + H → HCO	2.	OH	+	Н	\rightarrow	H_2O
5. HS + H \rightarrow H ₂ S 6. H ₂ S + S \rightarrow H ₂ S ₂ 7. CS + H \rightarrow HCS 8. HCS + H \rightarrow H ₂ CS 9. CO + S \rightarrow OCS 10. OCS + H \rightarrow HOCS 11. H ₂ S + CO \rightarrow OCS 12. H ₂ S + H ₂ S \rightarrow H ₂ S ₂ 13. H ₂ S ₂ + CO \rightarrow CS2 + O 14. H ₂ S + O \rightarrow SO ₂ 15. CS ₂ + O \rightarrow OCS 17. S + O \rightarrow SO 18. SO + O \rightarrow SO 19. SO + H \rightarrow HSO 20. HSO + H \rightarrow HCO 21. CO + HS \rightarrow H2CO	3.	CO	+	ОН	\rightarrow	CO_2
6. H_2S + S \rightarrow H_2S_2 7. CS + H \rightarrow HCS 8. HCS + H \rightarrow H_2CS 9. CO + S \rightarrow OCS 10. OCS + H \rightarrow $HOCS$ 11. H_2S + CO \rightarrow OCS 12. H_2S + H_2S \rightarrow H_2S_2 13. H_2S_2 + CO \rightarrow $CS2$ + O 14. OCS + OCS 15. OCS + OCS 16. OCS + OCS 17. OCS + OCS 18. OCS + OCS 19. OCS + OCS 19. OCS + OCS 19. OCS + OCS 19. OCS + OCS 20. OCS + OCS 21. OCS + OCS 22. OCS	4.	S	+	Н	\rightarrow	HS
7. $CS + H \rightarrow HCS$ 8. $HCS + H \rightarrow H_2CS$ 9. $CO + S \rightarrow OCS$ 10. $OCS + H \rightarrow HOCS$ 11. $H_2S + CO \rightarrow OCS$ 12. $H_2S + H_2S \rightarrow H_2S_2$ 13. $H_2S_2 + CO \rightarrow CS_2 + O$ 14. $H_2S + O \rightarrow SO_2$ 15. $CS_2 + O \rightarrow OCS + S$ 16. $CO + HS \rightarrow OCS$ 17. $S + O \rightarrow SO$ 18. $SO + O \rightarrow SO_2$ 19. $SO + H \rightarrow HSO$ 20. $HSO + H \rightarrow HCO$ 21. $CO + H \rightarrow HCO$	5.	$_{ m HS}$	+	Н	\rightarrow	H_2S
8. $HCS + H \rightarrow H_2CS$ 9. $CO + S \rightarrow OCS$ 10. $OCS + H \rightarrow HOCS$ 11. $H_2S + CO \rightarrow OCS$ 12. $H_2S + H_2S \rightarrow H_2S_2$ 13. $H_2S_2 + CO \rightarrow CS_2 + O$ 14. $H_2S + O \rightarrow SO_2$ 15. $CS_2 + O \rightarrow OCS + S$ 16. $CO + HS \rightarrow OCS$ 17. $S + O \rightarrow SO$ 18. $SO + O \rightarrow SO_2$ 19. $SO + H \rightarrow HSO$ 20. $HSO + H \rightarrow HSO$ 21. $CO + H \rightarrow HCO$ 22. $HCO + H \rightarrow H_2CO$	6.	H_2S	+	\mathbf{S}	\rightarrow	H_2S_2
9. $CO + S \rightarrow OCS$ 10. $OCS + H \rightarrow HOCS$ 11. $H_2S + CO \rightarrow OCS$ 12. $H_2S + H_2S \rightarrow H_2S_2$ 13. $H_2S_2 + CO \rightarrow CS2 + O$ 14. $H_2S + O \rightarrow SO_2$ 15. $CS_2 + O \rightarrow OCS + S$ 16. $CO + HS \rightarrow OCS$ 17. $S + O \rightarrow SO$ 18. $SO + O \rightarrow SO_2$ 19. $SO + H \rightarrow HSO$ 20. $HSO + H \rightarrow HSO$ 21. $CO + H \rightarrow HCO$ 22. $HCO + H \rightarrow H_2CO$	7.	CS	+	Н	\rightarrow	HCS
10. OCS + H \rightarrow HOCS 11. H ₂ S + CO \rightarrow OCS 12. H ₂ S + H ₂ S \rightarrow H ₂ S ₂ 13. H ₂ S ₂ + CO \rightarrow CS2 + O 14. H ₂ S + O \rightarrow SO ₂ 15. CS ₂ + O \rightarrow OCS + S 16. CO + HS \rightarrow OCS 17. S + O \rightarrow SO 18. SO + O \rightarrow SO ₂ 19. SO + H \rightarrow HSO 20. HSO + H \rightarrow HCO 21. CO + H \rightarrow HCO	8.	HCS	+	Н	\rightarrow	$\mathrm{H_{2}CS}$
11. H_2S + CO \rightarrow OCS 12. H_2S + H_2S \rightarrow H_2S_2 13. H_2S_2 + CO \rightarrow $CS2 + O$ 14. H_2S + O \rightarrow SO_2 15. CS_2 + O \rightarrow $OCS + S$ 16. CO + O	9.	CO	+	\mathbf{S}	\rightarrow	OCS
12. H_2S + H_2S \rightarrow H_2S_2 13. H_2S_2 + CO \rightarrow $CS2 + O$ 14. H_2S + O \rightarrow SO_2 15. CS_2 + O \rightarrow $OCS + S$ 16. CO + HS \rightarrow OCS 17. S + O \rightarrow SO 18. SO + O \rightarrow SO_2 19. SO + O \rightarrow OCS 20. O + O O O + O O +	10.	OCS	+	Н	\rightarrow	HOCS
13. $H_2S_2 + CO \rightarrow CS2 + O$ 14. $H_2S + O \rightarrow SO_2$ 15. $CS_2 + O \rightarrow OCS + S$ 16. $CO + HS \rightarrow OCS$ 17. $S + O \rightarrow SO$ 18. $SO + O \rightarrow SO_2$ 19. $SO + H \rightarrow HSO$ 20. $HSO + H \rightarrow HCO$ 21. $CO + H \rightarrow HCO$	11.	H_2S	+	CO	\rightarrow	OCS
14. H_2S + O \rightarrow SO_2 15. CS_2 + O \rightarrow $OCS + S$ 16. CO + HS \rightarrow OCS 17. S + O \rightarrow SO 18. SO + O \rightarrow SO_2 19. SO + H \rightarrow HSO 20. HSO + H \rightarrow HCO 21. CO + H \rightarrow HCO	12.	H_2S	+	$\mathrm{H}_2\mathrm{S}$	\rightarrow	H_2S_2
15. $CS_2 + O \rightarrow OCS + S$ 16. $CO + HS \rightarrow OCS$ 17. $S + O \rightarrow SO$ 18. $SO + O \rightarrow SO_2$ 19. $SO + H \rightarrow HSO$ 20. $HSO + H \rightarrow HO$ 21. $CO + H \rightarrow HO$	13.	H_2S_2	+	\mathbf{CO}	\rightarrow	CS2 + O
16. CO + HS \rightarrow OCS 17. S + O \rightarrow SO 18. SO + O \rightarrow SO ₂ 19. SO + H \rightarrow HSO 20. HSO + H \rightarrow HCO 21. CO + H \rightarrow HCO 22. HCO + H \rightarrow H ₂ CO	14.	$_{\mathrm{H_2S}}$	+	O	\rightarrow	SO_2
17. S + O \rightarrow SO 18. SO + O \rightarrow SO ₂ 19. SO + H \rightarrow HSO 20. HSO + H \rightarrow HCO 21. CO + H \rightarrow HCO 22. HCO + H \rightarrow H ₂ CO	15.	CS_2	+	O	\rightarrow	OCS + S
18. SO + O \rightarrow SO ₂ 19. SO + H \rightarrow HSO 20. HSO + H \rightarrow SO 21. CO + H \rightarrow HCO 22. HCO + H \rightarrow H ₂ CO	16.	CO	+	$_{\mathrm{HS}}$	\rightarrow	OCS
19. SO + H \rightarrow HSO 20. HSO + H \rightarrow SO 21. CO + H \rightarrow HCO 22. HCO + H \rightarrow H ₂ CO	17.	\mathbf{S}	+	O	\rightarrow	SO
20. HSO + H \rightarrow SO 21. CO + H \rightarrow HCO 22. HCO + H \rightarrow H ₂ CO	18.	SO	+	O	\rightarrow	SO_2
21. CO + H \rightarrow HCO 22. HCO + H \rightarrow H ₂ CO	19.	SO	+	H	\rightarrow	HSO
22. HCO + H \rightarrow H ₂ CO	20.	HSO	+	Н	\rightarrow	SO
	21.	CO	+	Н	\rightarrow	HCO
23. H_2CO + H \rightarrow CH_3OH	22.	HCO	+	Н	\rightarrow	${\rm H_2CO}$
	23.	${\rm H_2CO}$	+	Н	\rightarrow	$\mathrm{CH_{3}OH}$



Over a million chemical models



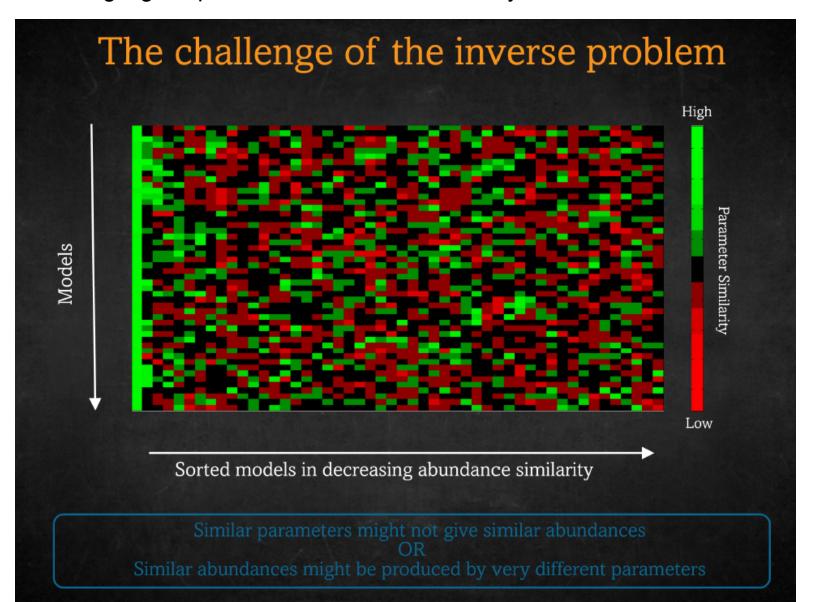
Only 8 out of 23 reactions show variability e.g. 15 out of 23 reactions may not be needed

i.e reactions of rates with uniform distribution do not impact on the outcome

No.			React	ions	
1.	O	+	Н	\rightarrow	OH
2.	OH	+	Н	\rightarrow	$_{\rm H_2O}$
3.	CO	+	ОН	\rightarrow	CO_2
4.	S	+	Н	\rightarrow	HS
5.	HS	+	Н	\rightarrow	H_2S
6.	H_2S	+	\mathbf{S}	\rightarrow	H_2S_2
7.	CS	+	Н	\rightarrow	HCS
8.	HCS	+	Н	\rightarrow	H_2CS
9.	CO	+	S	\rightarrow	OCS
10.	OCS	+	Н	\neq	HOCS
11.	H_2S	+	CO	/→	OCS
12.	H_2S	+	H_2S	\rightarrow	H_2S_2
13.	H_2S_2	+	co	\rightarrow	CS2 + O
14.	H_2S	+/	d	\rightarrow	SO_2
15.	CS_2	/+	0	\rightarrow	OCS + S
16.	CO	+	$_{ m HS}$	7	OCS
17.	Ś	+	O	\rightarrow	SO
18.	so	+	O	\rightarrow	SO_2
19/	SO	+	Н	\rightarrow	HSO
20.	HSO	+	Н	\rightarrow	so
21.	CO	+	H	\rightarrow	HCO
22.	HCO	+	Н	\rightarrow	${\rm H_2CO}$
23.	${\rm H_2CO}$	+	Н	\rightarrow	$\mathrm{CH_{3}OH}$

Small variations in the abundance of abundant molecules → large variations in the abundance of tracer molecules

The complexity of the chemical models is unavoidable: new approaches to disentangling simple conclusions are necessary





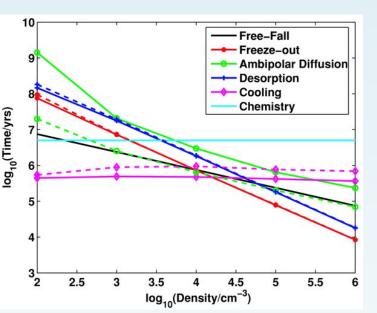
Star formation timescales

Low-mass star formation is relatively well-studied for Milky Way conditions and controlled by several critical timescales

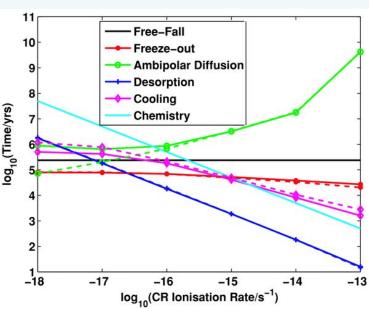
- 1. Collapse under gravity
- 2. Cooling by molecules and dust
- 3. Freeze-out of molecules on to dust
- 4. Desorption
- 5. Ambipolar diffusion
- 6. Chemistry

Timescales in galaxies: dependence on density

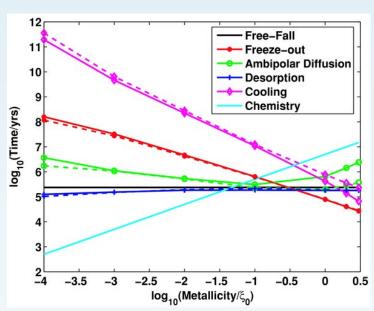
Banerji et al. 2009



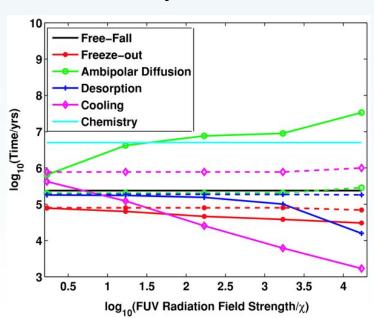
Timescales in galaxies: dependence on cosmic rays

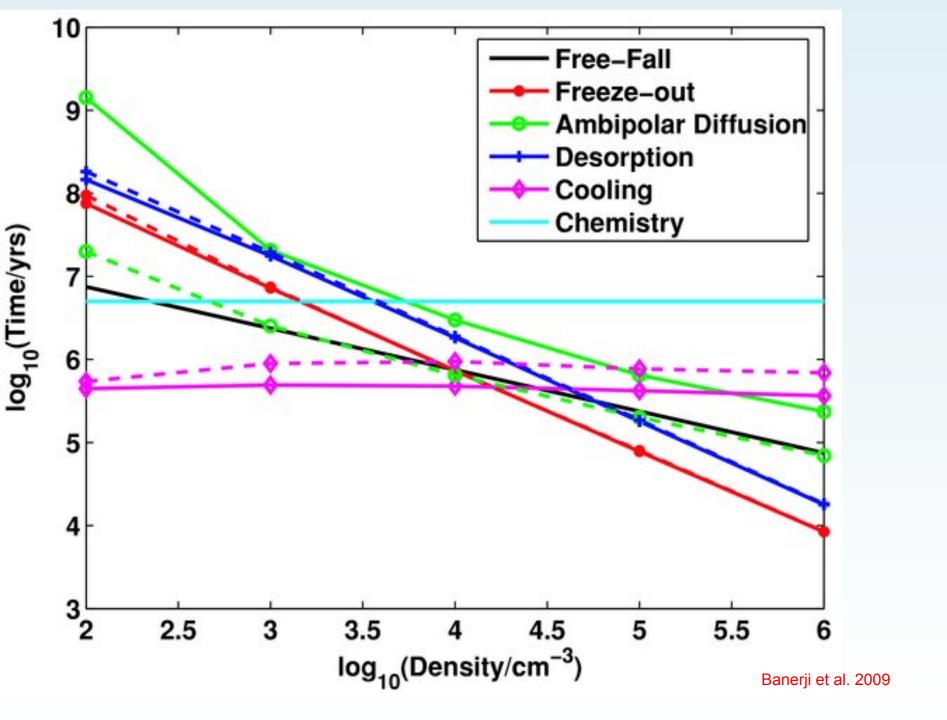


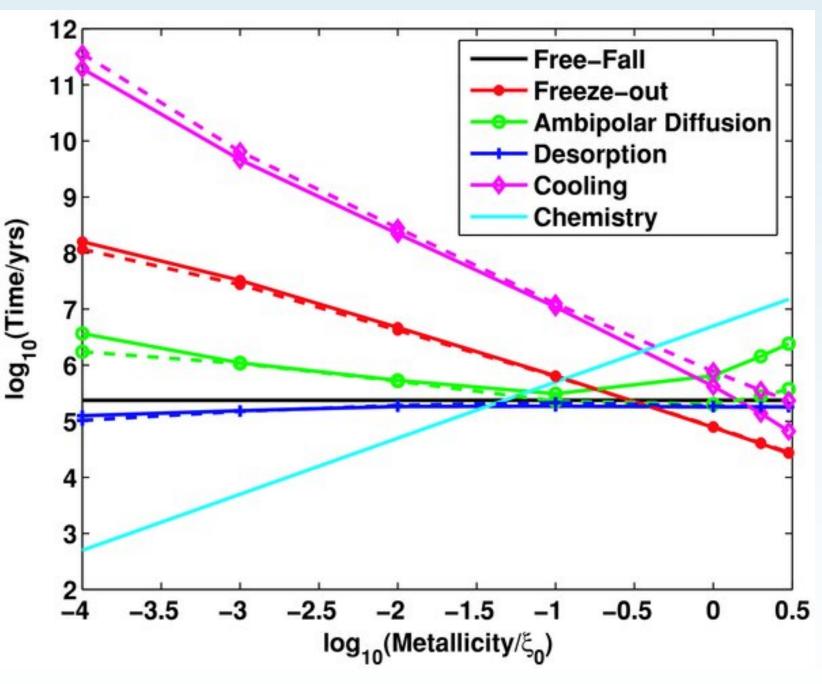
Timescales in galaxies: dependence on metallicity

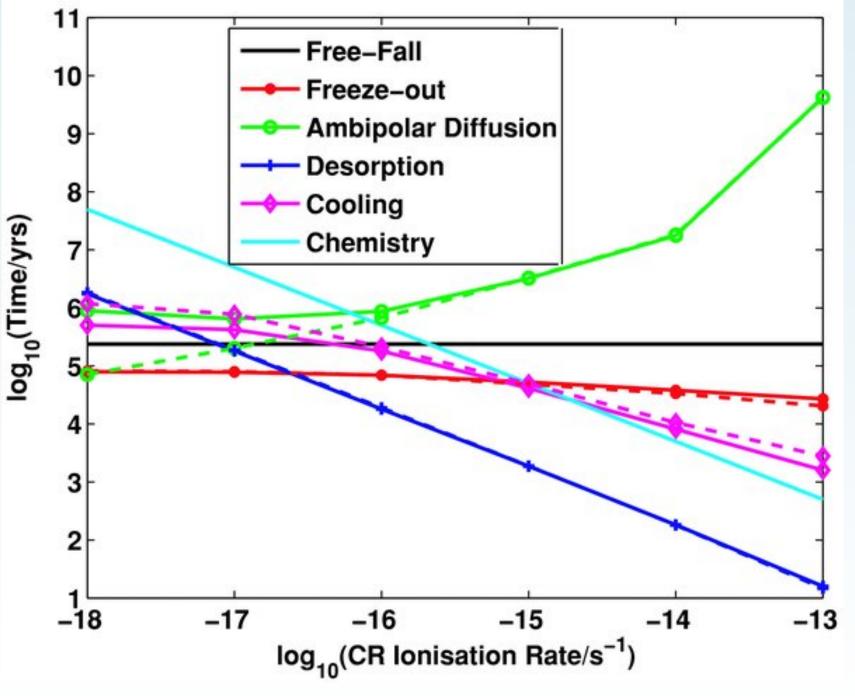


Timescales in galaxies: dependence on FUV intensity

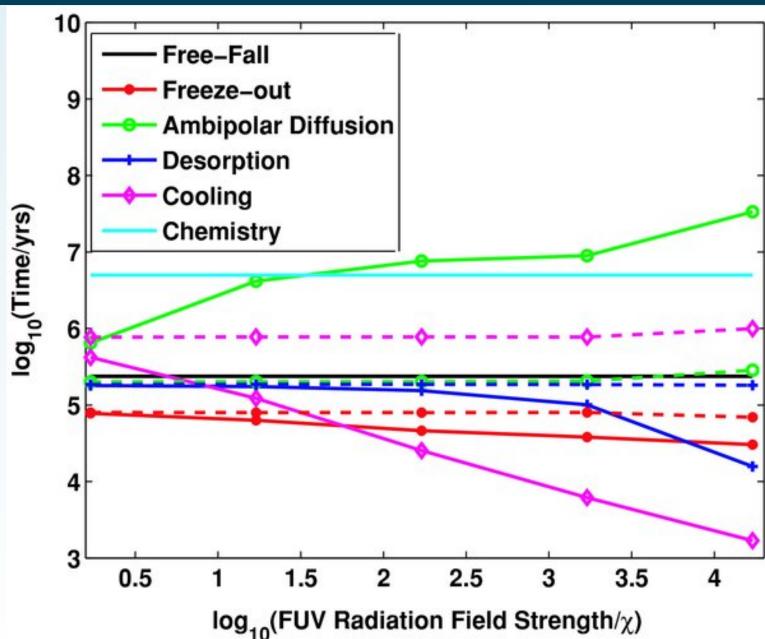














Timescales in external galaxies: some considerations

If low mass star formation inhibited → "top-heavy" IMF??

This may be true in:

- (i) Low metallicity systems
- (ii) Active galaxies (very high FUV and cosmic rays) with solar metallicity (collapse likely impeded by magnetic pressure)

High-redshift Models: Inputs to the Model as well as the Form of the IMF Associated with These Physical Condition

Model	Density (cm ⁻³)	Metallicity (ξ_{\odot})	CR Ionization Rate (s ⁻¹)	FUV Flux (Draine)
I	104	0.05	10^{-13}	1.7×10^{4}
II	10 ⁵	0.05	10-14	1.7×10^{3}
III	10 ⁵	1.00	10^{-14}	1.7×10^{3}

