

Astronomical Complex Organic Molecules (ACOMs) in Sun-type star forming regions

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With billions thanks to:

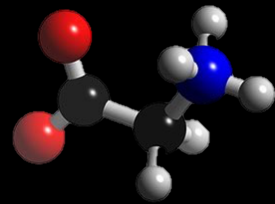
N.Balucani, P.Caselli, C.Codella, F.Fontani, A.Jaber, C.Kahane, B.Lefloch,
A. Lopez-Sepulcre, Y.Oka, L.Podio, A.Rimola, N.Sakai, V.Taquet,
P.Ugliengo, C.Vastel, S.Viti, S.Yamamoto

Note: They are here at KITP Cold Universe

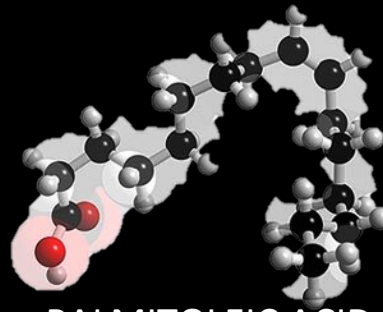
CONTEXT

1- Life on Earth & organic chemistry

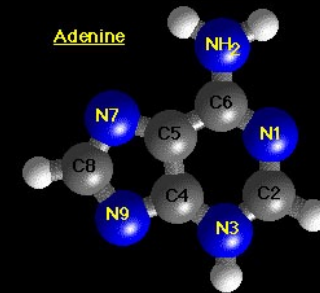
All living organisms, from microbes to men, are made up of the same basic components: amino acids, fatty acids, alcohols, and nitrogenous bases, etc. We are talking of about 50 small molecules of less than 100 atoms of **carbon**, with hydrogen, oxygen, nitrogen and other elements in smaller quantities.



GLYCINE



PALMITOLEIC ACID



LIFE ON EARTH IS BASED ON ORGANIC CHEMISTRY

CONTEXT

2- ISM largest detected molecules

It is not by chance that the longest molecules detected in the ISM are all carbon chains!

Actually, all molecules with more than 5 atoms contains C atoms.

6 atoms	7 atoms	8 atoms	9 atoms	10 atoms	11 atoms	12 atoms	>12 atoms
C ₅ H	C ₆ H	CH ₃ C ₃ N	CH ₃ C ₄ H	CH ₃ C ₅ N	HC ₉ N	<i>o</i> -C ₆ H ₆ *	HC ₁₁ N
<i>i</i> -H ₂ C ₄	CH ₂ CHCN	HC(O)OCH ₃	CH ₃ CH ₂ CN	(CH ₃) ₂ CO	CH ₃ C ₆ H	<i>n</i> -C ₃ H ₇ CN	C ₆₀ *
C ₂ H ₄ *	CH ₃ C ₂ H	CH ₃ COOH	(CH ₃) ₂ O	(CH ₂ OH) ₂	C ₂ H ₅ OCHO	<i>i</i> -C ₃ H ₇ CN 2014	C ₇₀ *
CH ₃ CN	HC ₅ N	C ₇ H	CH ₃ CH ₂ OH	CH ₃ CH ₂ CHO	CH ₃ OC(O)CH ₃		
CH ₃ NC	CH ₃ CHO	C ₆ H ₂	HC ₇ N				
CH ₃ OH	CH ₃ NH ₂	CH ₂ OHCHO	C ₈ H				
CH ₃ SH	<i>o</i> -C ₂ H ₄ O	<i>i</i> -HC ₆ H*	CH ₃ C(O)NH ₂				
HC ₃ NH ⁺	H ₂ CCHOH	CH ₂ CHCHO (?)	C ₈ H ⁻				
HC ₂ CHO	C ₆ H ⁻	CH ₂ CCHCN	C ₃ H ₆				
NH ₂ CHO		H ₂ NCH ₂ CN	CH ₃ CH ₂ SH (?)				
C ₅ N		CH ₃ CHNH					
<i>i</i> -HC ₄ H*							
<i>i</i> -HC ₄ N							
<i>o</i> -H ₂ C ₃ O							
H ₂ CCNH (?)							
C ₅ N ⁻							
HNCHCN							

CONTEXT

3- Organic chemistry in space & life



C. De Duve,
1997 Nobel laureate

The building block of life form naturally in our Galaxy and, most likely, also elsewhere in the Cosmos.

The chemical seeds of life are universal.

(“Singularities. Landmarks on the Pathways of life”, Cambridge University Press 2005)

Life is an obligatory manifestation of matter, written into the fabric of the universe, and that there must be many sites of life, perhaps even intelligent life sometimes, in many parts of our Galaxy and in others.

(Phyl.Trans.RoyalSoc. 2011).

CONTEXT

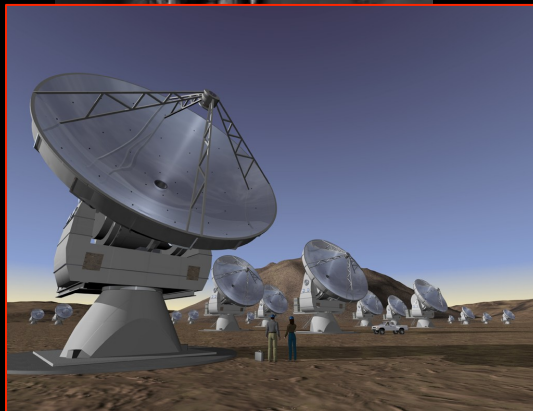
3- Organic chemistry in space & life



The building block of life form naturally in our Galaxy and, most likely, also elsewhere in the Cosmos.

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(“Singularities. Landmarks on the Pathways of life”, Cambridge University Press 2005)



ASTRONOMER JOBS:

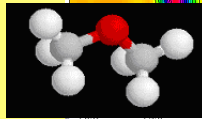
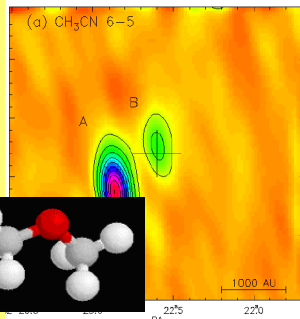
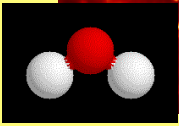
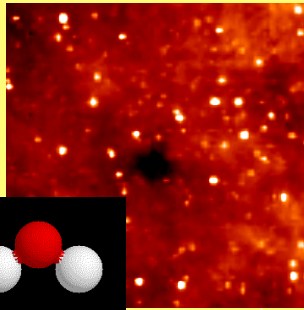
1. SEARCH FOR THE SEEDS
2. UNDERSTAND WHAT COMPLEXITY THEY CAN GROW TO IN SPACE

FROM A DIFFUSE CLOUD TO A PLANETARY SYSTEM

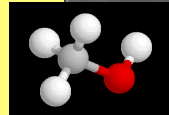
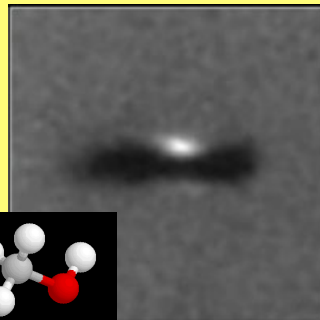
FROM ATOMS & SIMPLE MOLECULES TO LIFE

(adapted from Caselli & Ceccarelli 2012, A&ARev)

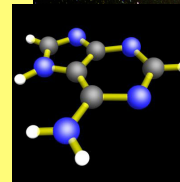
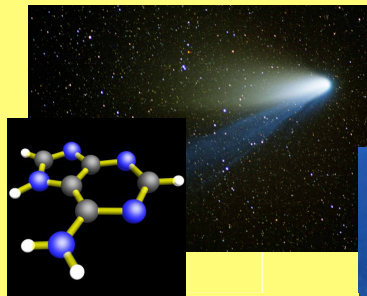
1- PRE-STELLAR PHASE: cold and dense gas
FORMATION OF SIMPLE/COMPLEX MOLECULES



2- PROTOSTELLAR PHASE: collapsing, warm dense gas
SUBLIMATION/FORMATION OF COMPLEX MOLECULES

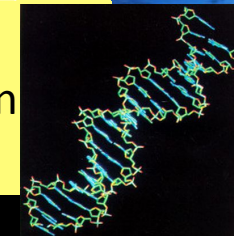


3- PROTOPLANETARY DISK PHASE:
cold and warm dense gas
SIMPLE & COMPLEX MOLECULES



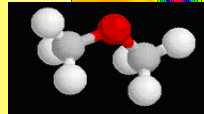
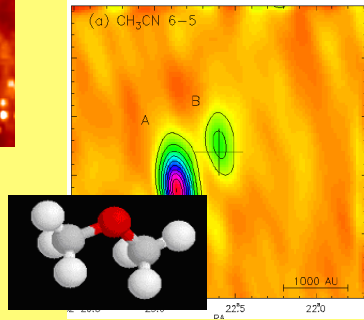
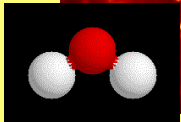
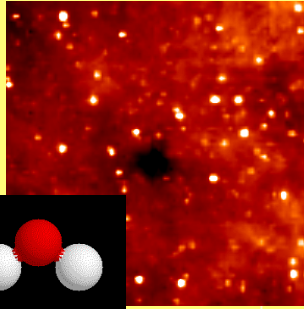
4- PLANETESIMAL FORMATION : grains agglomeration
STORAGE/REPROCESSING OF GRAIN MANTLE ICES

5- PLANET FORMATION AND THE "COMET/ASTEROID RAIN":
planet migration, small bodies scattering; Earth oceans formation
CONSERVATION/DELIVERY OF OLD MOLECULES + LIFE?



FOCUS ON THE FIRST TWO PHASES, WHERE THE RULES OF THE GAME ARE SET

1- PRE-STELLAR PHASE: cold and dense gas
FORMATION OF SIMPLE/COMPLEX MOLECULES



2- PROTOSTELLAR PHASE: collapsing, warm dense gas
SUBLIMATION/FORMATION OF COMPLEX MOLECULES

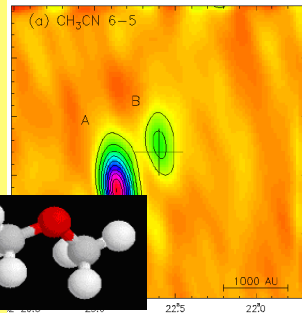
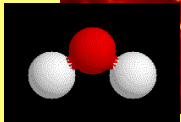
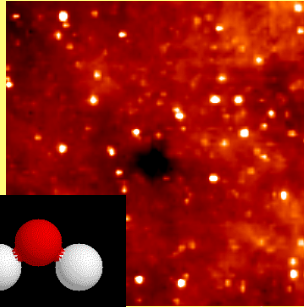
1- The same basic conditions applies to the other phases (*mutata mutandis*), but in these two phases processes are likely easier to characterize;

2- Once the processes at work are understood, we can extrapolate to molecules that we will never detect in the ISM;

3- In these phases, the “seeds” for the molecular complexity growth are planted.

FOCUS ON THE FIRST TWO PHASES, WHERE THE RULES OF THE GAME ARE SET

1- PRE-STELLAR PHASE: cold and dense gas
FORMATION OF SIMPLE/COMPLEX MOLECULES



2- PROTOSTELLAR PHASE: collapsing, warm dense gas
SUBLIMATION/FORMATION OF COMPLEX MOLECULES

➔ FOCUS ON ACOMs

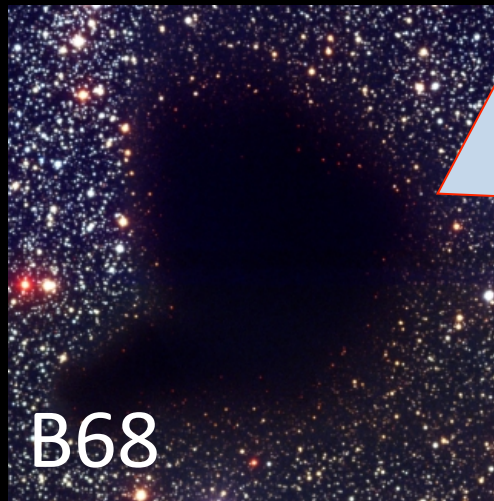
(Astronomical Complex Organic Molecules)

1. PRESTELLAR CORES (PSC)

BLACK SPOTS IN
OPTICAL PICTURES

PSC are dense and cold condensations in
molecular clouds:

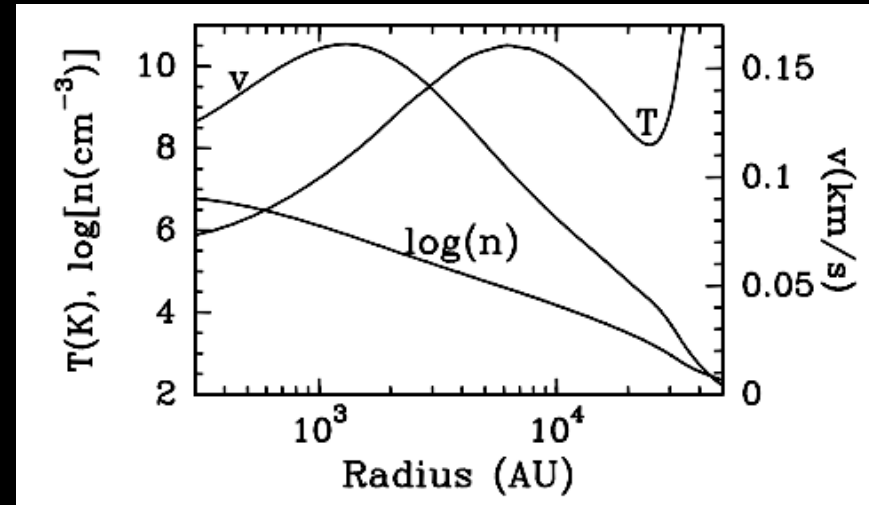
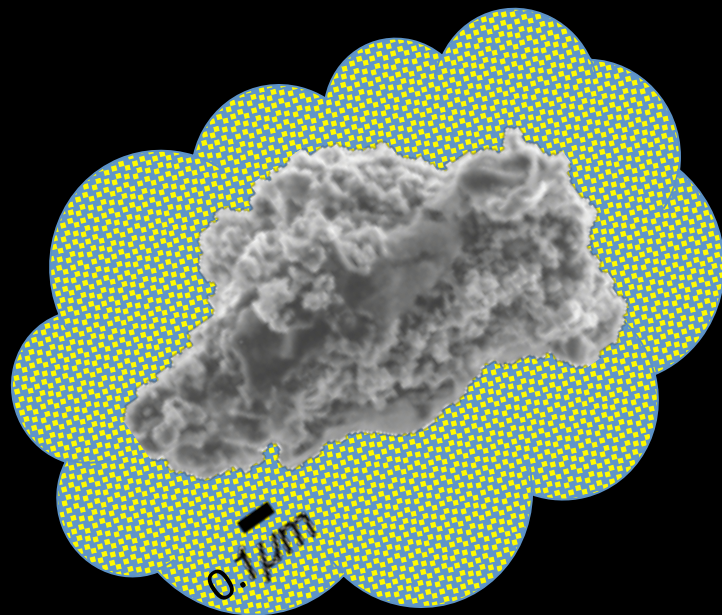
**PSC ARE THE SEEDS FROM WHICH
STARS ARE BORN**



B68

PRE-STELLAR CORES in a nutshell

THE CENTERS OF PSC ARE
COLD ($\leq 10\text{K}$) AND DENSE
($\geq 10^5\text{cm}^{-3}$)



Caselli et al. 2012

MOLECULES, e.g. CO FREEZE
OUT ONTO THE DUST GRAINS
FORMING **ICED MOLECULAR
MANTLES**

(reviews by Ceccarelli, Caselli, Herbst et al. 2005,
PP5; Bergin & Tafalla 2006; Caselli & Ceccarelli 2012)

MANTLE COMPOSITION: H₂O, CO, CO₂, NH₃, CH₃OH

HYDROGENATION ON THE GRAIN SURFACES

→ LABORATORY EXPERIMENTS + THEORY + ASTROMODELS



Example of the methanol formation on the grain surfaces

Rimola et al. 2014

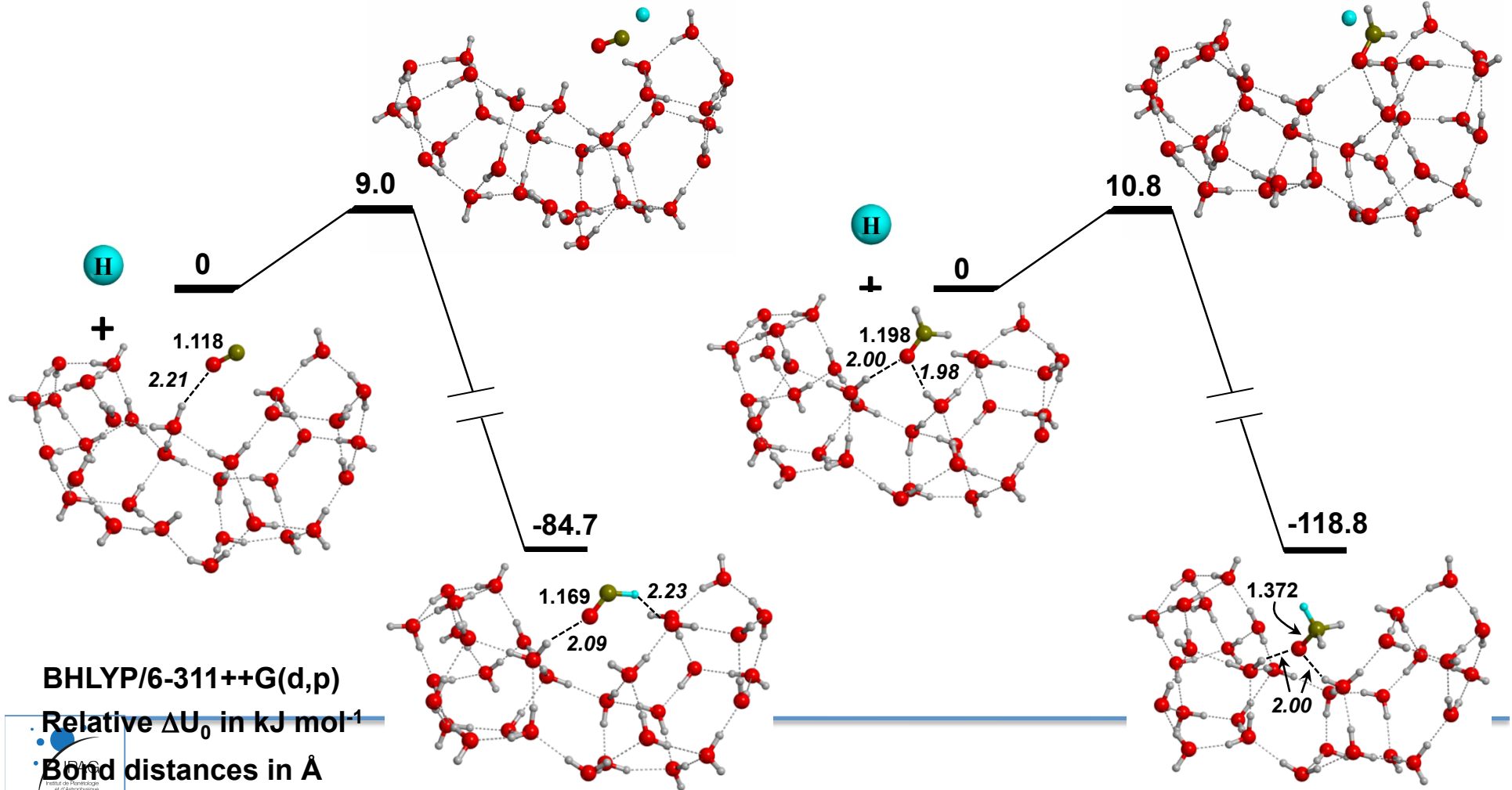
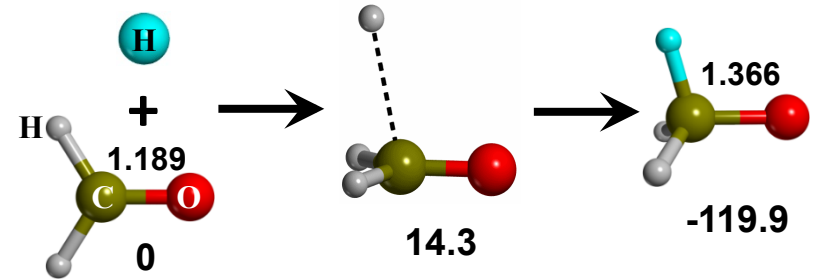
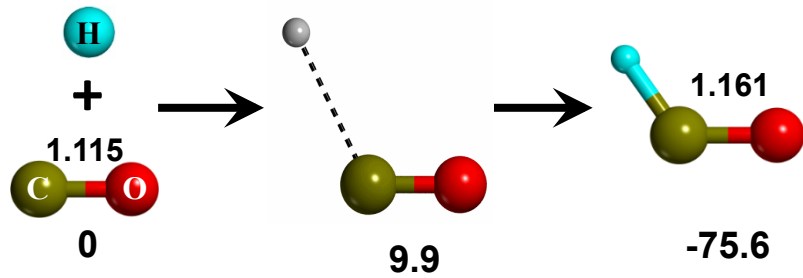
HYDROGENATION ON THE GRAIN SURFACES

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Example of the methanol formation on the grain surfaces

Interstellar H_2CO and CH_3OH Formation

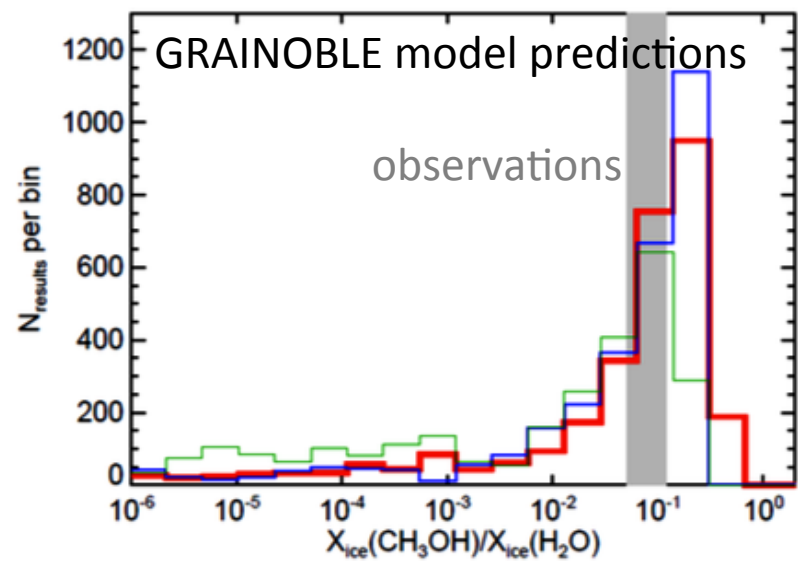


HYDROGENATION ON THE GRAIN SURFACES

→ LABORATORY EXPERIMENTS + THEORY + ASTROMODELS



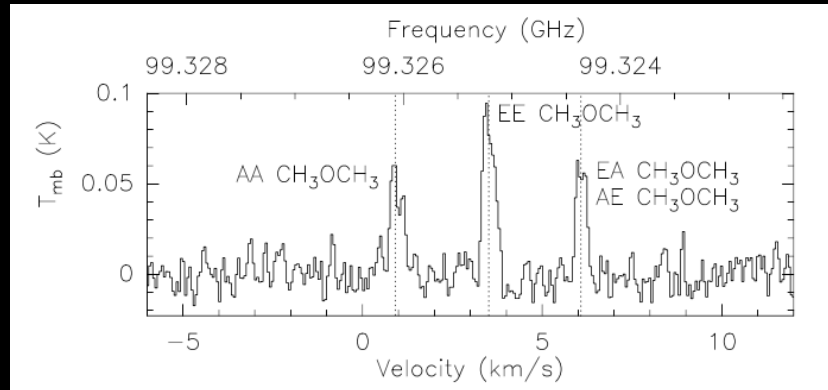
Example of the methanol formation on the grain surfaces



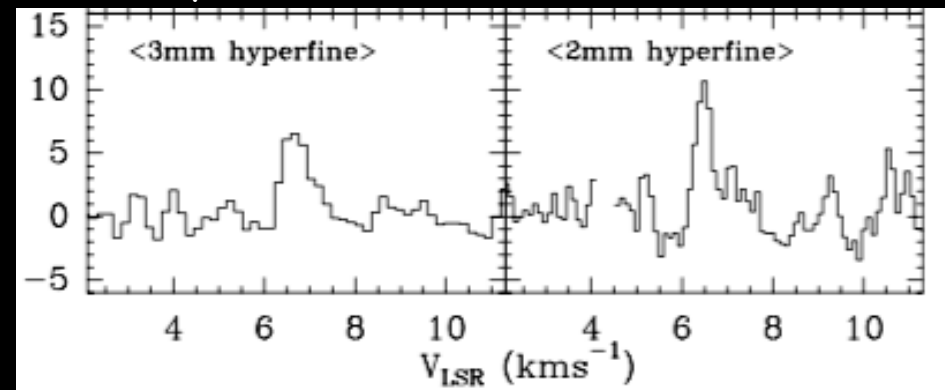
Rimola et al. 2014

2012-2014: DISCOVERY OF ACOMs in PSCs

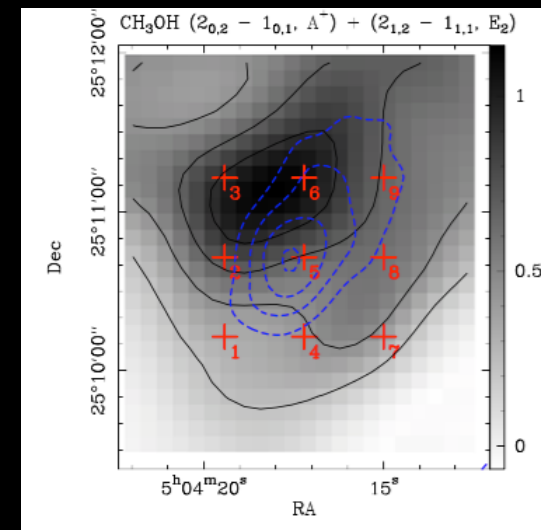
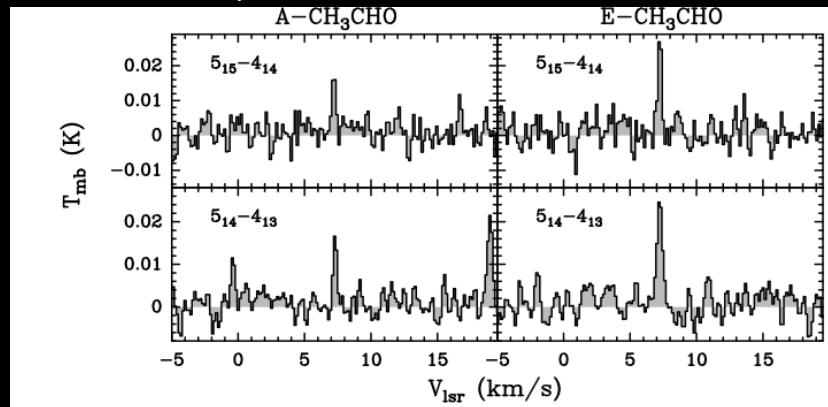
Dimethyl ether & methyl formate in L1689B;
Bacmann et al. 2012



Methoxy in B1-b; Cernicharo et al. 2012



Acetaldehyde in L1544; Vastel et al. 2014



Methanol in L1544; Bizzocchi et al. 2014

3. WHAT IS THE ACOMs ABUNDANCE?

SPECIES	ABUNDANCE (wrt H ₂)		
	L1544	L1689B*	B1-b*
CH ₃ OH	≈6x10 ⁻⁹		≈3x10 ⁻⁹
CH ₃ CHO	≈1x10 ⁻¹⁰	≈10 ⁻¹⁰	
HCOOH	≈1x10 ⁻¹⁰		
H ₂ CCO	≈1x10 ⁻⁹	≈10 ⁻¹⁰	
CH ₃ CCH	≈5x10 ⁻⁹		
CH ₃ OCH ₃	≤2x10 ⁻¹⁰	≈10 ⁻¹⁰	≈1x10 ⁻¹¹
HCOOCH ₃	≤2x10 ⁻⁹	≈10 ⁻¹⁰	≈1x10 ⁻¹¹
CH ₃ O	≤2x10 ⁻¹⁰		≈2x10 ⁻¹²

*Note: the abundances are computed assuming that the emission arises from the whole core

3. WHAT IS THE ACOMs ABUNDANCE?

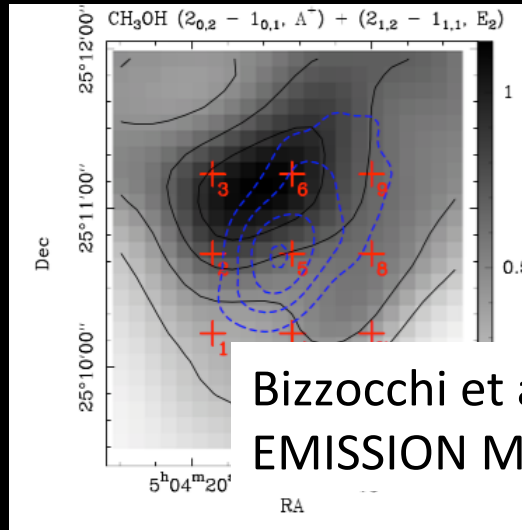
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HCOOH	≈1x10 ⁻¹⁰		
H ₂ CCO	≈10 ⁻¹⁰		
CH ₃ CCH	≈10 ⁻¹⁰		
CH ₃ OCH ₃	≤10 ⁻¹⁰	≈10 ⁻¹⁰	≈1x10 ⁻¹¹
HCOOCH	≈10 ⁻¹⁰		≈1x10 ⁻¹¹



→ PROBLEM

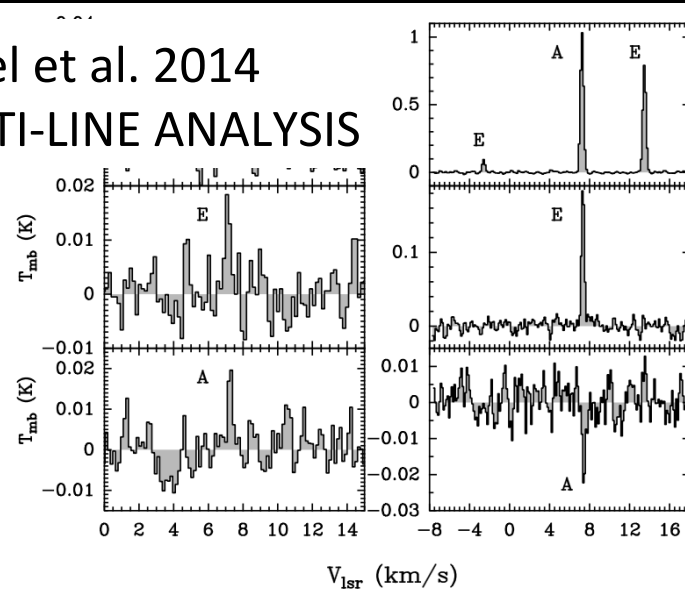
- 1) HOW THOSE LARGER MOLECULES ARE FORMED?? (no tunnel effect! No diffusion of heavy elements at 10K!)
- 2) WHY ARE THEY IN THE GAS PHASE??

1. WHERE DOES THE ACOMs EMISSION COME FROM? ANALYSIS OF METHANOL IN L1544



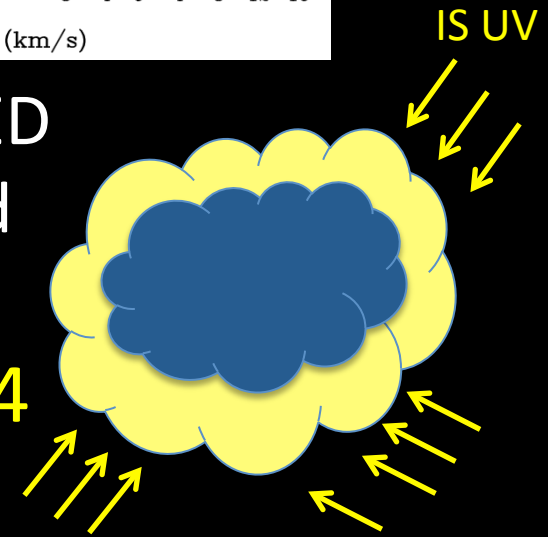
Bizzocchi et al. 2014
EMISSION MAP

Vastel et al. 2014
MULTI-LINE ANALYSIS

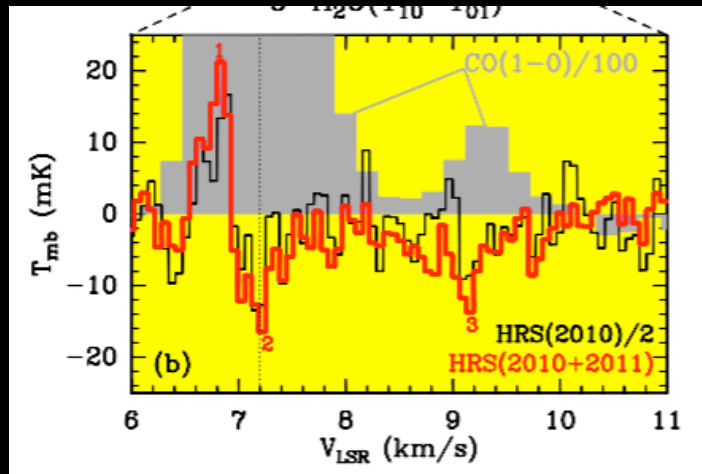


METHANOL EMISSION FROM AN EXTENDED
($>30''$) REGION WHERE $n_H \approx 3 \times 10^4 \text{ cm}^{-3}$ and
 $T \approx 10\text{K}$

→ FROM THE BORDER OF THE L1544
CONDENSATION



2. WHY ARE ACOMs IN THE GAS PHASE?



WATER IS ABUNDANT ($\approx 10^{-6}$) AT THE BORDER OF THE L1544 CONDENSATION
→ **PHOTODESORBED** BY THE IS FUV PHOTONS FROM THE ICE MANTLES?

Caselli et al. 2012; Quinard et al. 2015

a) IS METHANOL PHOTODESORBED FROM THE ICES ?

experiments (Minissale et al. 2016) suggest $< 1\%$ (when photodesorbed methanol is broken in parts...)

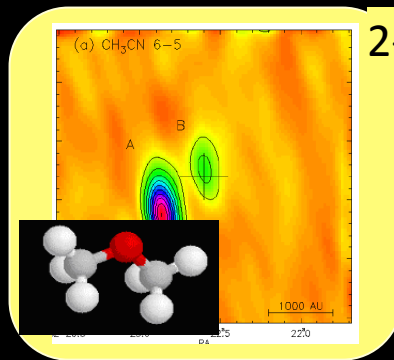
b) IS IT FORMED IN THE GAS FROM OTHER PHOTODESORBED SPECIES?

no known reactions



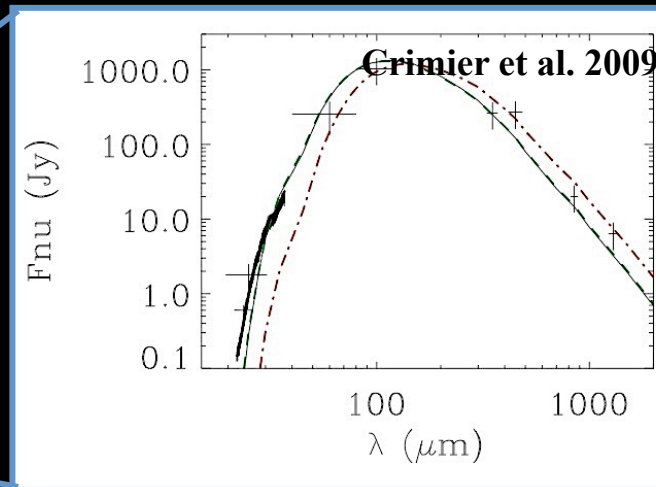
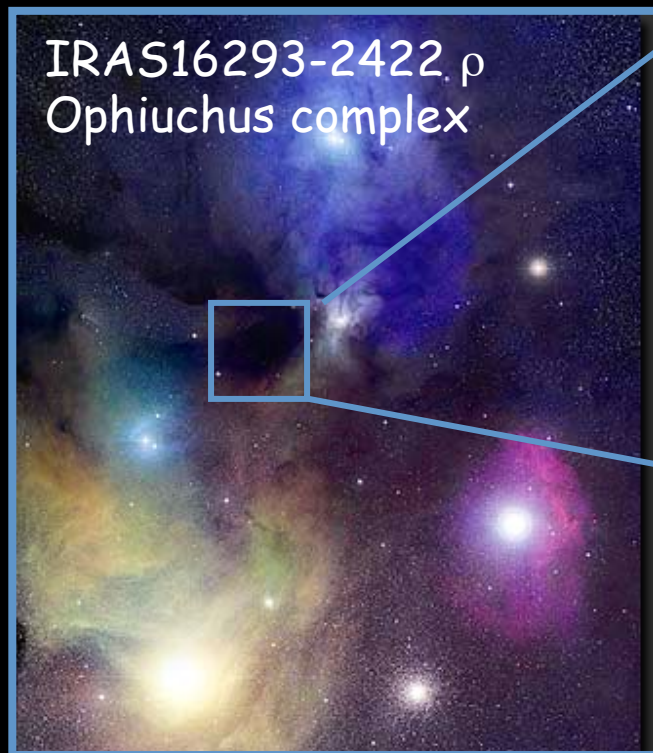
Vastel et al. 2014

CLASS 0 PROTOSTARS AND HOT CORINOS:



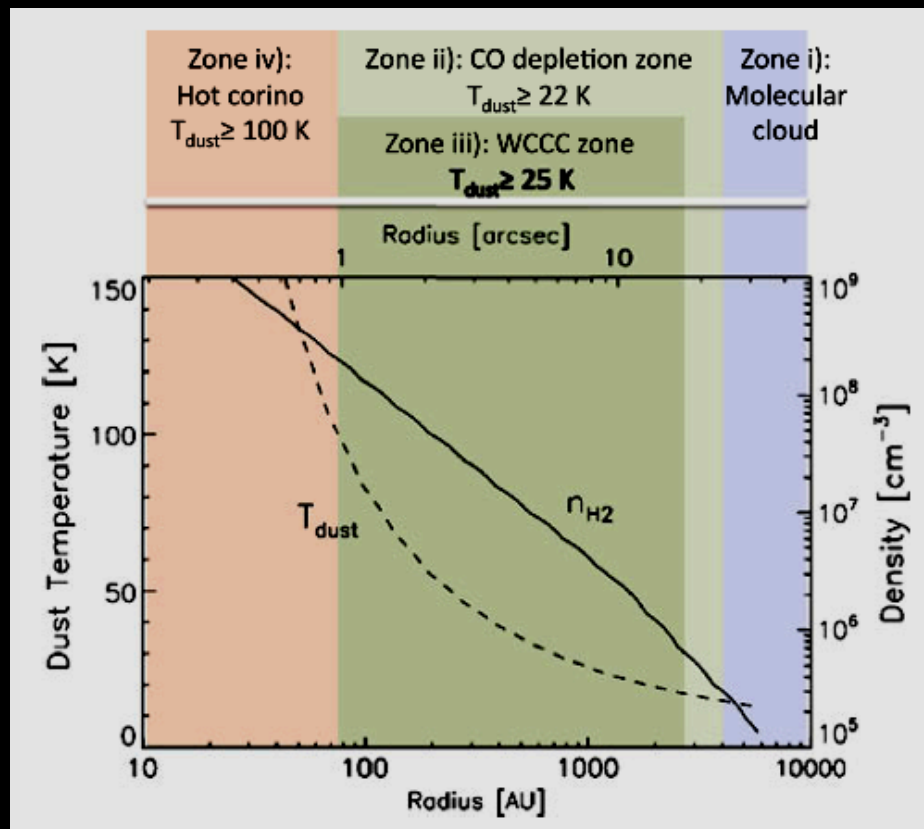
2- PROTOSTELLAR PHASE: collapsing, warm dense gas
FORMATION OF COMPLEX MOLECULES

CLASS 0 PROTOSTARS AND HOT CORINOS: the retail shops of ACOMS



CLASS 0 ARE COLD ($<30\text{K}$) SOURCES OF FEW M_{\odot} ,
EMITTING MOSTLY IN THE mm/submm

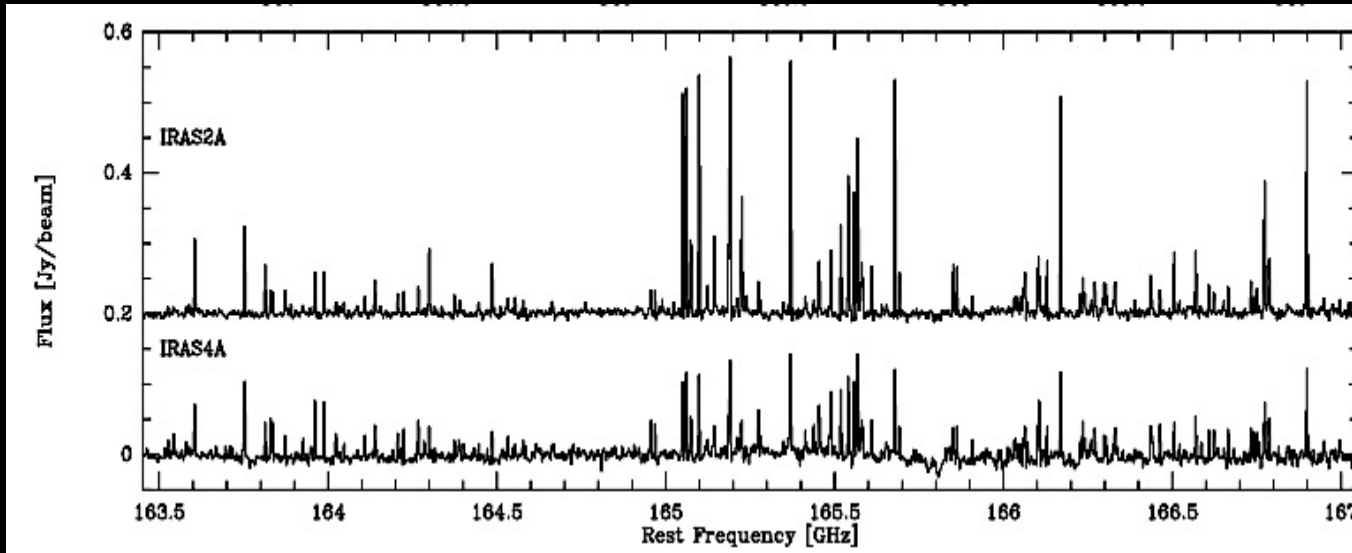
CLASS 0 PROTOSTARS AND HOT CORINOS: the retail shops of ACOMS



Caselli & Ceccarelli 2012

HOT CORINO: HOT (>100K) AND DENSE (10⁷cm⁻³) GAS
CHEMISTRY DOMINATED BY THE SUBLIMATION OF THE GRAIN
MANTLES => THE PREVIOUS OBJECT HISTORY

2003: DISCOVERY OF ACOMs IN HOT CORINOS



Taquet et al. 2015

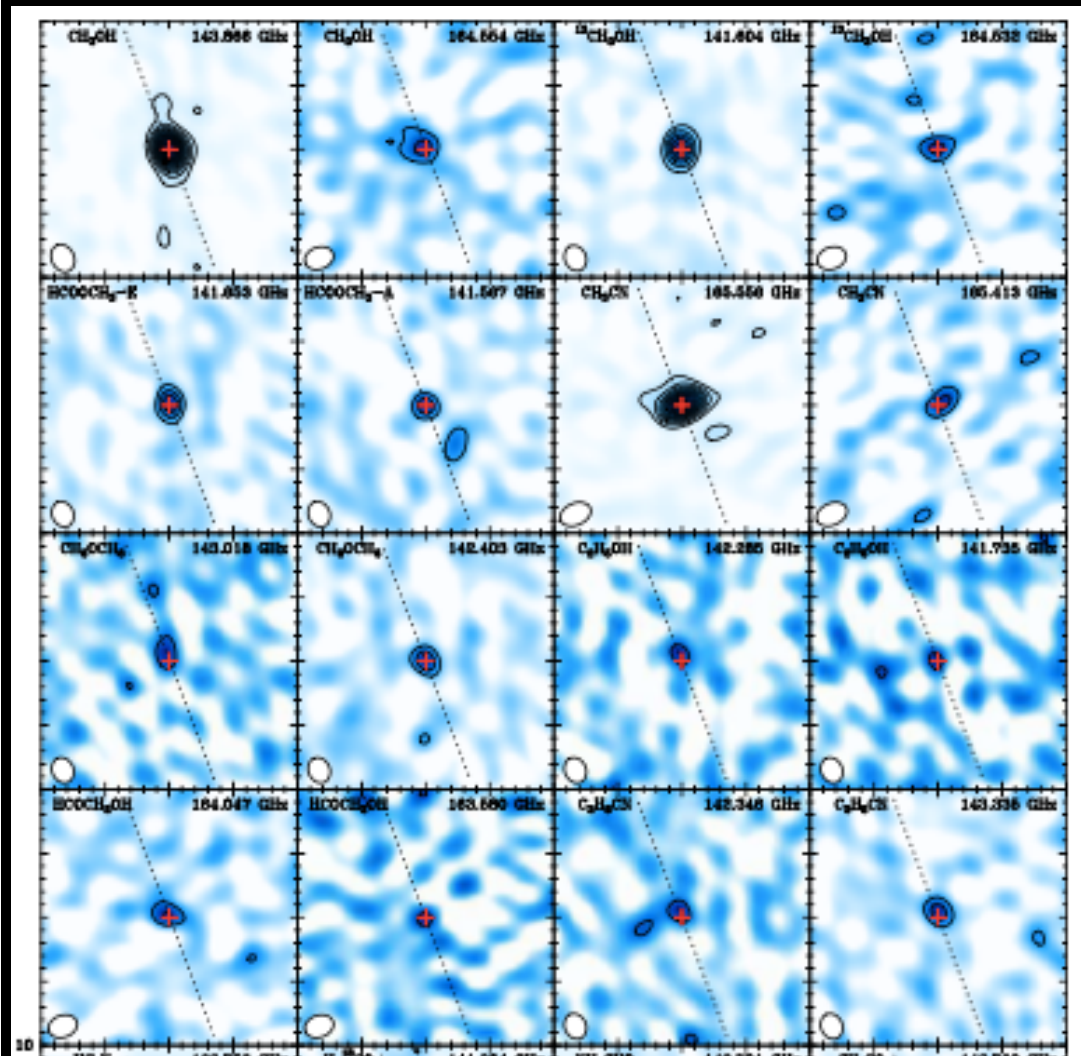
DETECTED ACOMs: METHYL FORMATE, DIMETHYL ETHER, FORMIC ACID, METHYL CYANIDE, ETHYL CYANIDE, GLYCOLALDEHYDE, FORMAMIDE

(Cazaux et al. 2003, Bottinelli et al. 2004, 2006; Jorgensen et al. 2012; Kahane et al. 2013; Coutens et al. 2015; Taquet et al. 2015...)

=> BUT NOTE: ONLY 5 HOT CORINOS KNOWN SO FAR <=

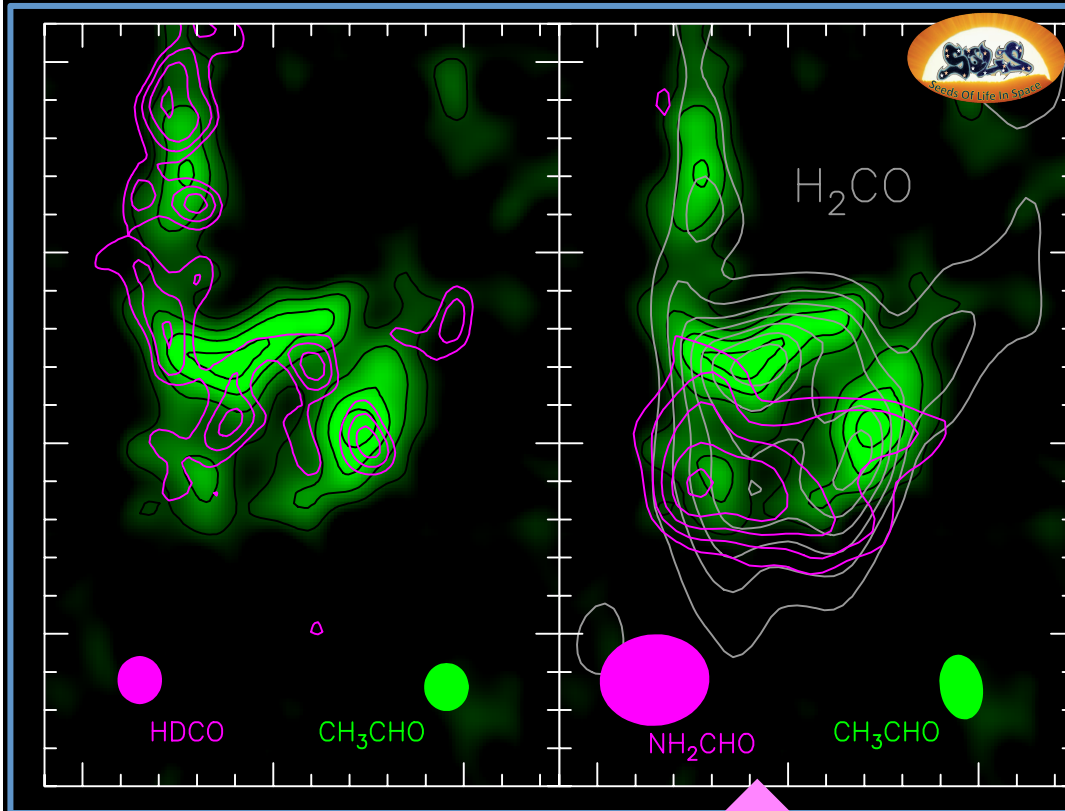
WHERE DOES THE ACOMs EMISSION COME FROM?

DEFINITELY FROM
THE SMALL (≤ 100 AU)
HOT CORINO REGION

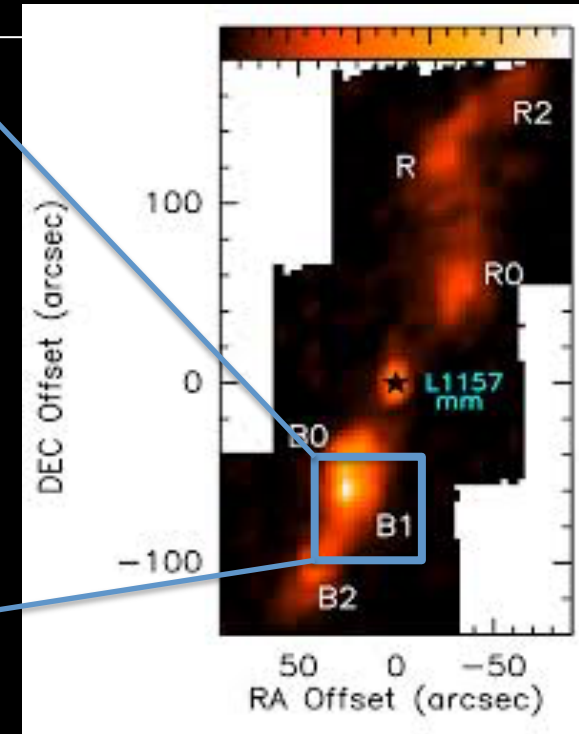


Taquet et al. 2015

WHERE DOES THE ACOMs EMISSION COME FROM?



SOMETIMES IN SHOCK SITES



H₂CO, HDCO: Fontani et al. 2015

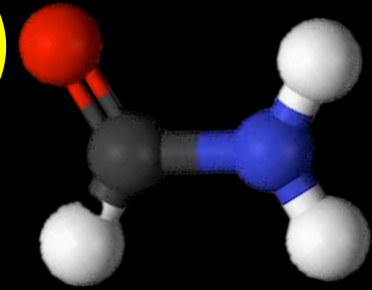
CH₃CHO: Codella et al. 2015

NH₂CHO: Codella et al. 2016

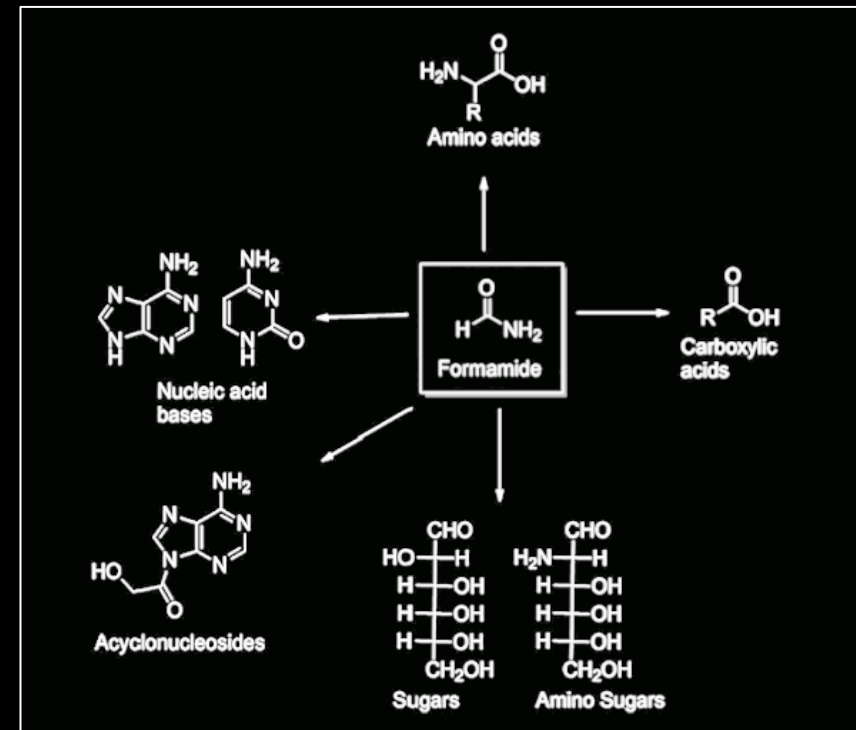
FORMAMIDE (NH₂CHO)

FORMAMIDE (NH_2CHO)

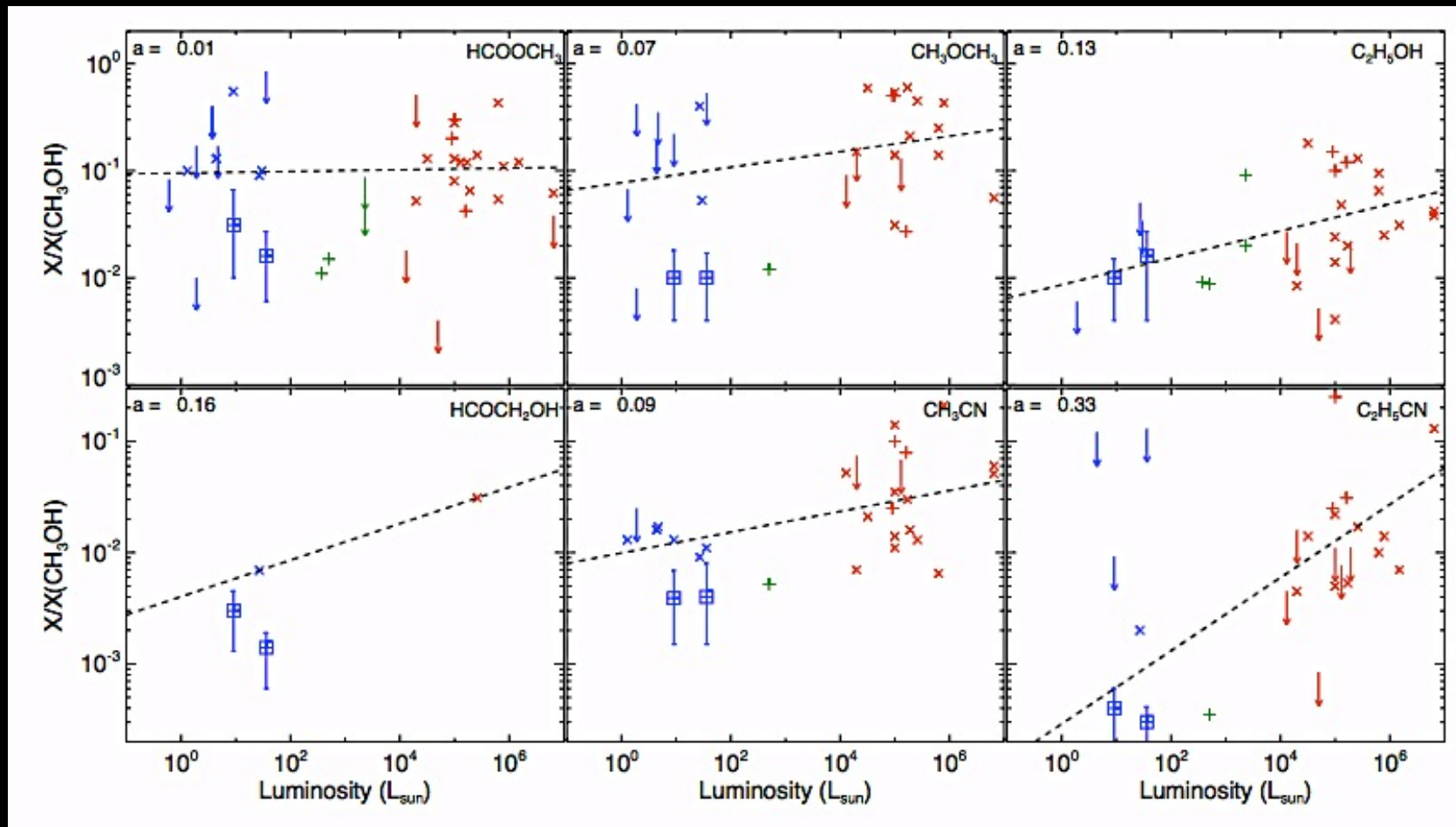
A KEY TO LIFE?



Formamide (NH_2CHO) may have been the **starting point** for the prebiotic synthesis of **both metabolic and genetic species** (Saladino et al. 2012).



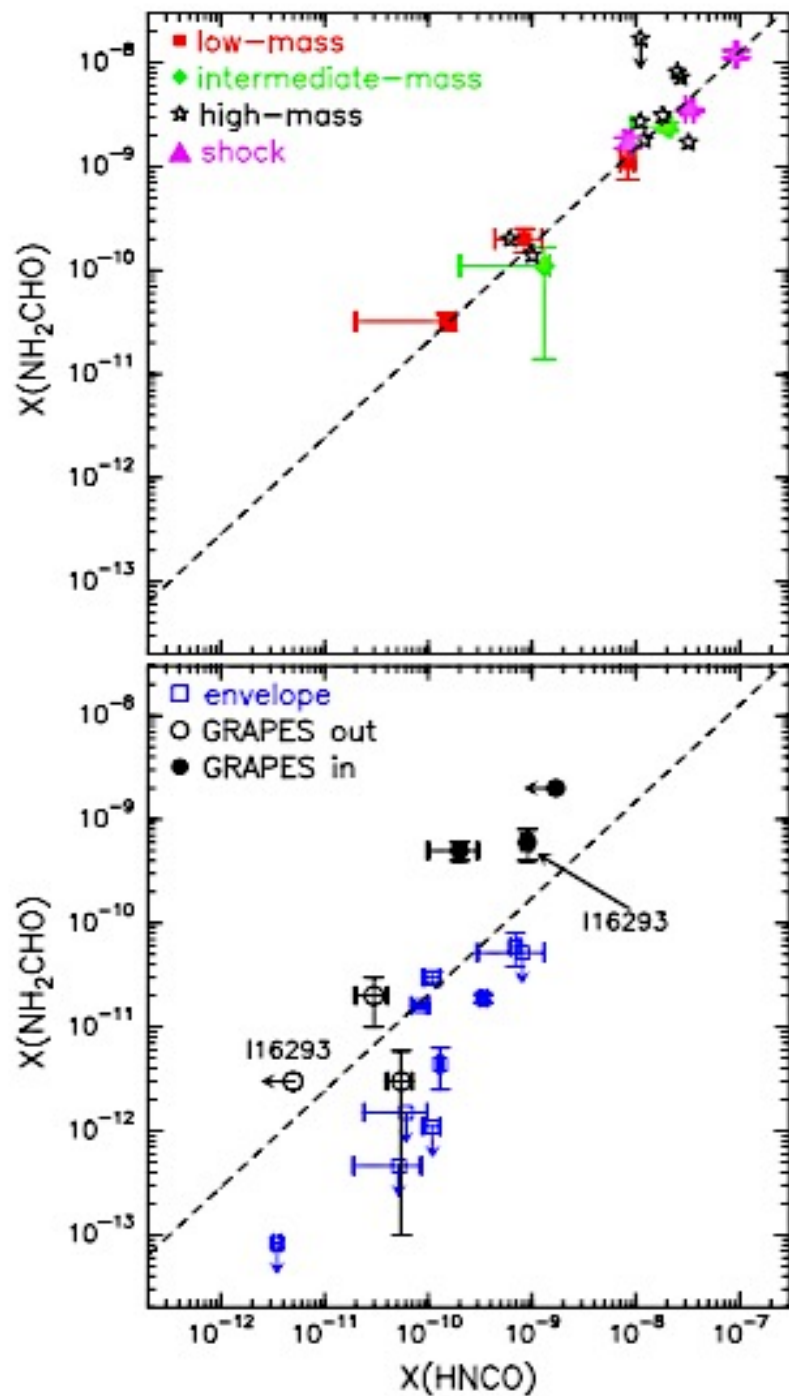
WHAT IS THE ACOMs ABUNDANCE?



A CONSIDERABLE FRACTION (few %) OF GASEOUS CARBON (Taquet et al. 2015; see also Jaber et al. 2014 and Oberg et al. 2014)

CORRELATIONS AMONG ACOMs ABUNDANCES

HELP TO INDIVIDUATE
MOTHER/DAUGHTER OR SISTER SPECIES
→ THEIR FORMATION MECHANISM



Lopez-Sepulcre et al. 2015

FORMAMIDE VS ISOCYANIC ACID

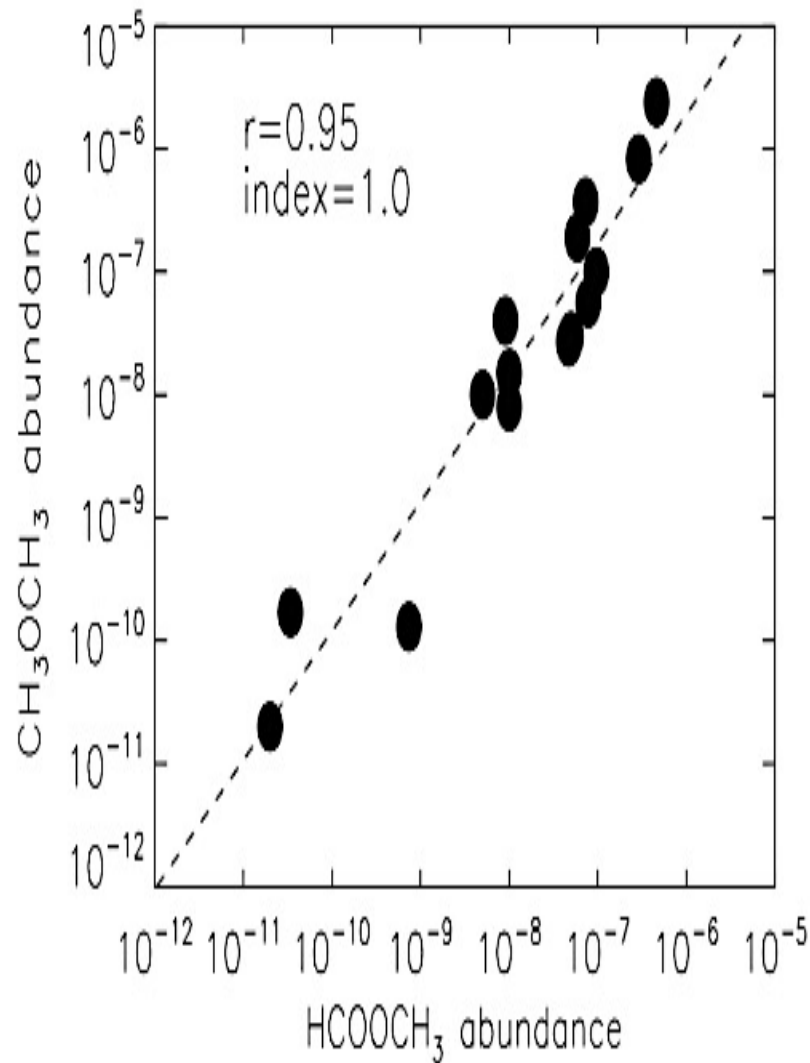
LABORATORY EXPERIMENTS
RULE OUT HYDROGENATION
OF HCNO \rightarrow NH₂CHO (Noble et al.
2015)



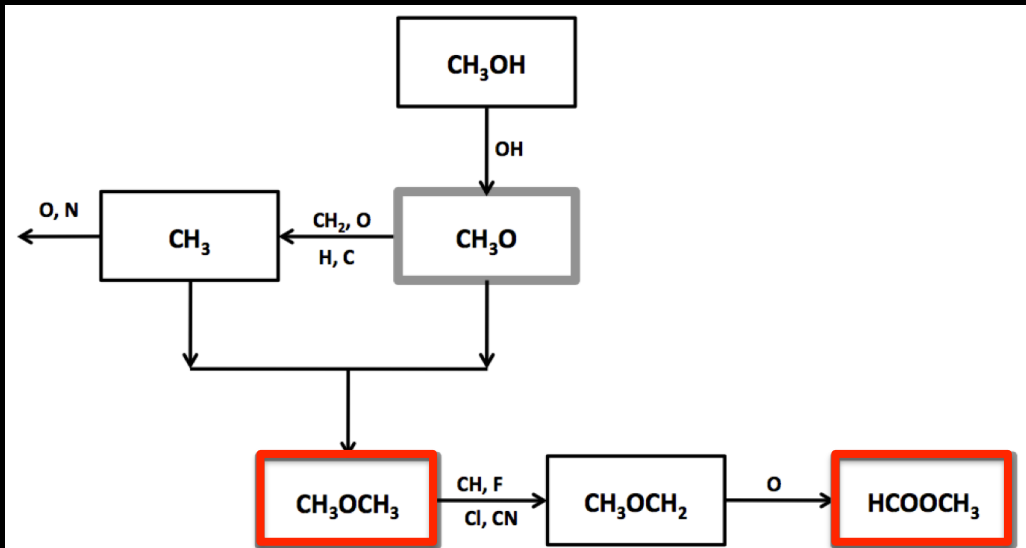
- POSSIBLE GAS PHASE ROUTE
NH₂ + H₂CO \rightarrow NH₂CHO (Barone
et al. 2015)

METHYL FORMATE VS DIMETHYL ETHER

PROPOSED GAS PHASE
ROUTE (Balucani et al. 2015)



Jaber et al. 2014



IN SUMMARY

ACOMs ARE PRESENT IN ENVIRONMENTS WITH TEMPERATURES RANGING FROM 10K TO >100K

HOW ARE ACOMs FORMED?

(and how are they injected into the gas phase?)



HOW ARE ACOMs FORMED?

IN VOGUE UNTIL 2003/5

GAS PHASE

reactions in the gas phase, often started by the injection of hydrogenated molecules formed on the grain surfaces

Step 1

H₂CO
CH₃OH NH₃

Step 2

ice
sublimation

Step 3

HCOCH₂OH
CH₃OCH₃
NH₂CHO

IN VOGUE UNTIL 2012/3

GRAIN SURFACES

reactions on the grain surfaces between radicals during the warm-up of the dust; radicals are formed in the cold phase

Step 1

H₂CO
CH₃OH NH₃

Step 2

HCOCH₂OH
CH₃OCH₃
NH₂CHO

Step 3

ice
sublimation

10 K

100 K

10 K

100 K

TEMPERATURE

IN BOTH SCHEMES MOLECULES ARE THERMALLY DESORBED FROM GRAIN MANTLES at 100K

THE GRAIN SURFACES THEORY

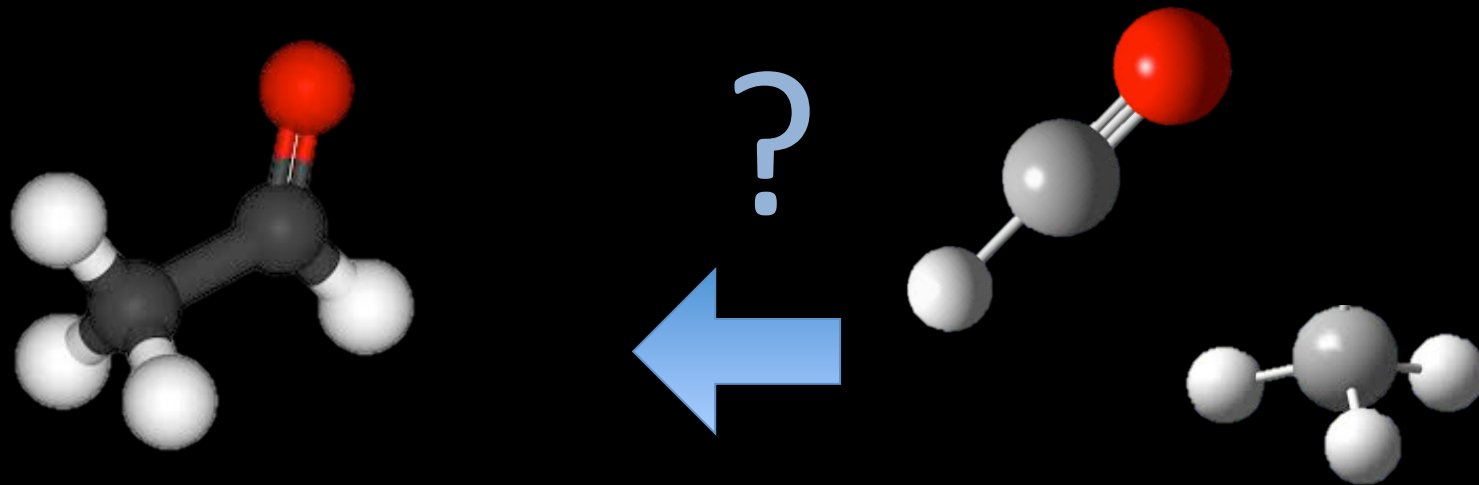
→ SEVERAL LABORATORY EXPERIMENTS SHOW THE FORMATION OF ACOMs IN FUV ILLUMINATED ICES
QUESTION: ARE THEY APPLICABLE TO THE ISM CONDITIONS?

CRITICAL ASSUMPTIONS IN THE ISM MODELS:

- ⇒ FUV CREATE RADICALS TRAPPED IN THE ICES
(current models assume gas-phase production rates)
- ⇒ AT $T_{\text{dust}} > 30\text{K}$ RADICALS HAVE ENOUGH ENERGY TO MOVE AROUND
ACOMs ARE DETECTED ALSO IN COLD 10K REGIONS
- ⇒ RADICALS COMBINE AND FORM ACOMs
IS THIS TRUE?

RADICAL S COMBINATION INTO ACOMs

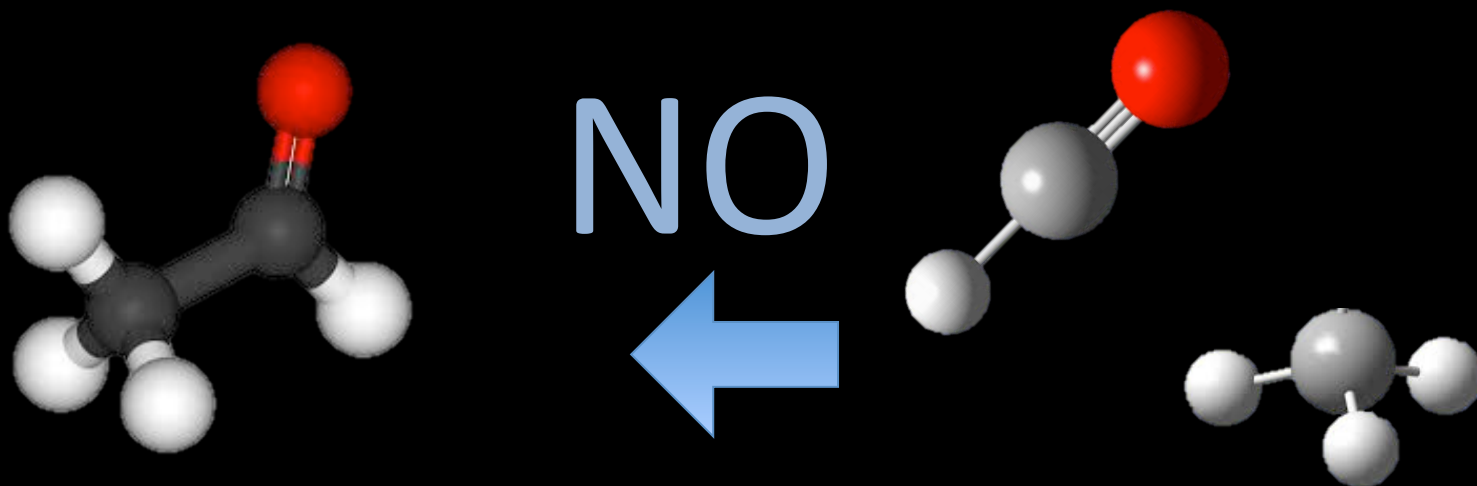
EXAMPLE OF ACETALDEHYDE



→ LABORATORY EXPERIMENTS + THEORY
ARE NECESSARY TO UNDERSTAND WHAT HAPPENS

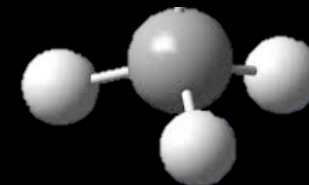
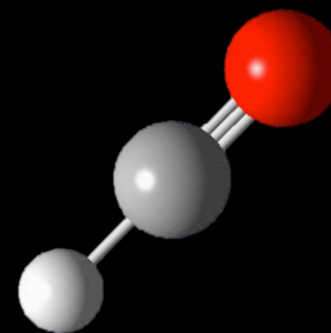
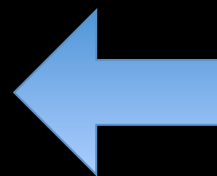
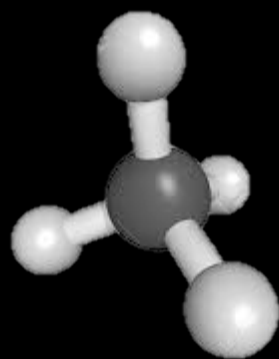
RADICAL S COMBINATION INTO ACOMs

EXAMPLE OF ACETALDEHYDE



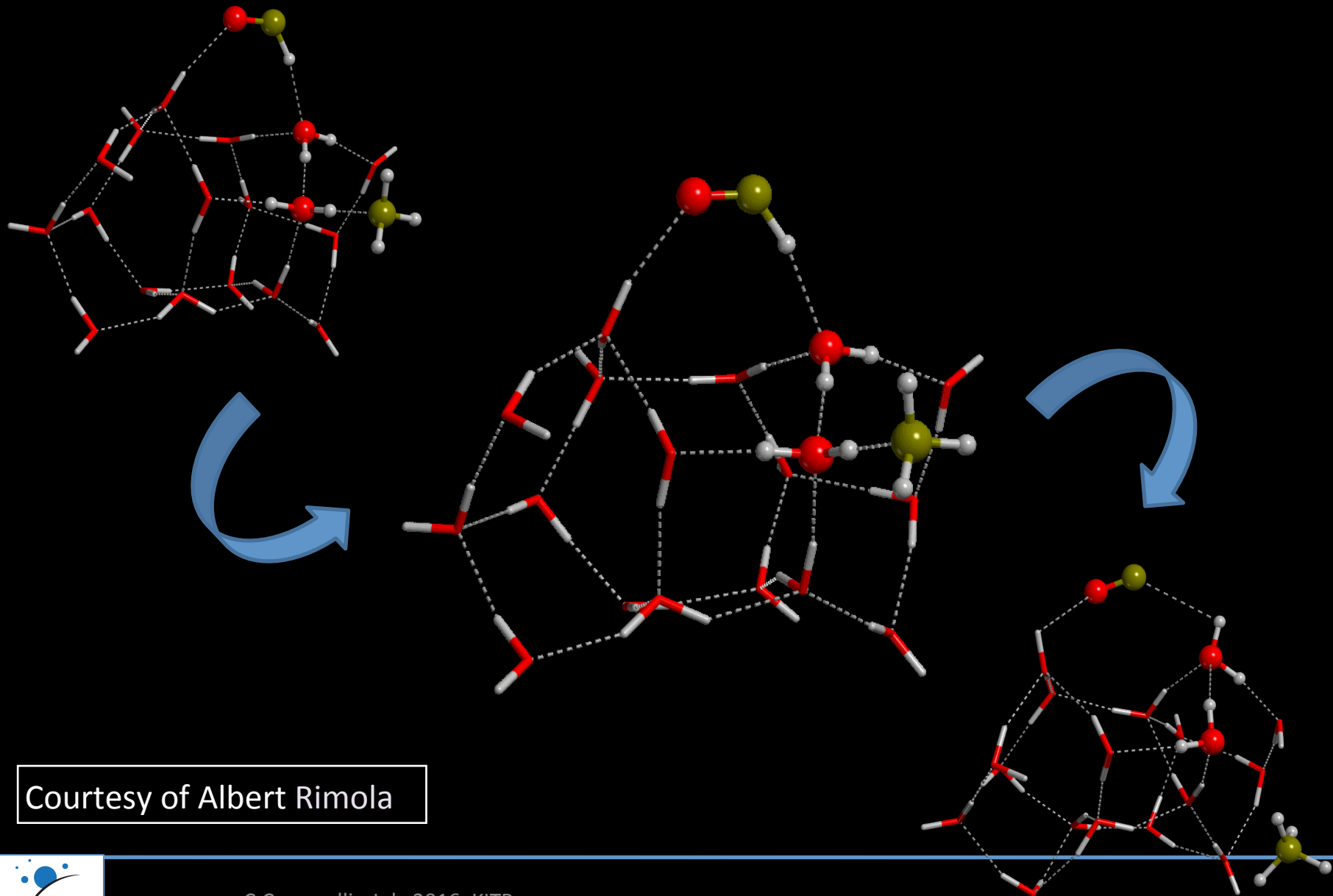
RADICAL S COMBINATION INTO ACOMs

EXAMPLE OF ACETALDEHYDE



Enrique-Romero, Rimola,
Ceccarelli, Balucani 2016

Acetaldehyde Formation. $\text{CH}_3 + \text{HCO}$ reaction on grain surfaces



Courtesy of Albert Rimola

THE GAS PHASE THEORY

HYPOTHESIS/ASSUMPTIONS:

1. AT $T_{\text{dust}} \leq 20\text{K}$, FROZEN O/CO/N/C ARE HYDROGENATED
=> SIMPLE MOLECULES FORMATION: H_2O , H_2CO ,
 CH_3OH , NH_3 , CH_4 ...
2. (A FRACTION OF) THESE ICED MOLECULES SUBLIMATE,
REACT IN THE GAS AND FORM ACOMs
IS IT TRUE? ..IN SOME CASES YES...
(BUT MUCH REMAINS TO BE DONE)

THE ROLE OF GAS PHASE REACTIONS IN THE ACOMs FORMATION, EVEN AT $\approx 10\text{K}$

ACOMs ARE CERTAINLY THE RESULTS OF THE
COMBINATION OF GAS PHASE AND SURFACE CHEMISTRY:

THE QUESTION IS:

WHO PLAYS WHEN?



SUMMARY

- ❖ ACOMs ARE PRESENT IN ENVIRONMENTS WITH TEMPERATURES RANGING FROM 10K TO >100K
- ❖ ACOMs ARE SURELY THE RESULT OF A COMBINED SURFACE + GAS PHASE CHEMISTRY
- ❖ VERY LIKELY, THE GAS PHASE CHEMISTRY ROLE HAS BEEN OVERLOOKED, THOUGH, AND NEEDS TO BE BETTER UNDERSTOOD
- ❖ **OBSERVATIONS + THEORY + EXPERIMENTS + MODELING ARE ABSOLUTELY NECESSARY TO UNDERSTAND WHAT'S GOING ON**

Take home message: ACOMs are easily synthesized in the ISM

WHAT NEXT (on our side) ?



IRAM NOEMA LARGE PROGRAM
(interferometer): **ACOMs**
PI: C.Ceccarelli & P.Caselli