

The Significance of RNA in Molecular Biology



Knowledge of RNA Structure



Causes diseases

**(RNA viruses: AIDS, colds
hepatitis, polio, etc.)**

**Catalyzes chemical reactions
(ribozymes)**

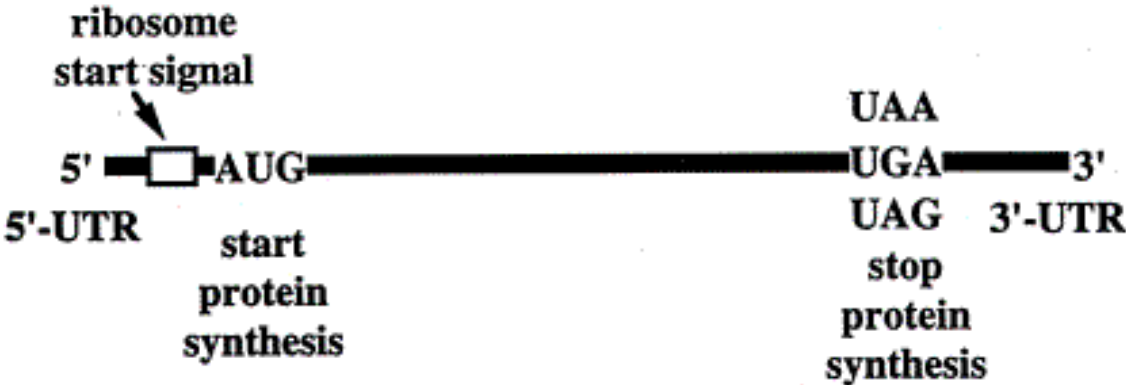
RNA

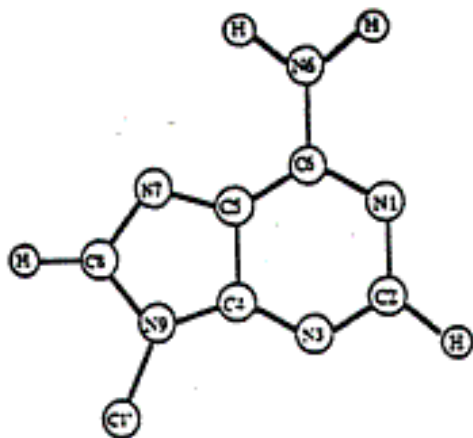
**Carries messages for
synthesis of all proteins.
(messenger RNAs)**

**Translates messages into protein
sequences.**

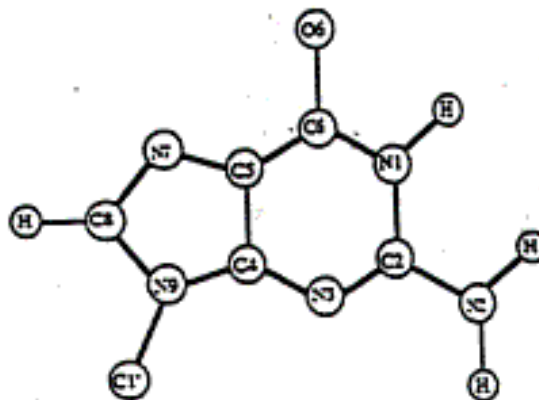
(transfer RNAs, ribosomal RNAs)

Translation of messenger RNA

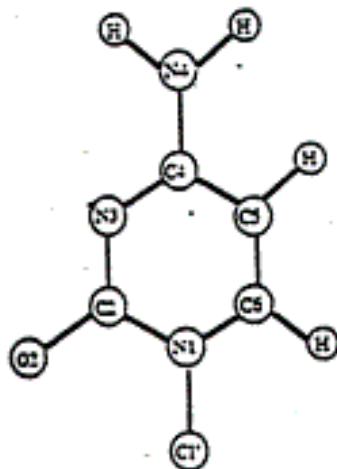




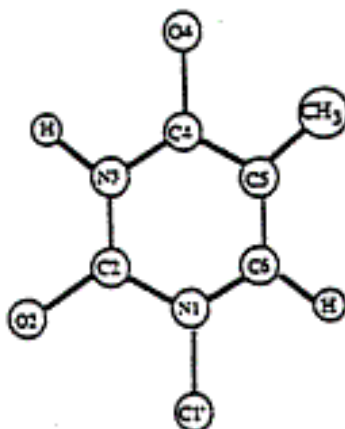
Adenine (A)



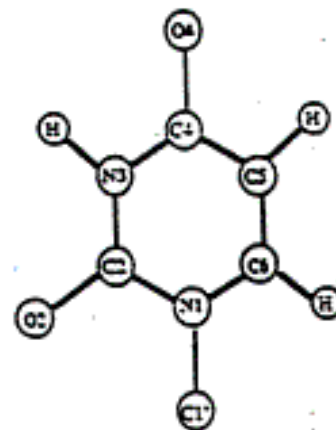
Guanine (G)



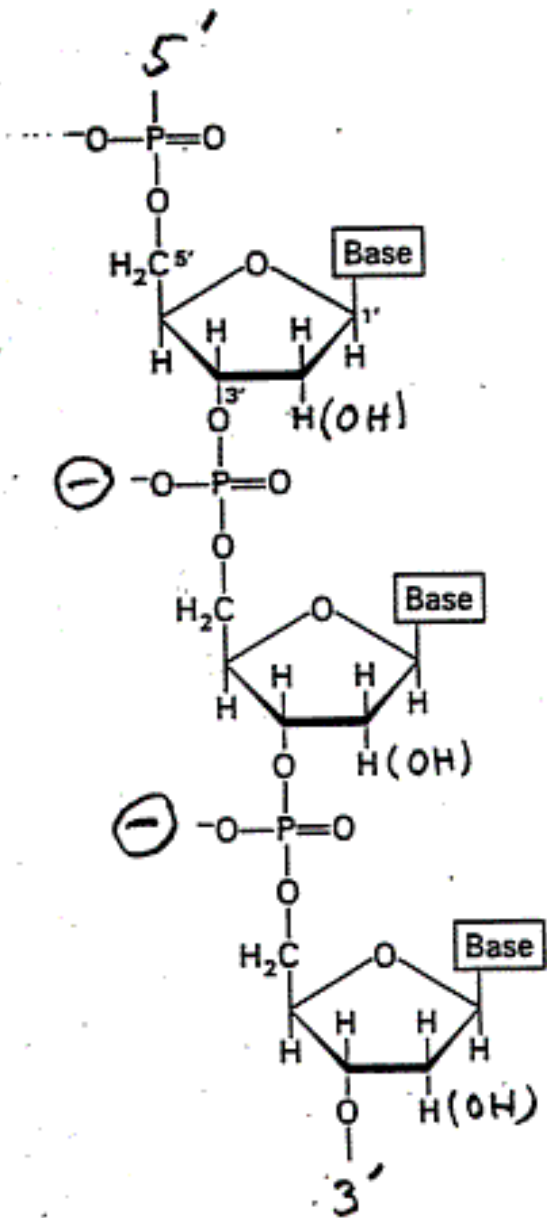
Cytosine (C)

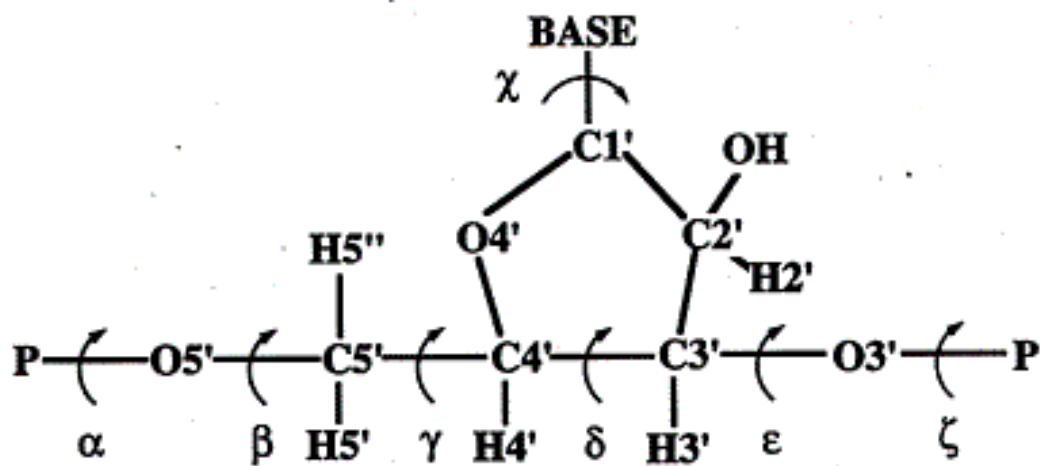


Thymine (T)



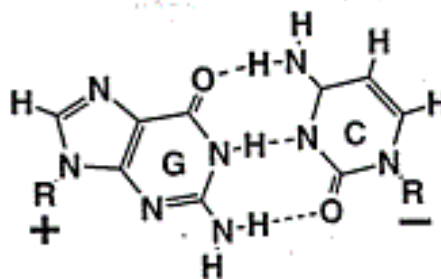
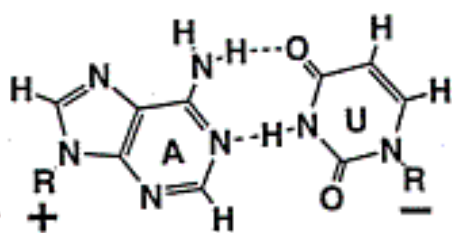
Uracil (U)



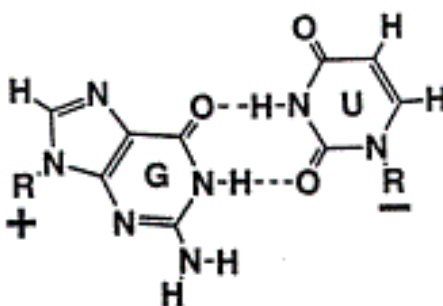


**Half the RNA folding problem is
already mainly solved**

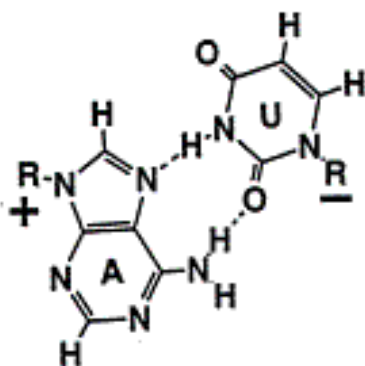
**The base pairing—the secondary structure—can
be predicted with good accuracy from
experimental thermodynamic data on double
strands, loops, and bulges.**



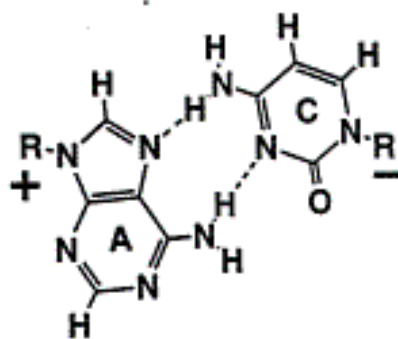
Watson-Crick



G•U Wobble



A•U Reverse Hoogsteen



A•C Reverse Hoogsteen

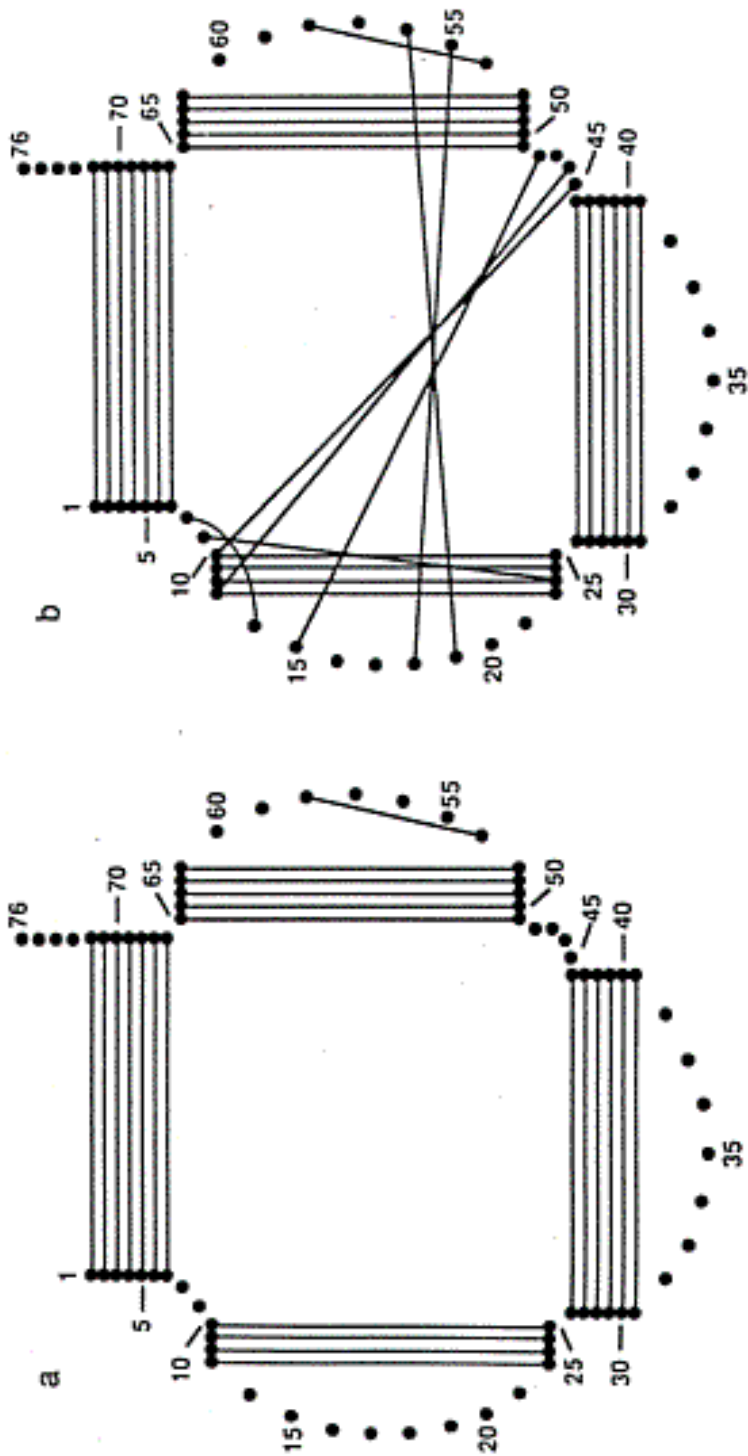
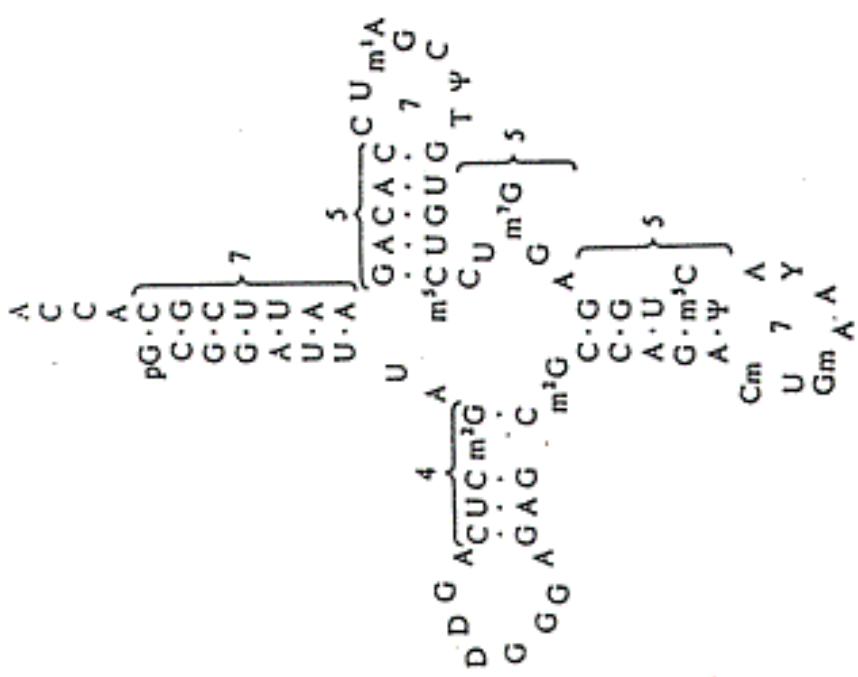
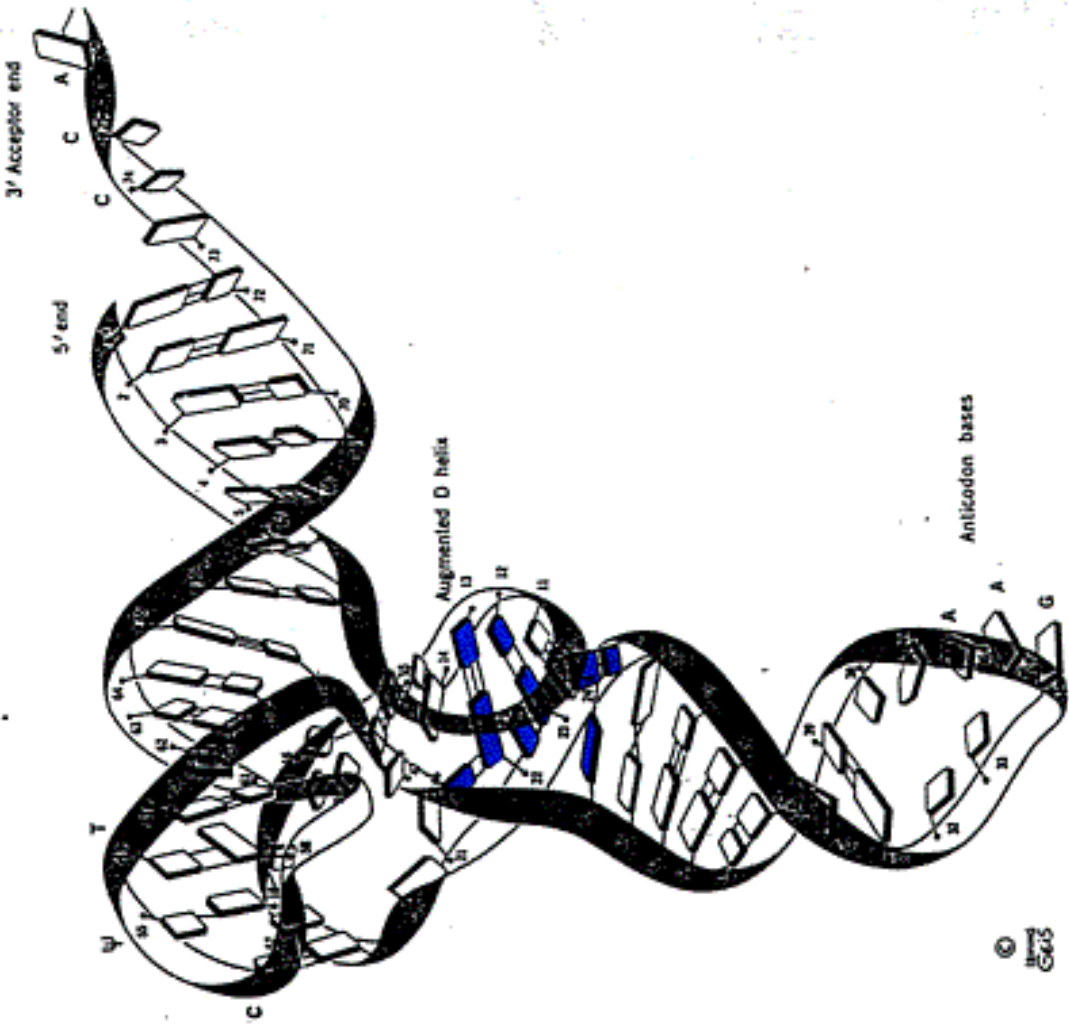


FIG. 1. The definition of secondary structure and tertiary structure in terms of chord crossing. The lines between points represent base-pairs. (a) The secondary structure of tRNA^{Phe}. (b) The secondary and tertiary structures of tRNA^{Phe}.



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Garavito

Secondary Structure → Tertiary Structure

The Gibbs free energy, G , of a closed system at constant temperature, T , and pressure, P , tends toward a minimum.

$$G = H - TS = E + PV - TS$$

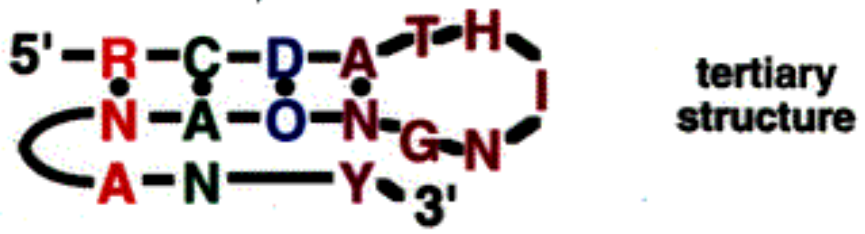
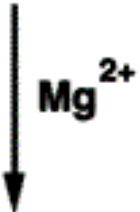
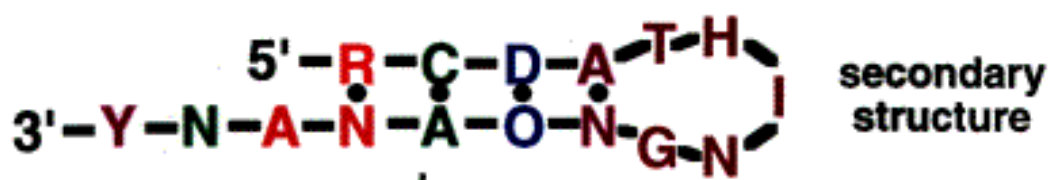
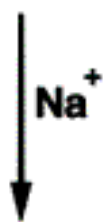
H = enthalpy

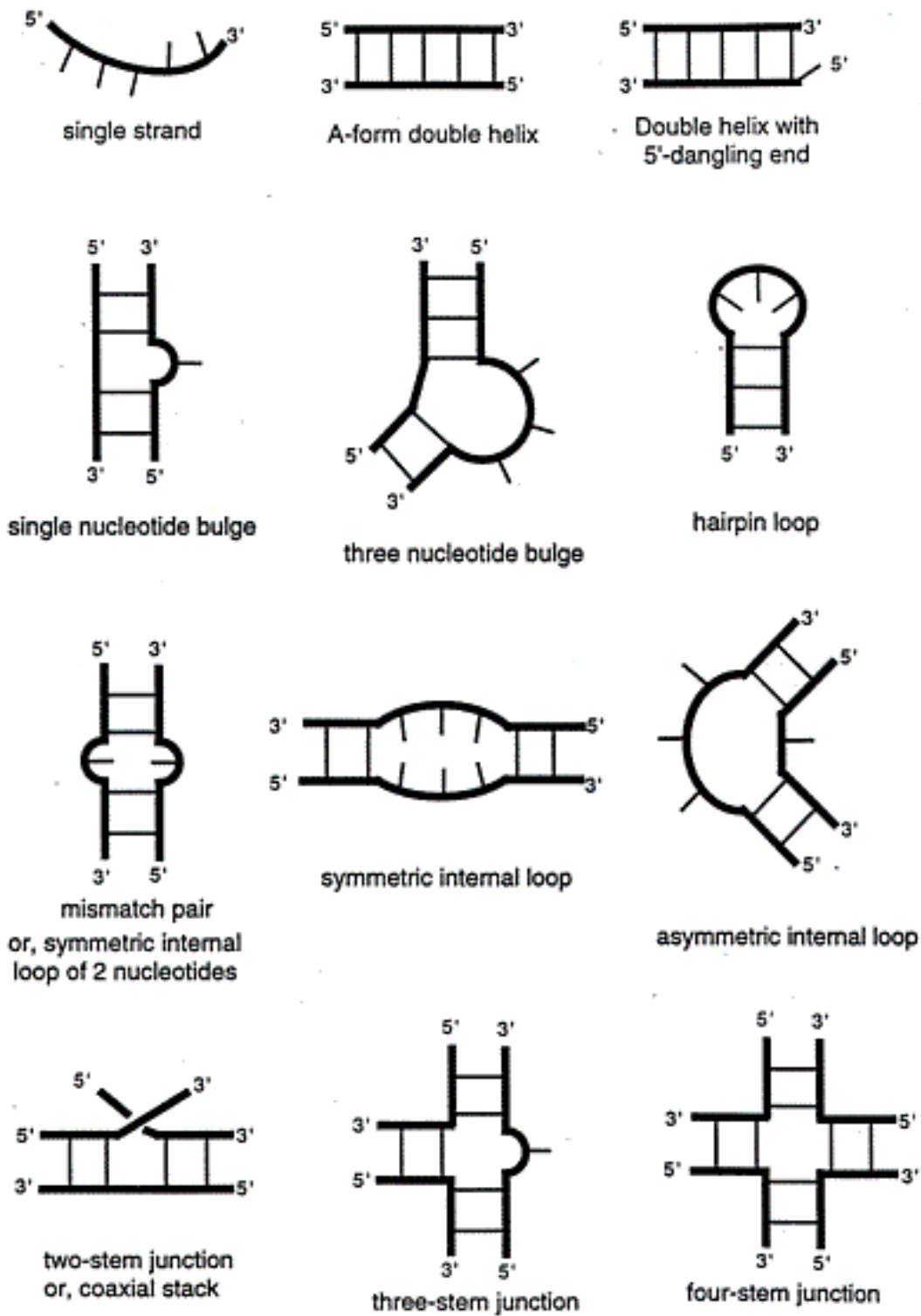
S = entropy

E = energy

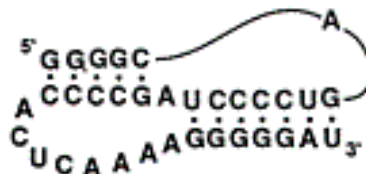
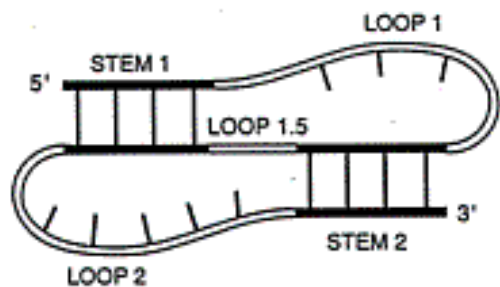
V = volume

5'-RCDATHINGNOANANY-3'



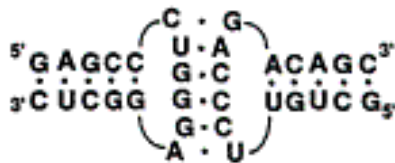
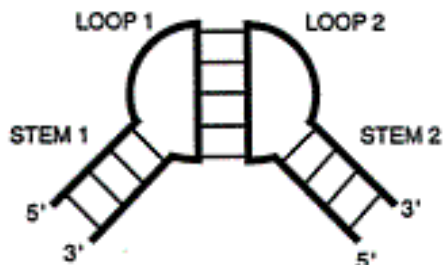


a)



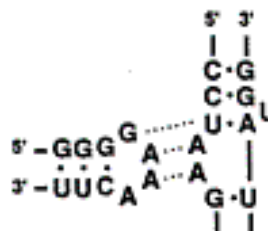
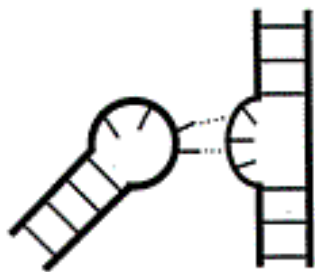
Pseudoknot

b)



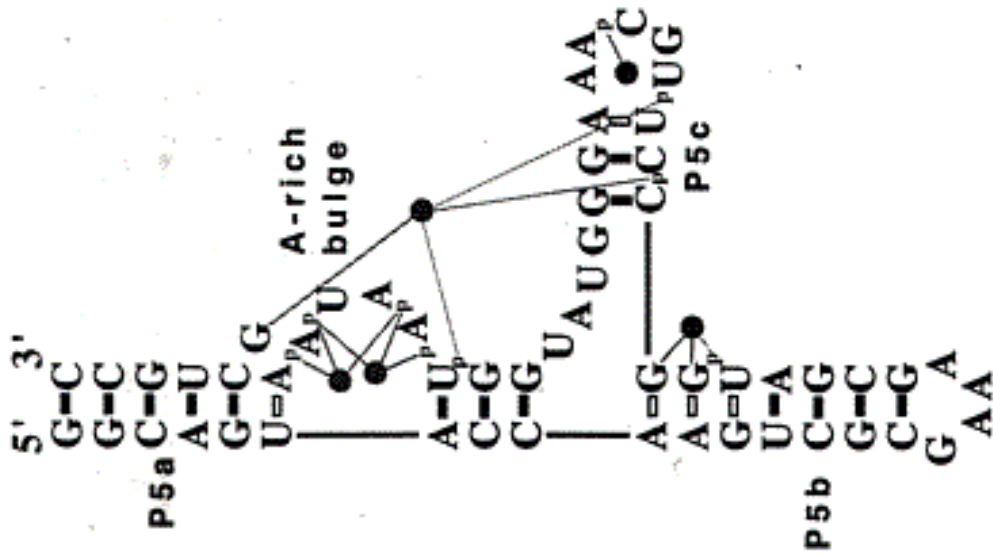
Kissing hairpins

c)



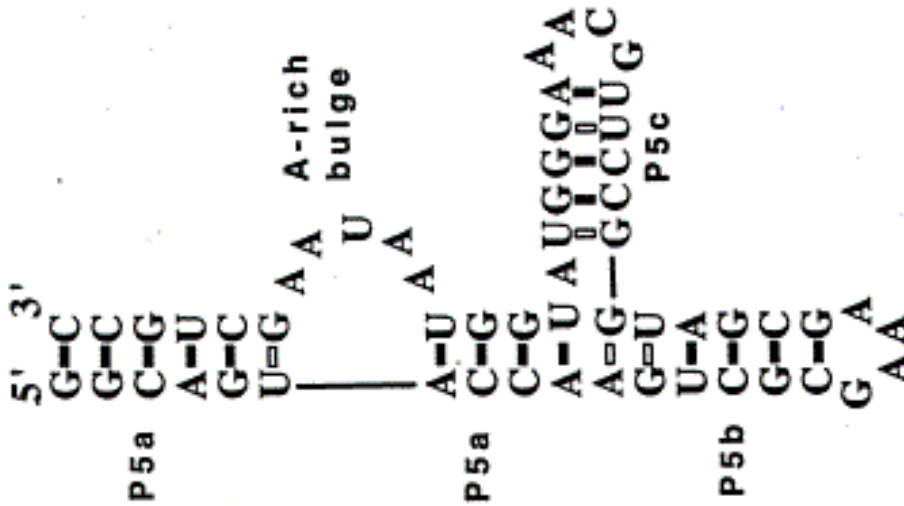
Hairpin loop - bulge contact

Secondary structure and ion binding by X ray

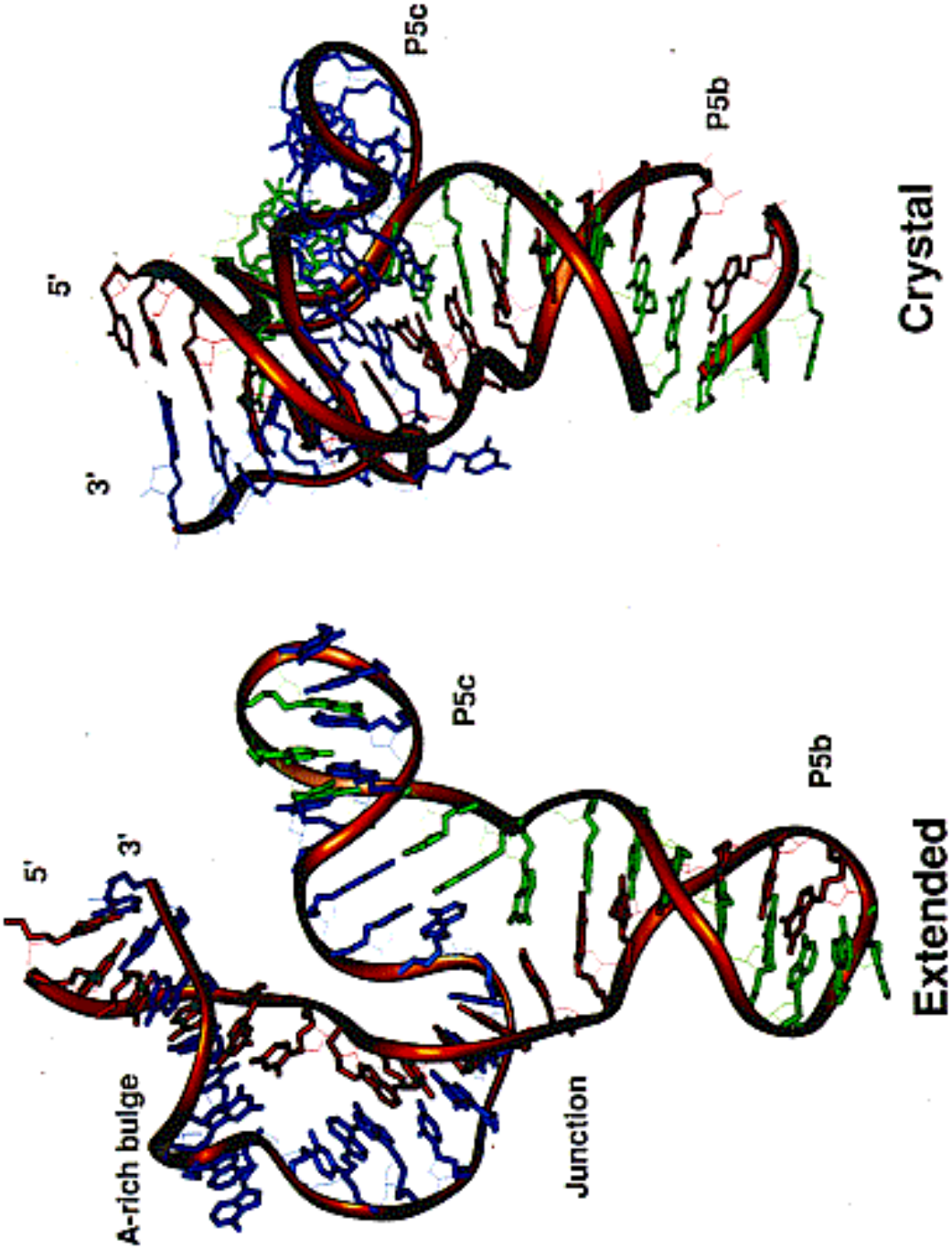


Phylogenetics prediction

p5abc secondary structure by NMR



Thermodynamics prediction



Coworkers

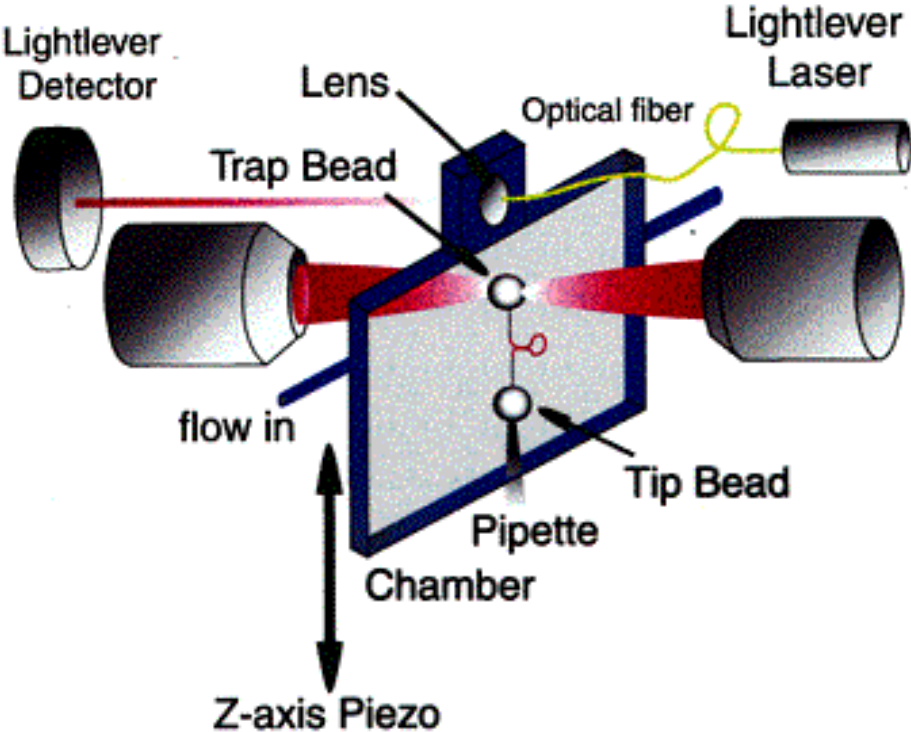
**Professor Carlos Bustamante, Departments of
Chemistry, Molecular and Cell Biology, and Physics,
and Howard Hughes Medical Institute, UCB**

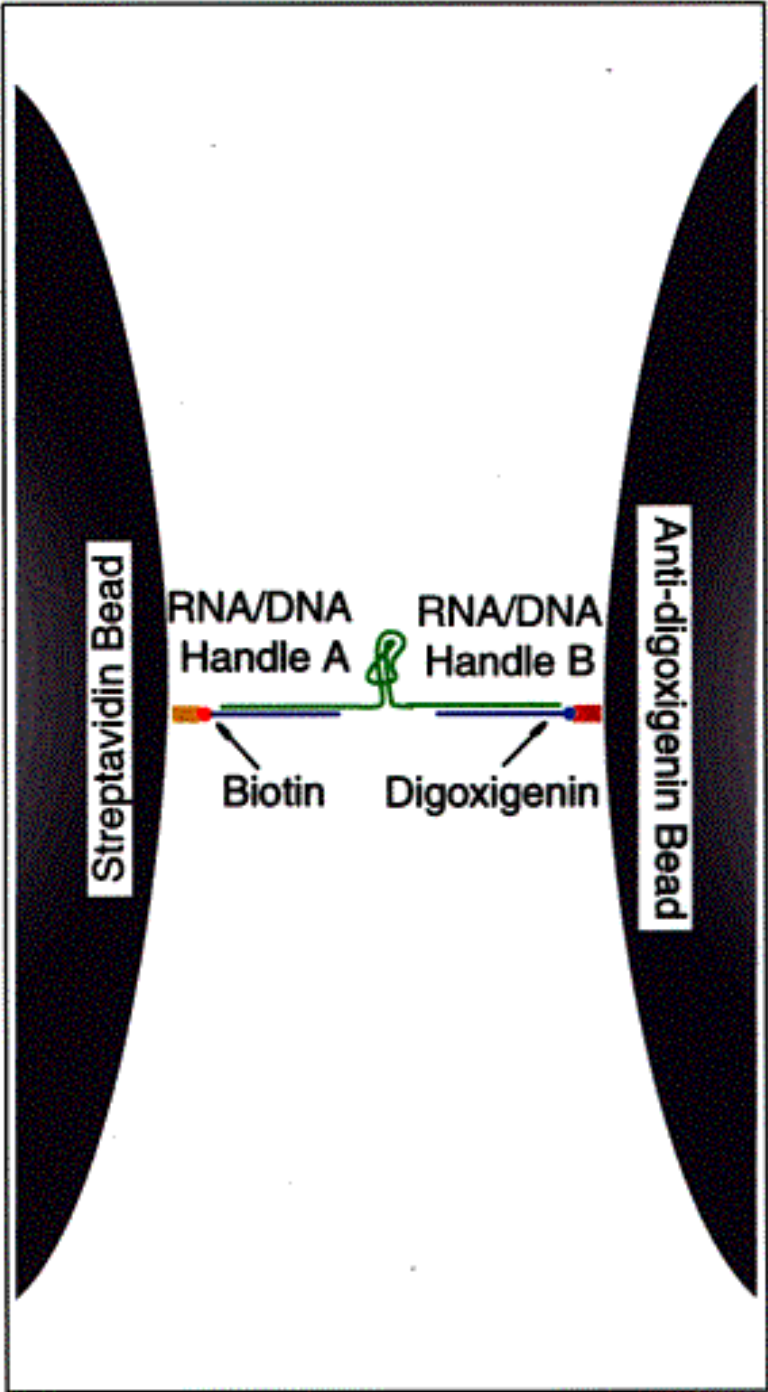
Dr. Delphine Collin, Department of Chemistry, UCB

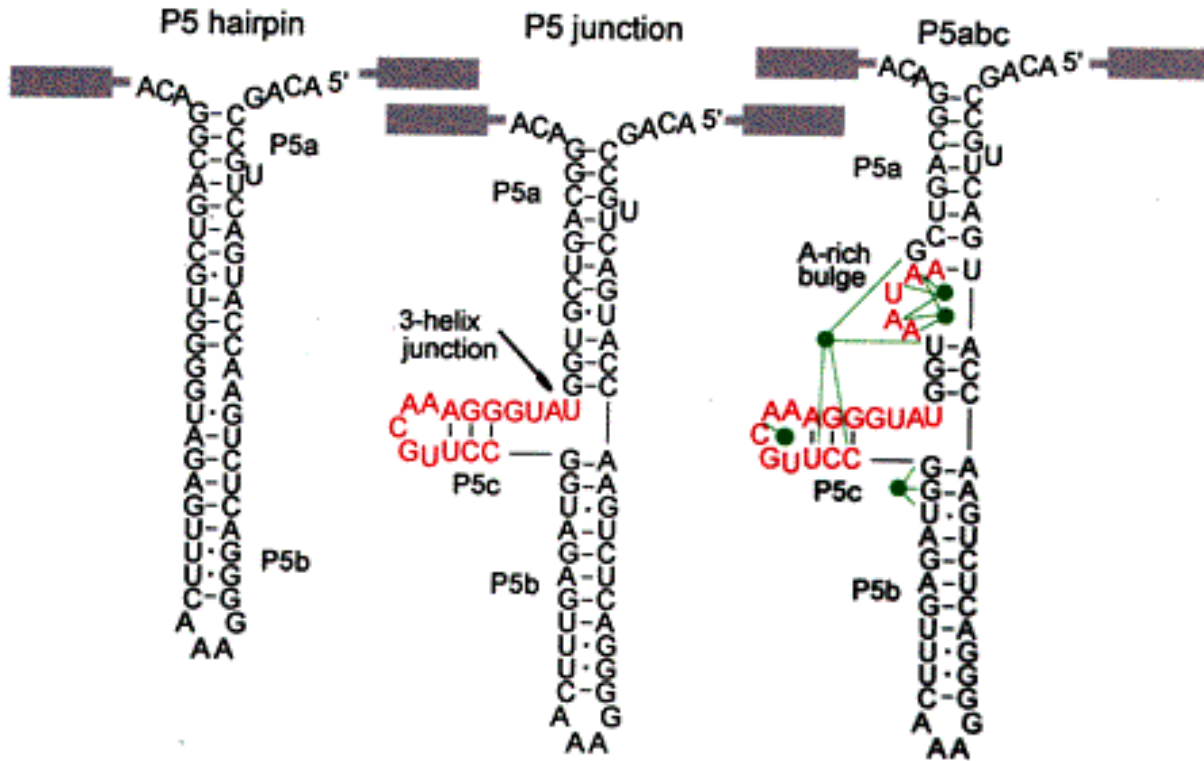
Dr. Jan Liphardt, Department of Chemistry, UCB

Dr. Bibiana Onoa, Department of Chemistry, UCB

Dr. Steven B. Smith, Department of Physics, UCB

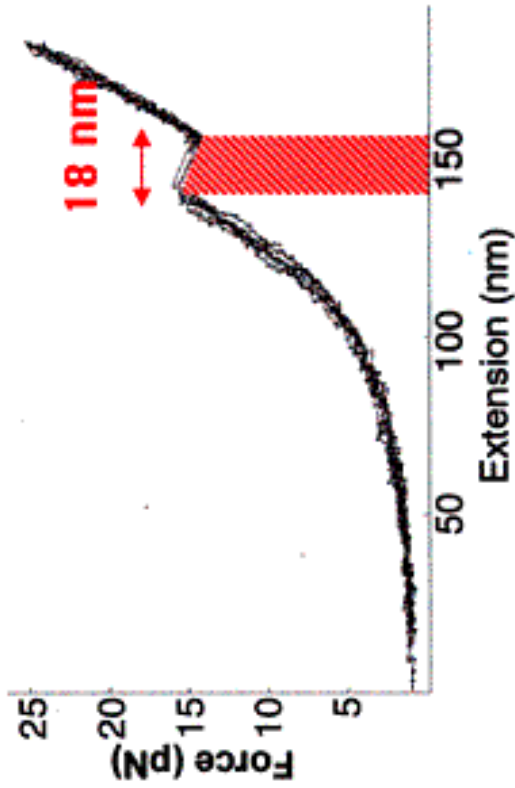




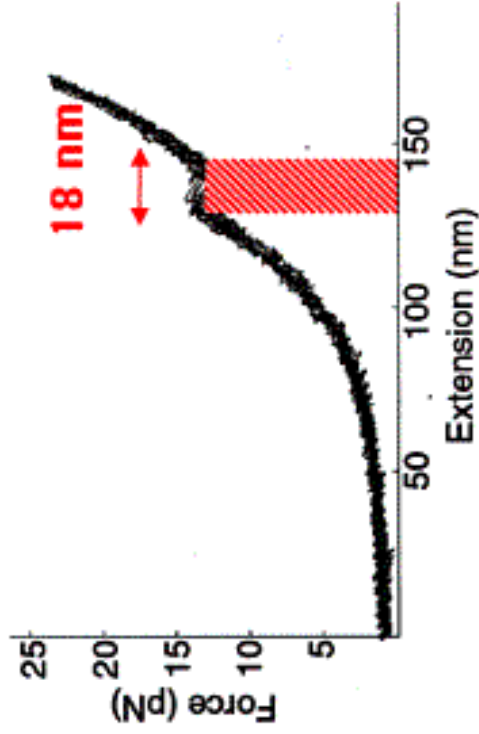


Mechanical unfolding of RNA

P5 hairpin in Mg^{++}



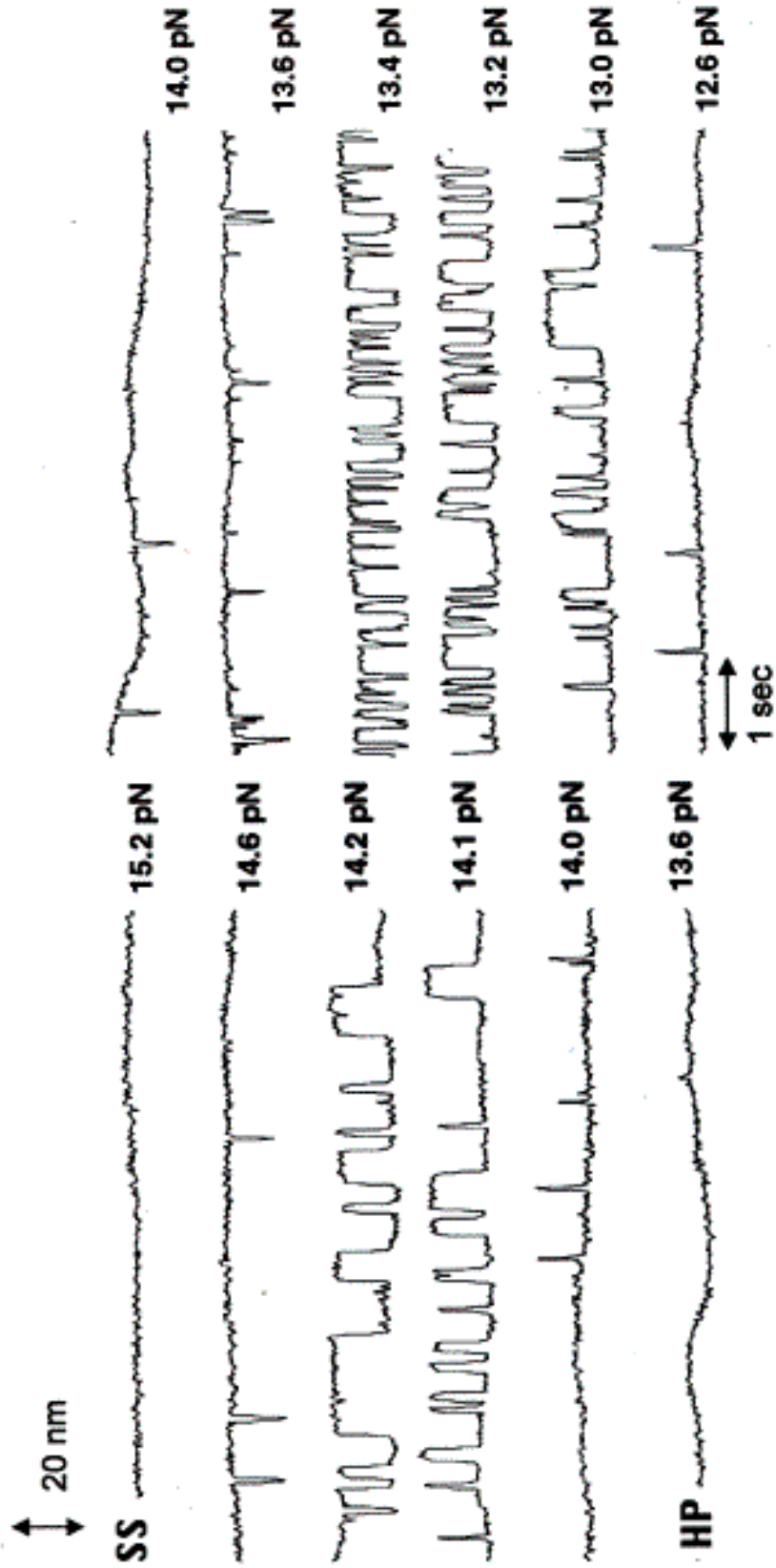
P5 hairpin in EDTA

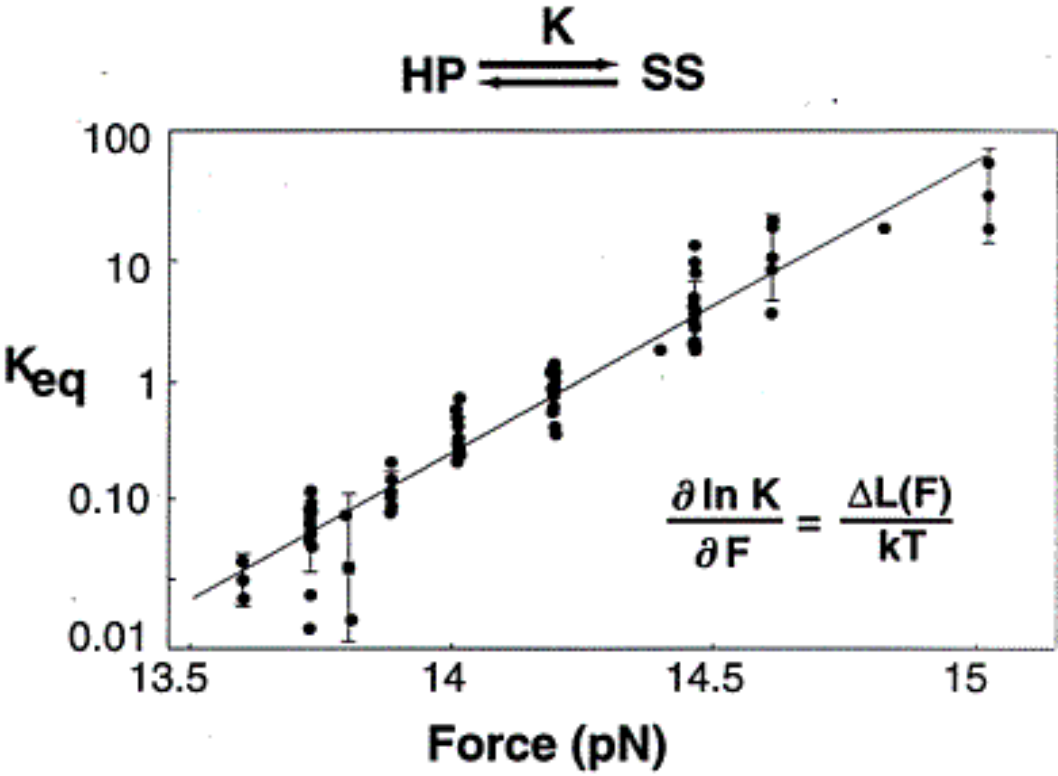


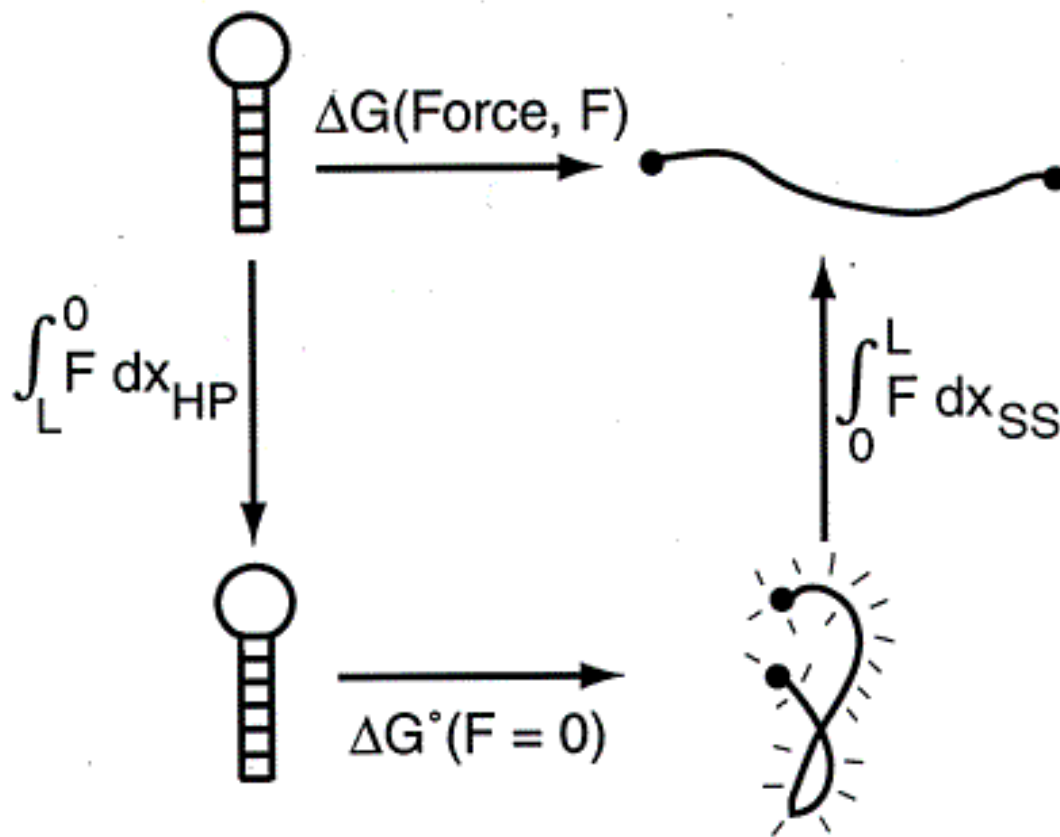
P5 hairpin constant force experiments

without Mg⁺⁺ (EDTA)

with Mg⁺⁺



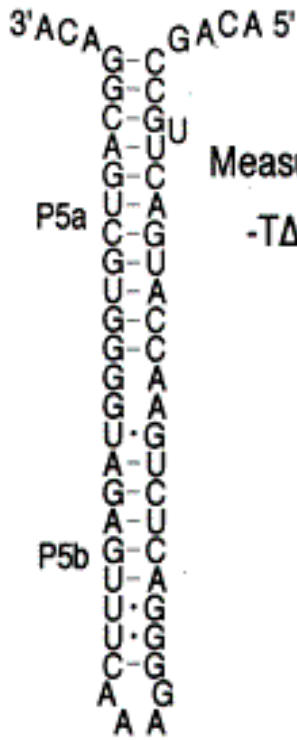




$$\Delta G(\text{Force}, F) = \Delta G^\circ(F=0) + \int_0^L F dx_{SS} - \int_0^L F dx_{HP}$$

$$\Delta G(\text{Force}, F) = \Delta G^\circ(F=0) + \Delta G(0 \rightarrow F)_{\text{stretching}}$$

$$\Delta G^\circ(F = 0) = F_{1/2} \Delta L - \Delta G(0 \rightarrow F_{1/2})_{\text{stretch}}$$



Measured $\Delta G(298)$ at 14.5 pN = 157 ± 20 kJ/mol = 37.6 kcal/mol

$-\Delta S(298)$ for tethered ends = -40 ± 10 kJ/mol = -9.6 kcal/mol

$\Delta G^\circ(298)$ free in solution = 117 ± 30 kJ/mol = 28.0 kcal/mol

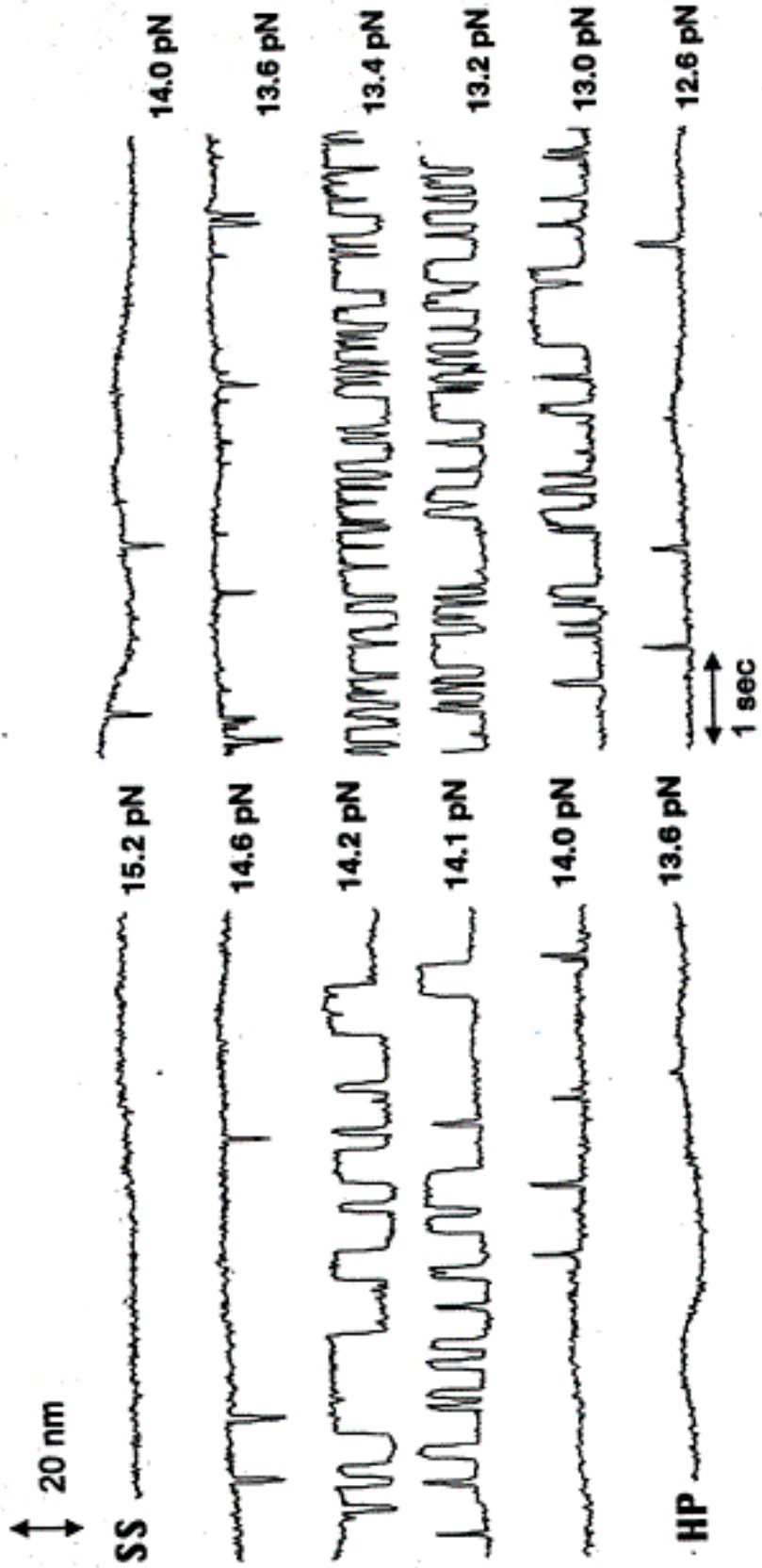
$\Delta G^\circ(298)$ Zuker's mfold = 141 ± 10 kJ/mol = 33.7 kcal/mol

P5 hairpin

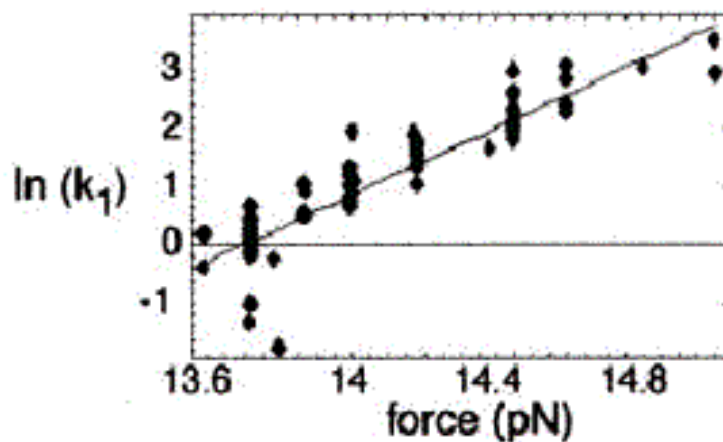
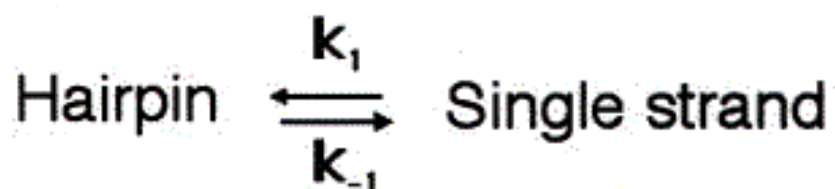
P5 hairpin constant force experiments

without Mg⁺⁺ (EDTA)

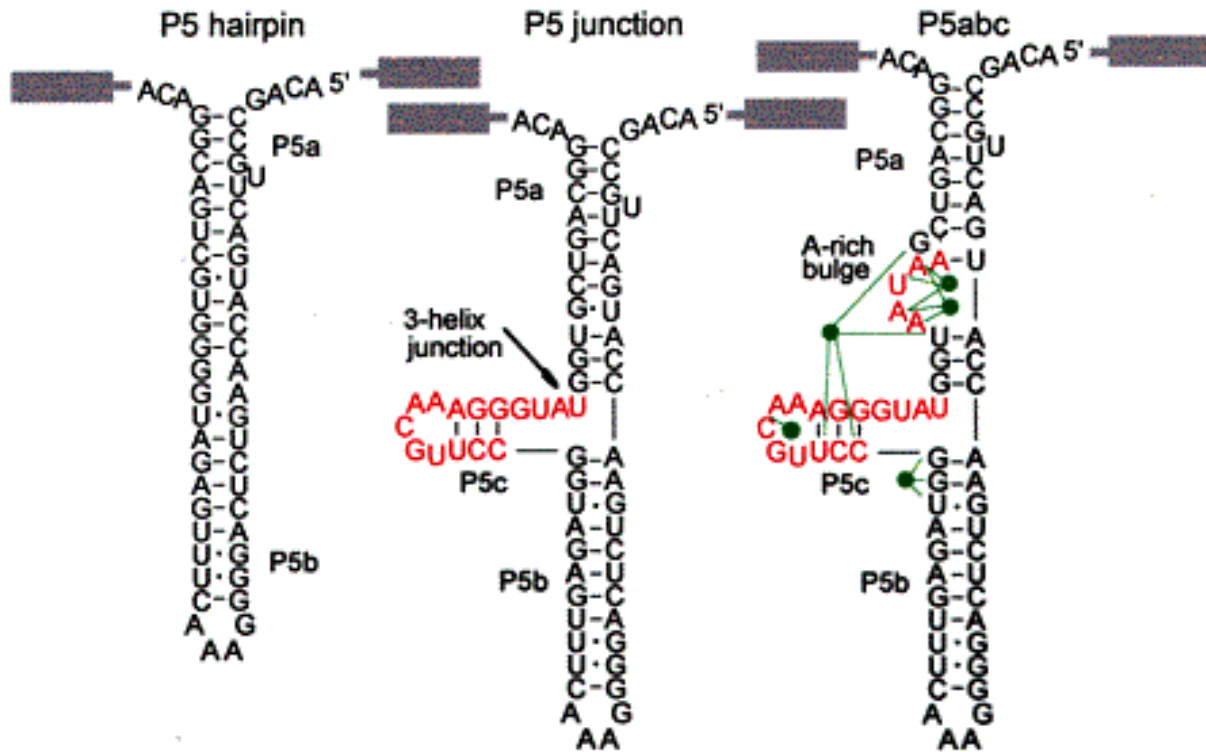
with Mg⁺⁺



Effect of force on the rate of unfolding P5 HP

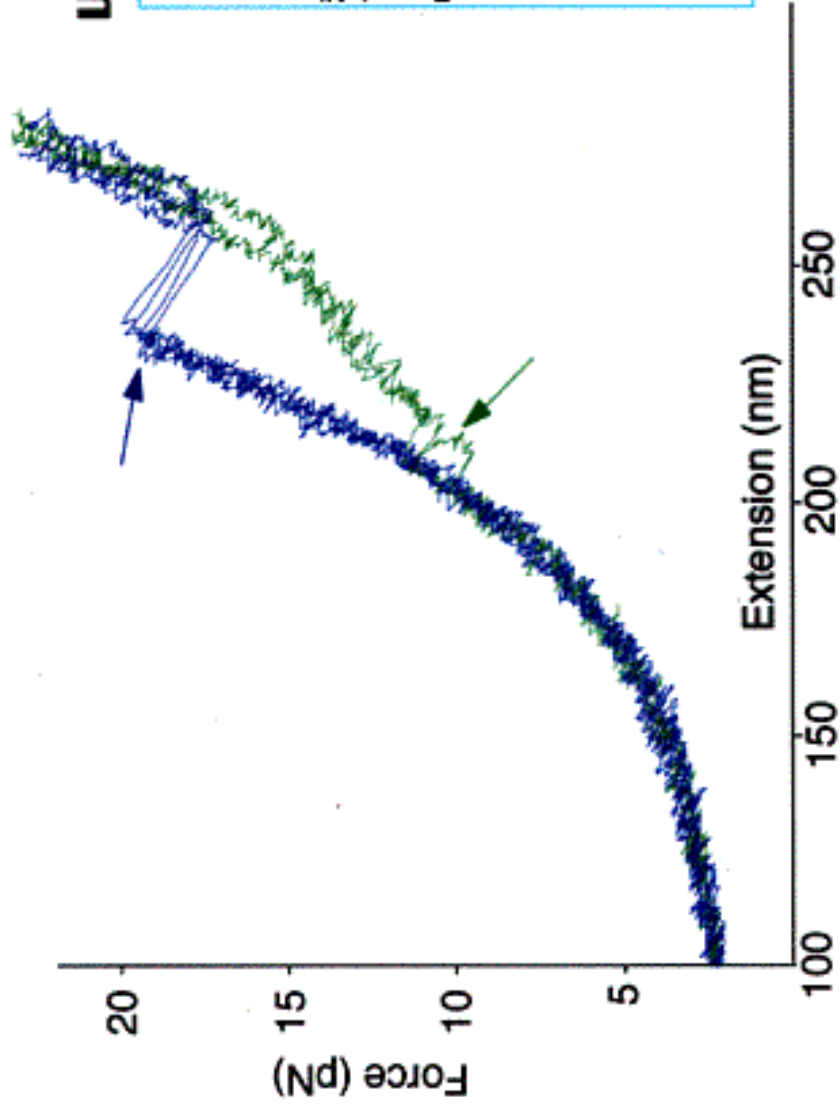


$$\frac{\partial \ln k_1}{\partial F} = \frac{\Delta l^*(F)}{kT} \quad \Delta l^* = 12 \text{ nm}$$

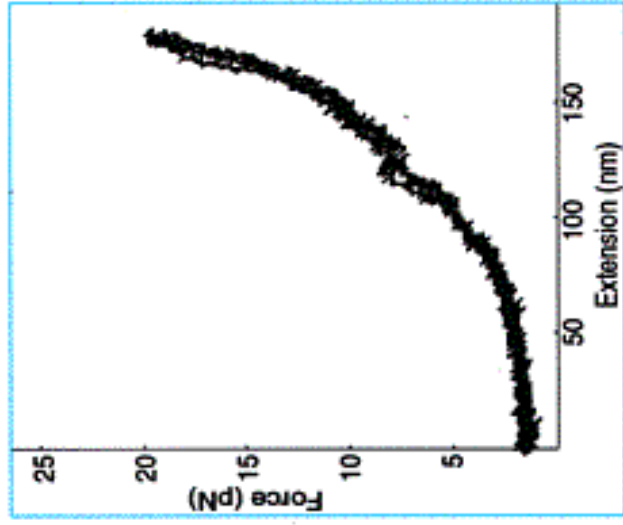


Metal-ion core: P5abc

in Mg^{++}

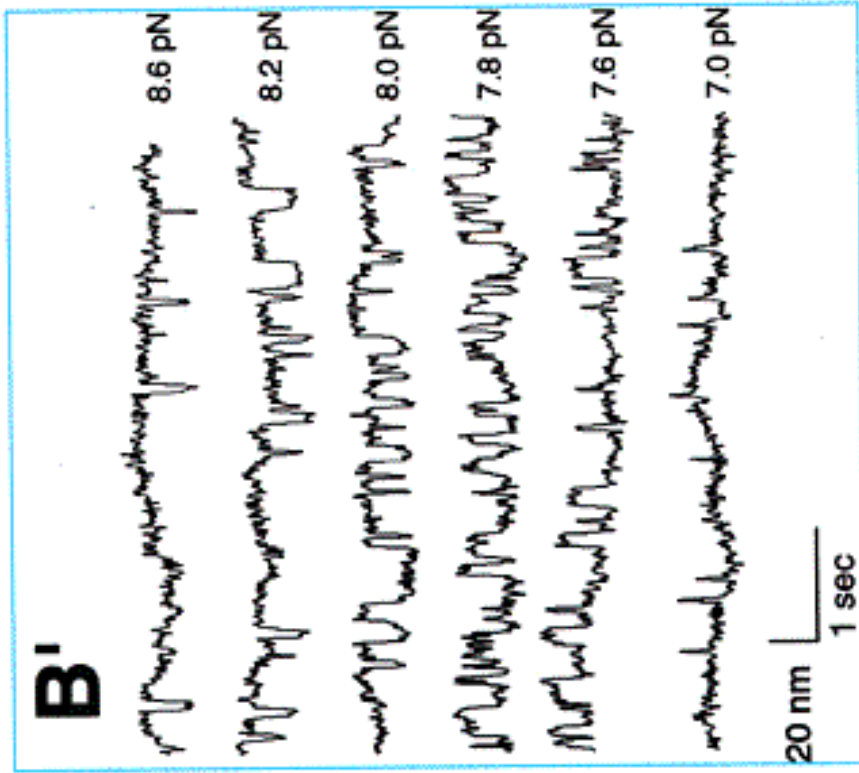


without Mg^{++} (EDTA)

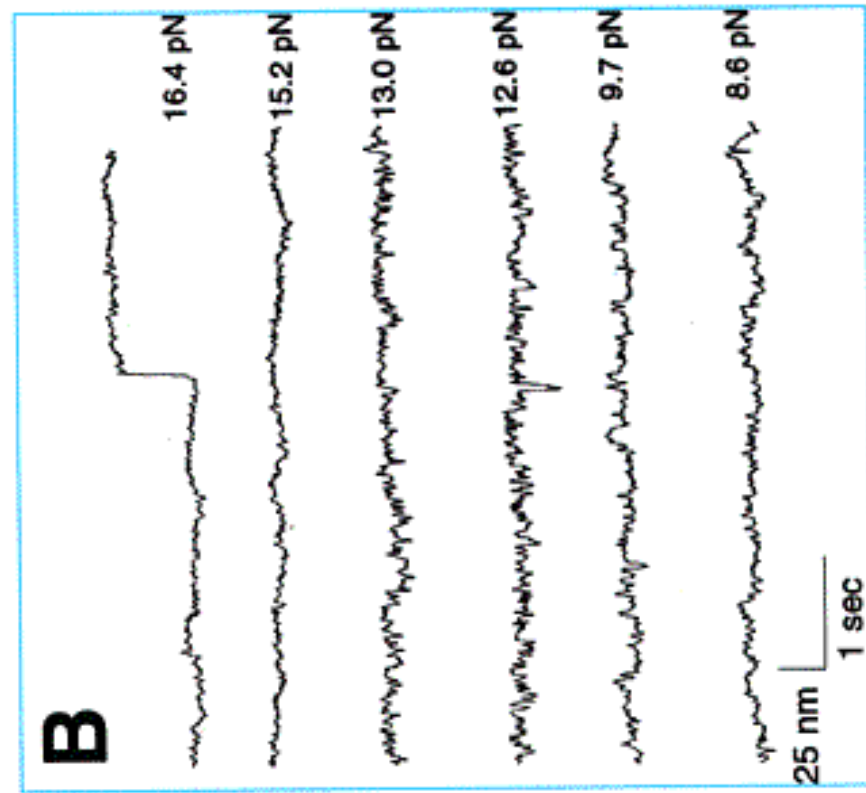


P5abc: constant force

without Mg^{++} (EDTA)

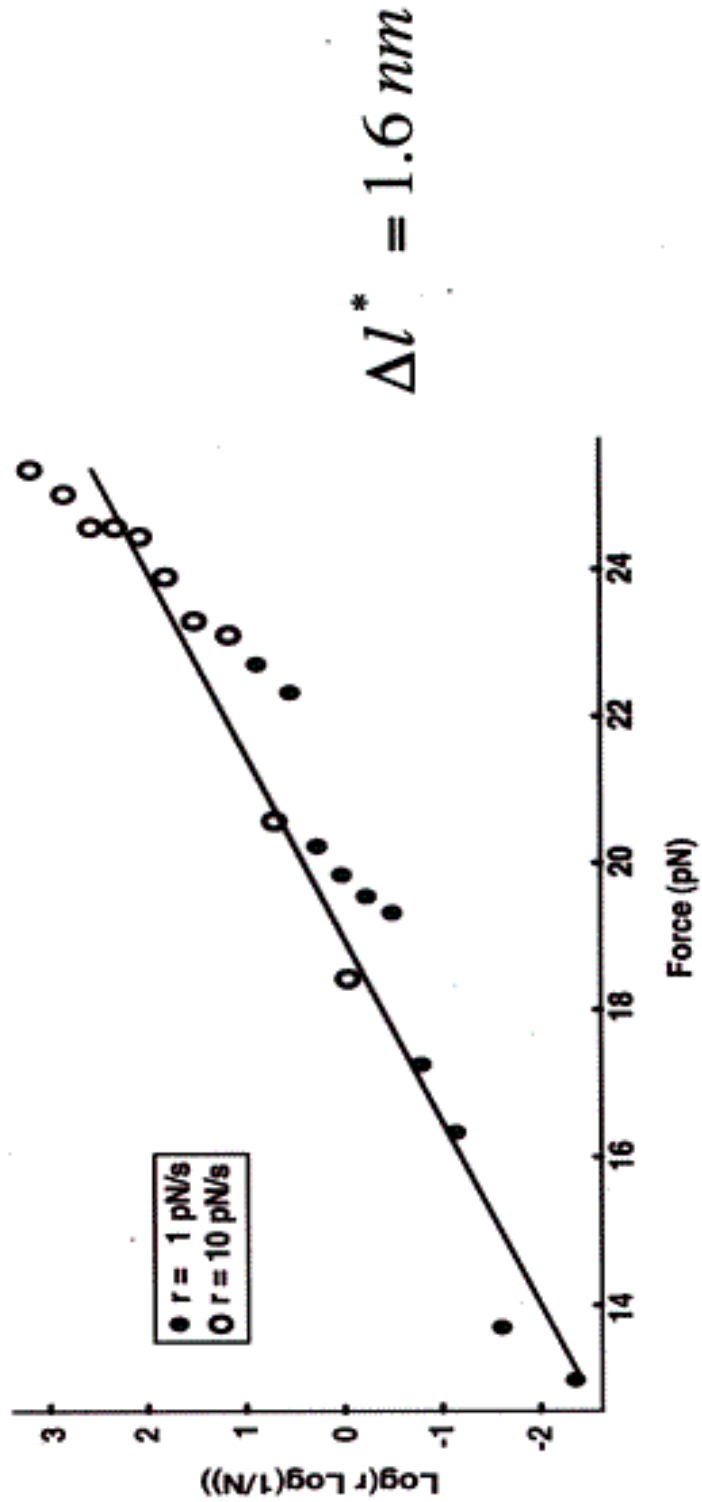


in Mg^{++}

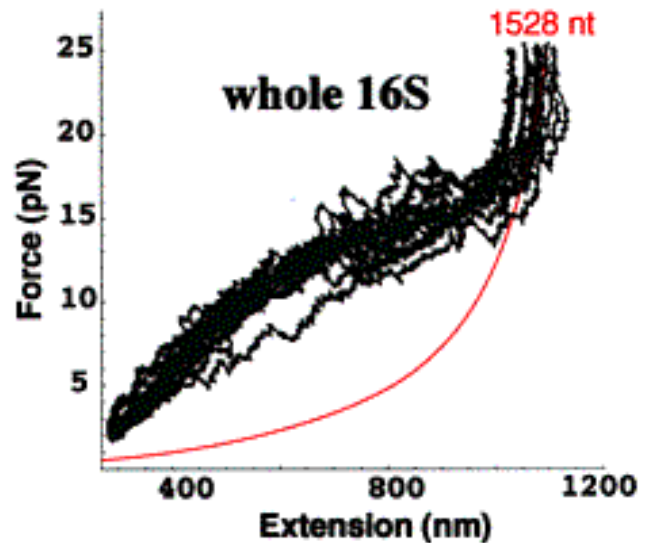
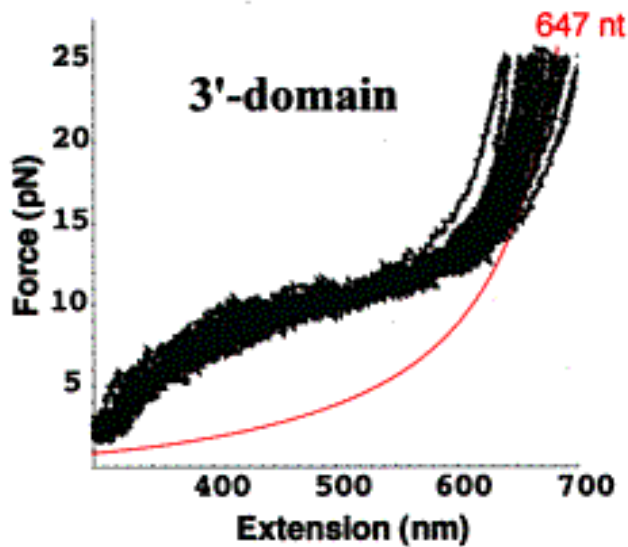
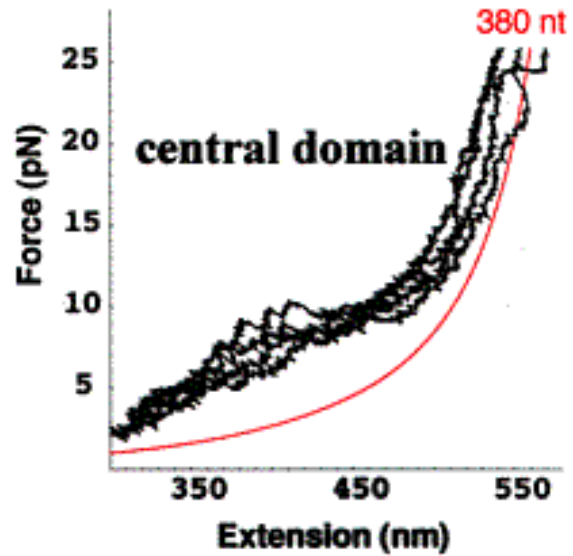
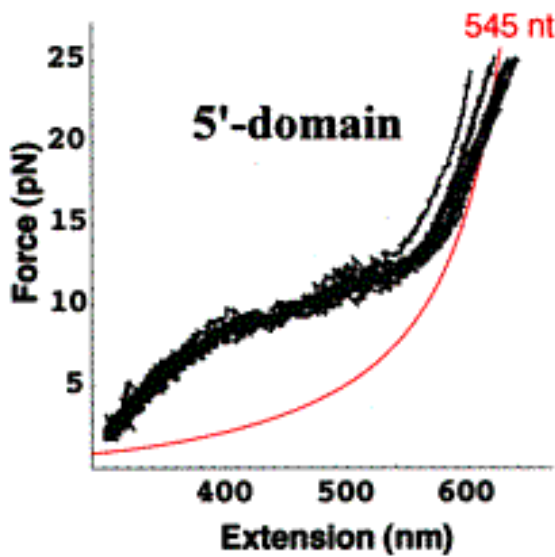


Non-equilibrium Unfolding

$$N(F, r) = e^{-\frac{k_0}{br}(e^{bF} - 1)} \quad \text{where} \quad b = \frac{\Delta l^*}{k_B T}$$



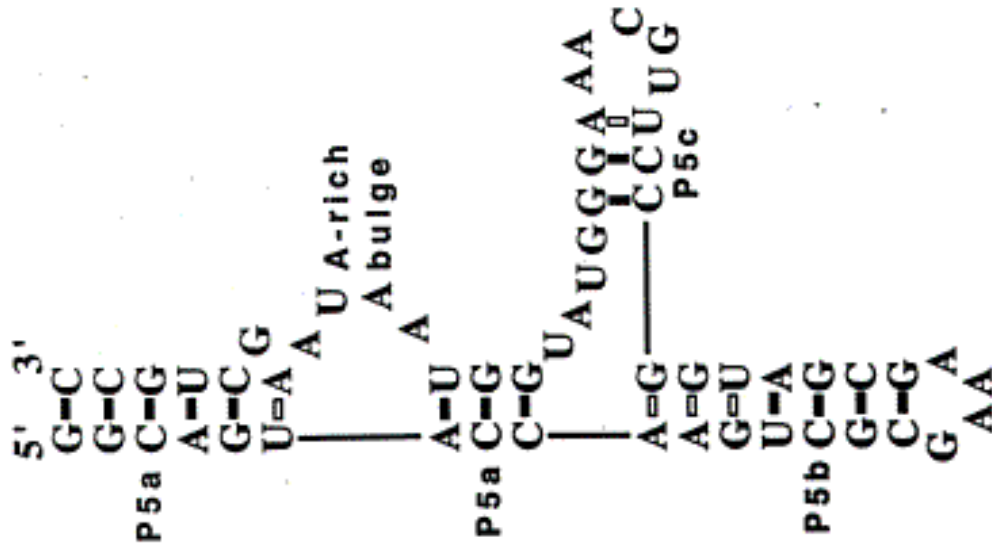
16S Ribosomal RNA in reconstitution buffer minus magnesium



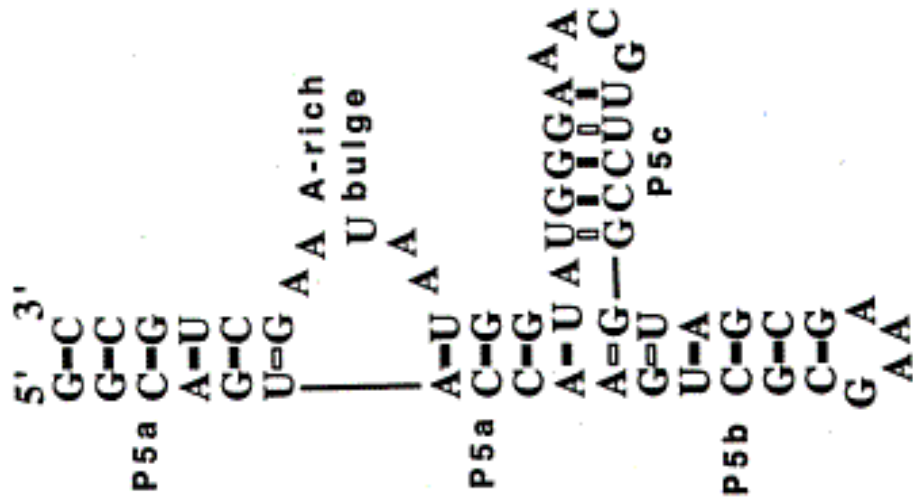
Summary

- 1. You can measure the work of unfolding a molecule from the product of force times extension; if the process is reversible the work is equal to the free energy change.**
- 2. You can change equilibrium constants and rate constants by applying force.**
 - a. The difference in end-to-end distance between initial and final states determines effect of force on equilibrium.**
 - b. The difference in length between initial state and transition state determines effect of force on kinetics.**
- 3. Mg^{2+} markedly decreases the kinetics of unfolding RNA molecules with tertiary interactions.**
- 4. Size is not important.**

p5abc



Phylogenetics



Thermodynamics