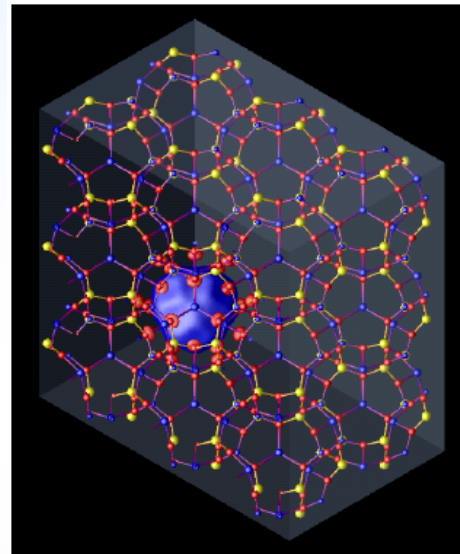


Toward “tailor-made correlations” in zeolites



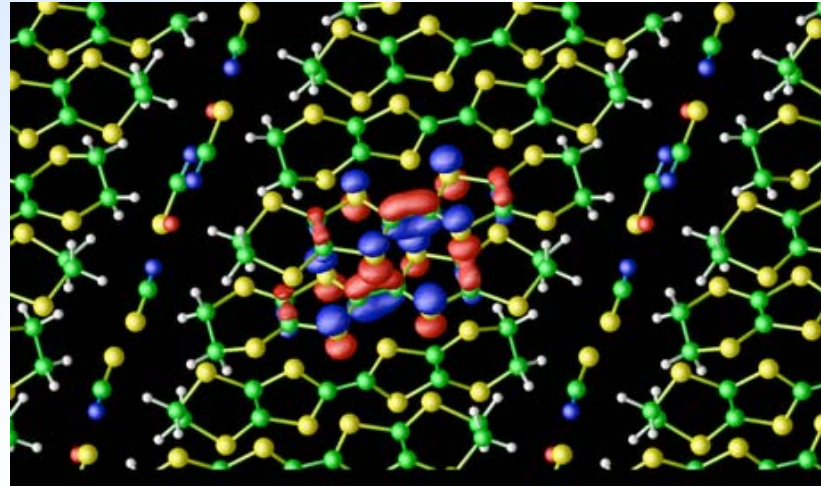
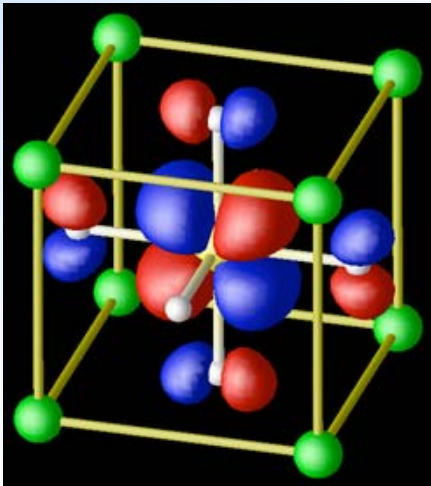
Ryotaro ARITA

Department of Applied Physics, University of Tokyo

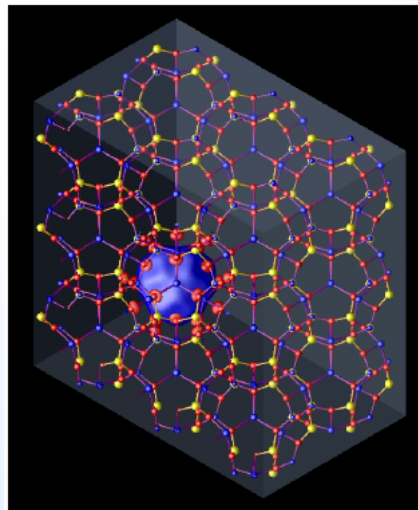
Strongly correlated electron systems

d electron system
(Transition metal oxides)

f electron system
(Heavy electron systems)



Correlated **p** electron system
(organic compounds, e.g., BEDT-TTF)



Zeolite =
Correlated **s** electron system

Collaborators

- Yoshiro Nohara (Dept. Phys., Univ. Tokyo)



- Kazuma Nakamura (Dept. Appl. Phys., Univ. Tokyo)



- Takashi Koretsune (Dept. Phys., Tokyo Inst. Tech.)

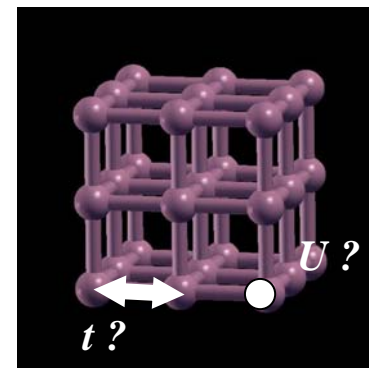
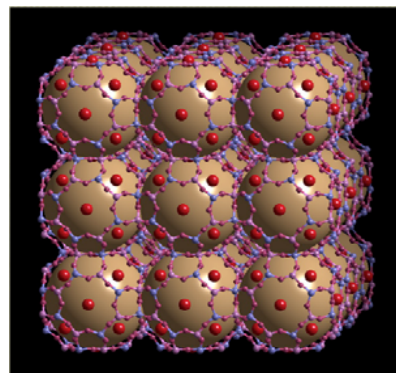
- References:

- Phys. Rev. B 80, 220410(R) (2009)
- Phys. Rev. B 80, 174420 (2009)

Outline

- Introduction to zeolite
 - nanoporous aluminosilicates + cluster of alkali atoms
 - Family of huge members, rich variety of physical/chemical properties
 - Strongly correlated electron system made from Al, Si, O, K (Na, Rb)

- Ferromagnetism in zeolite LTA
 - Ferromagnet comprising only non-magnetic elements
 - Mechanism of spin polarization: described by the multi-orbital Hubbard model



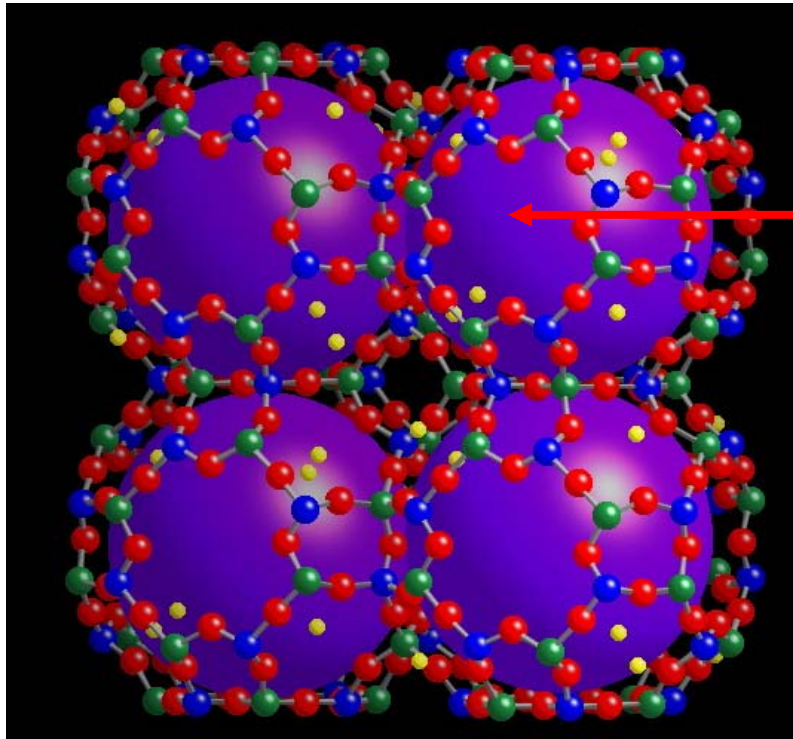
Supercrystal
of superatom

- “superatom” picture: materials design in terms of superatom ?
- Design of correlation effect (Taylor-made correlation)

- How large is U of superatom ?
 - Estimate of interaction parameters by cRPA

What is zeolite ?

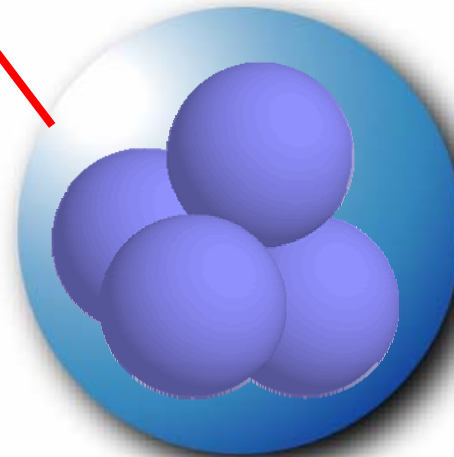
Host



● : Al ● : O ● : Si

cage made of
aluminosilicate

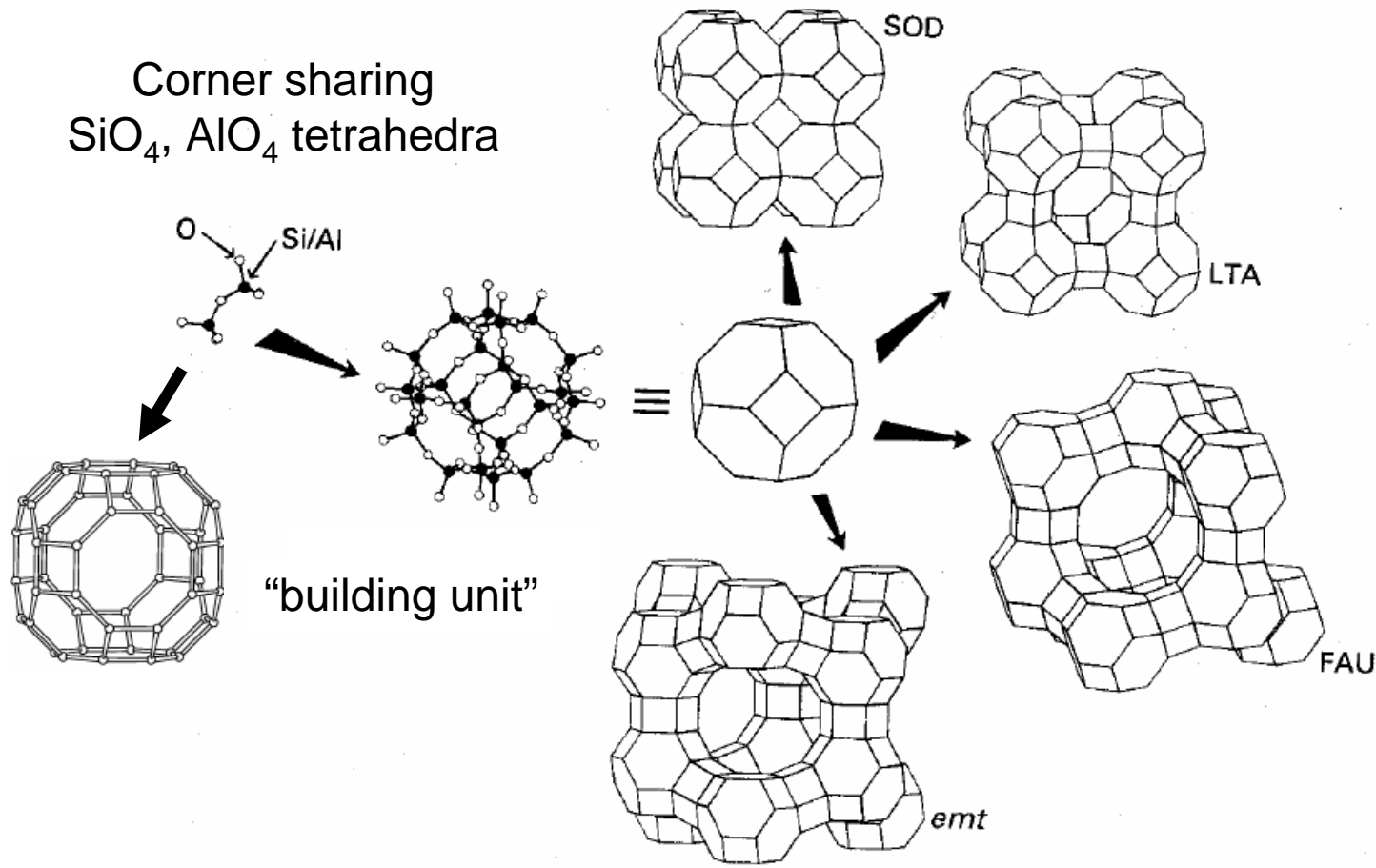
Guest



alkali cluster
(K, Na, Rb, etc.)

+

What is zeolite ? : rich variety of host-cage structures



various host-cage structure

What is zeolite ? : rich variety of host-cage structures



ABW	ACO	AEI	AEL	AEN	AET	AFG	AFI	AFN	AFO	AFR	AFS
AFT	AFX	AFY	AHT	ANA	APC	APD	AST	ASV	ATN	ATO	ATS
ATT	ATV	AWO	AWW	BCT	*BEA	BEC	BIK	BOF	BOG	BPH	BRE
BSV	CAN	CAS	CDO	CFI	CGF	CGS	CHA	-CHI	-CLO	CON	CZP
DAC	DDR	DFO	DFT	DOH	DON	EAB	EDI	EMT	EON	EPI	ERI
ESV	ETR	EUO	EZT	FAR	FAU	FER	FRA	GIS	GIU	GME	GON
GOO	HEU	IFR	IHW	IMF	ISV	ITE	ITH	ITR	ITW	IWR	IWS
IWV	IWW	JBW	JRY	KFI	LAU	LEV	LIO	-LIT	LOS	LOV	LTA
LTF	LTL	LTN	MAR	MAZ	MEI	MEL	MEP	MER	MFI	MFS	MON
MOR	MOZ	*MRE	MSE	MSO	MTF	MTN	MTT	MTW	MWW	NAB	NAT
NES	NON	NPO	NSI	OBW	OFF	OSI	OSO	OWE	-PAR	PAU	PHI
PON	RHO	-RON	RRO	RSN	RTE	RTH	RUT	RWR	RWY	SAO	SAS
SAT	SAV	SBE	SBN	SBS	SBT	SFE	SFF	SFG	SFH	SFN	SFO
SFS	SGT	SIV	SOD	SOF	SOS	SSF	SSY	STF	STI	*STO	STT
STW	-SVR	SZR	TER	THO	TOL	TON	TSC	TUN	UEI	UFI	UOS
UOZ	USI	UTL	VET	VFI	VNI	VSV	WEI	-WEN	YUG	ZON	

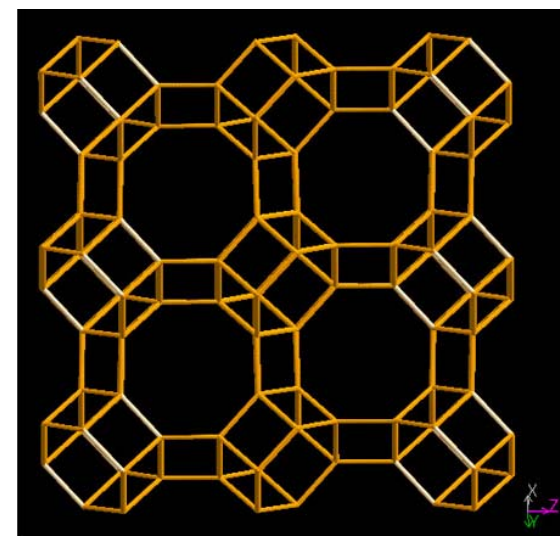
~200 kinds of structures

<http://www.iza-structure.org/databases/>

What is zeolite ? : rich variety of host-cage structures



ABW	ACO	AEI	AEL	AEN	AET	AFG	AFI	AFN	AFO	AFR	AFS
AFT	AFX	AFY	AHT	ANA	APC	APD	AST	ASV	ATN	ATO	ATS
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ESV	ETR	EUO	EZT	FAR	FAU	FER	FRA	GIS	GIU	GME	GON
GOO	HEU	IFR	IHW	IMF	ISV	ITE	ITH	ITR	ITW	IWR	IWS
IWV	IWW	JBW	JRY	KFI	LAU	LEV	LIO	-LIT	LOS	LOV	LTA
LTF	LTL	LTN	MAR	MAZ	MEI	MEL	MEP	MER	MFI	MFS	MON
MOR	MOZ	*MRE	MSE	MSO	MTF	MTN	MTT	MTW	MWW	NAB	NAT
NES	NON	NPO	NSI	OBW	OFF	OSI	OSO	OWE	-PAR	PAU	PHI
PON	RHO	-RON	RRO	RSN	RTE	RTH	RUT	RWR	RWY	SAO	SAS
SAT	SAV	SBE	SBN	SBS	SBT	SFE	SFF	SFG	SFH	SFN	SFO
SFS	SGT	SIV	SOD	SOF	SOS	SSF	SSY	STF	STI	*STO	STT
STW	-SVR	SZR	TER	THO	TOL	TON	TSC	TUN	UEI	UFI	UOS
UOZ	USI	UTL	VET	VFI	VNI	VSV	WEI	-WEN	YUG	ZON	



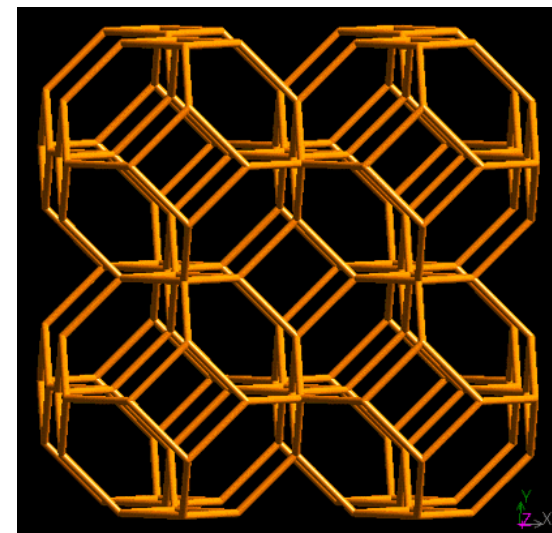
~200 kinds of structures

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What is zeolite ? : rich variety of host-cage structures



ABW	ACO	AEI	AEL	AEN	AET	AFG	AFI	AFN	AFO	AFR	AFS
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SFS	SGT	SIV	SOD	SOF	SOS	SSF	SSY	STF	STI	*STO	STT
STW	-SVR	SZR	TER	THO	TOL	TON	TSC	TUN	UEI	UFI	UOS
UOZ	USI	UTL	VET	VFI	VNI	VSV	WEI	-WEN	YUG	ZON	



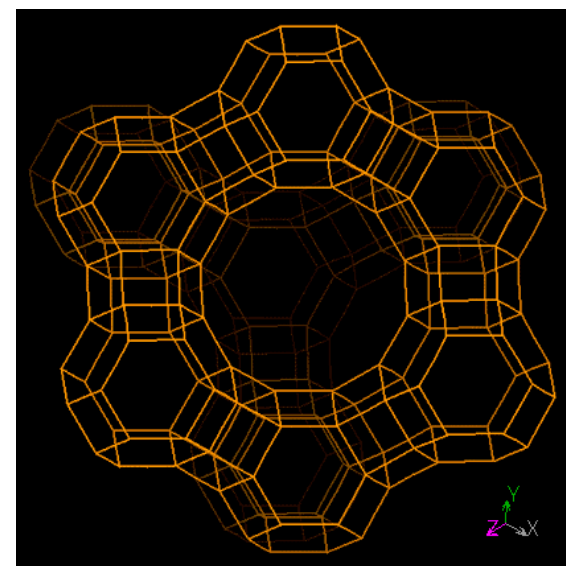
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What is zeolite ? : rich variety of host-cage structures



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SFS	SGT	SIV	SOD	SOF	SOS	SSF	SSY	STF	STI	*STO	STT
STW	-SVR	SZR	TER	THO	TOL	TON	TSC	TUN	UEI	UFI	UOS
UOZ	USI	UTL	VET	VFI	VNI	VSV	WEI	-WEN	YUG	ZON	



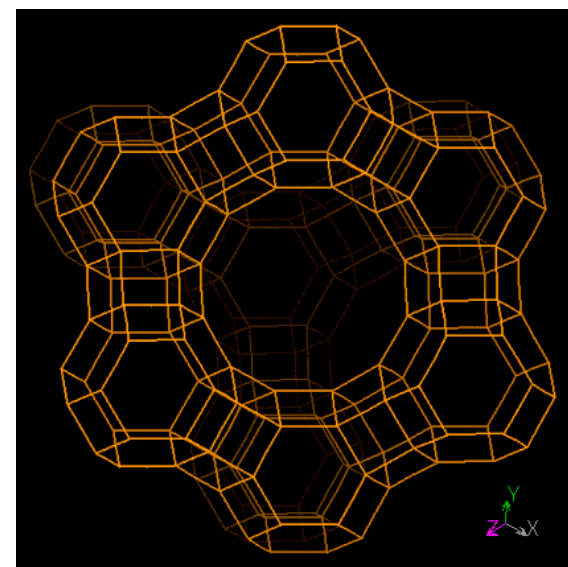
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What is zeolite ? : rich variety of host-cage structures



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SFS	SGT	SIV	SOD	SOF	SOS	SSF	SSY	STF	STI	*STO	STT
STW	-SVR	SZR	TER	THO	TOL	TON	TSC	TUN	UEI	UFI	UOS
UOZ	USI	UTL	VET	VFI	VNI	VSV	WEI	-WEN	YUG	ZON	



~200 kinds of structures

+ Al/Si ratio (controls acidity)

+ size/species (Na, K, Rb) of guest clusters

= Family of huge members / various physical & chemical properties

What is zeolites?: many applications

- Detergents: taking advantage of **iron-exchange capability**
 - Cations in the host-cage such as Na^+ , K^+ , Ca^{2+} , Mg^{2+} etc., are rather loosely held so that they can readily be exchanged for others in a contact solution

- Molecular sieves: takeing advantage of **regular pore structure of molecular dimensions**
 - Zeolites can selectively sort molecules in a size-exclusion process. (Molecules small enough to pass through the pores are absorbed while larger molecules are not.)

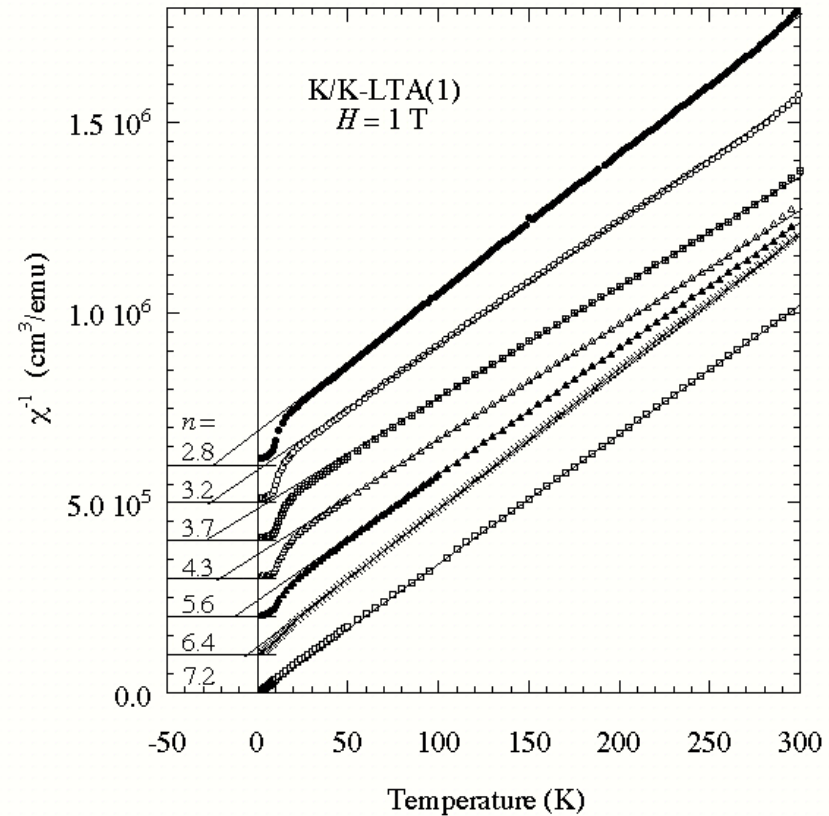
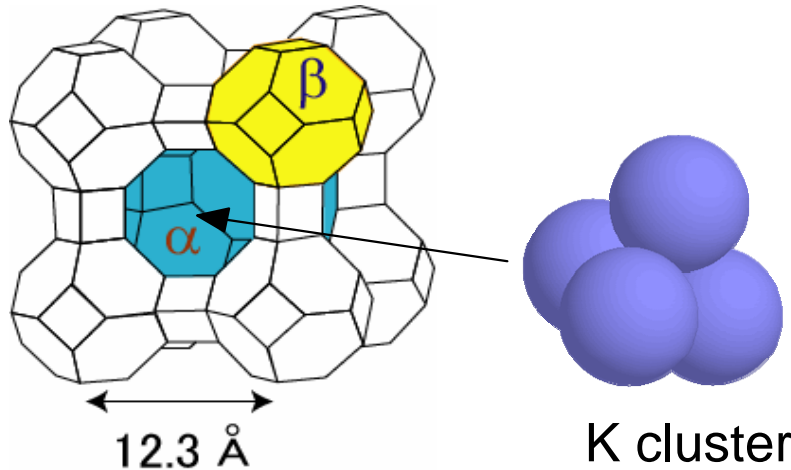
- Catalysis: taking advantage of **high surface area** and **acidity**

- ...

What is zeolite ?

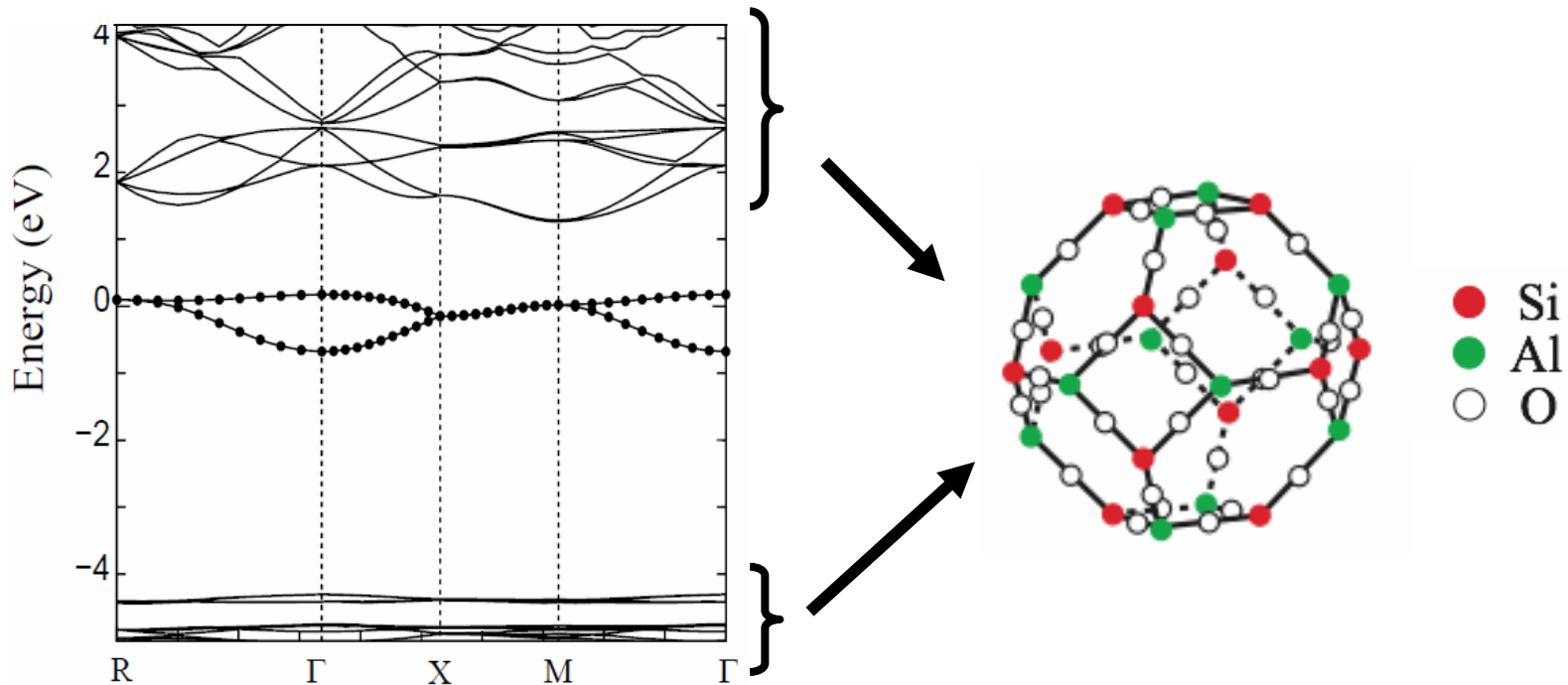
Strongly correlated electron system
made from Al, Si, O, K (Na, Rb)

LTA



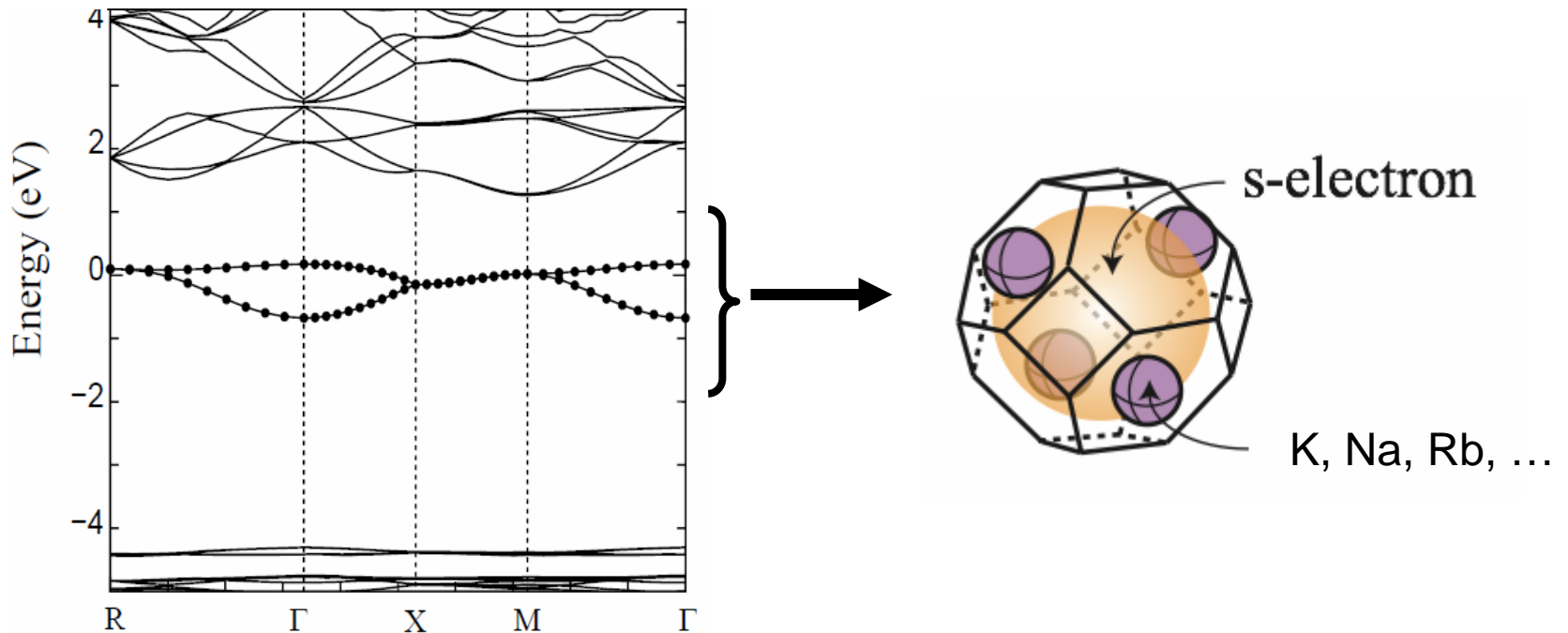
Ferromagnetism in a system
comprising only non-magnetic
elements (Nozue et al., 92)

Electronic structure of zeolite



“host” (= aluminosilicate) cage forms a gap $\sim 6\text{eV}$

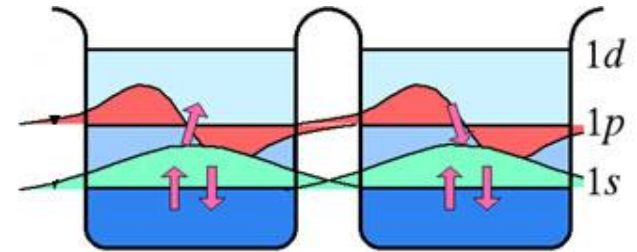
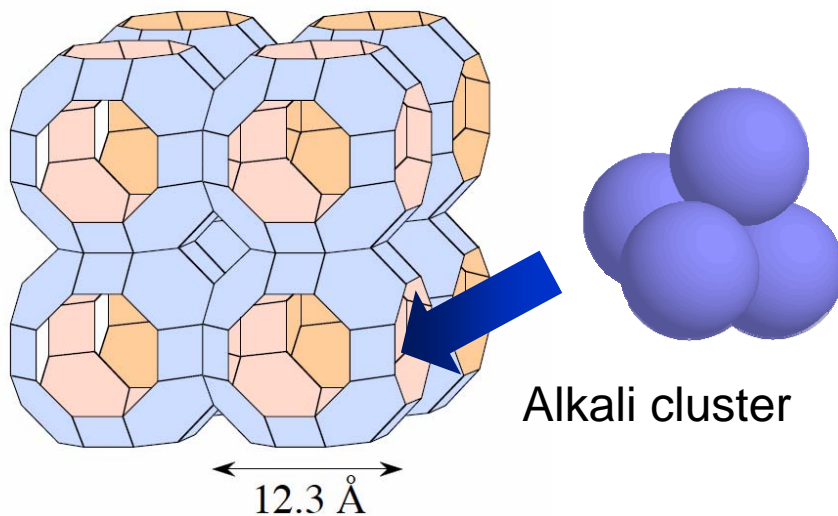
Electronic structure of zeolite



“guest” (= alkali cluster) makes states inside the gap

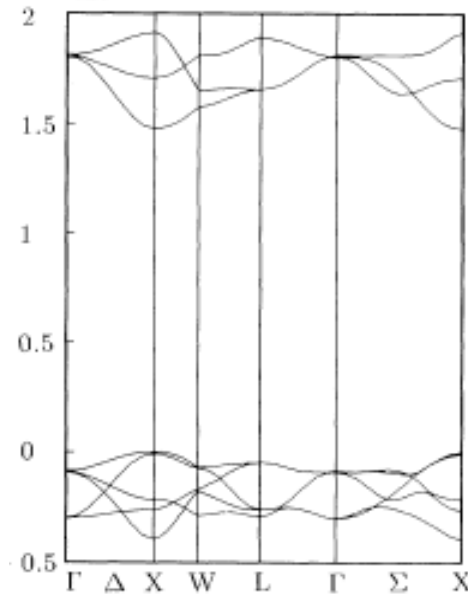
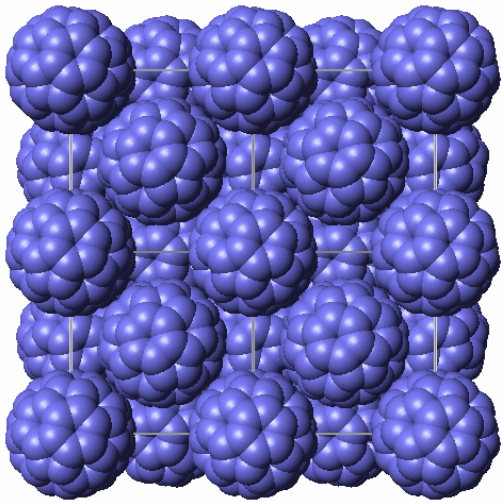
Electronic structure of zeolite: superatom picture

- potential formed by cages = **atomic potential** of superatom
- s-electron systems of alkali cluster = **valence electrons** of superatom



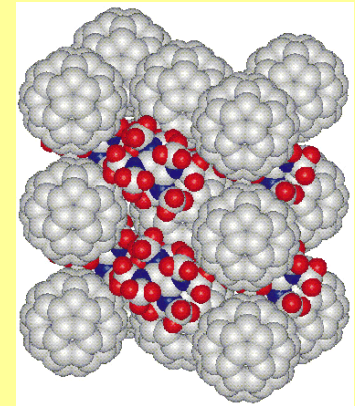
Correlation effects explained in terms of “superatom” ?
 Materials design in terms of “superatom” ?

Electronic structure of C_{60}

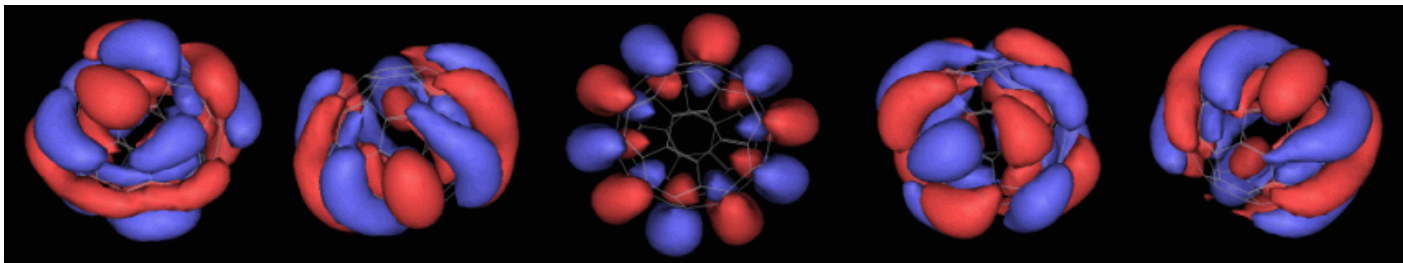


C_{60} HOMO

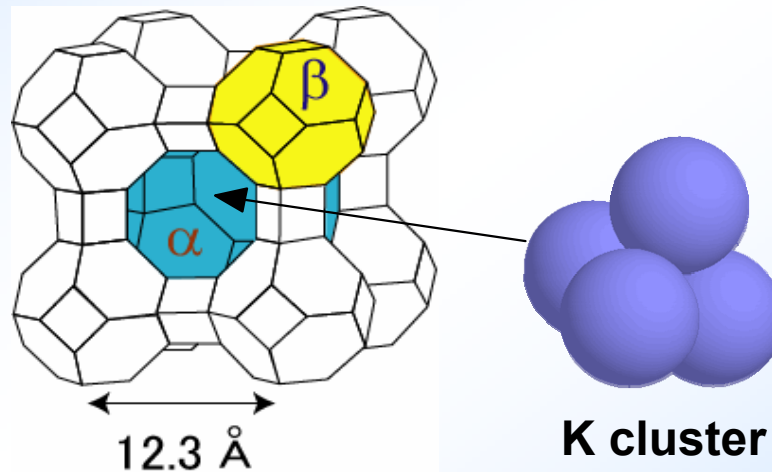
Ferromagnet
made from C, H, N
 C_{60} -TDAE



$T_c \sim 16K$



Ferromagnetism in potassium-loaded zeolite LTA

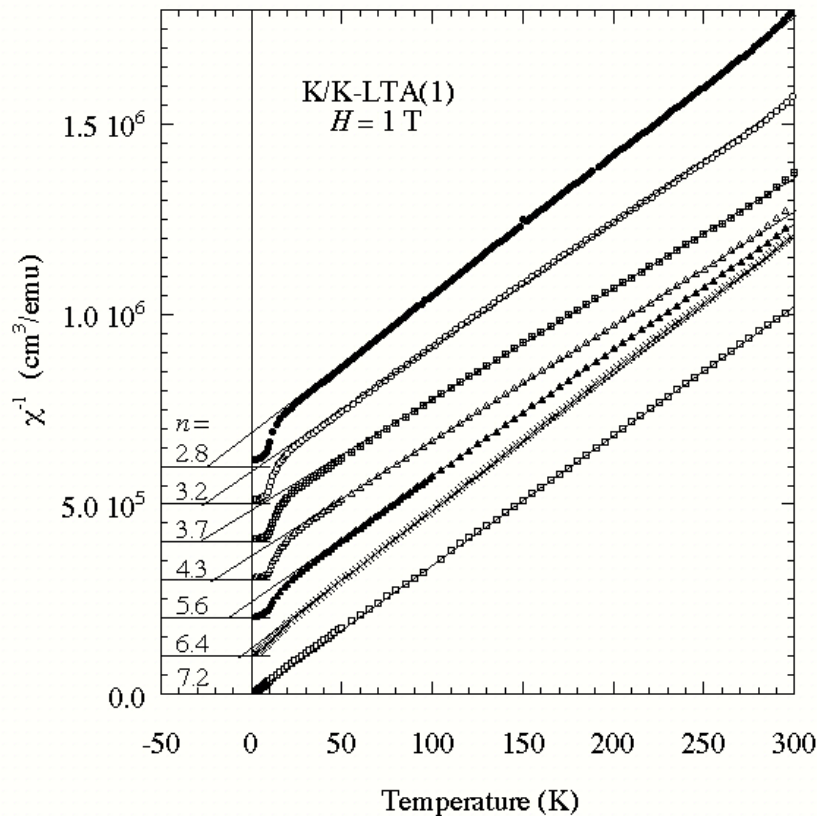


Y. Nohara, K. Nakamura & RA
PRB 80 220410(R) (2009)

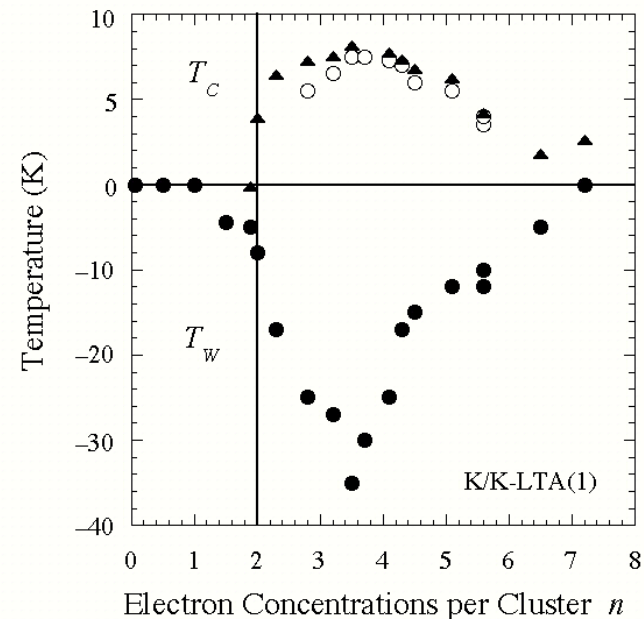
K-loaded zeolite LTA $(K_n + K_{12}Al_{12}Si_{12}O_{48})$

Ground state = ferromagnetic

Nozue et al, PRL68 3789(1992)

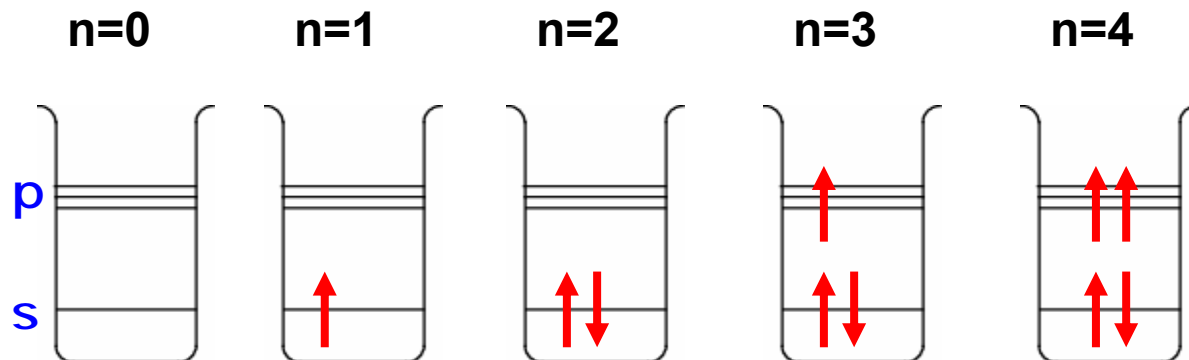
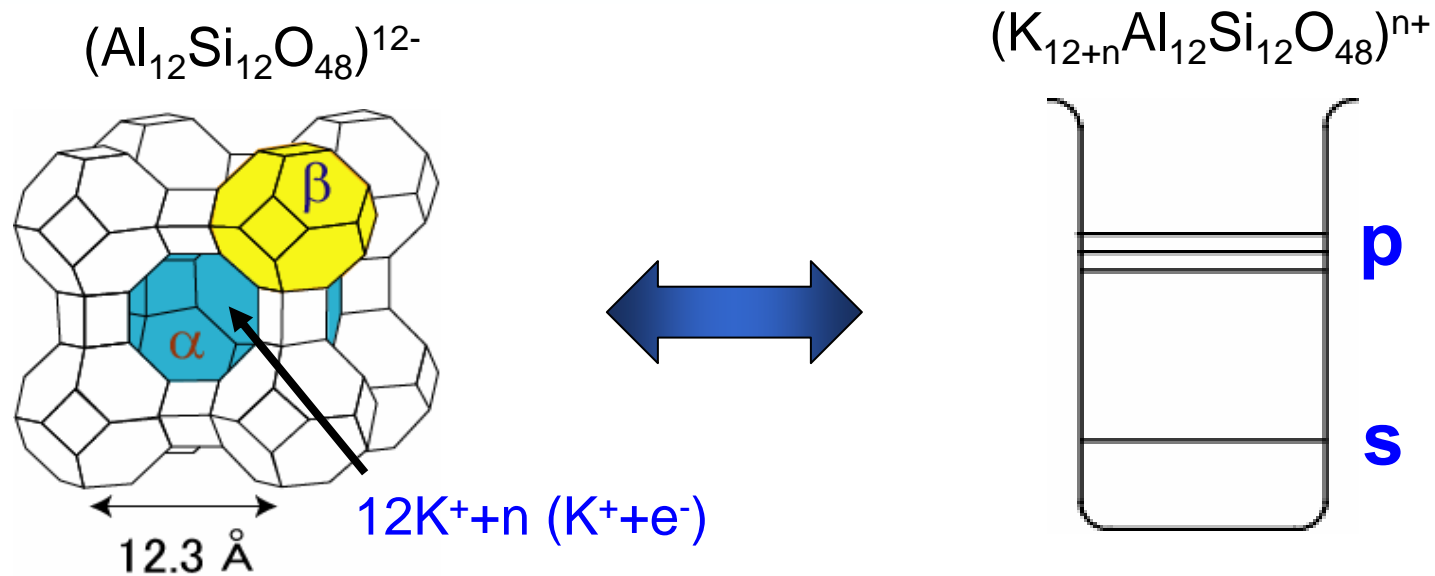


Curie temperature & Weiss temperature



(Nozue *et al*, 1992,
Nakano *et al*, 2002)

Description of magnetism in terms of “superatom”



F appears

Highest T_c

Nozue et al.

Mechanism of spin polarization?

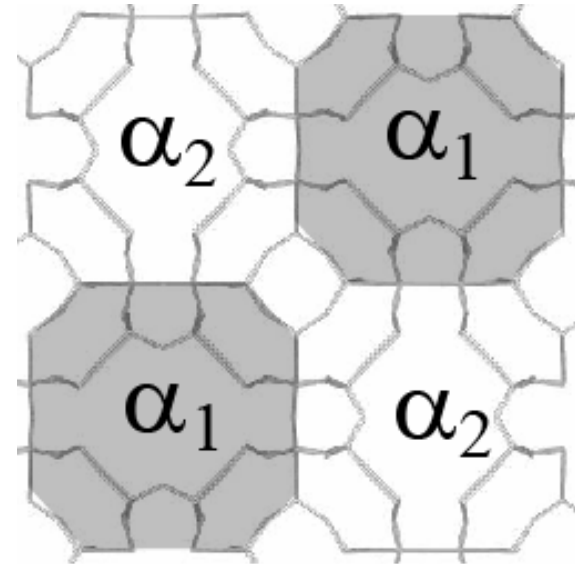
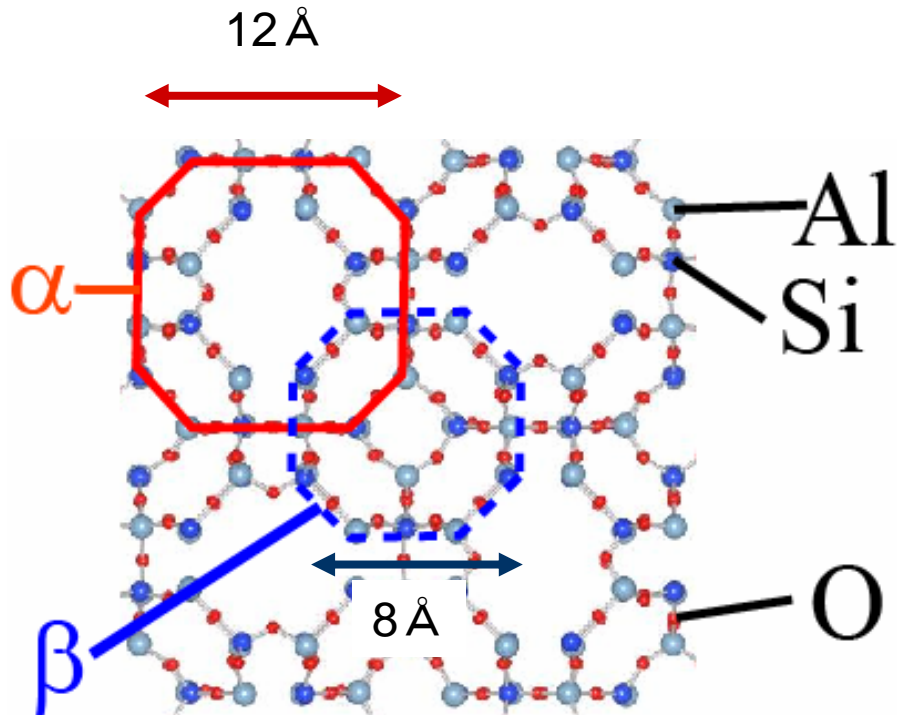
Ab initio study for zeolite LTA has been limited, because

- Position of K not determined
 - Detailed Neutron scattering measurement
T. Ikeda et al, Chem. Phys. Lett. 318, 93 (2000)

- Large unit cell: ~200 atoms in the unit cell
 - LDA for a “simplified” unit cell with < 100 atoms
RA et al, Phys. Rev. B. 69, 195106 (2004)

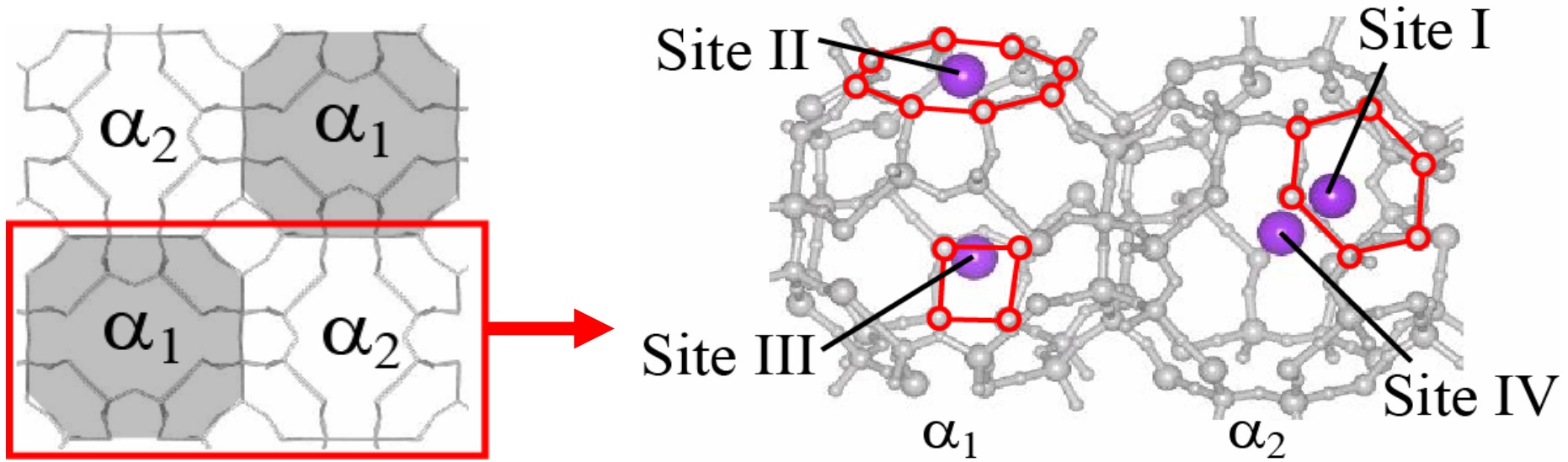
- LSDA calculation for LTA (n=4)
 - Ferromagnetic ground state ?
Nohara, Nakamura, RA 09

Atomic configuration: host cage

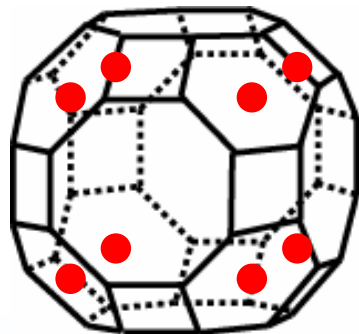


Atomic configuration: positions of K

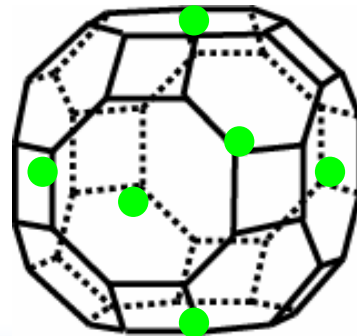
K are in α , not β (Ikeda et al, 2000)



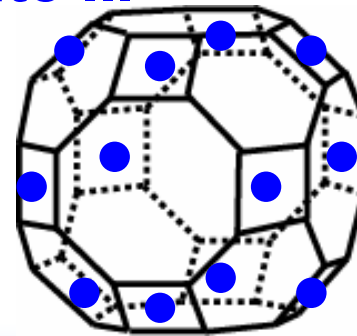
site I



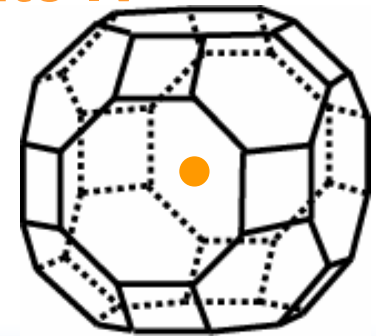
site II



site III

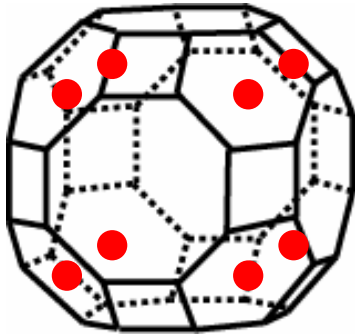


site IV

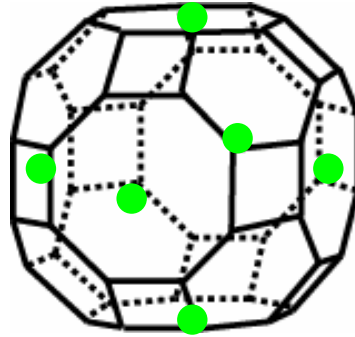


Atomic configuration: positions of K

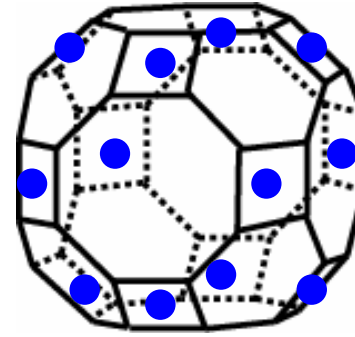
site I



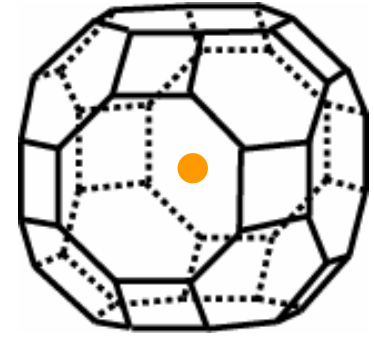
site II



site III



site IV



Occupancies ($K_4+K_{12}Al_{12}Si_{12}O_{48}$)

Ikeda et al, 2000

	I ($\alpha 1$)	I ($\alpha 2$)	II	III ($\alpha 1$)	III ($\alpha 2$)	IV ($\alpha 1$)	IV ($\alpha 2$)	
# of sites	8	8	6	12	12	1	1	
Expt	8	8	6.4	5.9	3.5	0	0.5	(Ikeda et al, 2000)
Our model	8	8	6	6	3	0	1	

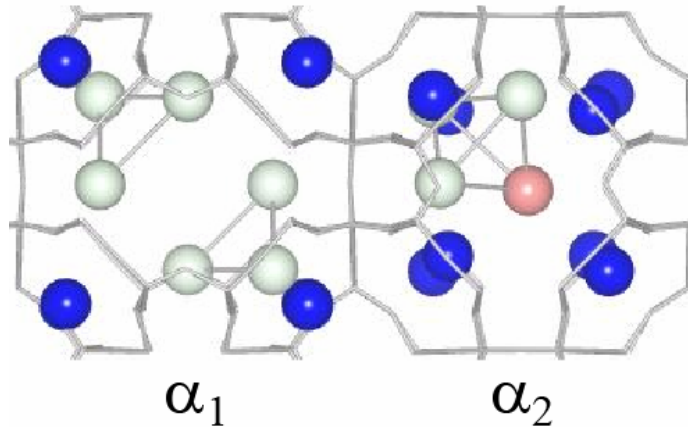


Partially filled:

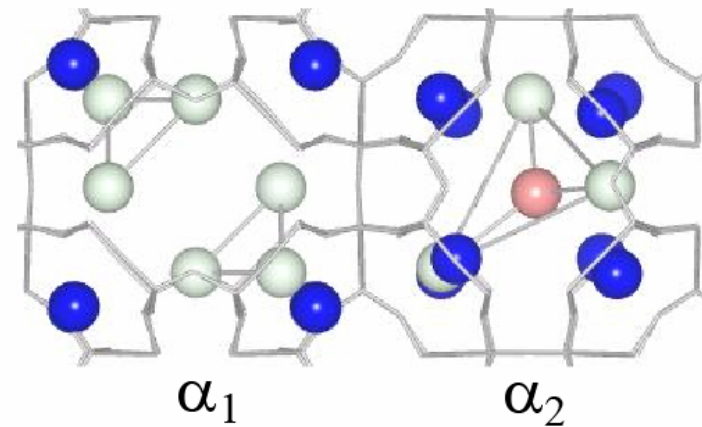
Many possible configurations = ${}_{12}C_6 \times {}_{12}C_3$

Configurations with trigonal symmetry

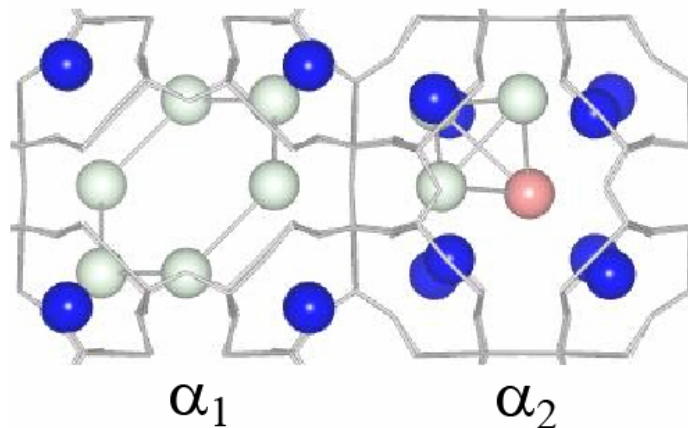
Geometry I



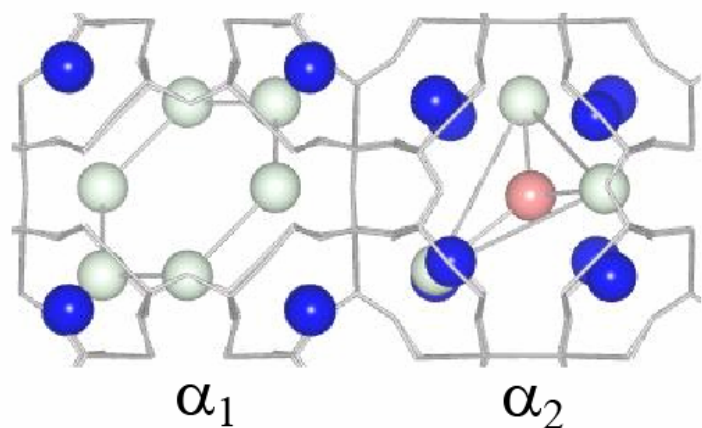
Geometry II



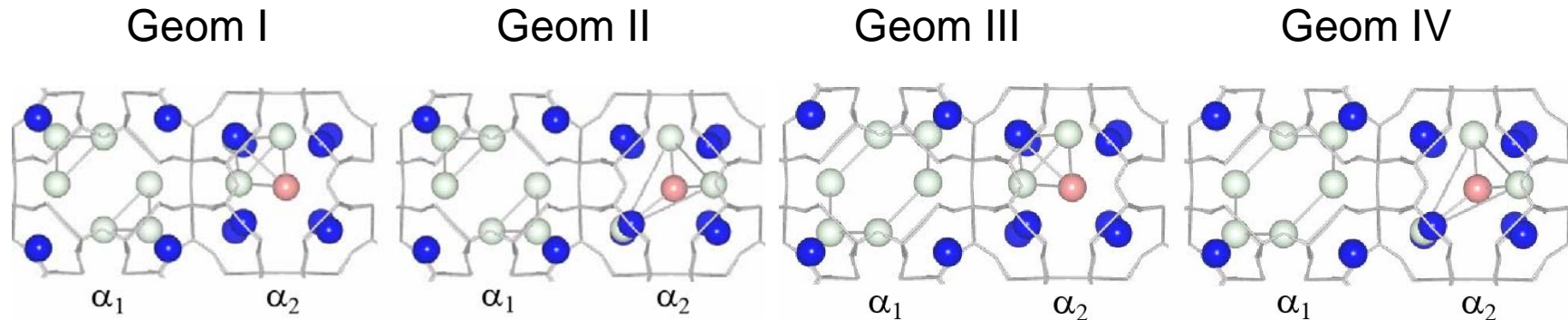
Geometry III



Geometry IV



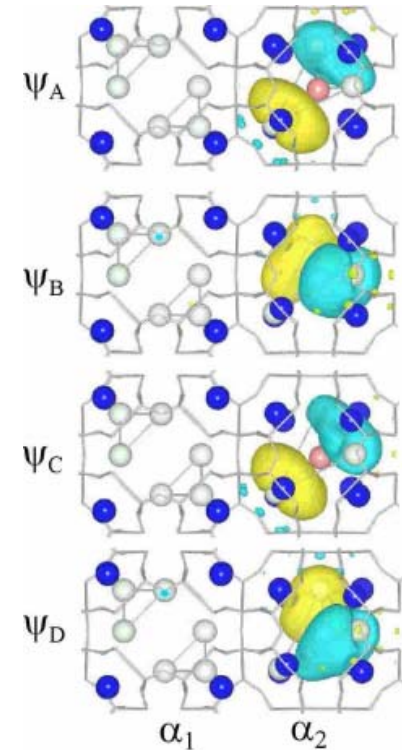
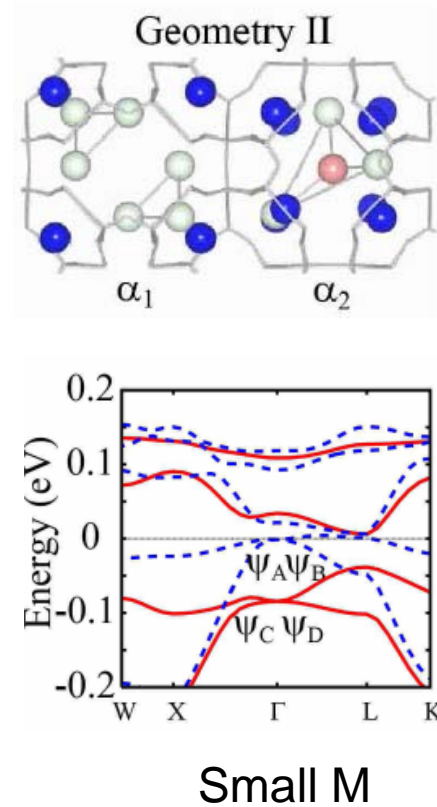
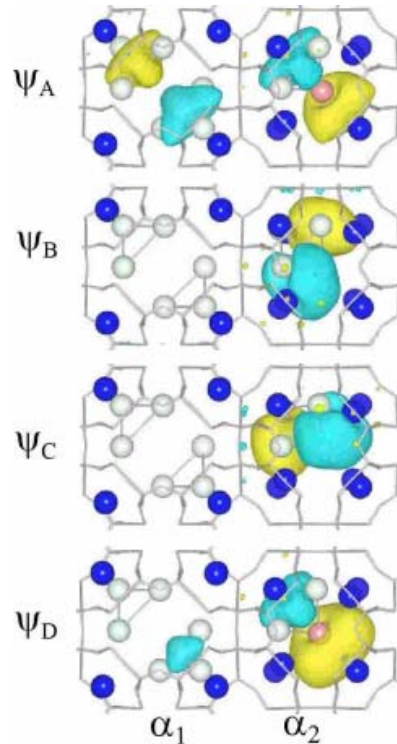
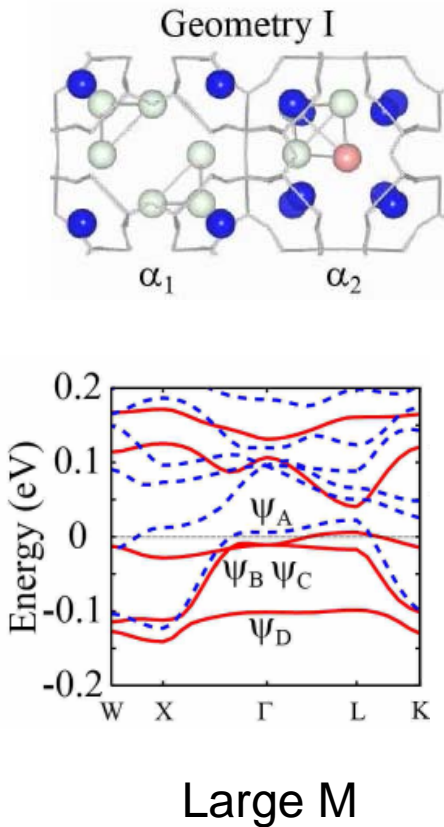
Total energy & magnetic moment



	Geom I	Geom II	Geom III	Geom IV
$\Delta E_{\text{tot}} (\text{eV})$	0	1.02	1.49	0.12
$M_1 (\mu_B)$	-0.02	-0.20	0.06	-0.16
$M_2 (\mu_B)$	1.92	0.33	1.36	1.00

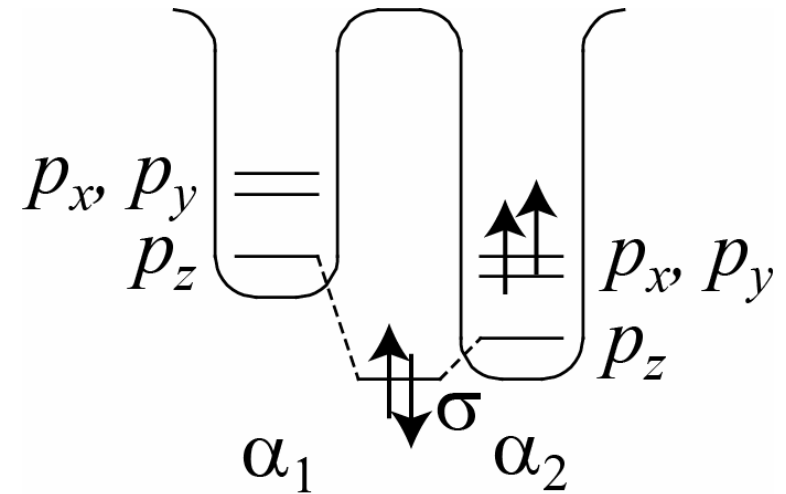
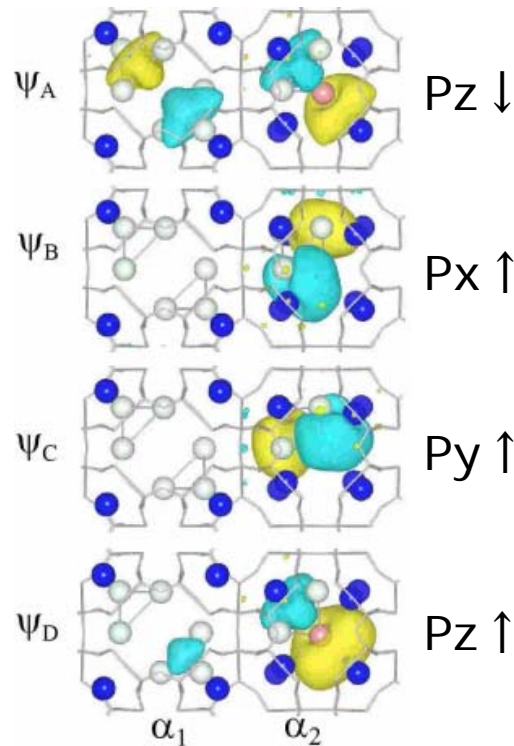
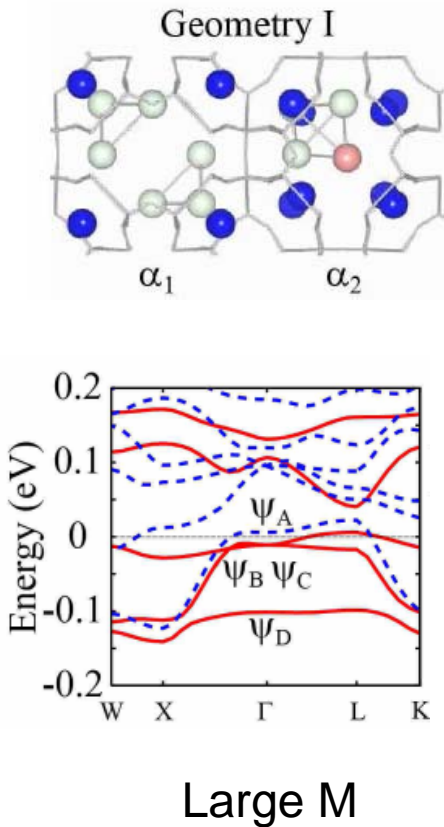
LSDA gives finite magnetic moment

Band dispersion & wave function



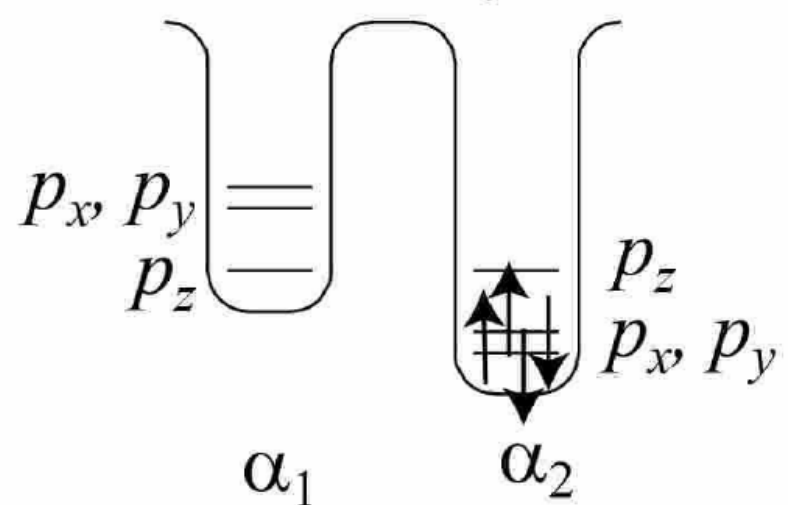
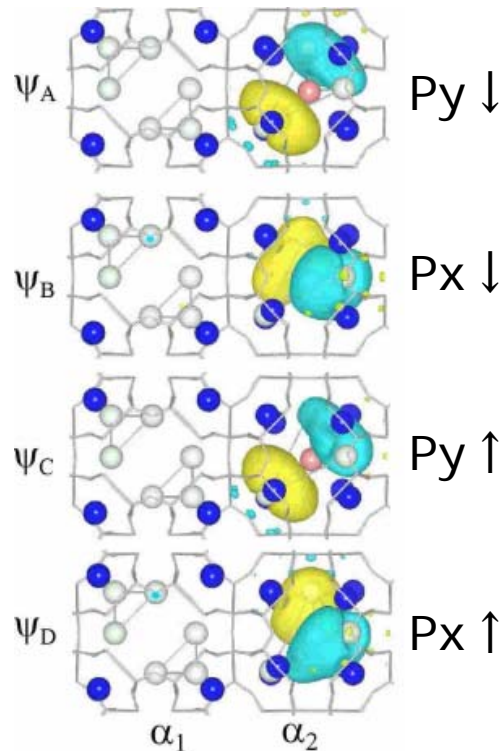
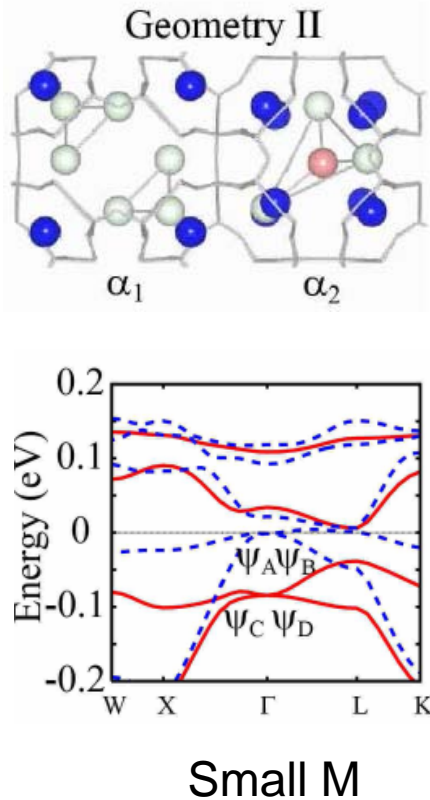
“p-states” are responsible for magnetism

Band dispersion & wave function



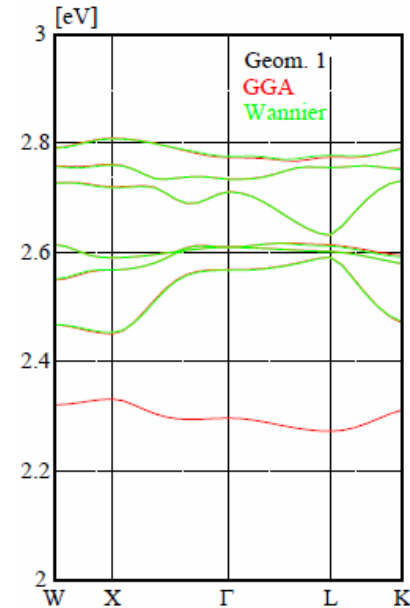
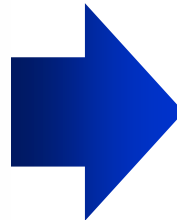
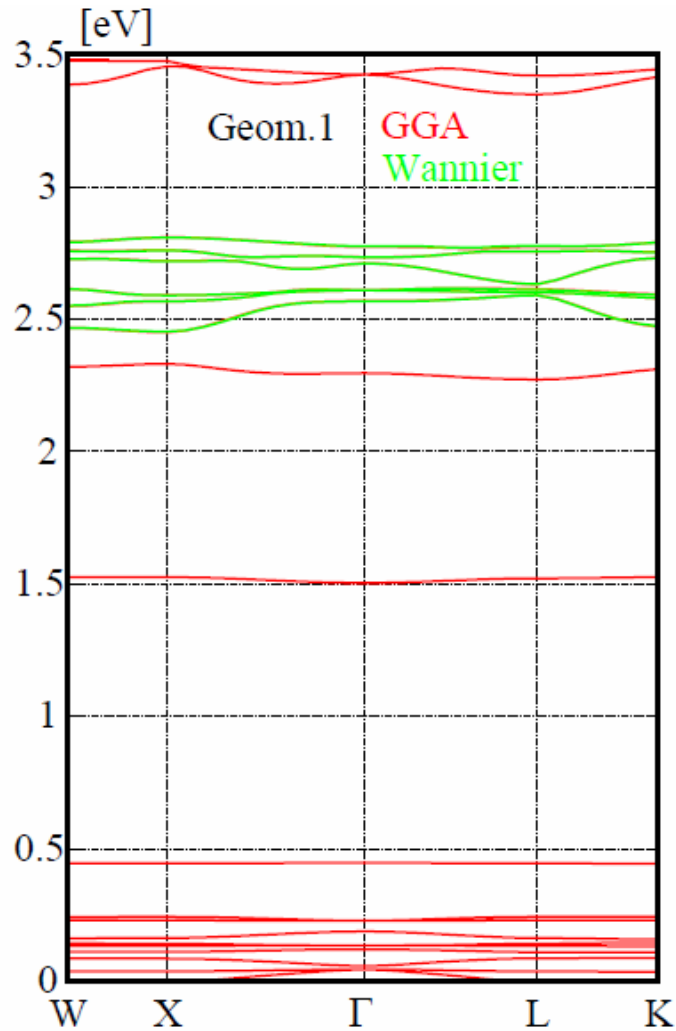
“p-states” are responsible for magnetism

Band dispersion & wave function

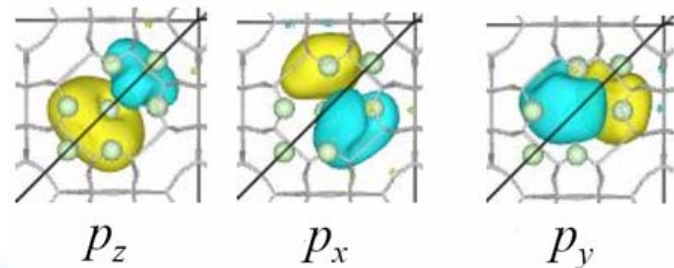


“p-states” are responsible for magnetism

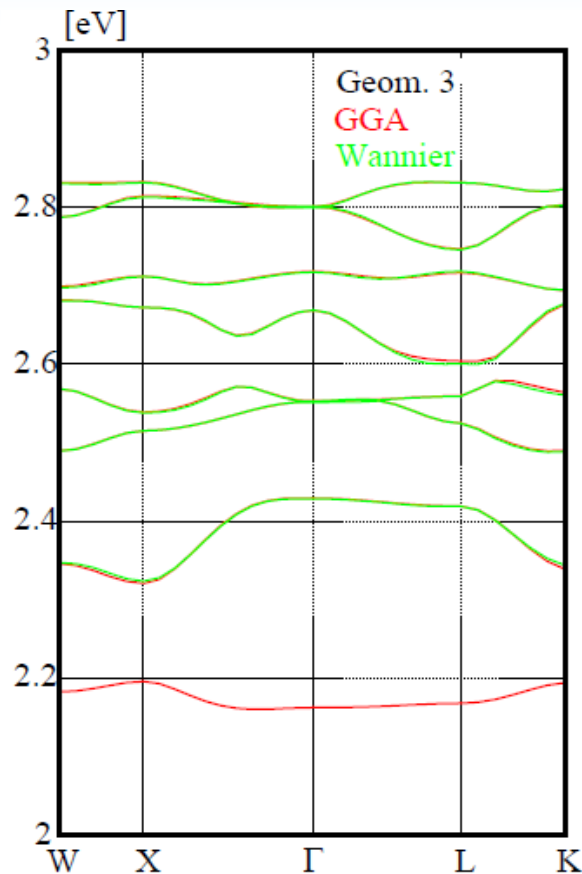
Effective low-energy model



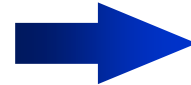
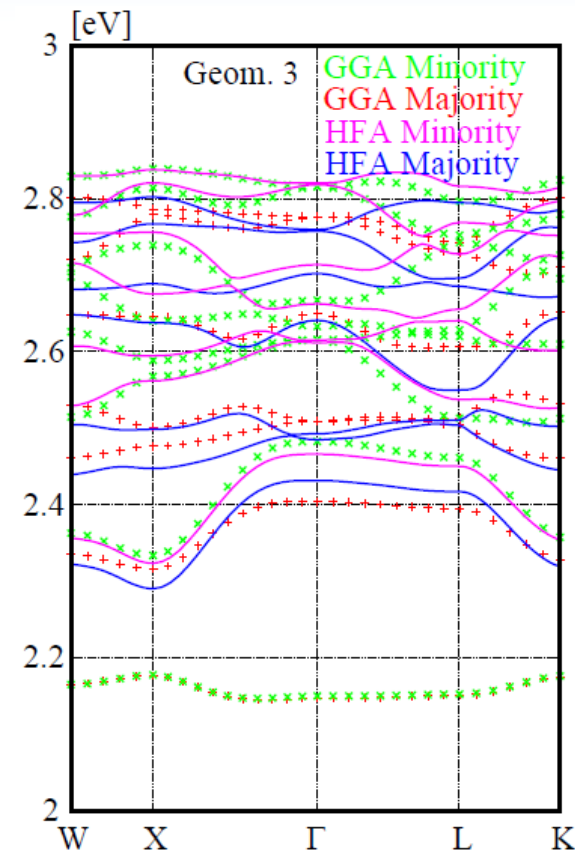
Maximally localized Wannier fn.



Hartree-Fock calculation for the effective model



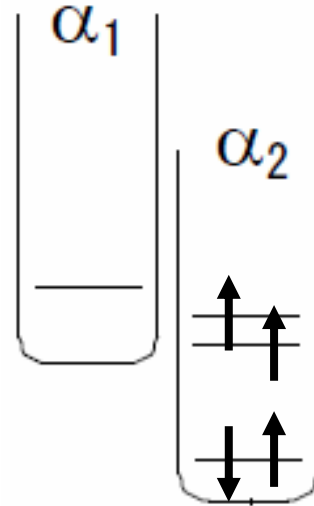
U, J

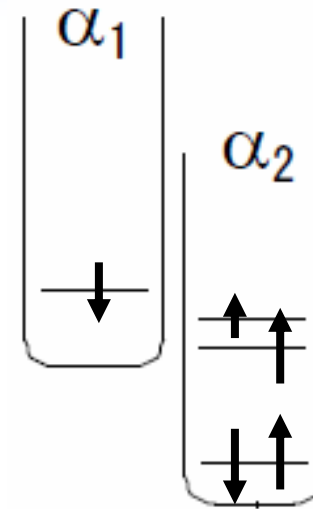
Multi-orbital Hubbard model for superatom p states describes the magnetism

How large is U ?

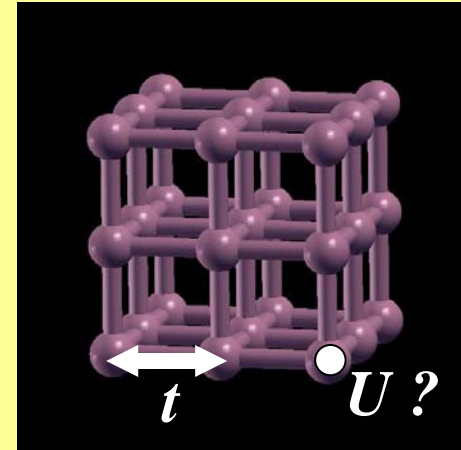
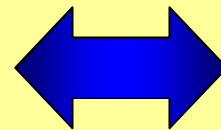
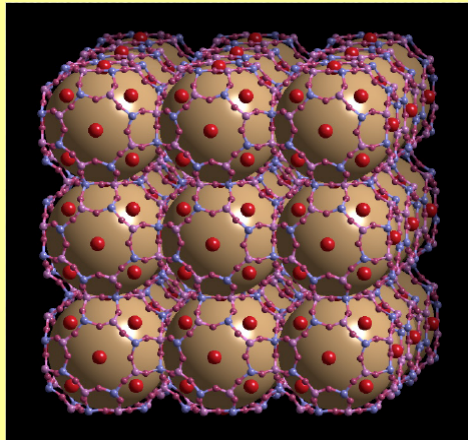
LSDA
(small U)



Large U



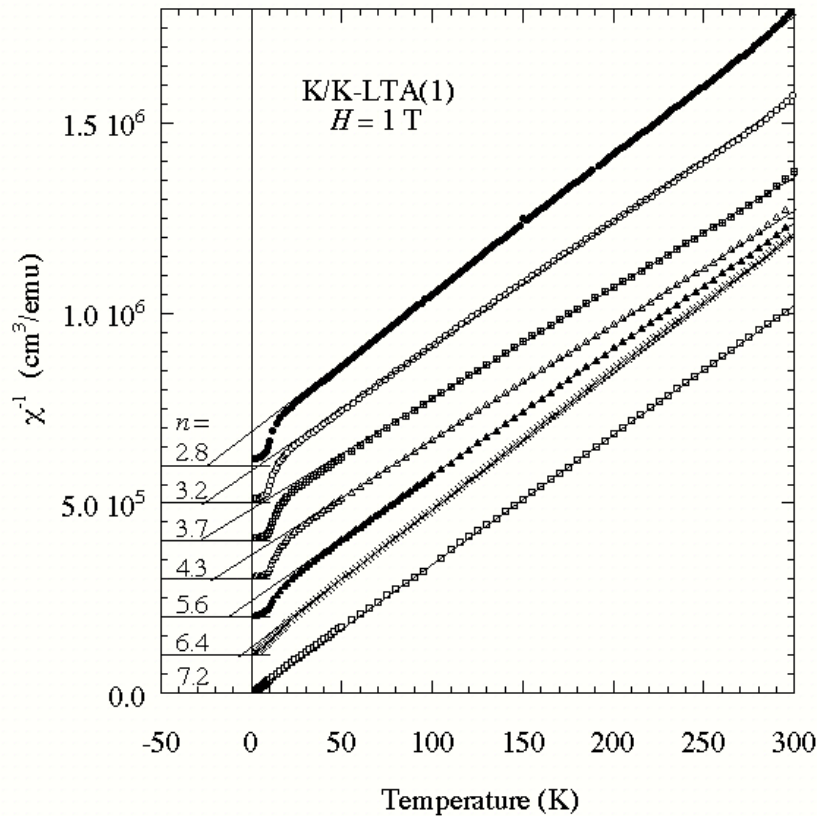
Ferrimagnetic GS



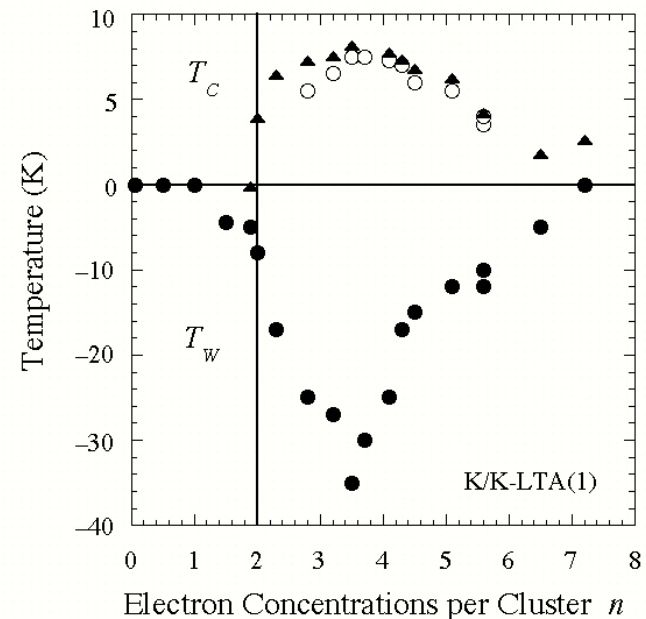
K-loaded zeolite LTA $(K_n + K_{12}Al_{12}Si_{12}O_{48})$

Ground state = ferromagnetic

Nozue et al, PRL68 3789(1992)



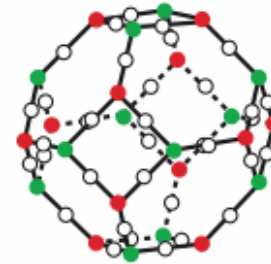
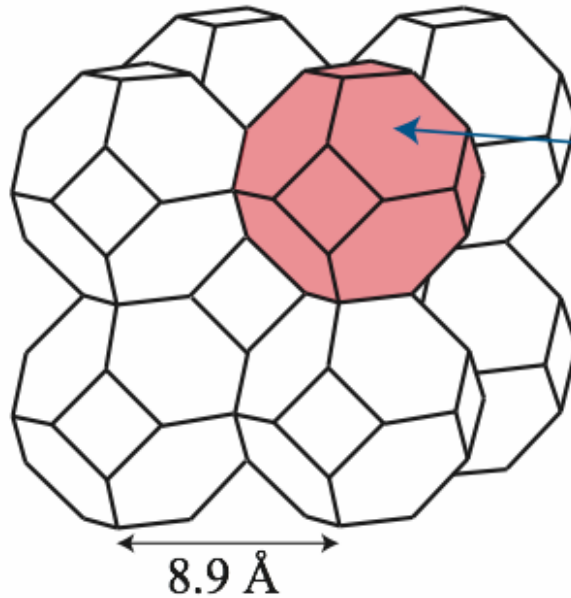
Curie temperature & Weiss temperature



(Nozue *et al*, 1992,
Nakano *et al*, 2002)

Hydrogen superatom in zeolite

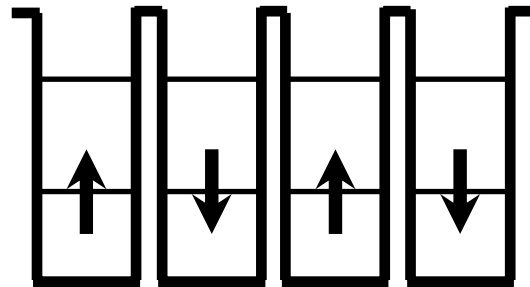
Sodalite (SOD-type structure)



46 atoms / unit cell

β -cage (sodalite cage)

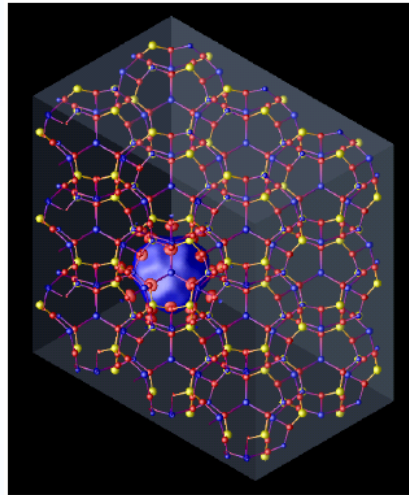
inside diameter: ~ 7 Å



1 electron in 1s state

U of 1s ?

Ab initio estimate of Hubbard U in zeolites



K. Nakamura, T. Koretsune & R. Arita
PRB 80 174420 (2009)

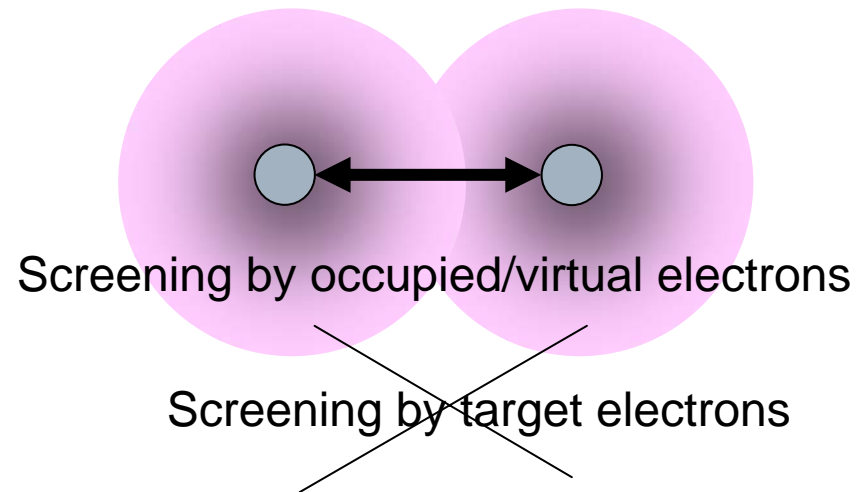
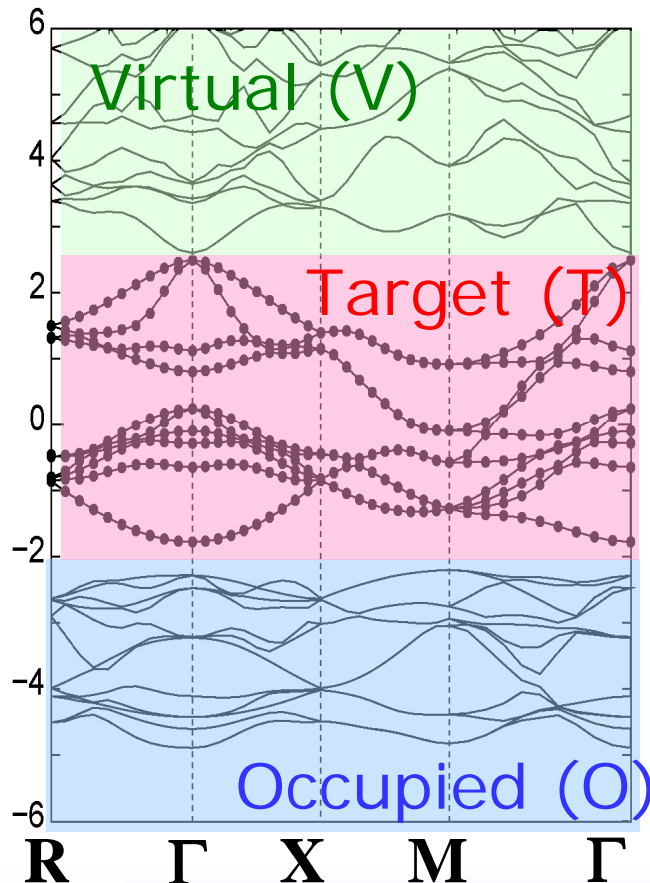
Estimate of interaction parameters by constrained RPA

Aryasetiawan et al, PRB 70, 195104 (2004)
 Solovyev-Imada, PRB 71, 045103 (2005)

$$W = (1 - v\chi)^{-1} v$$

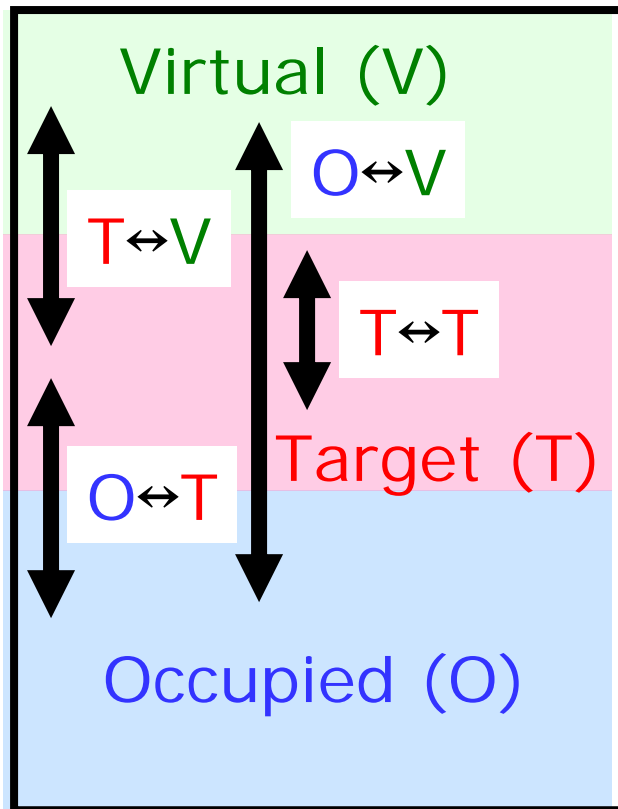
Full RPA polarizability:

$$\chi = \sum_i^{occ} \sum_j^{unocc} \frac{\psi_i(r)\psi_j^*(r)\psi_i^*(r')\psi_j(r')}{\omega - \varepsilon_j + \varepsilon_i \pm i\delta}$$



Estimate of interaction parameters by constrained RPA

$$W = (1 - v\chi)^{-1} v$$



Full RPA polarizability:

$$\chi = \sum_i^{occ} \sum_j^{unocc} \frac{\psi_i(r)\psi_j^*(r)\psi_i^*(r')\psi_j(r')}{\omega - \varepsilon_j + \varepsilon_i \pm i\delta}$$



$$\chi = \sum_{O \leftrightarrow T} + \sum_{T \leftrightarrow V} + \sum_{O \leftrightarrow V} + \sum_{T \leftrightarrow T}$$

$$\chi_r = \sum_{O \leftrightarrow T} + \sum_{T \leftrightarrow V} + \sum_{O \leftrightarrow V} \quad \chi_d = \sum_{T \leftrightarrow T}$$

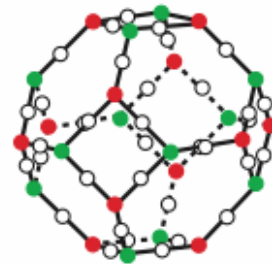
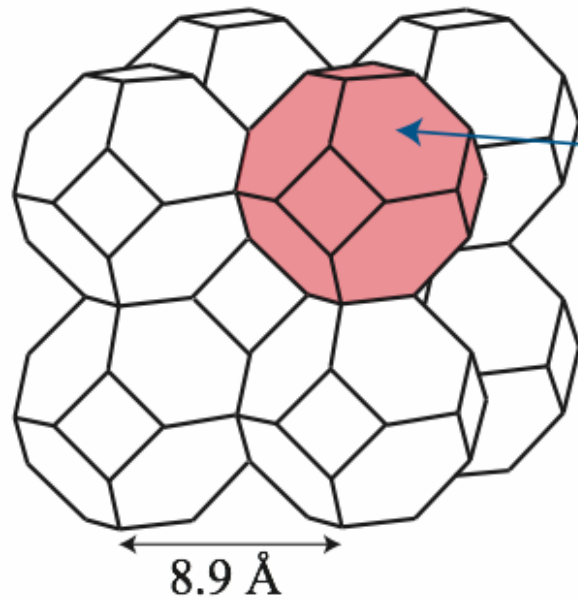
$$W_{eff} = (1 - v\chi_r)^{-1} v$$

$$W = \frac{v}{1 - v\chi} = \frac{W_{eff}}{1 - W_{eff}\chi_d}$$

$$U_{\mathbf{R}} = \langle w_{\mu\mathbf{0}} w_{\mu\mathbf{0}} | W_{eff} | w_{\nu\mathbf{R}} w_{\nu\mathbf{R}} \rangle$$

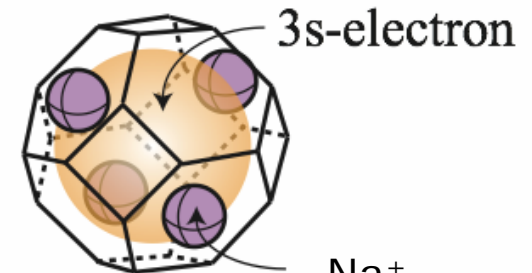
Sodalite

Sodalite (SOD-type structure)

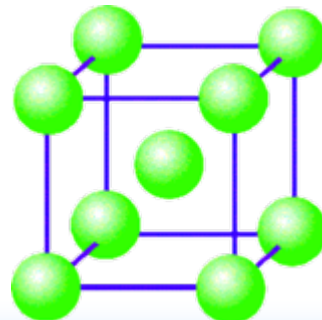


β -cage (sodalite cage)

inside diameter: $\sim 7 \text{ \AA}$

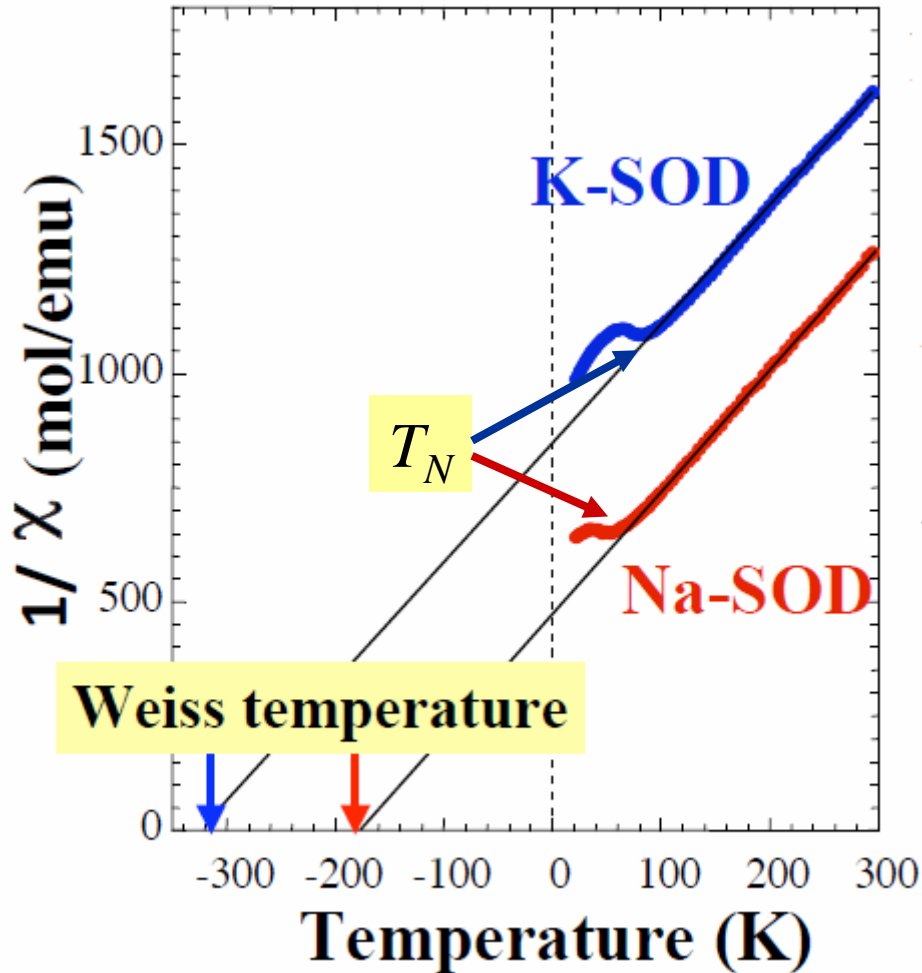


Na⁺
 K⁺
 Rb⁺

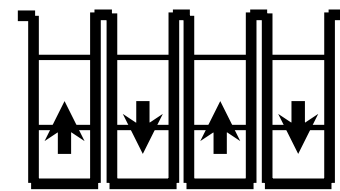


“BCC lattice” of β -cage

Black sodalite: spin susceptibility



Nozue et al.



Slope of $1/\chi$
 → $s = 1/2$ spins
 in each cage

$T_N < 1$: frustrated J_1 - J_2 Heisenberg model
 on BCC lattice

High temperature expansion for χ

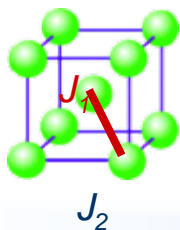
S=1/2 Heisenberg model $H = \sum_{\langle kl \rangle} J_{kl} S_k \cdot S_l = \sum_{\langle kl \rangle} h_{kl}$

Power series expansion around $\beta=1/k_B T=0$

$$\begin{aligned} \langle S_i^z \cdot S_j^z \rangle &= \frac{\text{Tr} e^{-\beta H} S_i^z \cdot S_j^z}{\text{Tr} e^{-\beta H}} \\ &= \langle S_i^z \cdot S_j^z \rangle_c - \beta \left(\sum_{kl} \langle S_i^z \cdot S_j^z h_{kl} \rangle_c \right) + \frac{\beta^2}{2} \left(\sum_{klmn} \langle S_i^z \cdot S_j^z h_{kl} h_{mn} \rangle_c \right) + \dots \end{aligned}$$

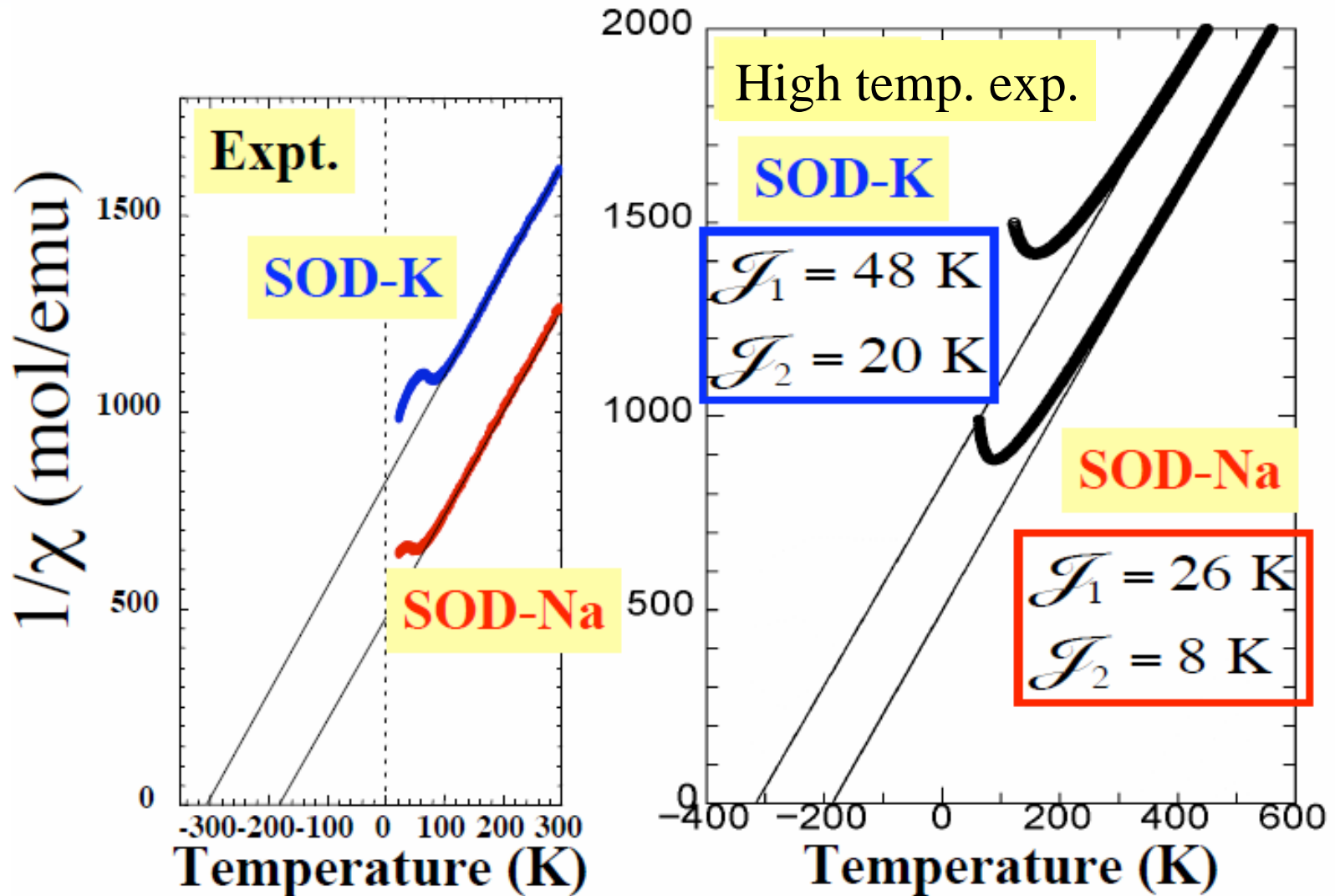
- Computer-aided analytical method
- Calculation for thermodynamic limit (Free from finite-size effect)

For frustrated BCC lattice:



$$\begin{aligned} \chi &= \frac{N(g\mu_B)^2 \beta}{4} (1 + (4J_1 + 3J_2)\beta + (12J_1^2 + 24J_1J_2 + 6J_2^2)\beta^2 + \\ &\quad \left(\frac{104}{3} J_1^3 + 105J_1^2J_2 + 84J_1J_2^2 + 11J_2^3 \right) \beta^4 \dots) \end{aligned}$$

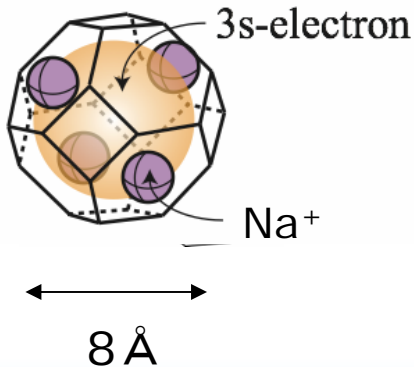
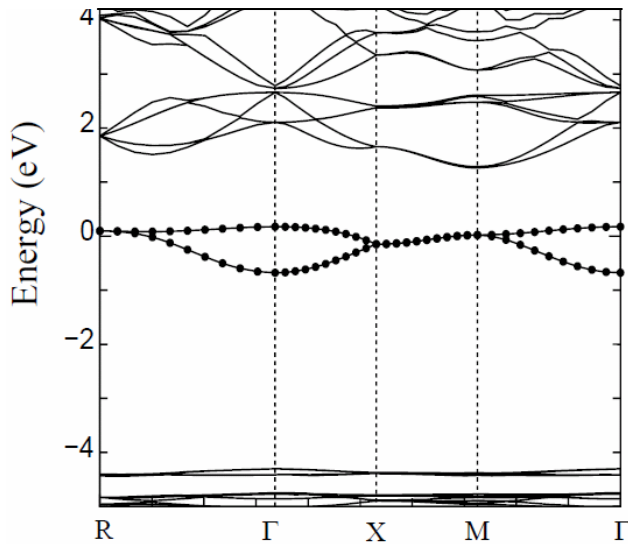
Estimate of J_{exp}



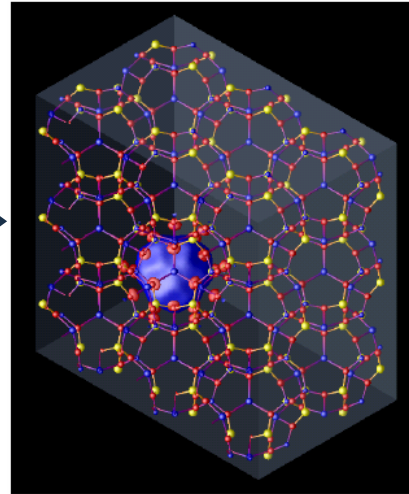
Comparison between J_{exp} and J_{theory}

Procedure of downfolding

LDA



Maximally localized Wannier functions



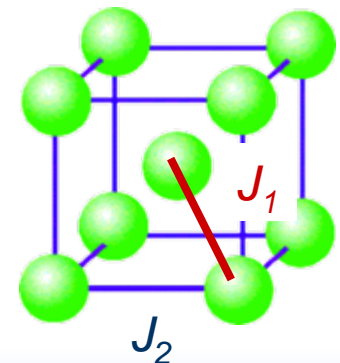
Estimate of
int. parameters
by cRPA

cRPA

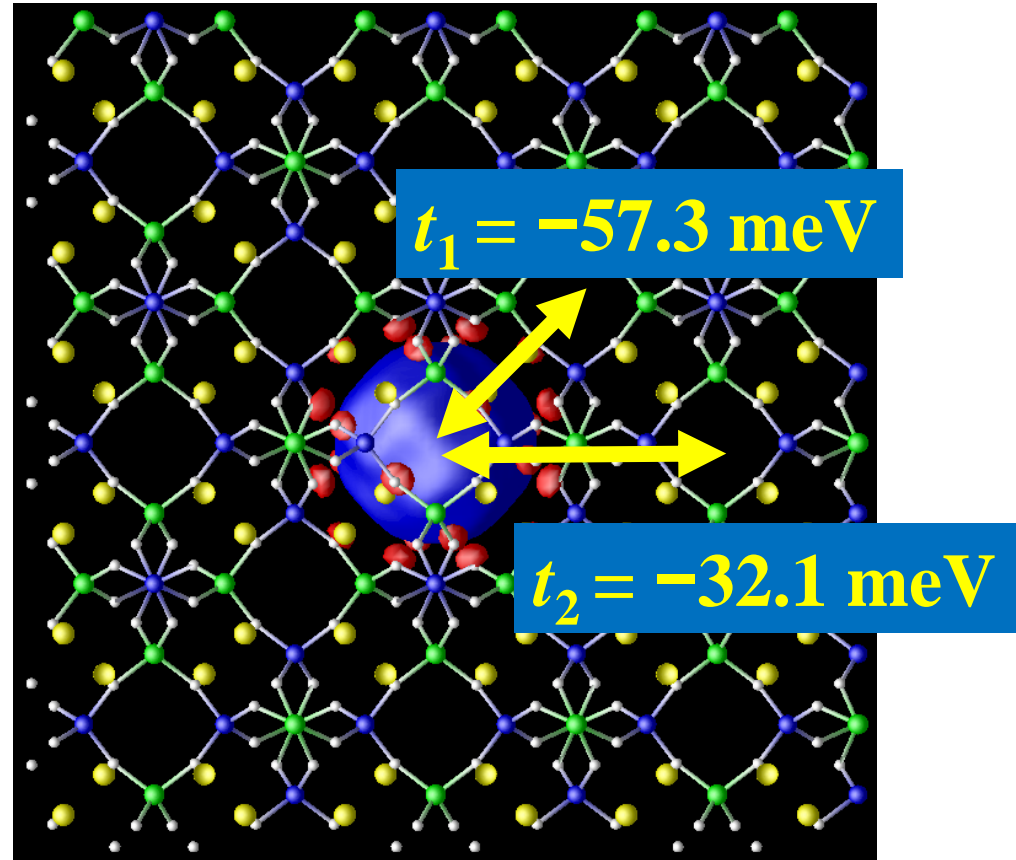
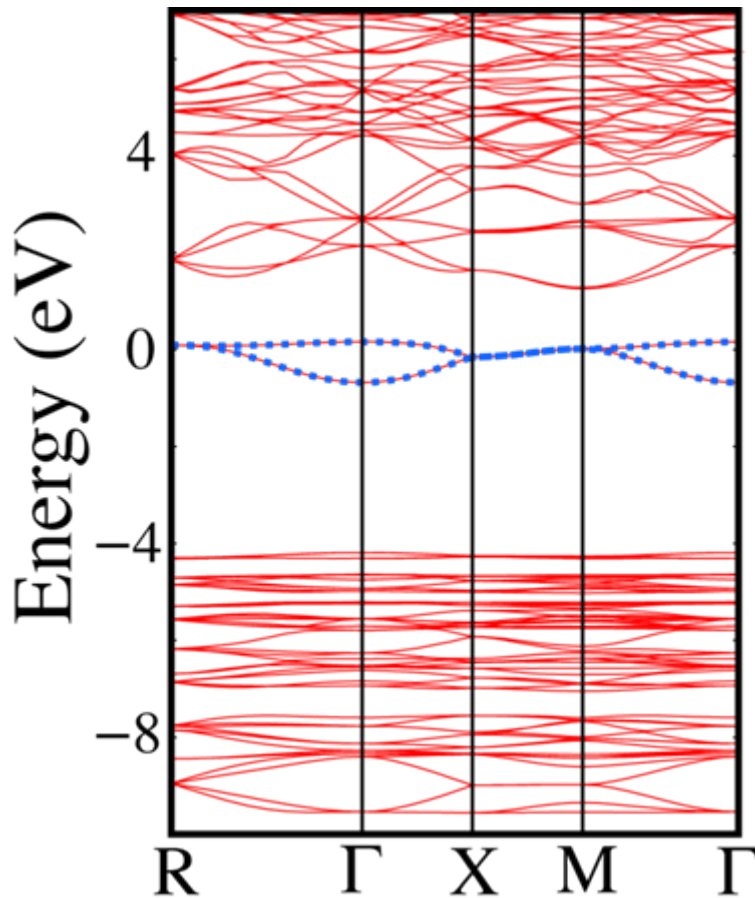
Extended Hubbard model

$$\mathcal{H} = \sum_{ij\sigma} t_{ij} c_{i\sigma}^\dagger c_{j\sigma} + \frac{1}{2} \sum_{ij\sigma\tau} V_{ij} c_{i\sigma}^\dagger c_{j\tau}^\dagger c_{j\tau} c_{i\sigma} + \frac{1}{2} \sum_{ij\sigma\tau} J_{ij} c_{i\sigma}^\dagger c_{j\tau}^\dagger c_{i\tau} c_{j\sigma}$$

Heisenberg model

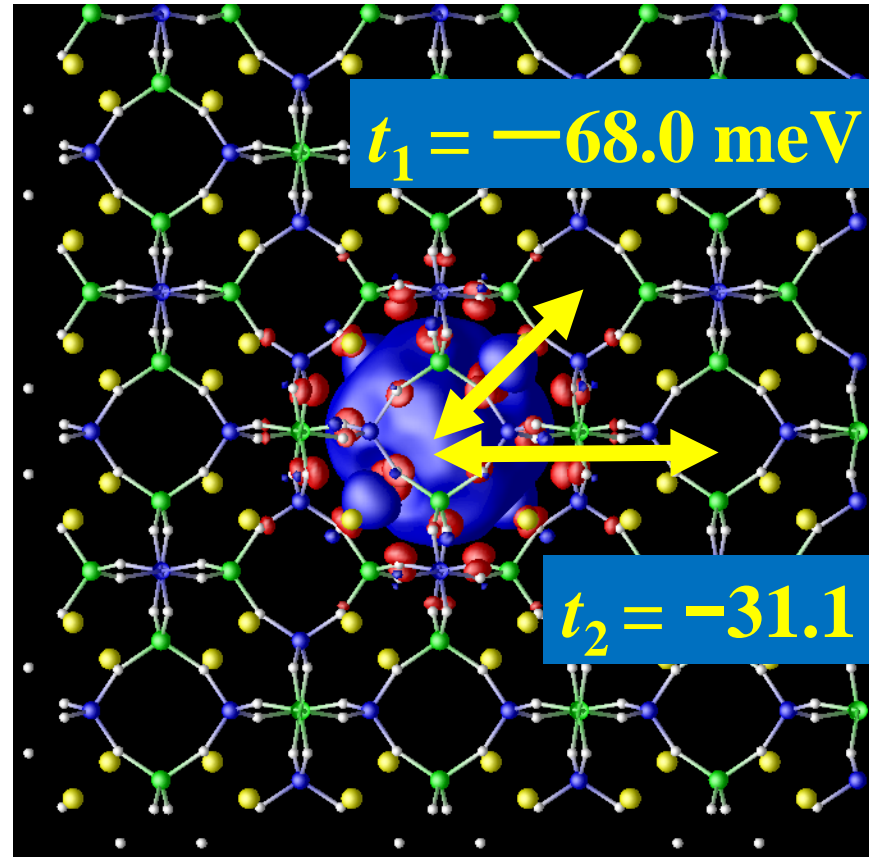
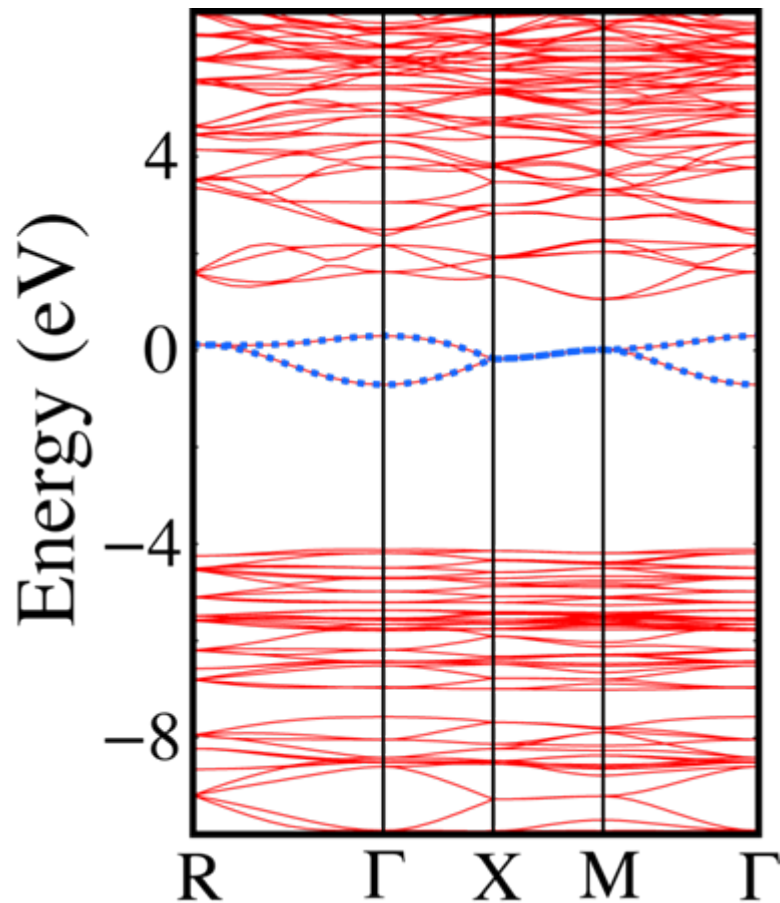


Downfolding (1) MaxLoc of Na-SOD



Superatom s state

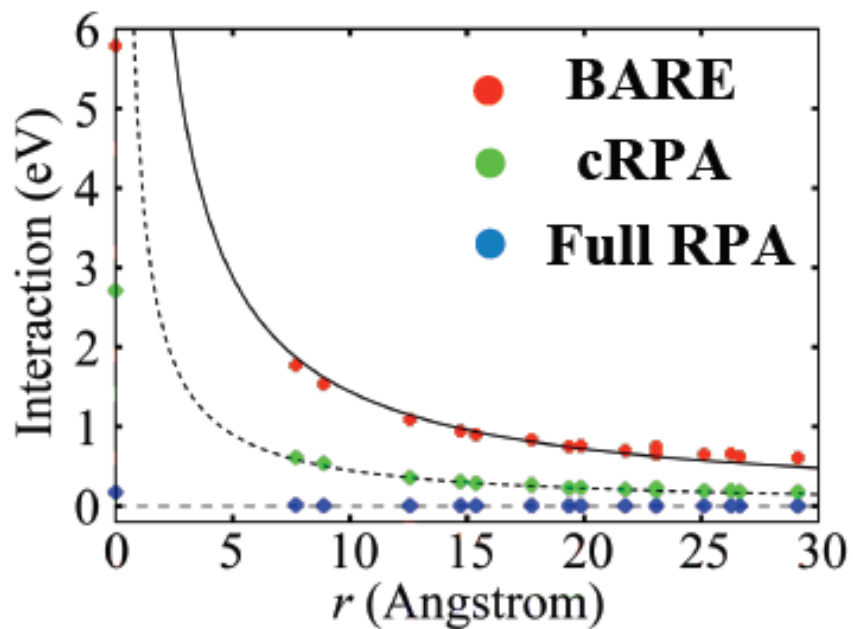
Downfolding (1) MaxLoc of K-SOD



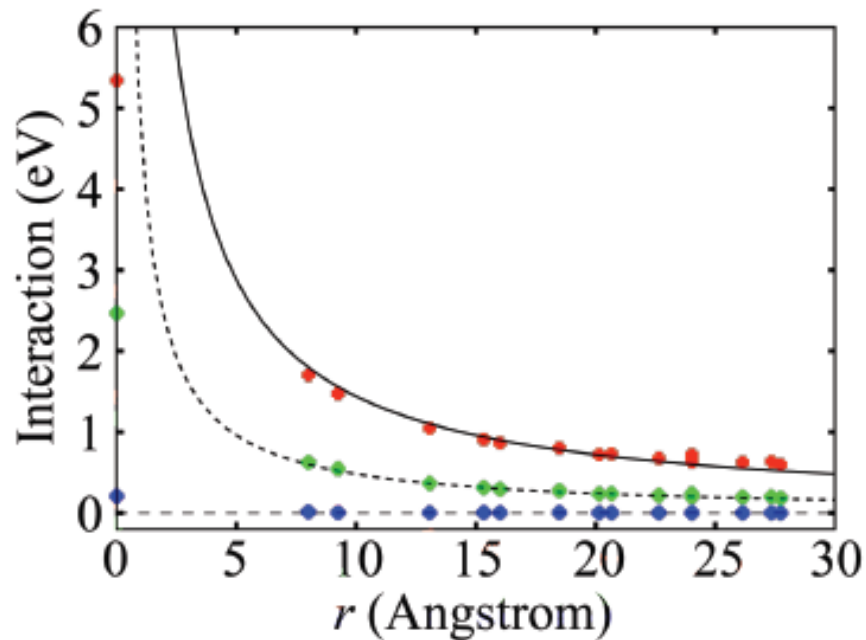
Superatom s state

Downfolding (2): cRPA

Na-SOD



K-SOD



Downfolding (2): cRPA

	sodium electrosodalite			potassium electrosodalite			
	bare	cRPA	RPA	bare	cRPA	RPA	
U	5.79	2.71	0.17	5.34	2.47	0.21	} Unit: eV
V_1	1.77	0.61	0.01	1.70	0.63	0.01	
V_2	1.54	0.54	0.00	1.47	0.54	0.00	
J_1	56.9	27.0	22.4	97.2	44.5	34.8	} Unit: K
J_2	22.4	10.6	8.8	20.9	9.9	8.3	
K_1	18.9	36.3	476.1	29.4	58.2	535.6	
K_2	5.6	11.0	143.7	5.8	11.7	109.2	

$$K_{ij} = \frac{2t_{ij}^2}{U - V_{ij}}$$

Comparison with experiments

$$\mathcal{H}_{ex} = \sum_{\langle NN \rangle} \mathcal{J}_1 \mathbf{S}_i \cdot \mathbf{S}_j + \sum_{\langle NNN \rangle} \mathcal{J}_2 \mathbf{S}_i \cdot \mathbf{S}_j$$

$$\mathcal{J}_{ij} = K_{ij} - J_{ij}$$

$$K_{ij} = \frac{2t_{ij}^2}{U - V_{ij}}$$

\mathcal{J}_1 \mathcal{J}_2 : Result of subtle balance between J & K

Unit: K

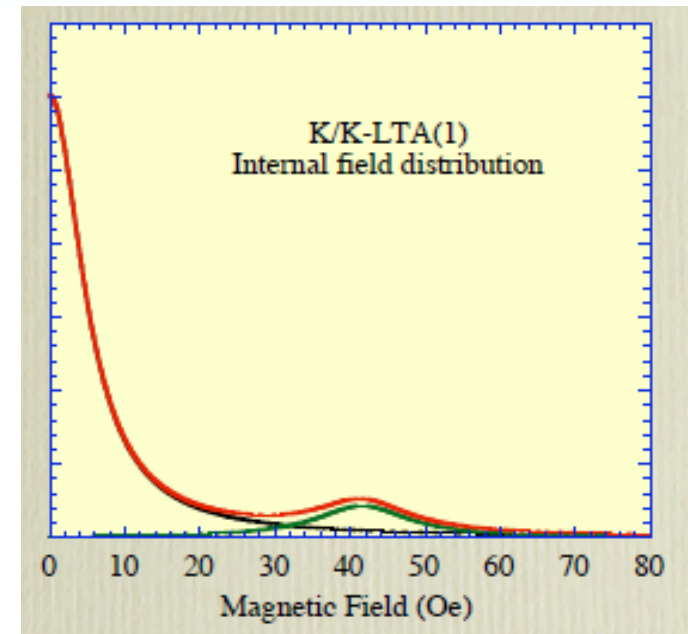
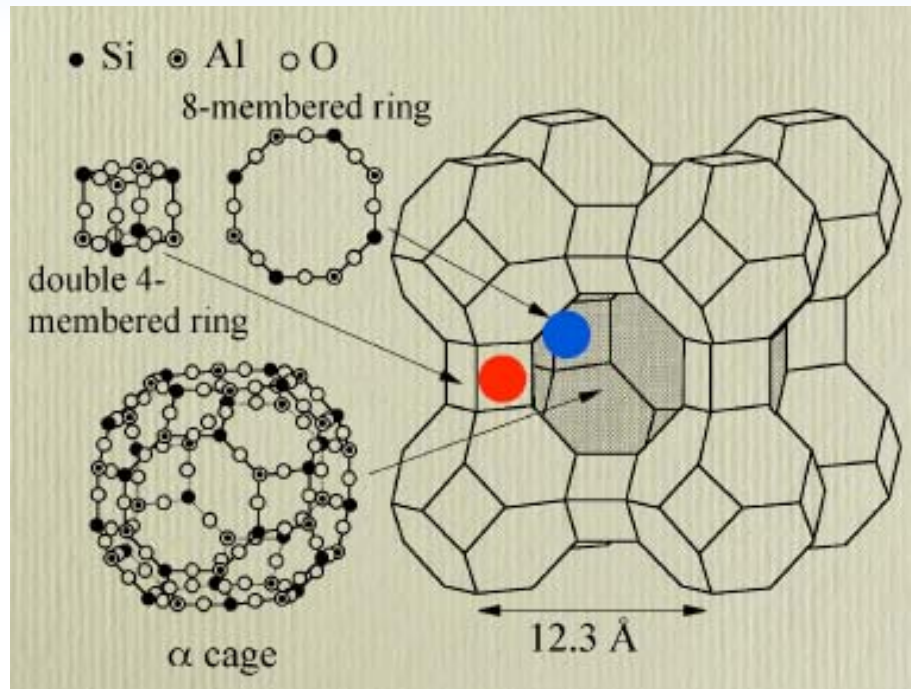
	sodium electrosodalite				potassium electrosodalite			
	bare	cRPA	RPA	Expt.	bare	cRPA	RPA	Expt.
\mathcal{J}_1	-37.9	9.3	453.7	26	-67.8	13.8	500.8	48
\mathcal{J}_2	-16.8	0.4	134.9	8	-15.1	1.8	100.9	20

Summary

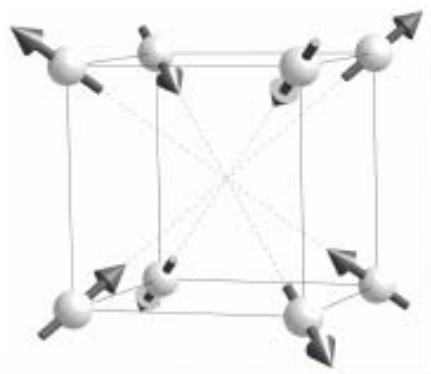
- Ferromagnetism in zeolite LTA
 - Magnetism described by “superatom” picture

- *ab initio* derivation of low-energy model
 - Estimate of interaction parameters by cRPA
 - First step to “tailor-made correlation”

Future problem: relativistic effect in zeolite



Nozue et al.



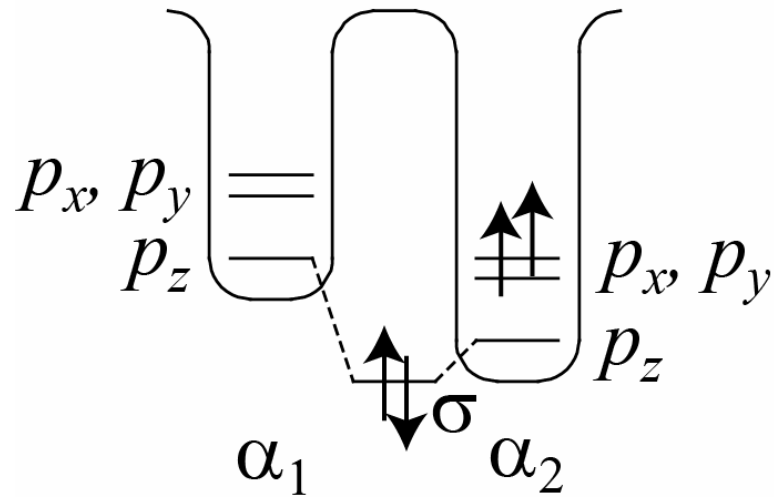
Canted AF due to the Dzyaloshinskii-Moriya int. ?

Large LS coupling ?

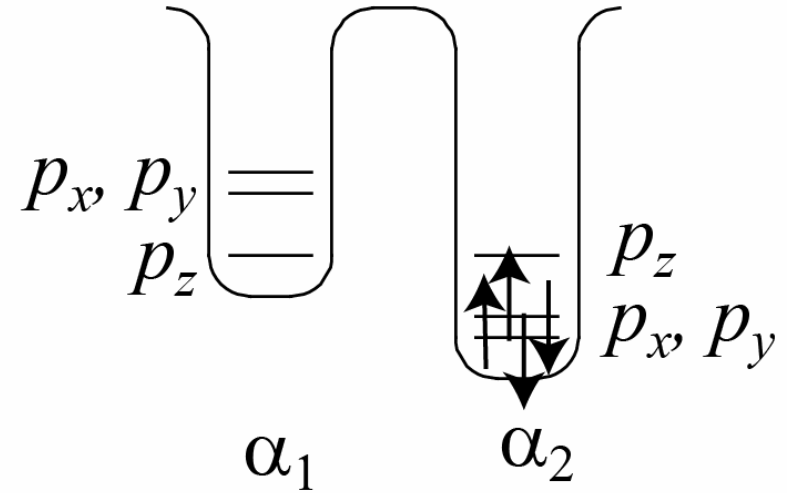
Relativistic effect in materials of light elements



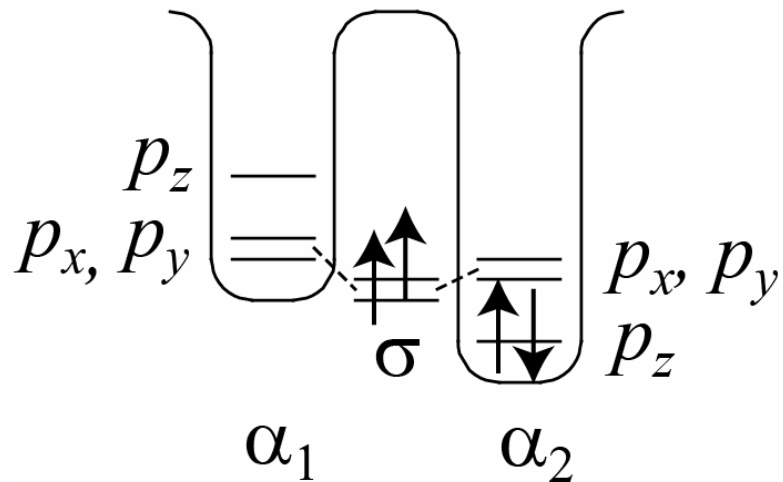
Geometry I



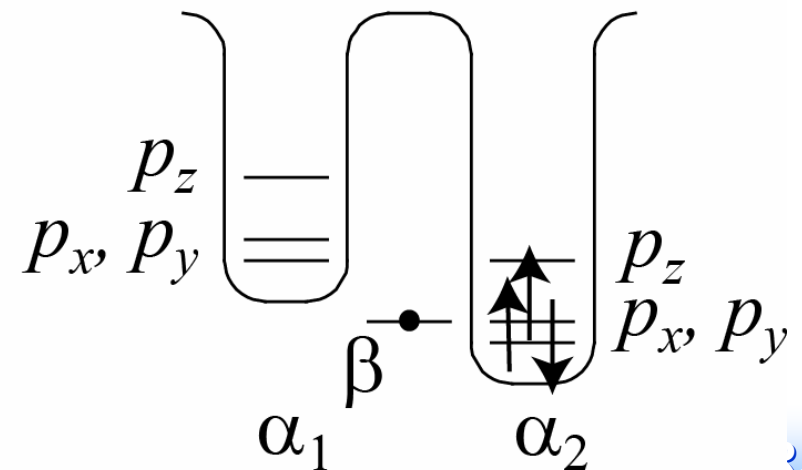
Geometry II



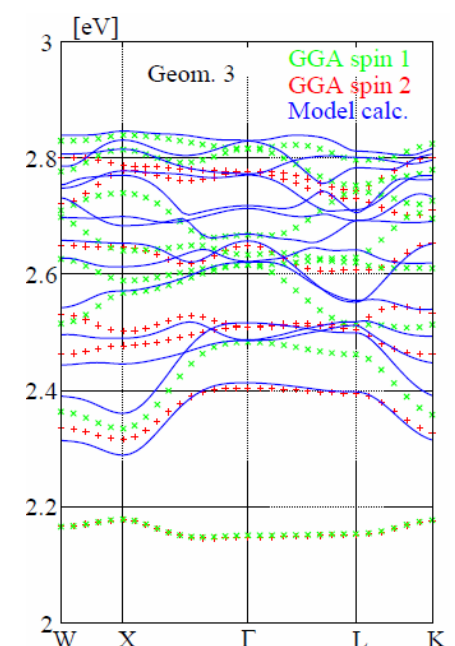
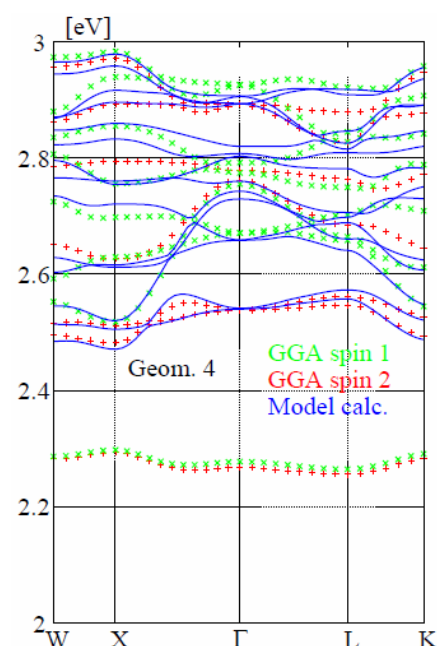
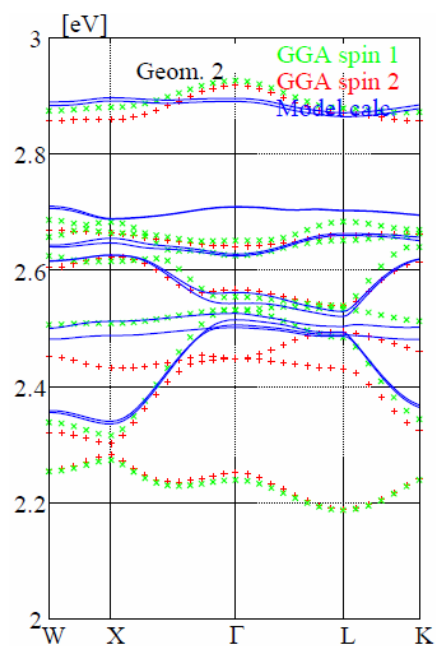
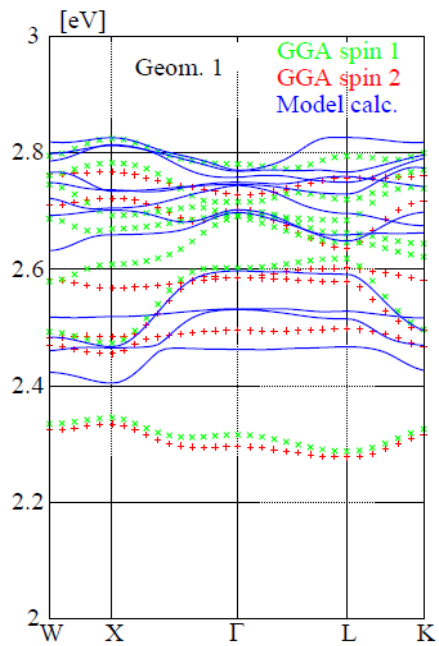
Geometry III



Geometry IV

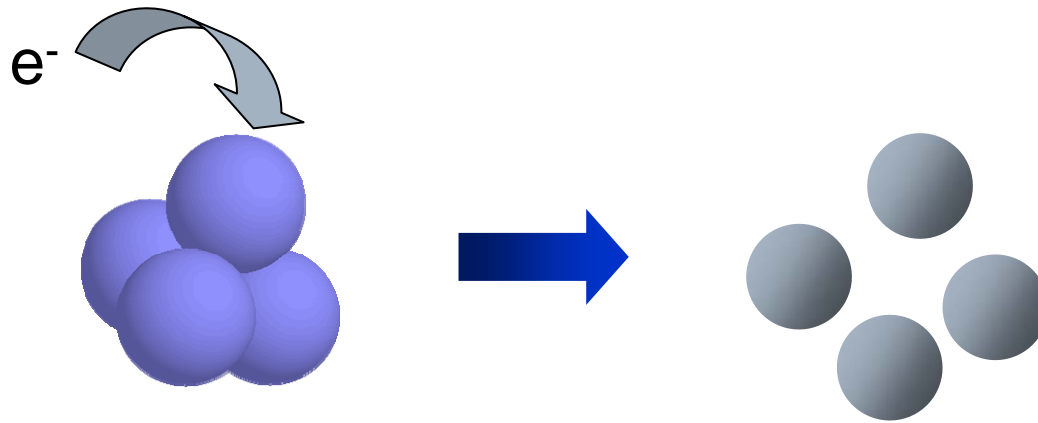


Mean Field calculation for the effective model



Attractive interaction due to ele-phonon coupling

Onsite Coulomb is overestimated ?



$$U_{\text{eff}} = U - S \quad (S: \text{attractive int. due to ele-ph})$$

S can be as large as $0.2 \sim 0.4 \text{ eV}$
 $\rightarrow J_1$ can be $\sim 20 \text{ K}$