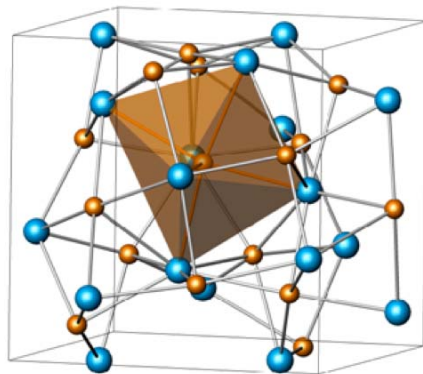




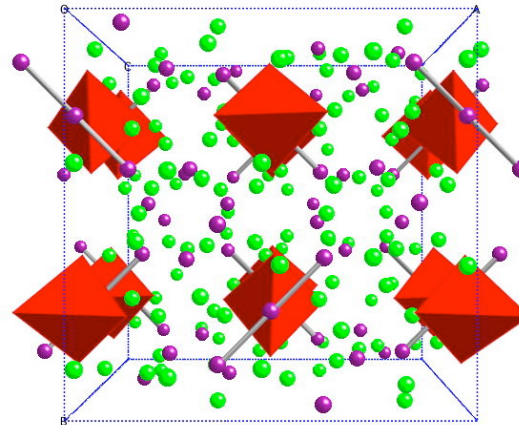
Challenges and Opportunities for Condensed Matter Physics of Thermoelectric Materials

G. Jeffrey Snyder

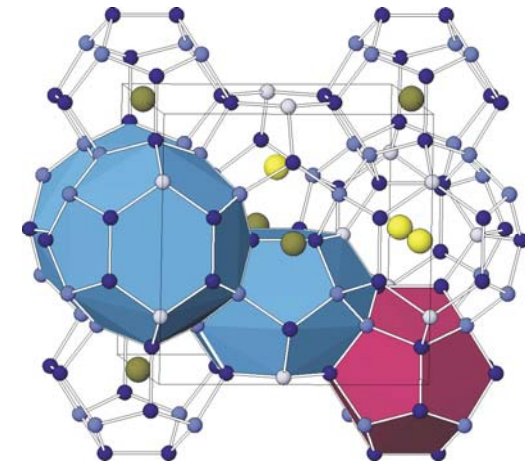
California Institute of Technology
Pasadena, California, USA
<http://thermoelectrics.caltech.edu>



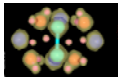
La_3Te_4



$\text{Yb}_{14}\text{MnSb}_{11}$



$\text{Ba}_8\text{Ga}_{16}\text{Ge}_{30}$
Type I Clathrate



THERMOELECTRICS

Toberer, May, Snyder *Chem. Mat.*, (DOI: 10.1021/cm901956r)

Thermoelectric Generator



Thermoelectrics
Convert Heat into Electricity

Heat Flow drives free
electrons and holes
from hot to cold

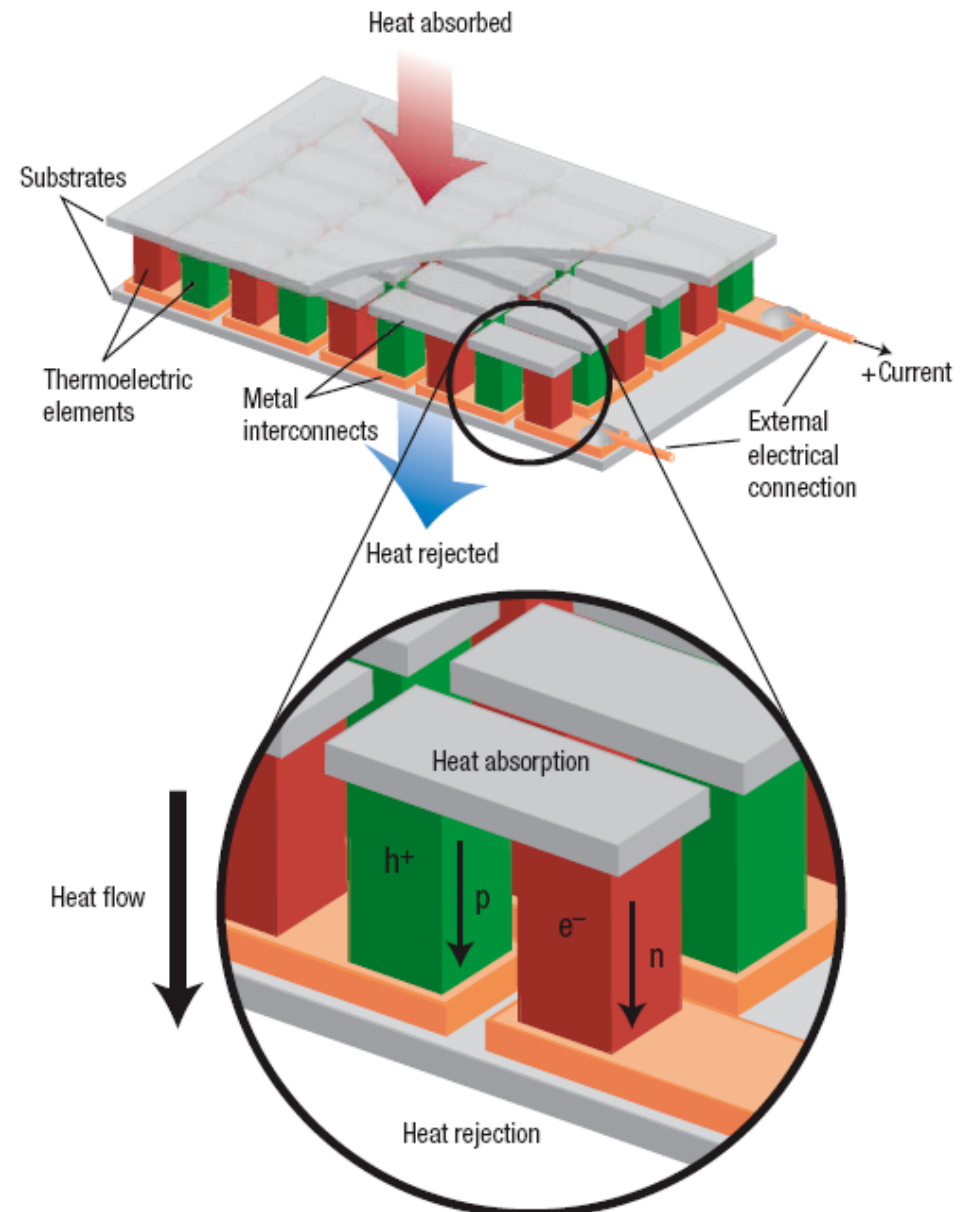
Voltage Produced
Seebeck effect
or Thermoelectric Power

$$V = \alpha \Delta T$$

Seebeck Coefficient α

Efficiency $\sim zT$

$$zT = \frac{\alpha^2 \sigma T}{\kappa}$$



Thermoelectric Applications



Solid State Advantage

- No moving parts
- No maintenance
- Long life
- Scalability



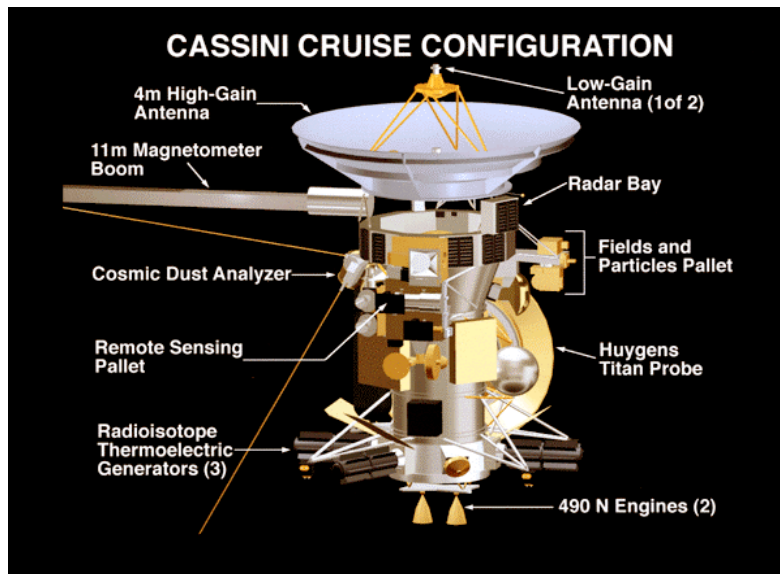
Cooling - Thermal Management

- Small Refrigerators
- Optoelectronics
- Detectors



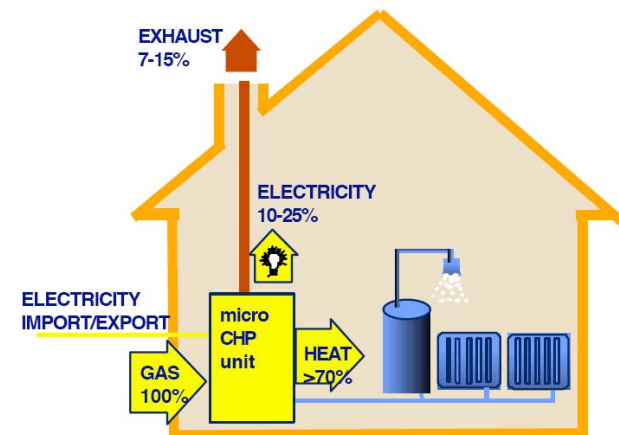
Power Generation (heat to electricity)

- Spacecrafts
- Voyager over 30 years!
- Remote power sources



Future Possibilities

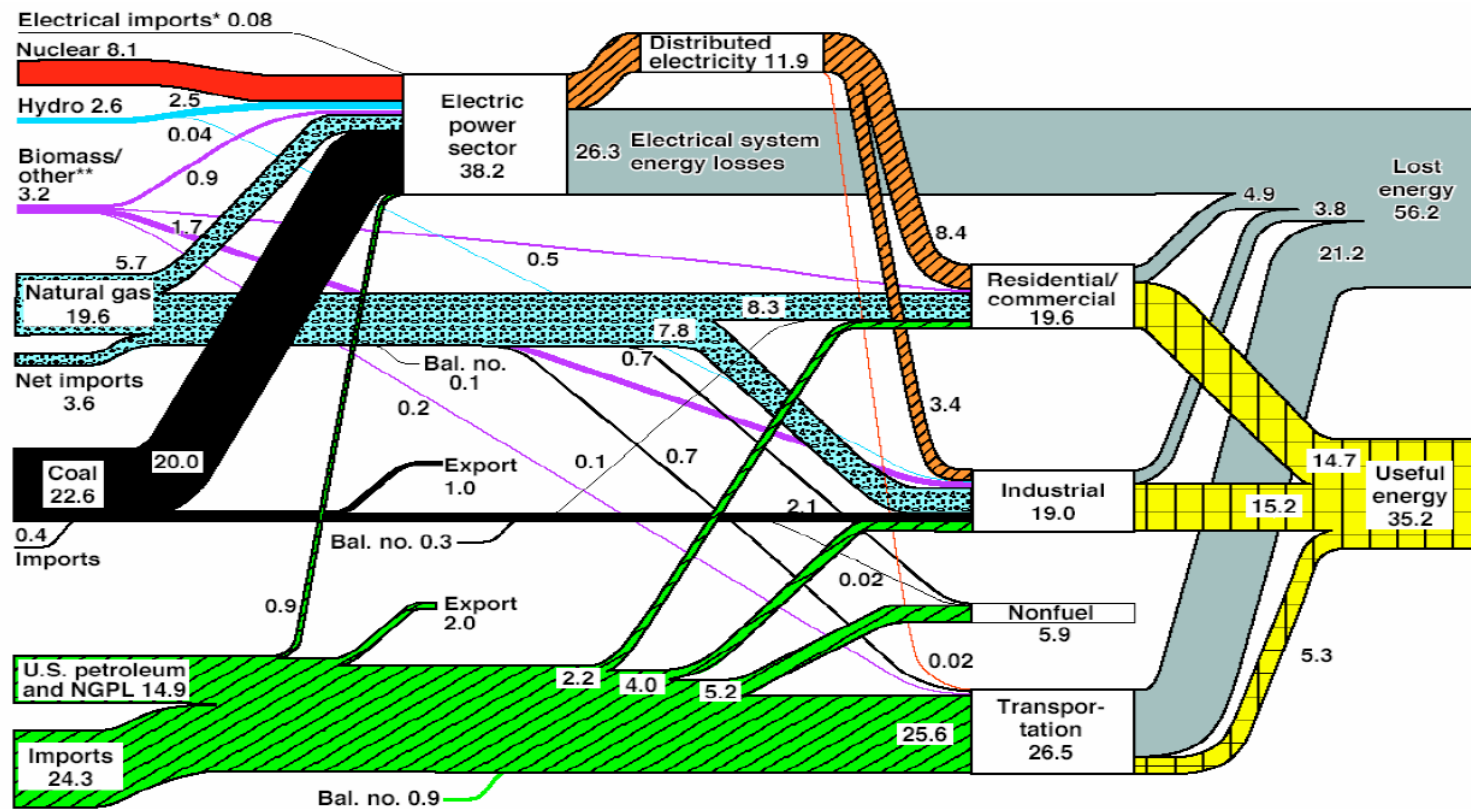
- Waste Heat Recovery
- Automobiles
- CoGeneration
- Distributed Thermal Management



Heat in Energy Use



USA



As in End Product
Home, Business heating systems

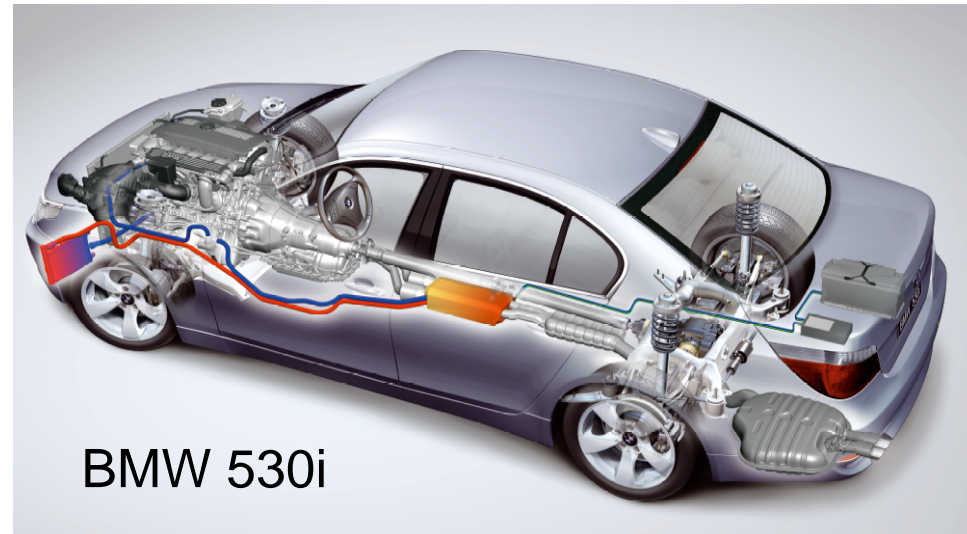
As waste by product
In production of useful energy
In consumption of useful energy

Automotive TE Generator

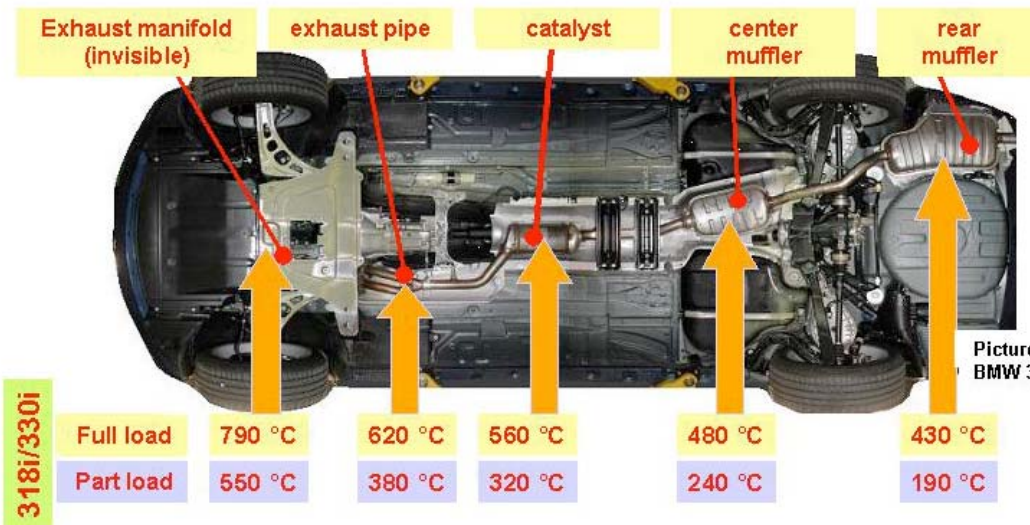


Convert Exhaust heat to electricity

- Replace alternator
- Improve fuel economy by 10%,



BMW 530i



Picture: BMW 318i MJ 04



KITP Outline



Transport Physics of Complex Thermoelectric Materials

Semi-empirical Approach

Solid-State Chemistry Inspired

- Single Parabolic Band
- Rigid Bands
- Acoustic Phonon Scattering

Successful examples of approach

- Predict trends in
 - Seebeck
 - mobility
 - zT
- Optimize material
- Simplify the physics

Challenges - Opportunities

Band Structure Calculations

- Band Gap (within $\sim 0.05\text{eV}$)
- Band Edge Energy (within $\sim 0.01\text{eV}$)
- Effective Mass (within $\sim 0.1m_e$)

Disorder

- Doping at $\sim 1\%$
- Alloying $\sim 10\%$
- Vacancies
- Anti-Site Defects
- Interstitial atoms

Transport Calculations

- Relaxation Time Approximation
- Constant RTA (?)
- Thermal Transport

Exotic TE Materials

- Resonant States
- Nano-structures/e- filtering
- Heavy Fermion
- Kondo
- Correlated Electron Systems



Ideal Thermoelectric Material

Desire High zT Figure of Merit

$$zT = \frac{\alpha^2 \sigma T}{\kappa}$$

“Electron Crystal - Phonon Glass”

α Seebeck Coefficient
High in Crystalline Semiconductor

$$\alpha = \frac{8\pi^2 k_B^2}{3eh^2} m^* T \left(\frac{\pi}{3n} \right)^{2/3}$$

σ Electrical Conductivity
High In Crystalline Metal

$$\sigma = ne\mu$$

κ Thermal Conductivity
Low for Glass

$$\kappa \approx \kappa_l + L\sigma T$$



Carrier Concentration

Desire High zT Figure of Merit

$$zT = \frac{\alpha^2 \sigma T}{\kappa}$$

Conflicting Materials Requirements

α Seebeck Coefficient

Need small n , large m^*

- Semiconductor (Valence compound)

$$\alpha = \frac{8\pi^2 k_B^2}{3eh^2} m^* T \left(\frac{\pi}{\delta n} \right)^{2/3}$$

σ Electrical Conductivity

Need large n , high μ

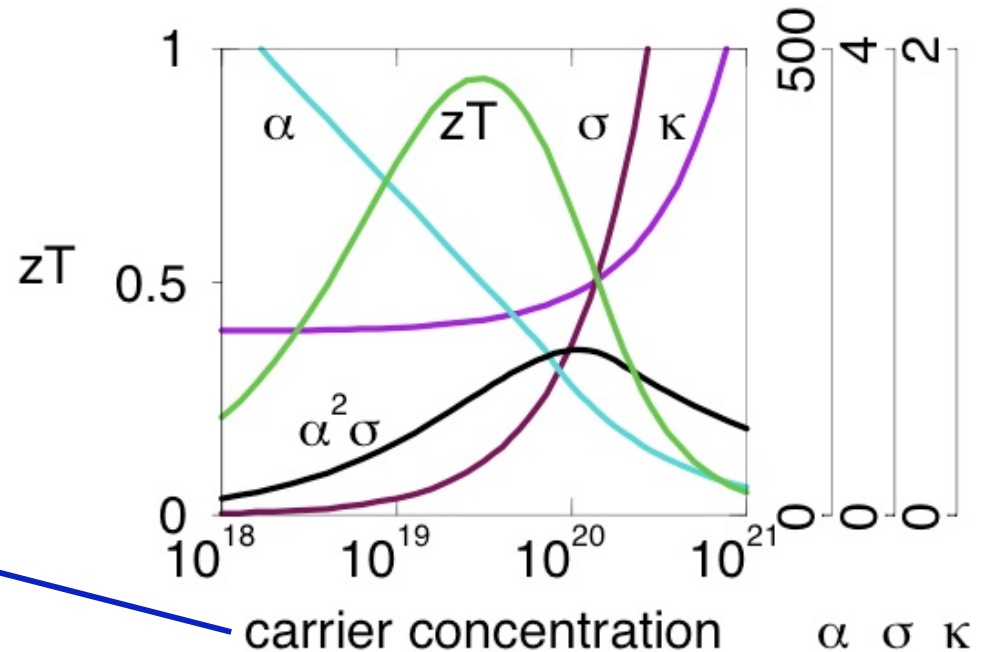
- Metal

$$\sigma = n e \mu$$

κ Thermal Conductivity

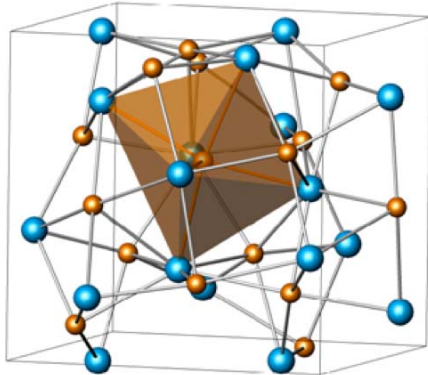
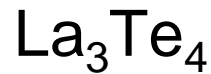
Desire small κ_l , small n

$$\kappa \approx \kappa_l + L T n e \mu$$

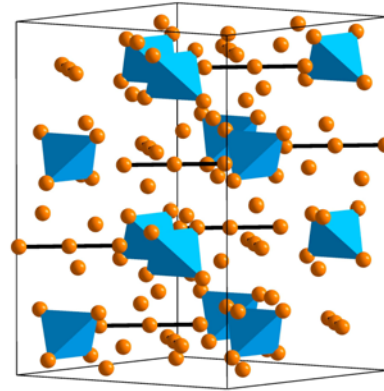


“Electron Crystal - Phonon Glass”

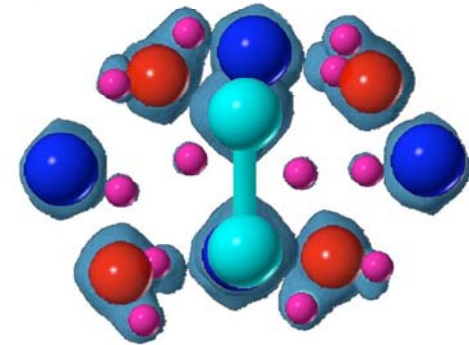
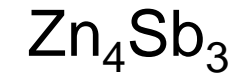
Good Thermoelectric Materials



Ionic



Zintl



Intermetallic

Very different materials but with some common features

- Structural complexity - good for low thermal cond.
- Structurally related to intrinsic semiconductors
near valence balanced compounds using Zintl concepts
- Transport properties of metals

Are Thermoelectric Materials Semiconductors or Metals ?

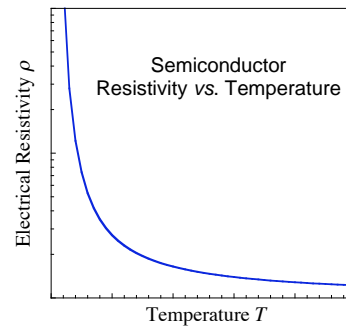
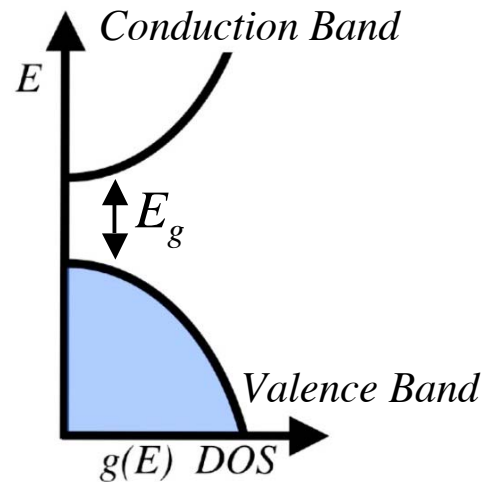
Semiconductors vs. Metals



Semiconductors

Filled Valence Band & Empty
Conduction band Separated by
Band Gap E_g

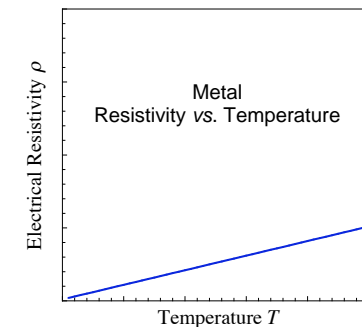
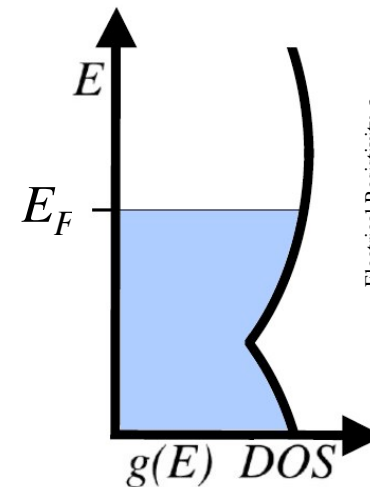
- Elec. resistivity ρ , *decreases* as
Temperature increases
 $\rho \rightarrow \infty$ as $T \rightarrow 0$ K
- charge carriers n_H
increase with Temperature



Metals

no band edges near Fermi Level E_F

- Elec. resistivity ρ , *increases* as
Temperature increases
- charge carriers n_H
constant with Temperature





Valence Semiconductors

Semiconductors

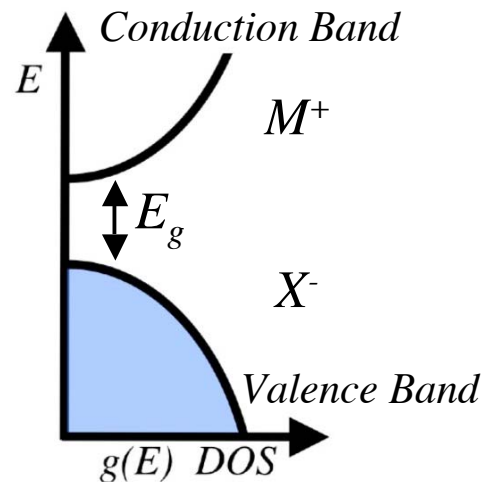
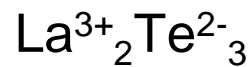
Filled Valence Band & Empty Conduction band Separated by Band Gap E_g

Valence Compounds

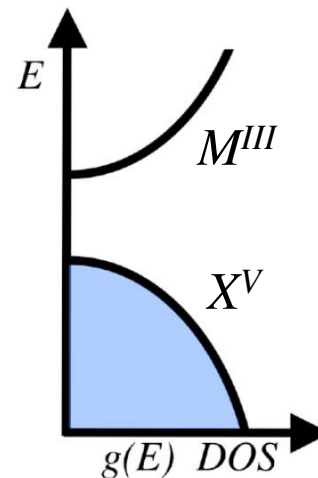
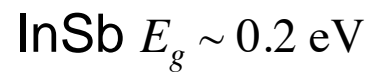
Stoichiometric balance of valences \Rightarrow Semiconductor

- Ionic Compound: Valence Band = filled anion states; Conduction = cation states
- Covalent Compound: Valence Band = filled bonding states; Conduction = antibonding states

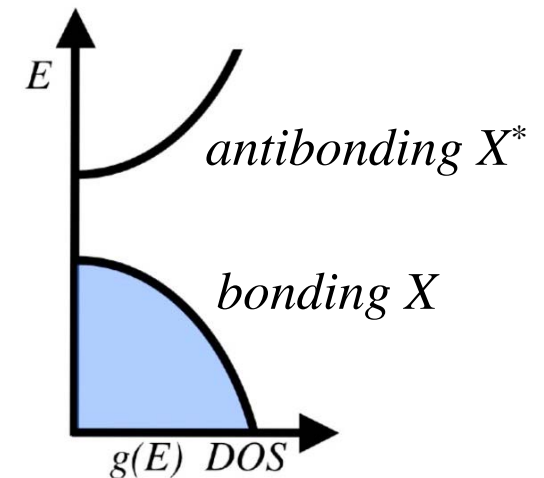
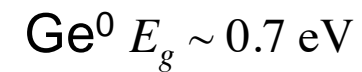
Ionic MX



Polar SC MX



Semicond. X



Zintl Electron Counting

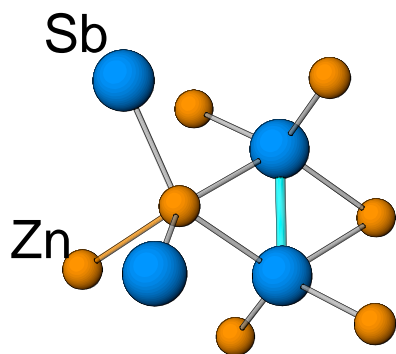


Sb Zintl Electron Counting

- Sb^{3-} discrete ions = Same as N^{3-}
 - CaZn_2Sb_2 closest Sb-Sb is 4.4 Å

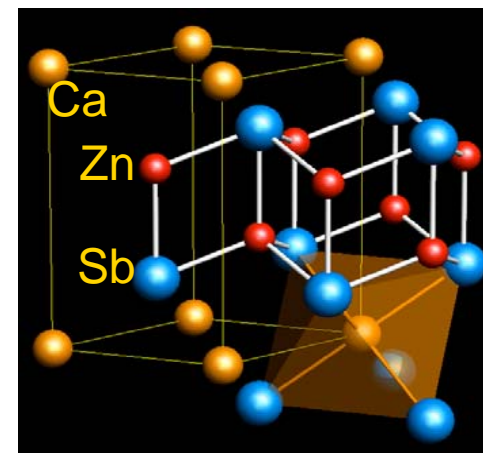
Each bond ($\sim 2.9\text{\AA}$) reduces valence by 1

- Sb^{2-} dimer = one bond
 - ZnSb Sb-Sb is 2.84Å
- Sb^{1-} chain or ring = two bonds
 - SrZnSb_2 Sb^{1-} chains
 - CoSb_3 Sb^{1-} square ring
- Sb^0 sheet = three bonds
 - Elemental Sb (α -Sb) Sb^0 sheet

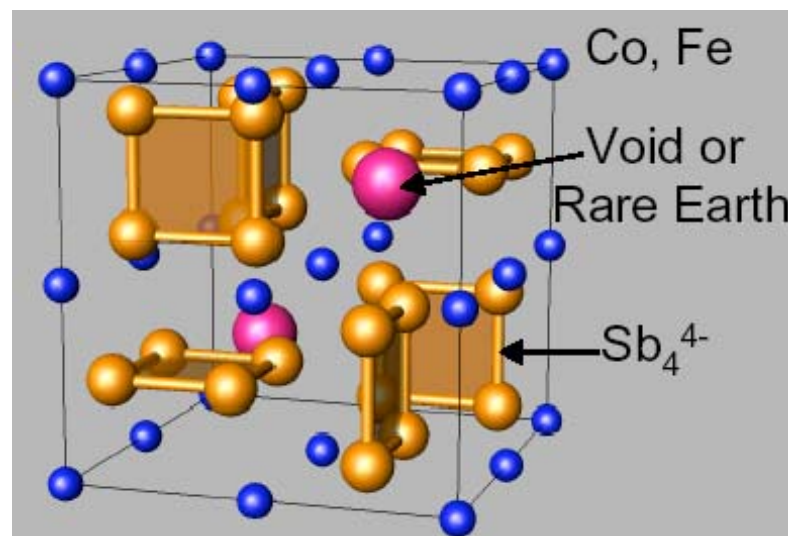


ZnSb: Sb^{2-} - Sb^{2-} dimers

Group 15
7 N Nitrogen 14.006 74
15 P Phosphorus 30.973 761
33 As Arsenic 74.921 60
51 Sb Antimony 121.760
83 Bi Bismuth 208.980 38



CaZn_2Sb_2 : Sb^{3-}
No Sb-Sb bonds



Skutterudite: Sb^{1-} two bonded rings



“Zintl Metals”

“Zintl Phases” = Semiconductors (G. Miller in *Chemistry, Structure, and Bonding of Zintl Phases and Ions* S. Kauzlarich ed.)

“Metallic Zintl Phases” (J. Corbett in *Chemistry, Structure, and Bonding of Zintl Phases and Ions* S. Kauzlarich ed.)

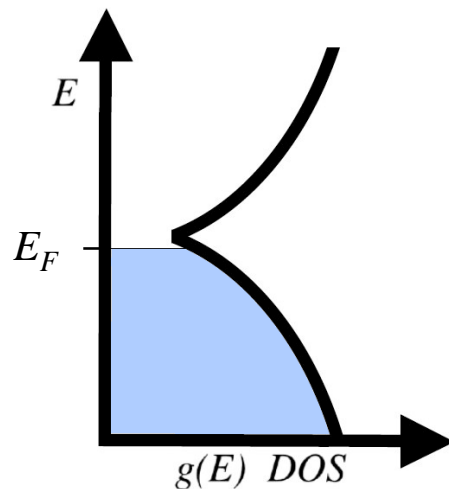
Semi-Metallic Zintl Phases

- Valence Precise according to Zintl electron counting
- but Band Gap $E_g < 0$

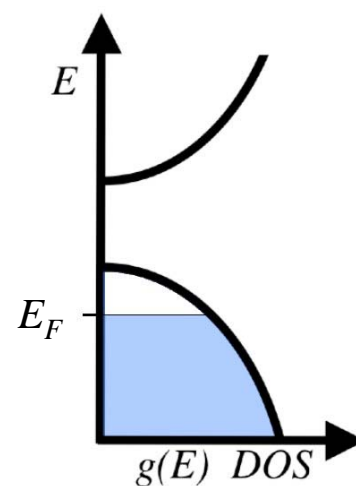
Valence Imbalanced Zintl-like Phases ($\sim 1 e^-/\text{Formula}$)

- Zintl valence count is slightly off
- giving electron-rich or electron-poor phases that are metallic (heavily doped semiconductor)

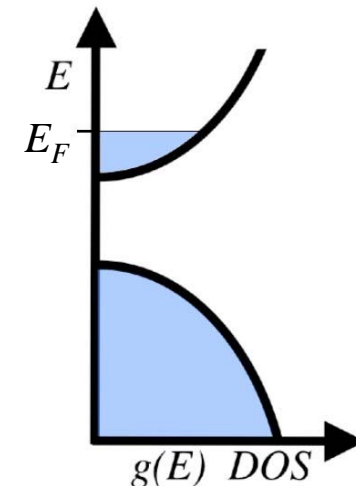
Valence Balanced Semi-Metal



Valence Imbalanced Metal with band gap (heavily doped semiconductor)



Electron Poor



Electron Rich



Valence Metals with Band Gap

Thermoelectric materials are (postulate ?)

Off Valence Balance compounds

Where **concentration of valence imbalance = free carrier concentration**

- free Carrier Concentration measured by Hall Effect n_H

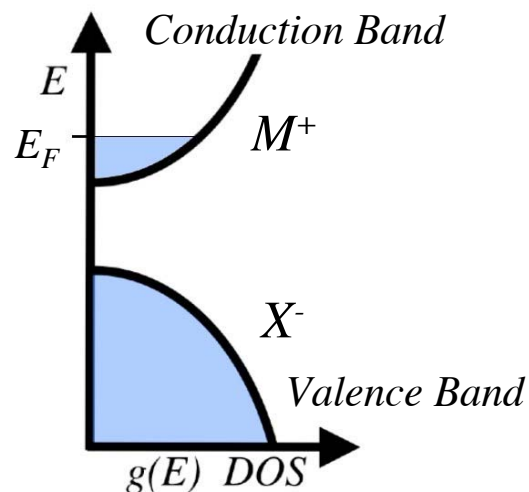
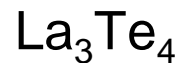
Transport properties are metallic

- Heavily doped, degenerate semiconductors

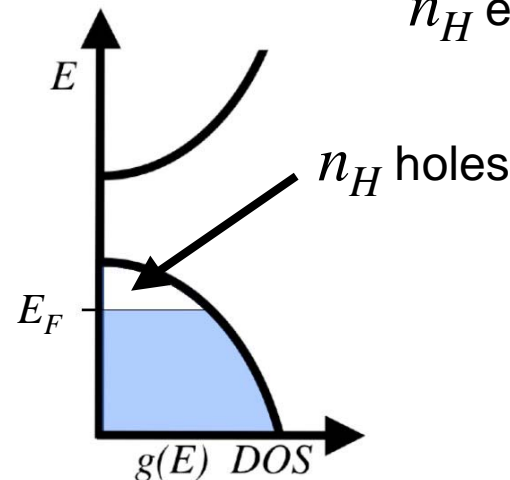
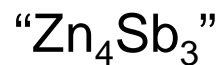
Zintl electron counting rules apply

- Poly anions, Metal-Metal bonding

Ionic MX

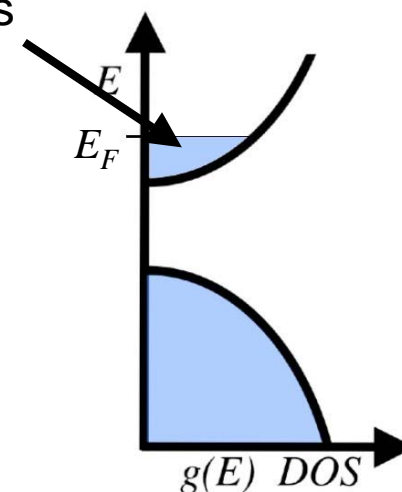


Polar MX



n_H electrons

Semicond. X



Hall Effect



Hall Effect

Magnetic Field deflects mobile charges

Hall Effect measurements give:

Sign of Charge Carrier

- n (electron) or p (hole) type

Carrier concentration

- $n_H = 1/R_H e$

Mobility

- $\mu_H = \sigma/n_H e$

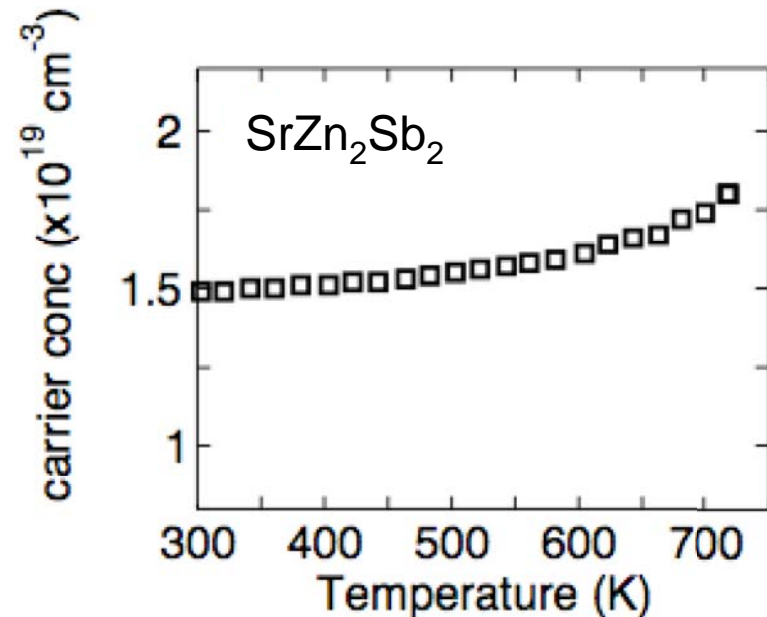
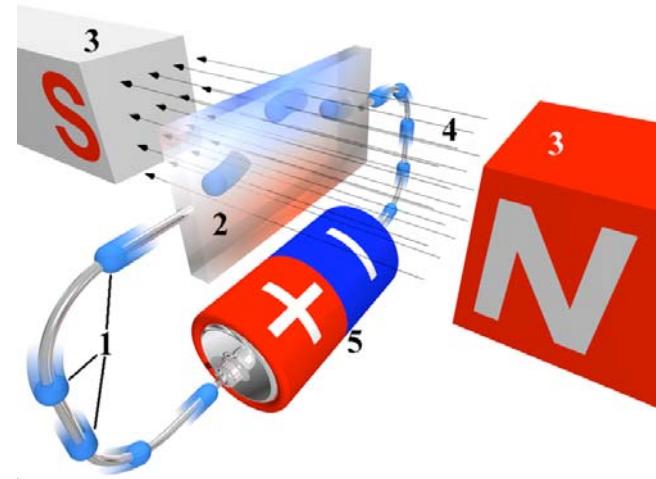
Hall Effect of Extrinsic Semicond.

Constant n_H at low temp

- $n_H = \text{dopant concentration}$

Rises at high temp

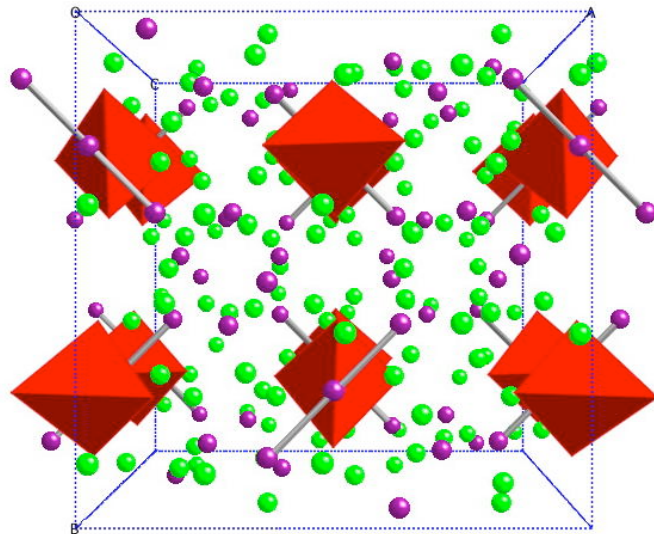
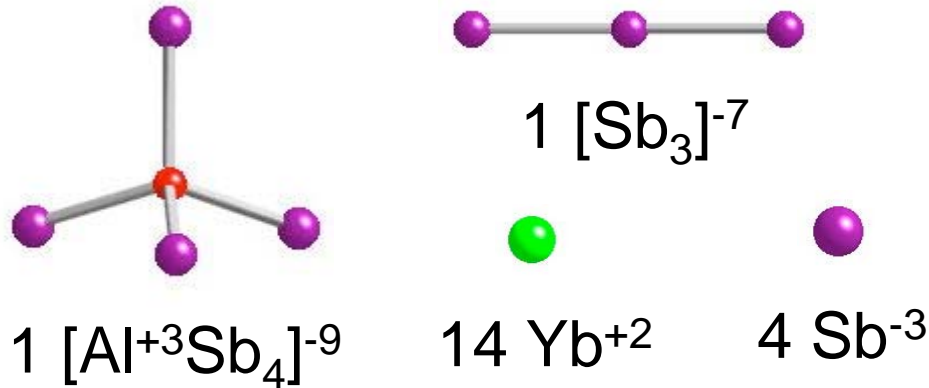
- minority carriers activated across Band Gap



Yb₁₄AlSb₁₁

Complex structure

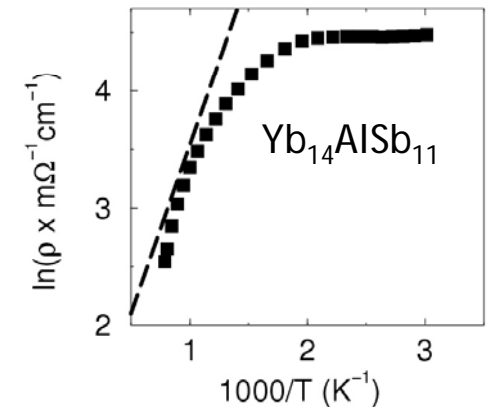
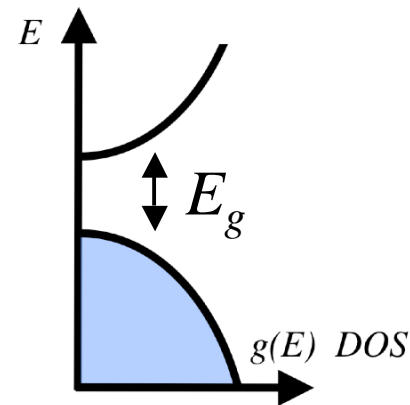
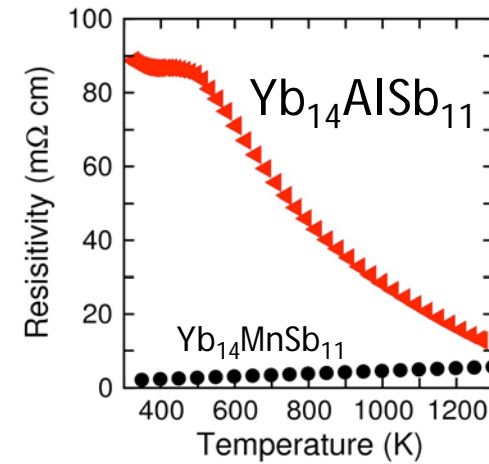
Contains variety of structural elements



Valence Balance

Semiconductor

Band Gap ~ 0.5 eV



Cation doping $\text{Yb}_{14}\text{MnSb}_{11}$



Mn^{2+} (for Al^{3+}) gives $1h^+$ per $\text{Yb}_{14}\text{MnSb}_{11}$
delocalized hole $n \sim 1.3 \times 10^{21}/\text{cc}$

Linear resistivity with T

Metal (Heavily doped semiconductor)

Linear p-type Seebeck with T

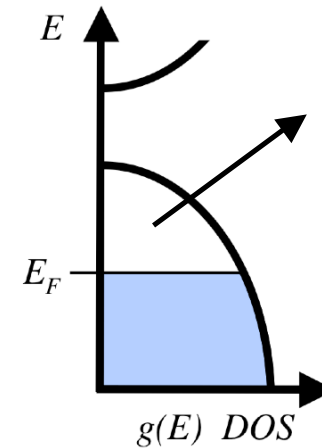
$m^* \sim 3 m_e$, $\mu \sim 3 \text{ cm}^2/\text{Vs}$, $E_g \sim 0.5 \text{ eV}$

Mn^{2+} (h.s. d^5) + spin paired hole (anti-ferro.)

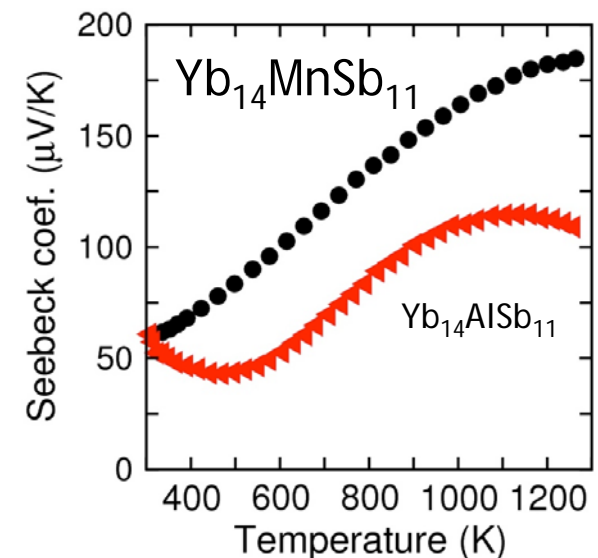
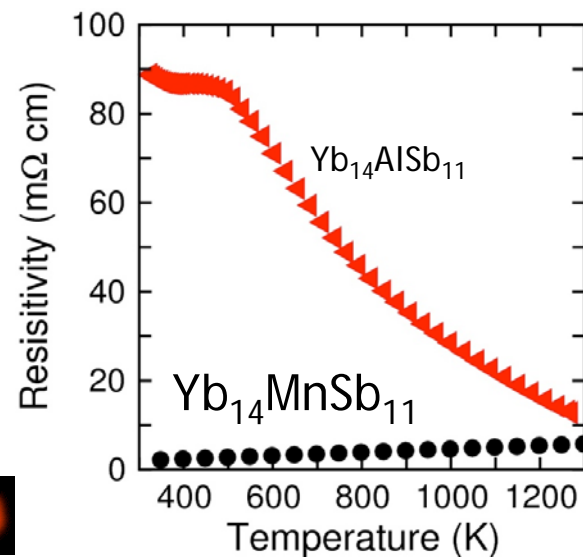
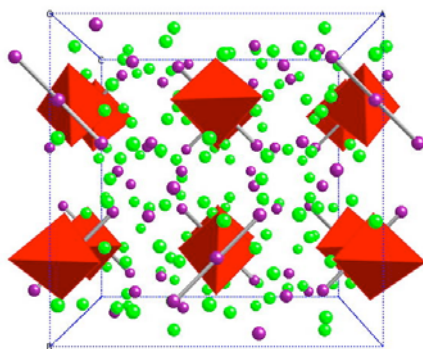
net spin = $5/2 - 1/2 = 2$

Under-screened Kondo lattice

Sales, et al. *PRB* **72** 205207 (2005)



Mn^{2+} gives
 $1h^+$ per
 $\text{Yb}_{14}\text{MnSb}_{11}$



Thermoelectric $\text{Yb}_{14}\text{MnSb}_{11}$



High Efficiency, zT

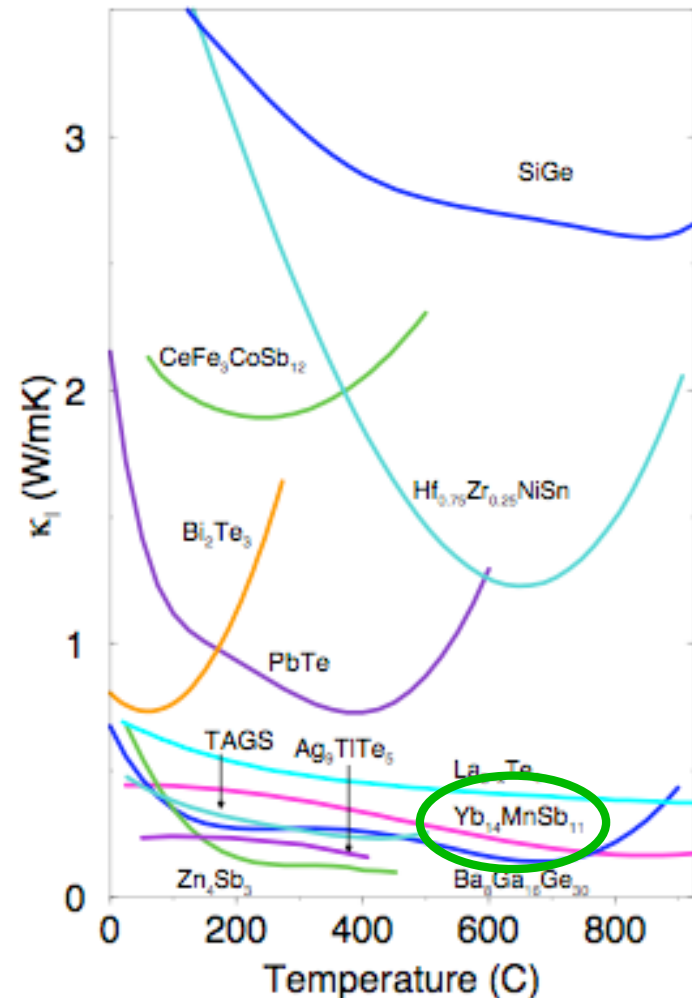
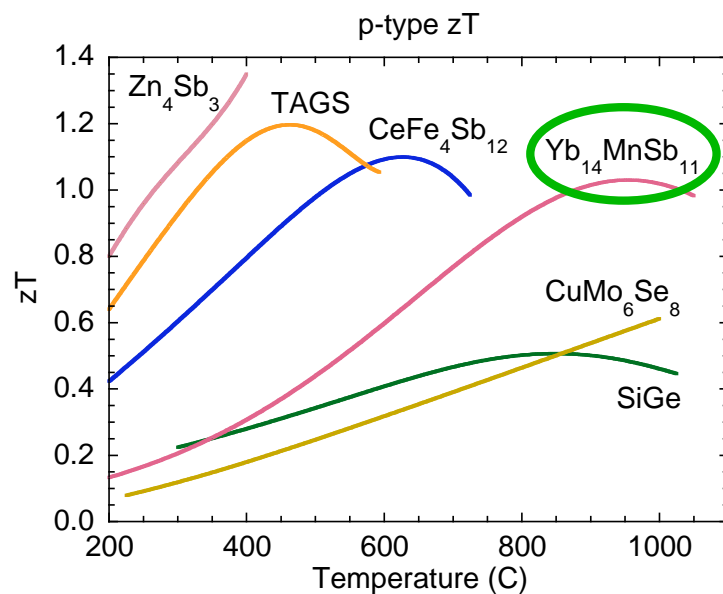
zT maximum > 1.0

Avg zT 0.95

- for $700 < T < 1000$ C
- 2 x better than SiGe
- Due to low lattice thermal conductivity
 - Complex structure ?

Currently being developed by NASA

- Next generation RTG



Snyder, *Nature Materials* **7**, 105 (2008)

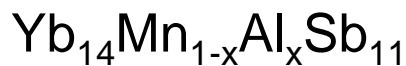


Brown, Kauzlarich, Gascoin, Snyder, *Chem. Materials* **18**, 873 (2006)

Carrier Concentration Tuning

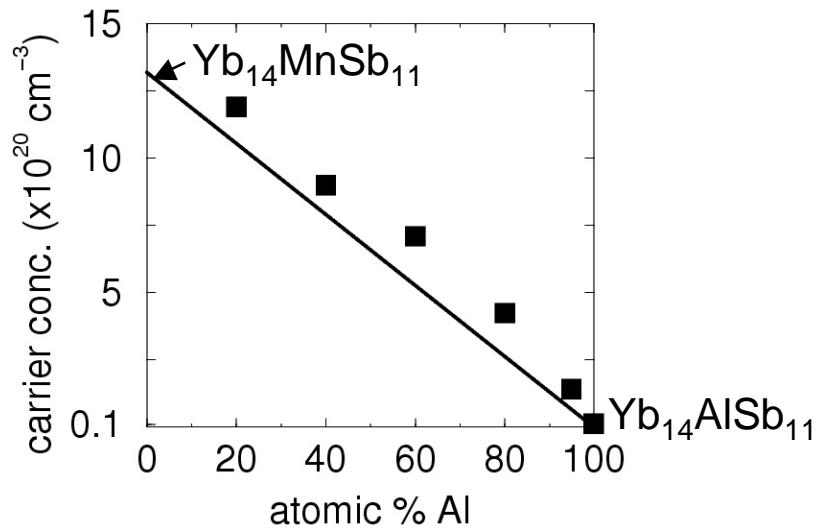


zT maximizes at optimum free carrier concentration

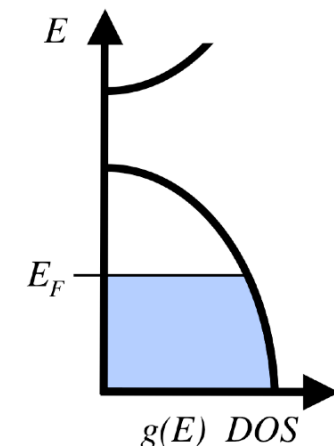
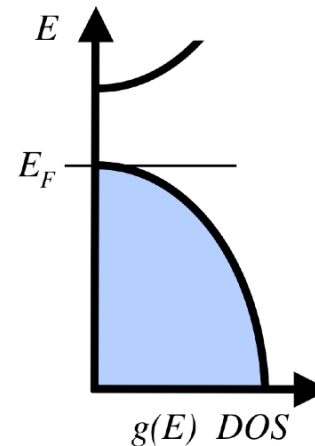
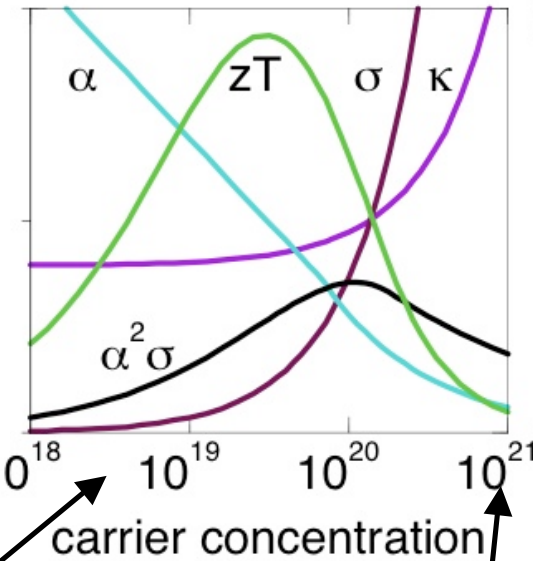


Al^{3+} for $\text{Mn}^{2+} + h^+$

- reduces hole concentration



From Room Temperature Hall Effect

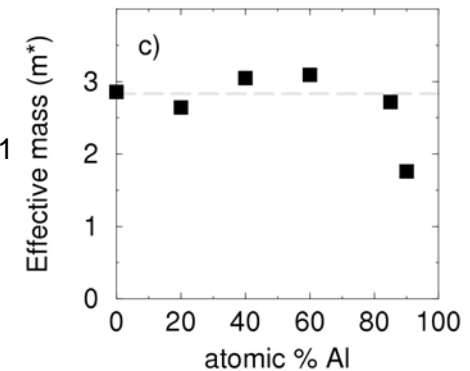
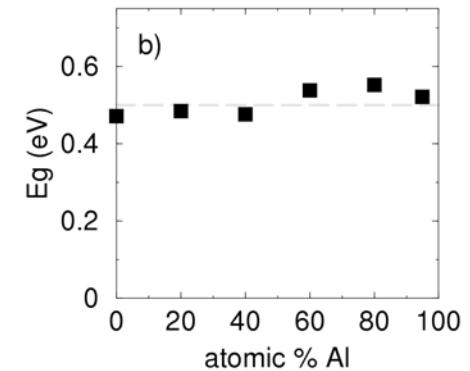
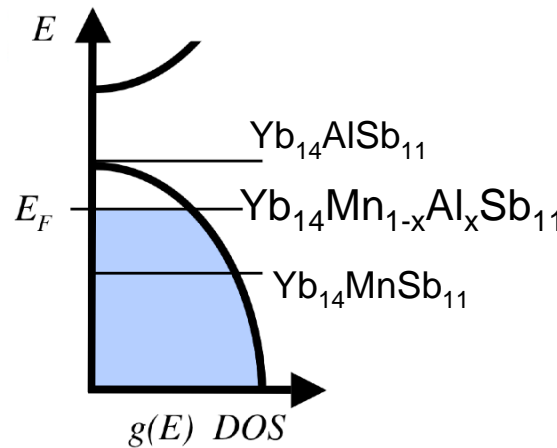
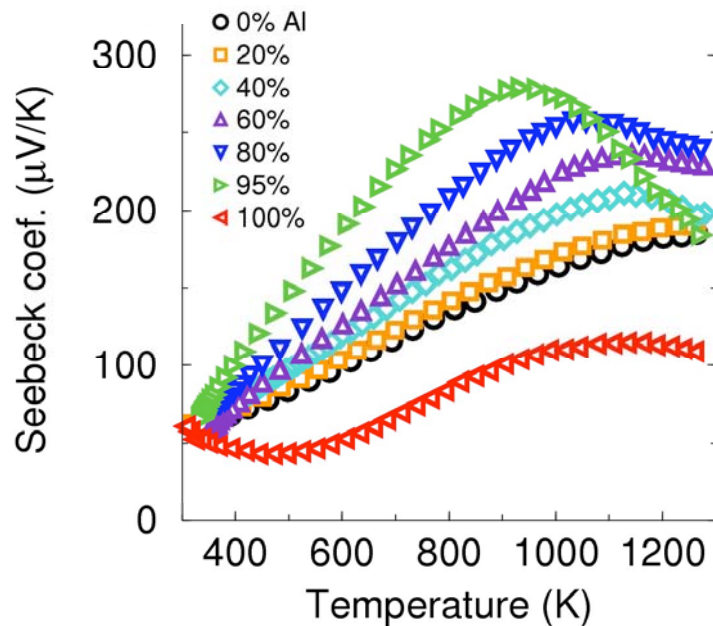


Rigid Bands in $\text{Yb}_{14}\text{Mn}_{1-x}\text{Al}_x\text{Sb}_{11}$

Properties Follow simple rigid, parabolic-band model as Fermi Level changes

- Constant effective band gap
 - From peak in Thermopower
Sharp, et al., *J. Elec. Mater.* **28** 869 (1999)
– additional h^+ and e^- reduce thermopower
- Constant effective mass
 - From $d\alpha/dT$ in linear region

$$E_g \sim 2e\alpha_{max} T_{max}$$



$$\alpha = \frac{8\pi^2 k_B^2}{3eh^2} m^* T \left(\frac{\pi}{3n} \right)^{2/3}$$

Complex Crystal Structures

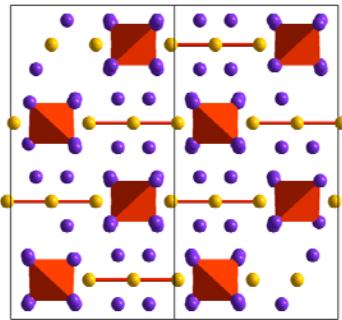


Low Lattice Thermal Conductivity

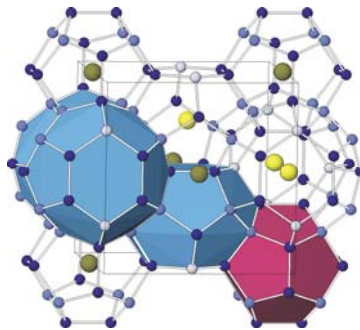
From complex structure

- $\text{Yb}_{14}\text{MnSb}_{11}$
- Clathrate $\text{Ba}_8\text{Ga}_{16}\text{Ge}_{30}$
- Zn_4Sb_3

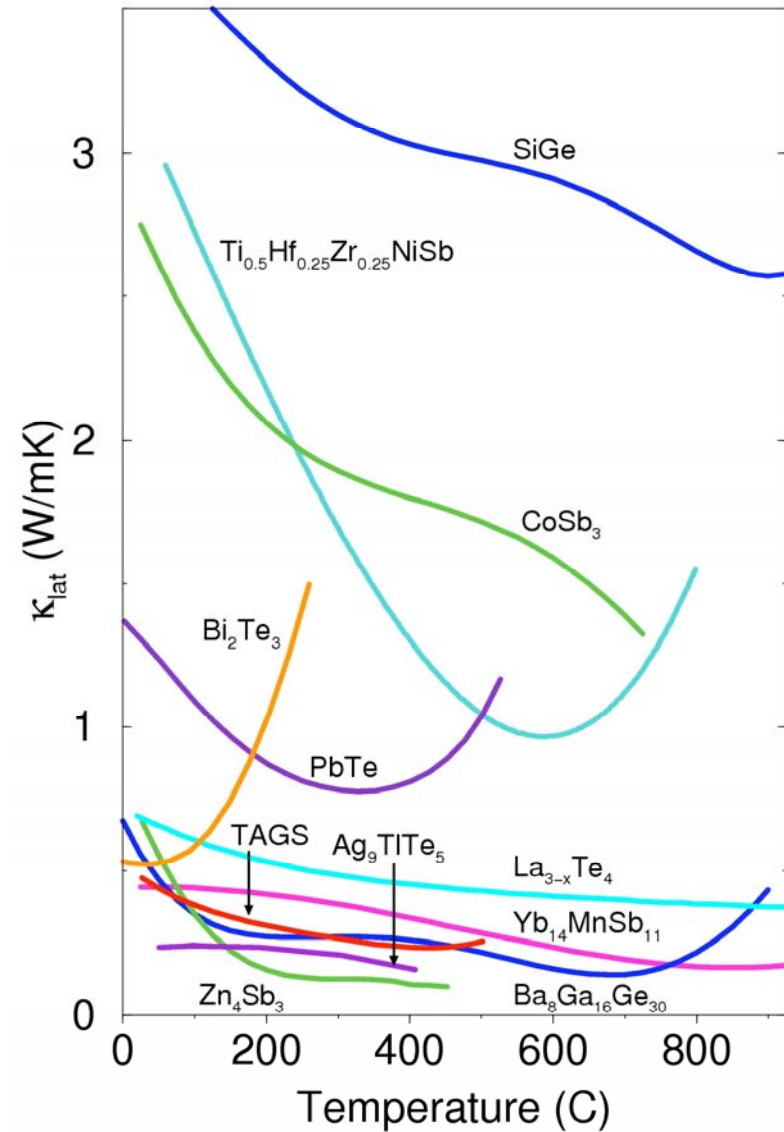
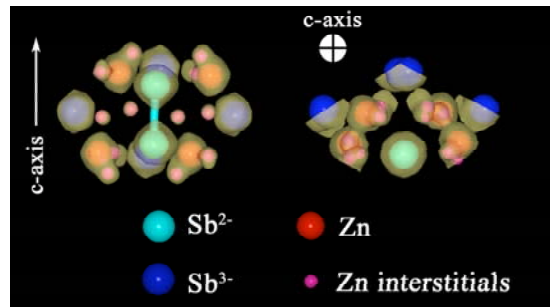
$\text{Yb}_{14}\text{MnSb}_{11}$



$\text{Ba}_8\text{Ga}_{16}\text{Ge}_{30}$



Zn_4Sb_3





Large Cells with low thermal cond.

Heat primarily carried by acoustic modes

$$\kappa_{lattice} = \frac{1}{3} C v l$$

C - heat capacity

v - speed of sound

l - phonon mean free path

N - atoms per cell

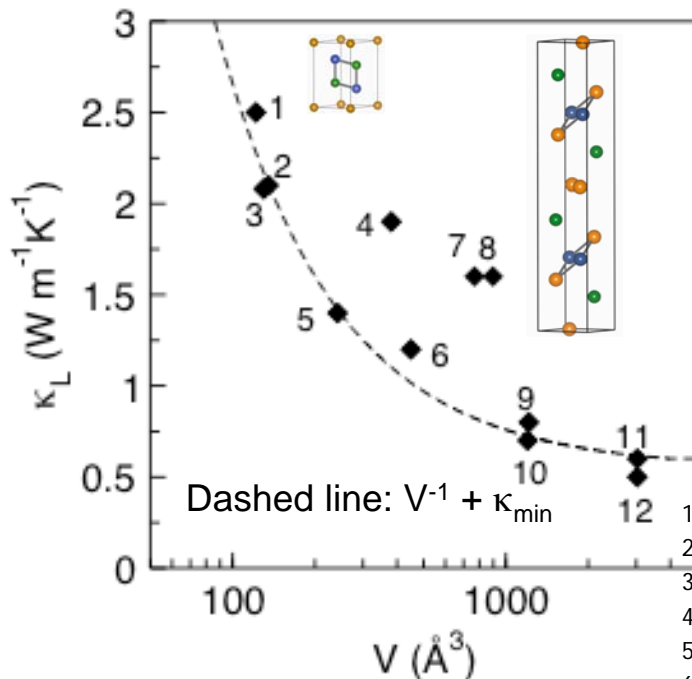
V - volume of cell

$$\text{acoustic } C = \frac{3k_B}{cell} = \frac{3k_B}{V} \quad \text{optic } C = \frac{3(N-1)k_B}{cell} = \frac{3(N-1)k_B}{V}$$

Low lattice thermal conductivity

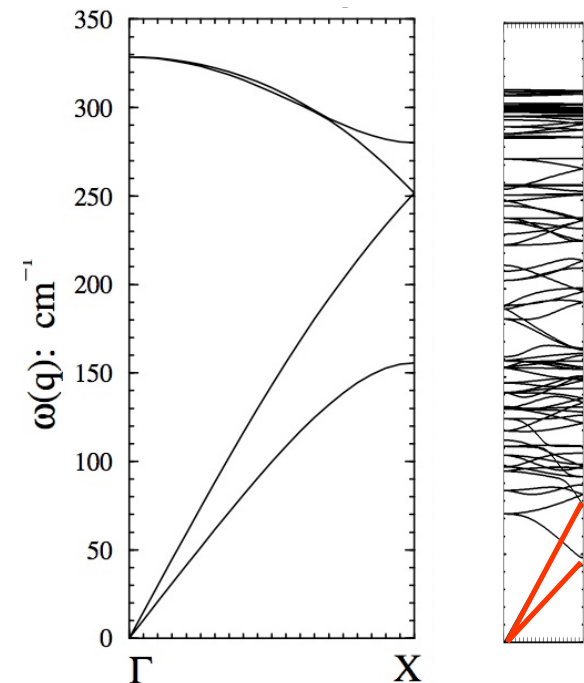
for large primitive unit cell volume (V)

$$\kappa_l \approx \frac{k_B v l}{V}$$



- 1 - LiZnSb
- 2 - SrZn₂Sb₂
- 3 - Mg₃Sb₂
- 4 - CeFe₄Sb₁₂
- 5 - BaZn₂Sb₂
- 6 - SrZnSb₂
- 7 - Yb₅In₂Sb₆
- 8 - Ba₄In₈Sb₁₆
- 9 - Yb₁₁Sb₁₀
- 10 - Yb₁₁GaSb₉
- 11 - Yb₁₄AlSb₁₁
- 12 - Yb₁₄MnSb₁₁

Diamond Ge₂ Empty clathrate Ge₄₆



Dong et al, Phys. Rev. Lett. 2001



Toberer, May, Snyder *Chem. Mat.*, (DOI: 10.1021/cm901956r)

Clathrates

Large Unit Cell

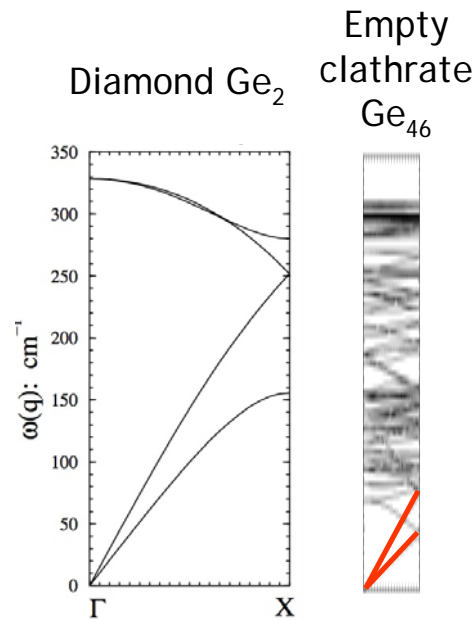
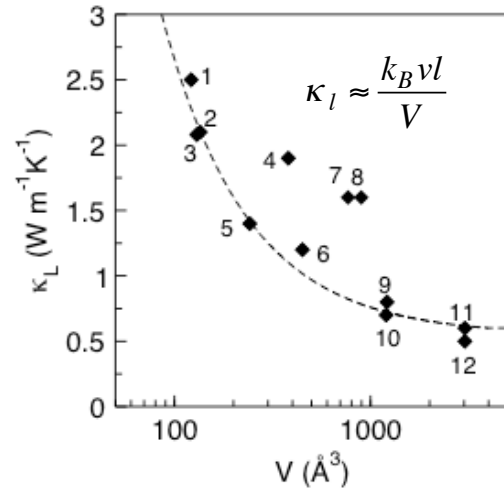
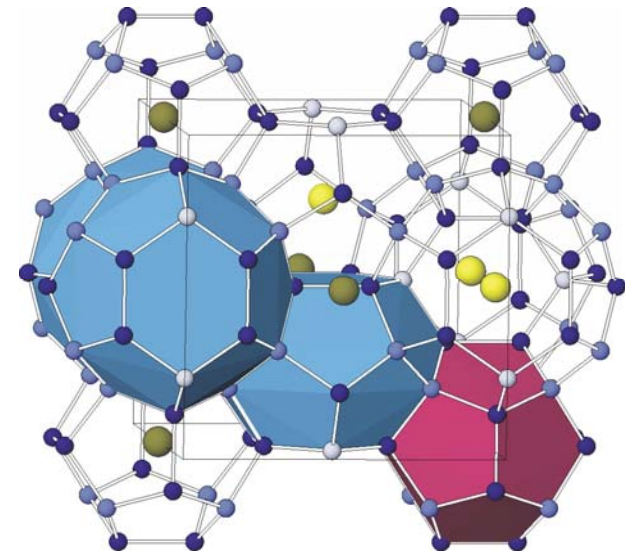
expect low κ_l

Covalent bonding

high m^* and μ like elemental Si or Ge?

can we modify n ?

$\text{Ba}_8\text{Ga}_{16}\text{Ge}_{30}$
Type I Clathrate



Dong et al, Phys. Rev. Lett. 2001

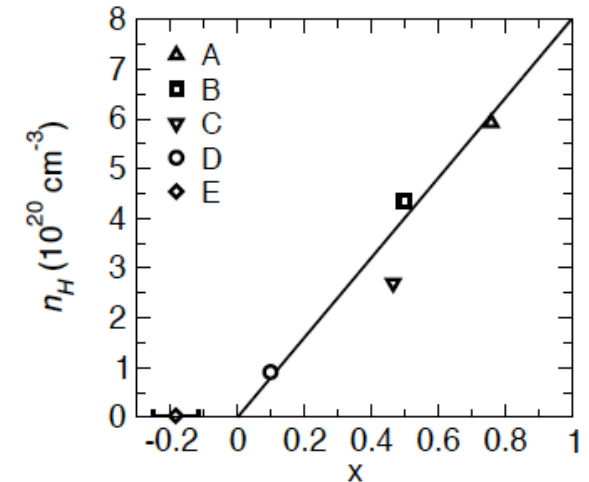
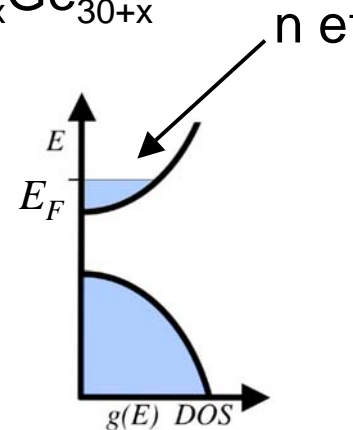
Thermoelectric $\text{Ba}_8\text{Ga}_{16-x}\text{Ge}_{30+x}$



Zintl Electron Counting for $\text{Ba}_8\text{Ga}_{16-x}\text{Ge}_{30+x}$
no vacancies for $x < 3$

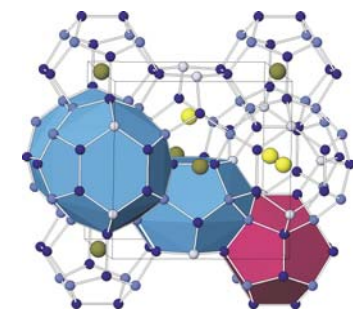
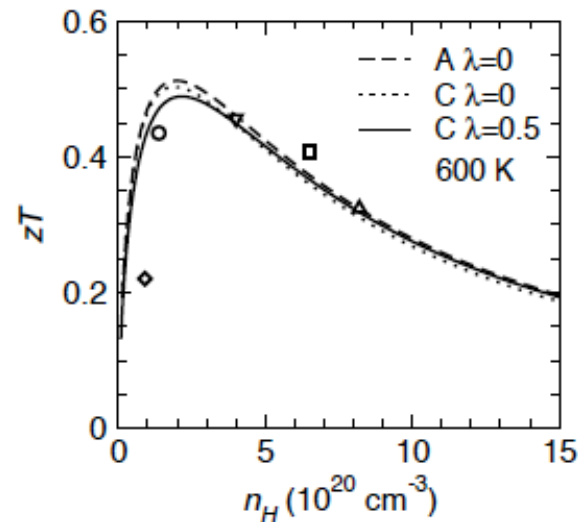
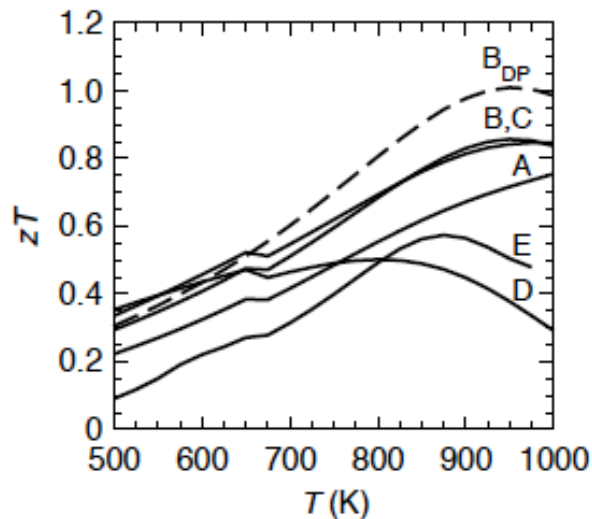
$$n = 2 \times [\text{Ba}^{2+}] - 1 \times [4b\text{-Ga}^{-1}]$$

$$n/\text{FU} = 16 - (16 - x) = x$$



Observed n_{Hall} = valence imbalance
 zT optimized by tuning n

Transport fits Single Parabolic Band model

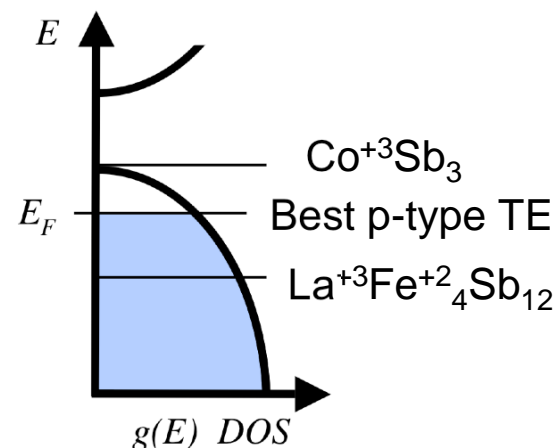
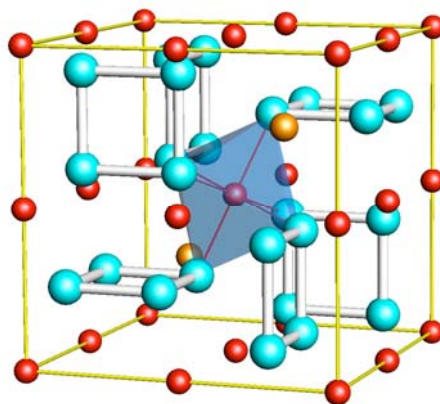


Other Thermoelectric Zintl Metals



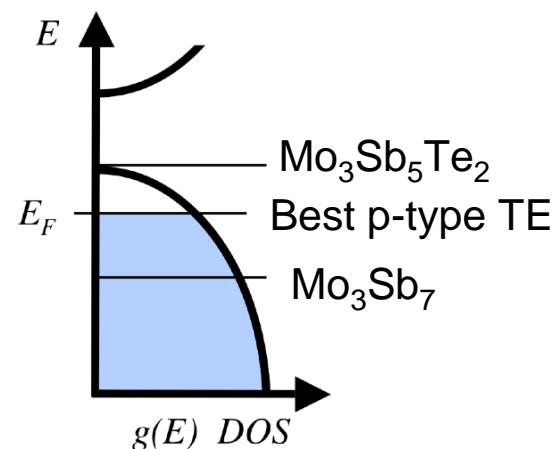
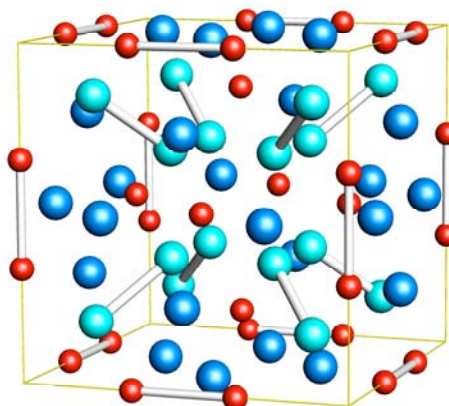
Filled Skutterudites

Sb square rings $2b\text{-Sb}^{-1}$
 $\text{Co}^{+3}\text{Sb}_3$ valence semicond.
 $\text{La}^{+3}\text{Fe}_4^{+2}\text{Sb}_{12}$ Zintl Metal
 1 hole/fu



Mo_3Sb_7

4x Sb dimers $1b\text{-Sb}^{-2}$
 3x Sb isolated $0b\text{-Sb}^{-3}$
 Mo-Mo dimer $1b\text{-Mo}^{+5}$
 2 holes/FU = metal
 • $\text{Mo}_3\text{Sb}_5\text{Te}_2$ for SC

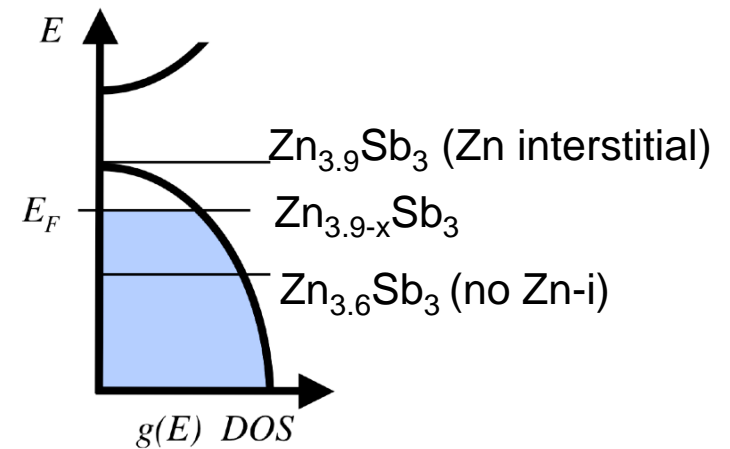
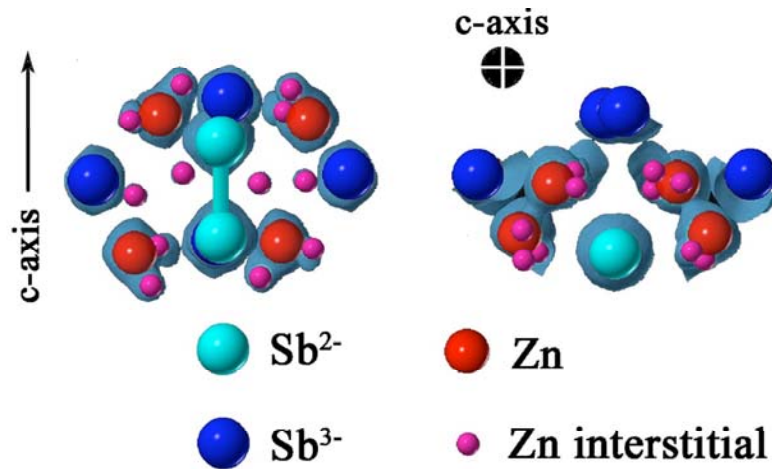


Gascoin, et al *J. Alloys. Compounds* **427**, 324 (2007)
 Kauzlarich, Snyder et al, *Dalton Trans.* p. 2099 (2007)

Zn₄Sb₃



Extra Zn found in interstitial sites

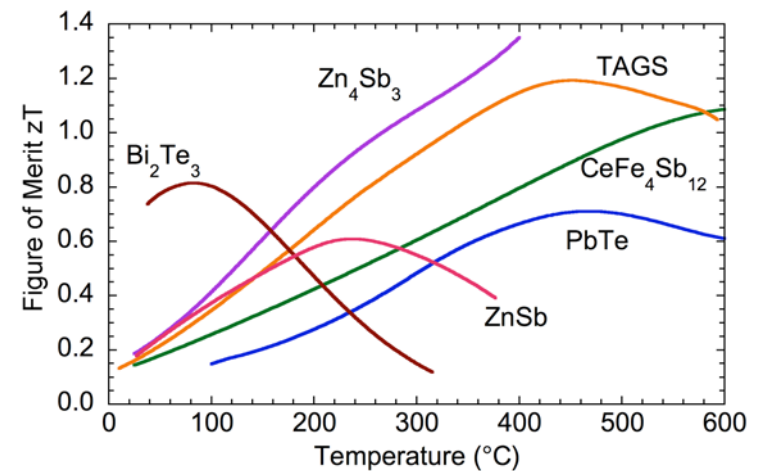


Zn-i adds 2e⁻ (Zn²⁺)

Moves E_f up in valence band

- rigid band model
- Zn_{3.9}Sb₃ should be Semiconducting

Does not add new states

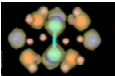
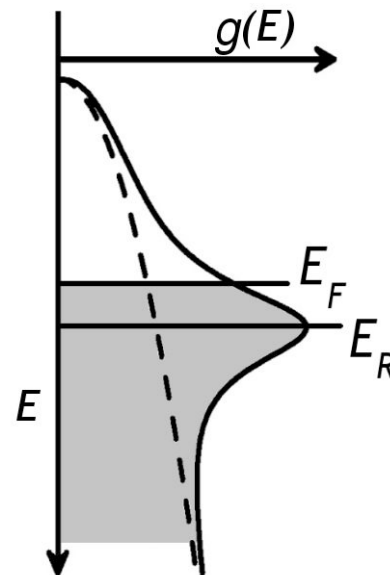


Snyder et al, *Nature Materials* **3**, 458 (2004)

See Häusserman et al, *Chem. Eur. J.*, **11**, 4912 (2005)



High Efficiency from Band Structure Engineering





Enhancing Thermopower

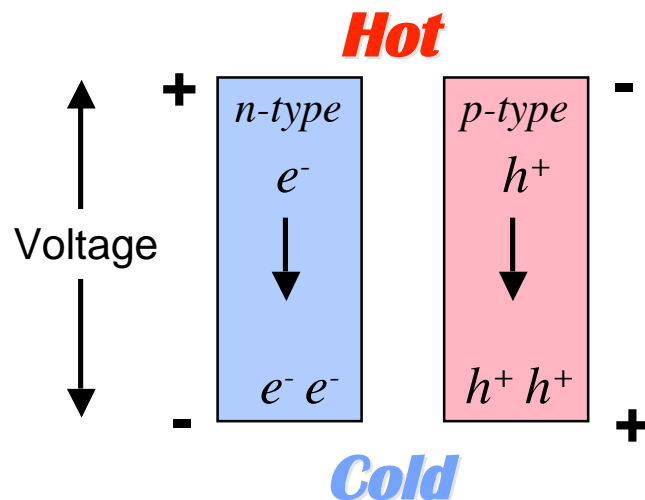
Mott Equation

Thermopower of metals depends on Energy dependent Conductivity

- $\sigma(E)$ is conductivity σ when $E = E_F$

$$\alpha = \frac{\pi^2}{3} \frac{k_B}{q} k_B T \left\{ \frac{d[\ln(\sigma(E))]}{dE} \right\}_{E=E_F}$$

- if $\sigma(E_{hot}) \neq \sigma(E_{cold})$ then charge will build up at the cold end



$$\alpha = \frac{\pi^2 k_b^2 T}{3q} \left(\frac{d \ln N(E)}{dE} + \frac{d \ln \tau(E) v(E)^2}{dE} \right)_{E=E_F}$$

Large Thermopower from rapidly changing density of states

$$\frac{dN(E)}{dE}$$

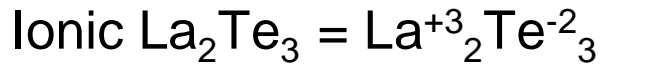
or Scattering

- relaxation time τ
- group velocity v
- effective mass m^*
- mobility μ
- mean free path $\lambda \propto \tau$
- carrier concentration n

$$\mu = \frac{e\tau}{m^*}$$

$$\sigma = ne\mu$$

La₂Te₃



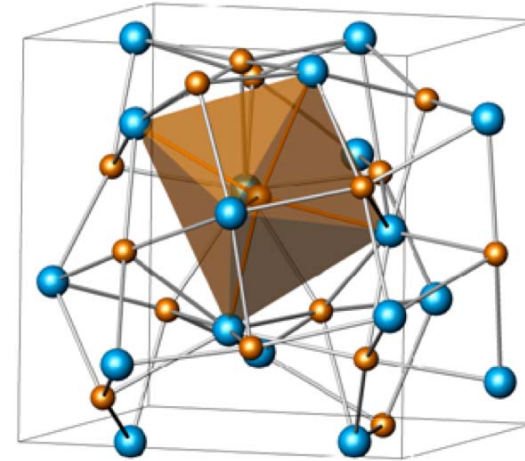
valence balanced insulator

$$E_g \sim 1.0 \text{ eV}$$

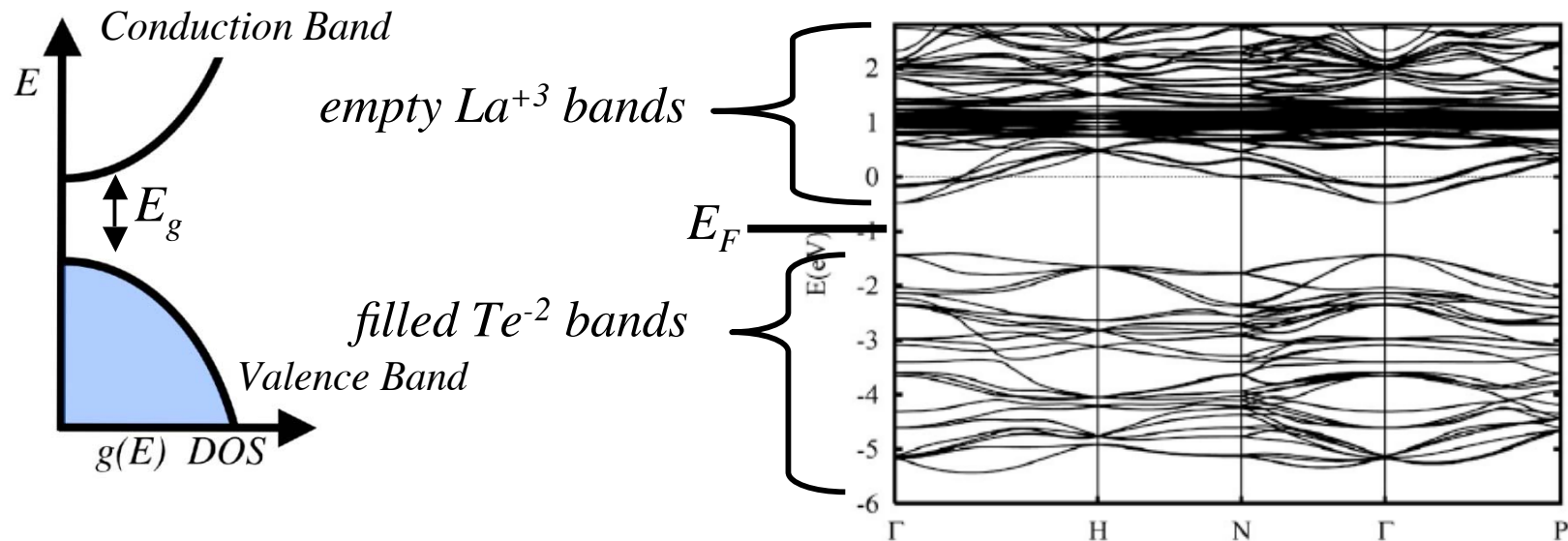
Defect Th₃P₄ structure type

- La_{3-x}Te₄
- La vacancies: x = 1/3 for La₂Te₃

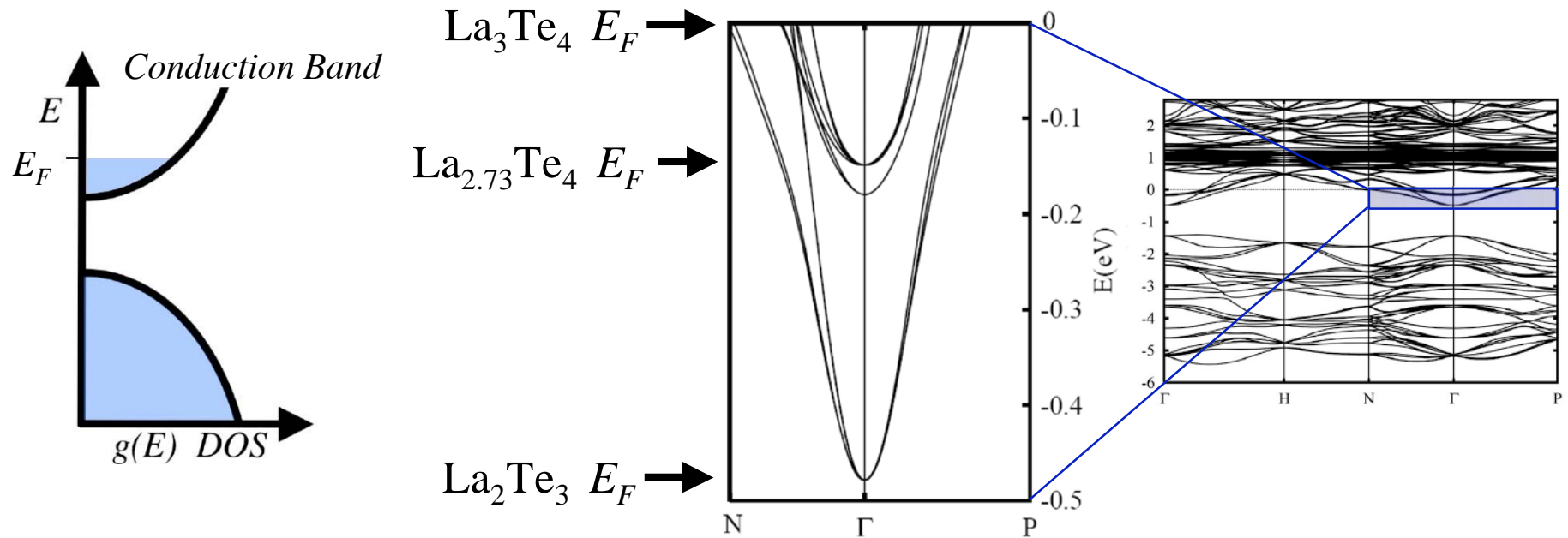
Many RE₂X₃ chalcogenides have related structure



La₃Te₄ distorted octahedron around Te



La_{3-x}Te₄



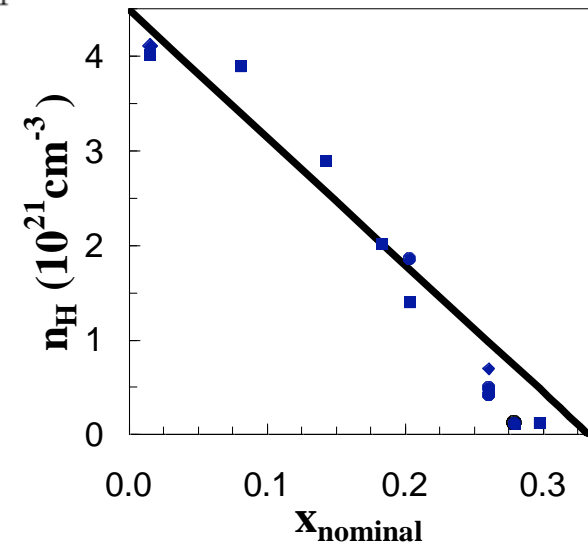
Valence-Imbalanced Metal La_{3-x}Te₄ (Ionic)
 electron concentration = valence imbalance

$$n = n_H = 3 [\text{La}^{3+}] - 2 [\text{Te}^{2-}]$$

removing La removes e⁻ (E_F drops)

not available states

states determined by unit cell



Ab initio vs Semi-Emperical

Ab Initio calculations

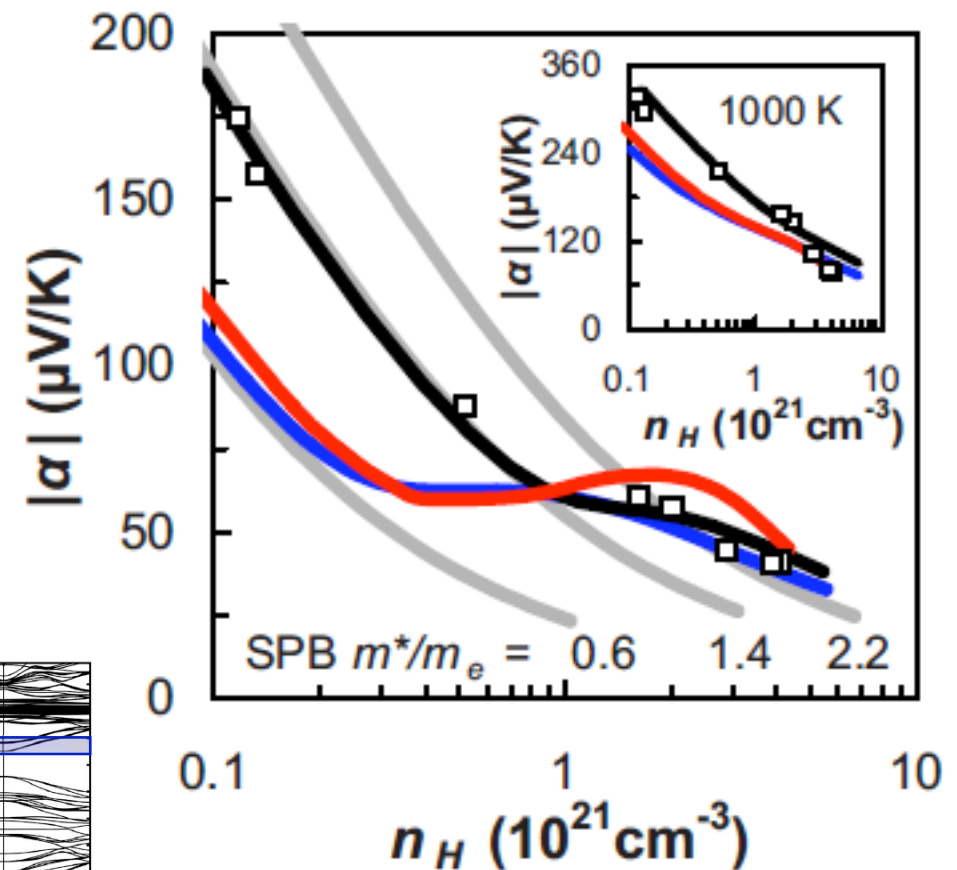
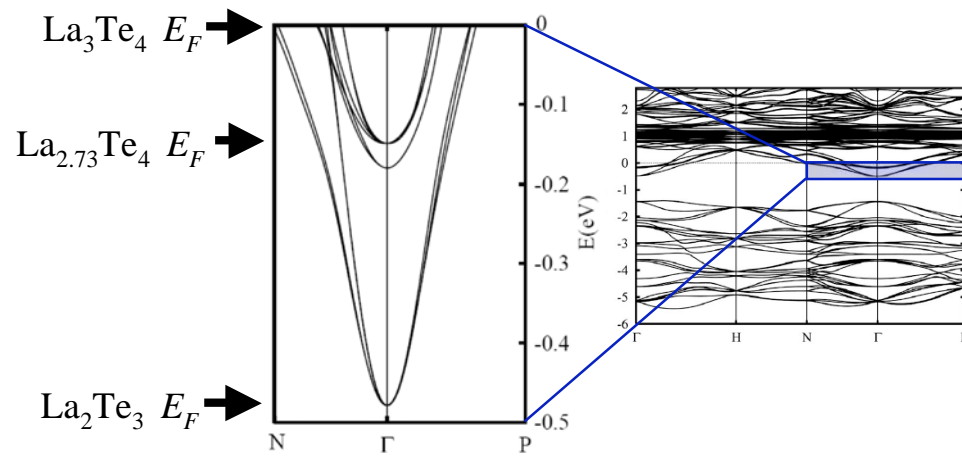
Explain qualitative trends

- D. Singh, Oak Ridge NL

Semi-emperical

assume parabolic bands

provides better fit



Thermoelectric $\text{La}_{3-x}\text{Te}_4$

$\text{La}_{3-x}\text{Te}_4$ has high zT at high T
Higher than SiGe used by NASA

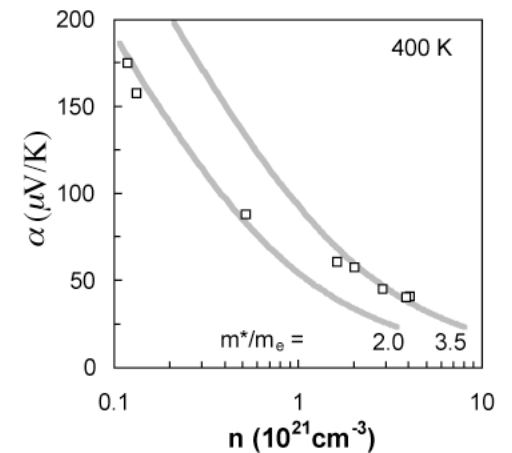
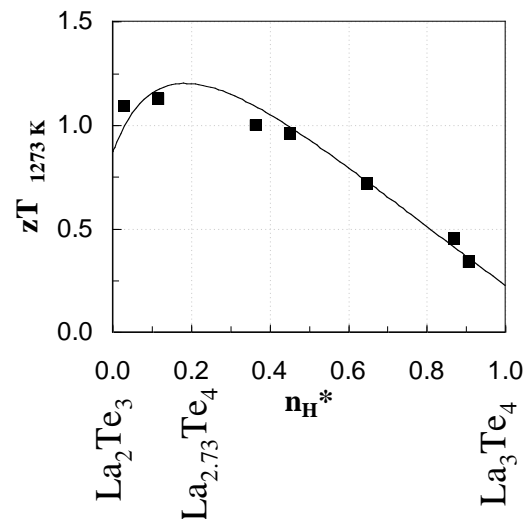
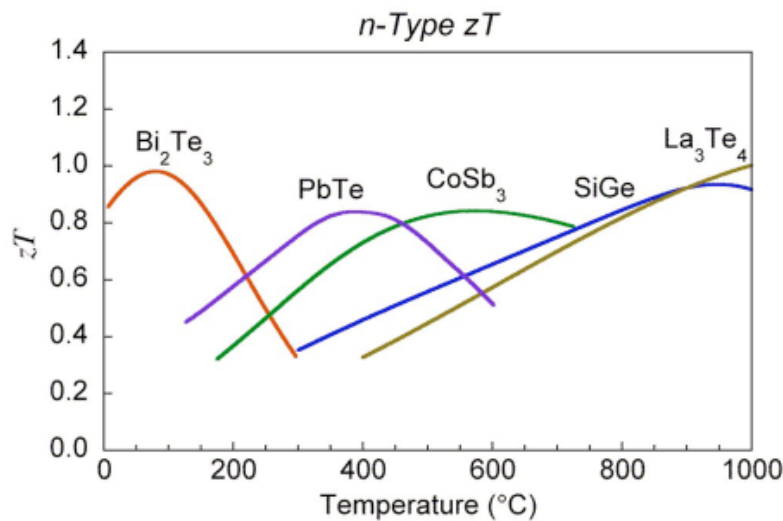
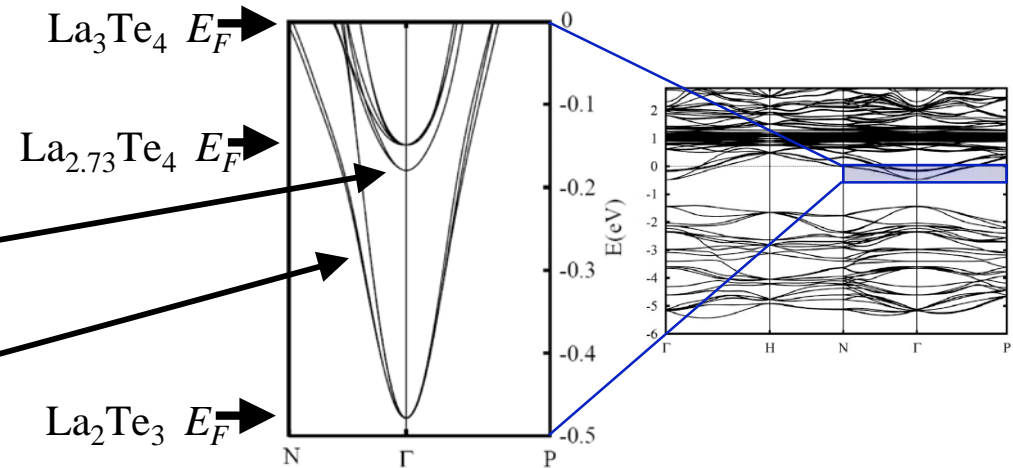
Highest zT for $\text{La}_{2.73}\text{Te}_4$

E_F at edge of heavy band

- $m^* \sim 3.5 m_e$

E_F deep inside light band

- $m^* \sim 2 m_e$

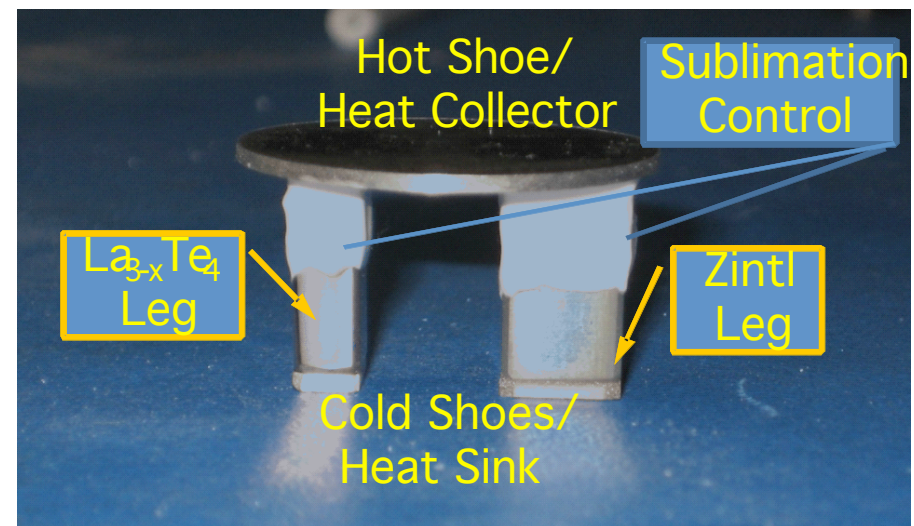
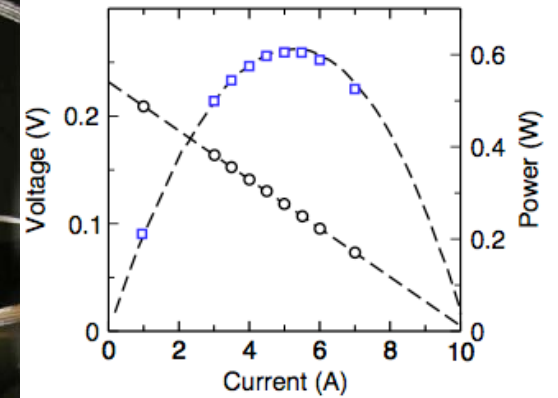
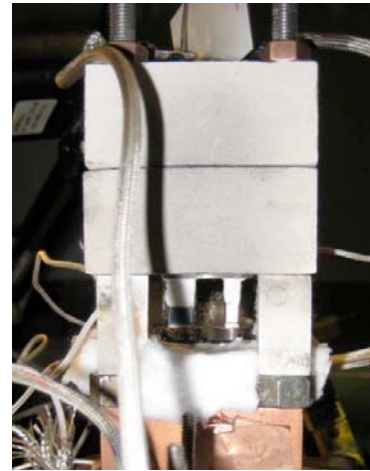
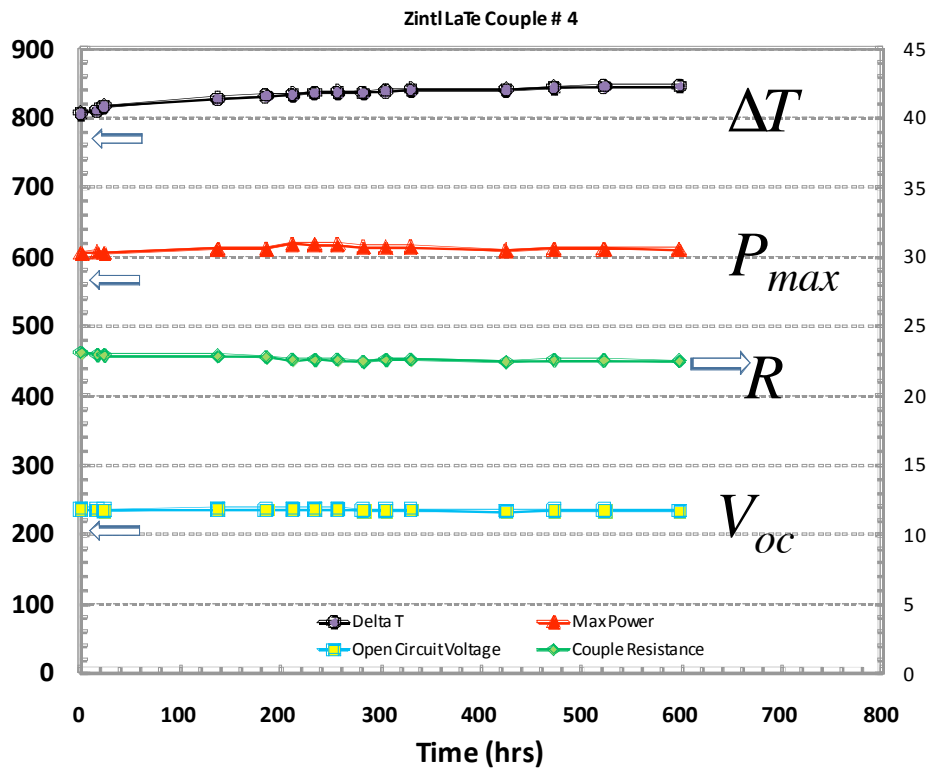


May et. al. *Phys. Rev. B*, **78**, p. 125205, (2008)
May et. al. *Phys. Rev. B*, **79**, p. 153101, (2009)



$\text{Yb}_{14}\text{MnSb}_{11}$ - $\text{La}_{3-x}\text{Te}_4$ Couple Testing

10% efficiency with 750°C ΔT (200°C - 950°C)
2 month lifetime test successful, 1 year underway



Jean-Pierre Fleurial (JPL)



Resonant States



Goal:

Increase Density of States (DOS)

Without ruining mobility

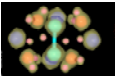
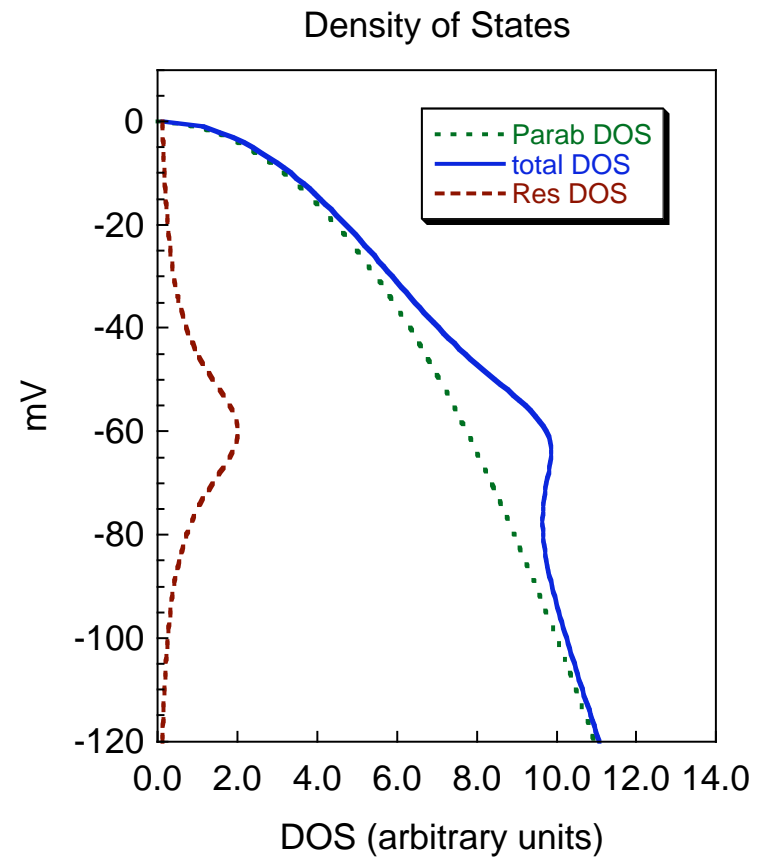
Isolated impurity state

adds delta function in DOS

but hopping conduction

Resonant Impurity State

Adds DOS to existing bands



TI Resonant States in PbTe



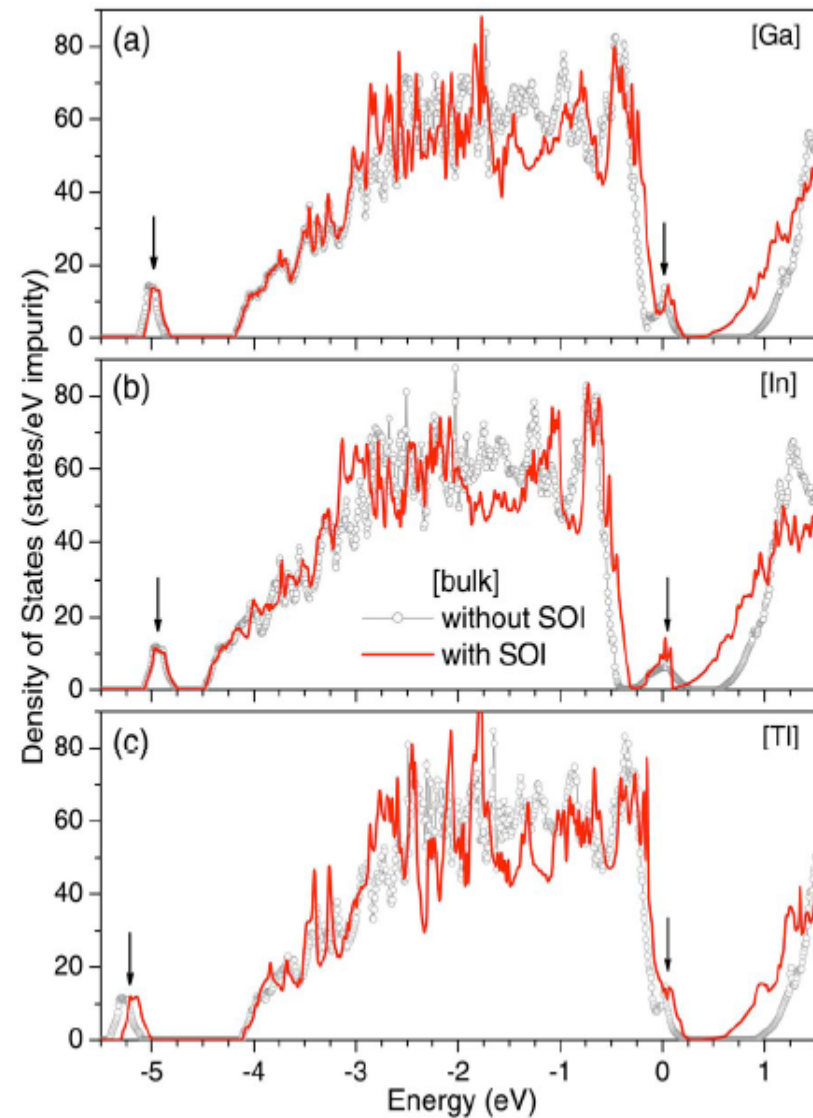
TI impurity in PbTe

add DOS to top of valence band

- Ga and In states are in the gap
 - Huang, Mahanti and Jena, *PRB* **76** 115432 (2007)

Increased DOS of holes observed

- larger electronic heat capacity
- superconducting T_c
 - Y. Matsushita, *et al.*, *Phys. Rev. B* **74** 134512 (2006)



PbTe:TI Pisarenko Plot



Thermopower depends on carrier concentration

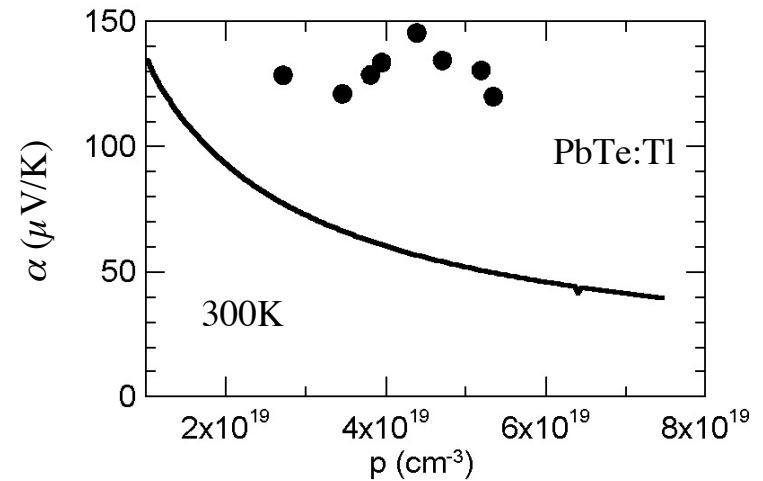
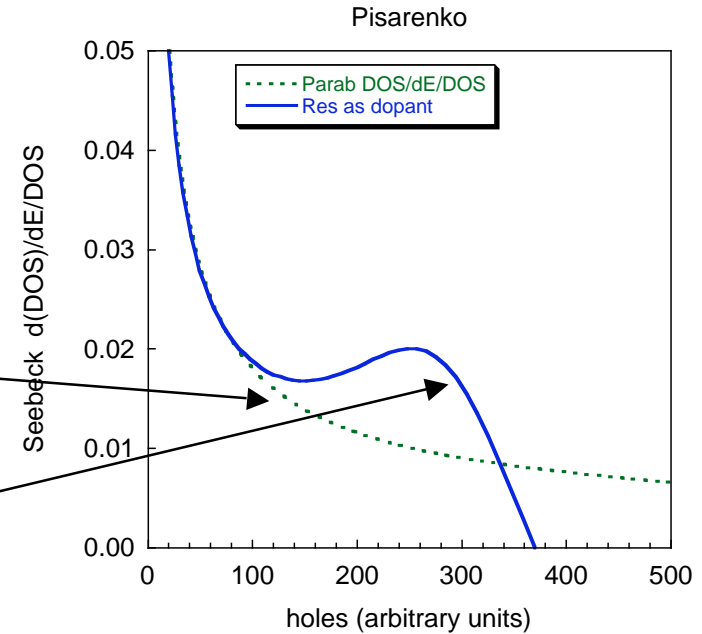
parabolic band metal $\alpha \propto n^{-2/3}$

$$\alpha = \frac{8\pi^2 k_B^2}{3eh^2} m^* T \left(\frac{\pi}{3n} \right)^{2/3}$$

Resonant state should alter thermopower from increased DOS

PbTe:TI shows higher thermopower than other p-type dopants

J. P. Heremans, et al.. *Science* **321**, p 554 (2008).



Improved zT from impurity states



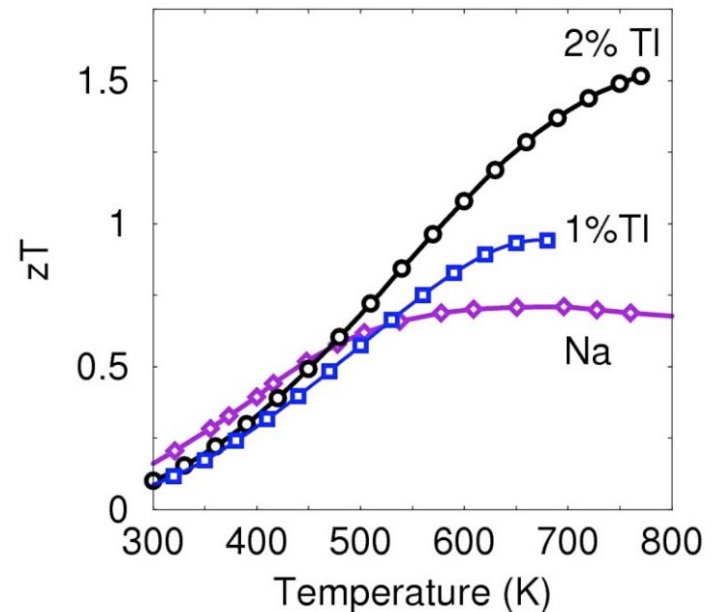
PbTe:TI shows higher thermopower than p-type Na-doped at same carrier concentration

Results in a higher zT about twice that of PbTe:Na

Mobility and effective mass are also different but net effect is positive

Thermal conductivity is about the same
Would benefit from reduction in lattice thermal conductivity using other methods

J. P. Heremans, et al.. *Science* **321**, p 554 (2008).



Theory Challenges = Opportunities



Band Structure Calculations

Need Accuracy near E_F

Transport is derivative property

- Band Gap (within $\sim 0.05\text{eV}$)
- Band Edge Energy (within $\sim 0.01\text{eV}$)
 - location of resonant states, other bands
- Effective Mass (within $\sim 0.1m_e$)

Imperfect crystals = Disorder

- Doping at $\sim 1\%$
- Alloying $\sim 10\%$
- Vacancies
- Anti-Site Defects
- Interstitial atoms

Transport Calculations

Relaxation Time Approximation

- $\tau \sim E^\lambda$

Constant RTA (?)

Thermal Transport

- diamond structure is 2009

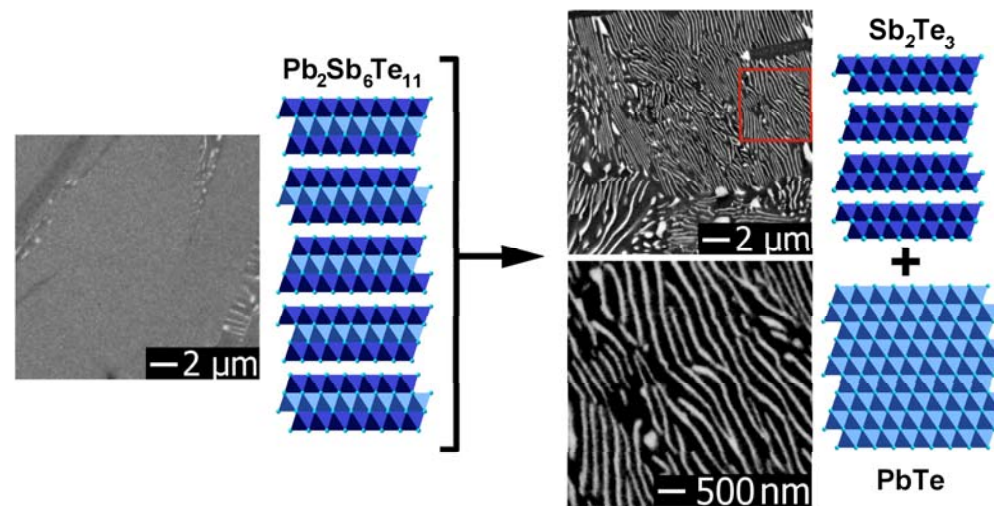
Exotic TE Materials

- Resonant States
- e- filtering
- Heavy Fermion
- Kondo
- Correlated Electron Systems

Nano-structures

- nanowires
- superlattices
- composites
- interfaces

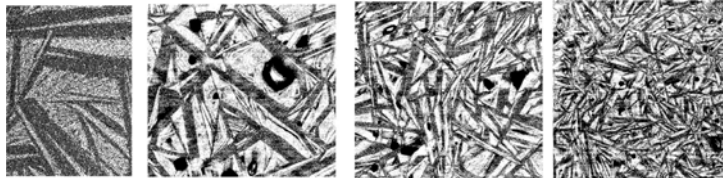
Complex Thermoelectrics from Nanosized Microstructures



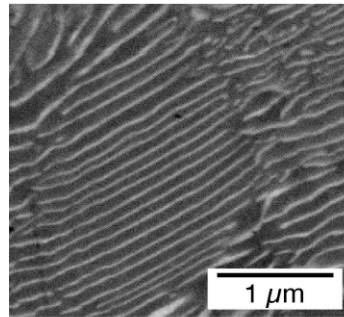
PbTe - Sb₂Te₃ Composites



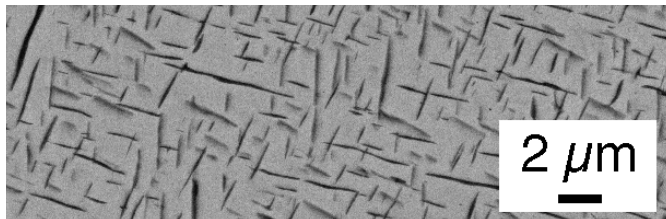
- 1) Crystallization from liquid
fast diffusion in liquid
control by cooling rate



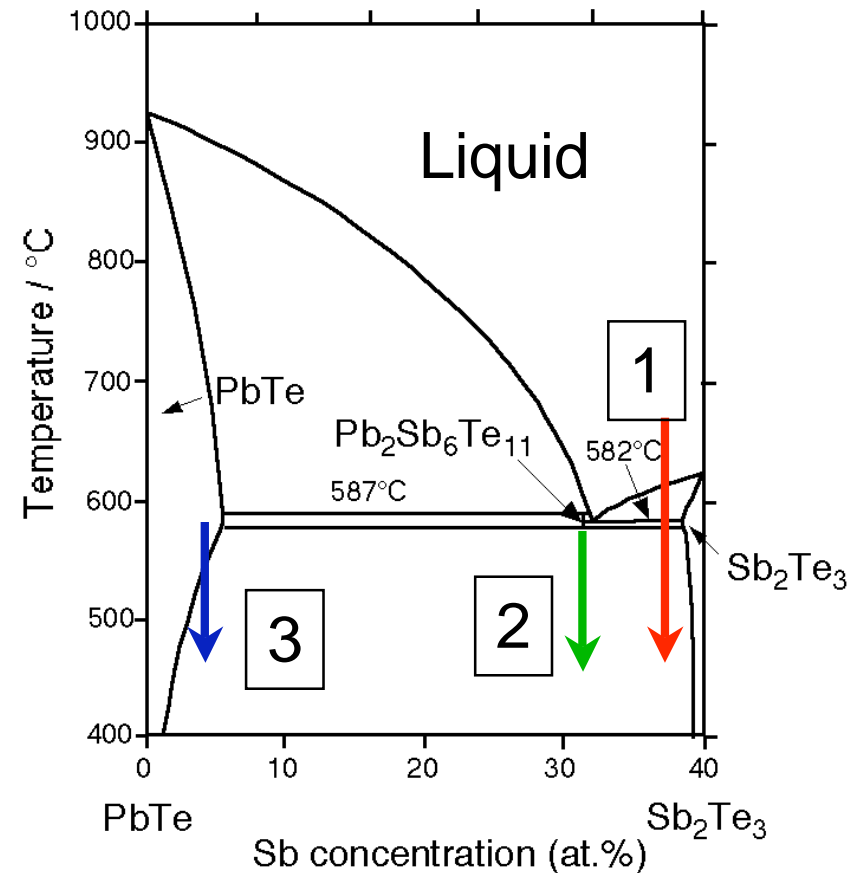
- 2) Solid-to-Solid eutectoid decomposition
slow diffusion in solid
control by quench and anneal



- 3) Solid-state Precipitation
slow diffusion in solid
control by composition and anneal



Three Methods to Create Microstructure in Sb₂Te₃ - PbTe System



Epitaxy-like Orientation

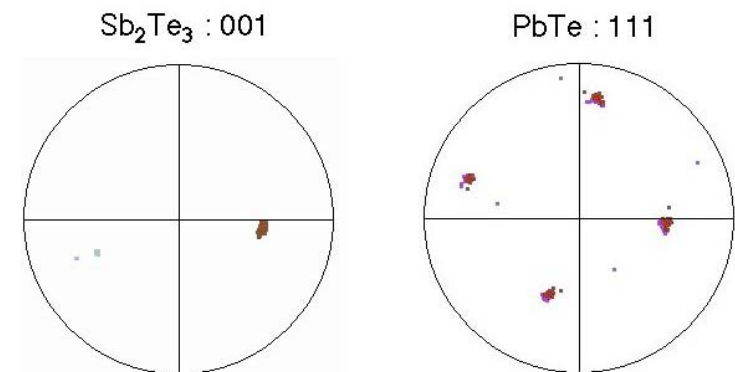
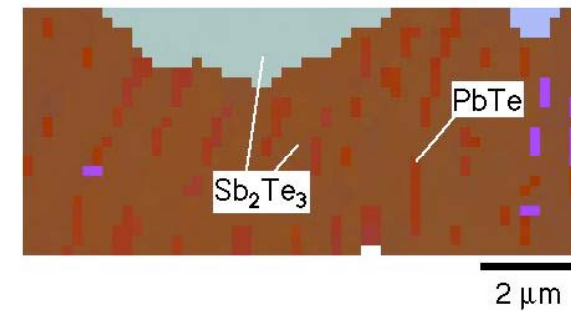
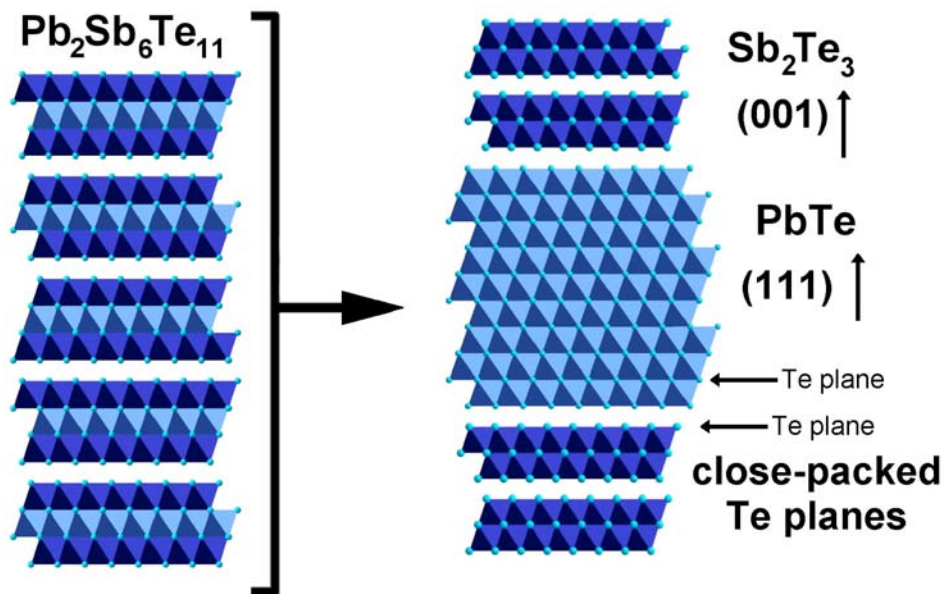
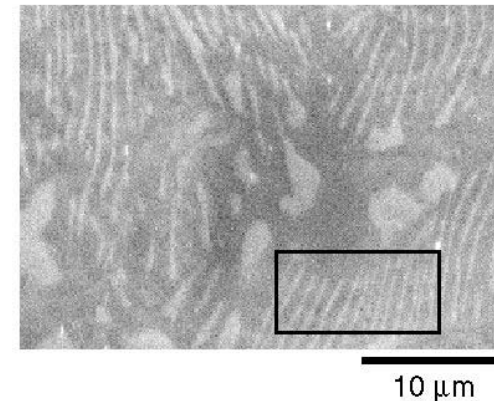
Sb₂Te₃ (001) aligned with PbTe (111)
SEM Electron Backscatter Diffraction (EBSD)

close packed Te plane

Forms epitaxy-like interface

May be critical for High zT

- Allow high e⁻ mobility
- Disrupt phonon at interface



Ikeda, et al. *Chem. Materials* **19**, p 763 (2007)

Semi-coherent interfaces



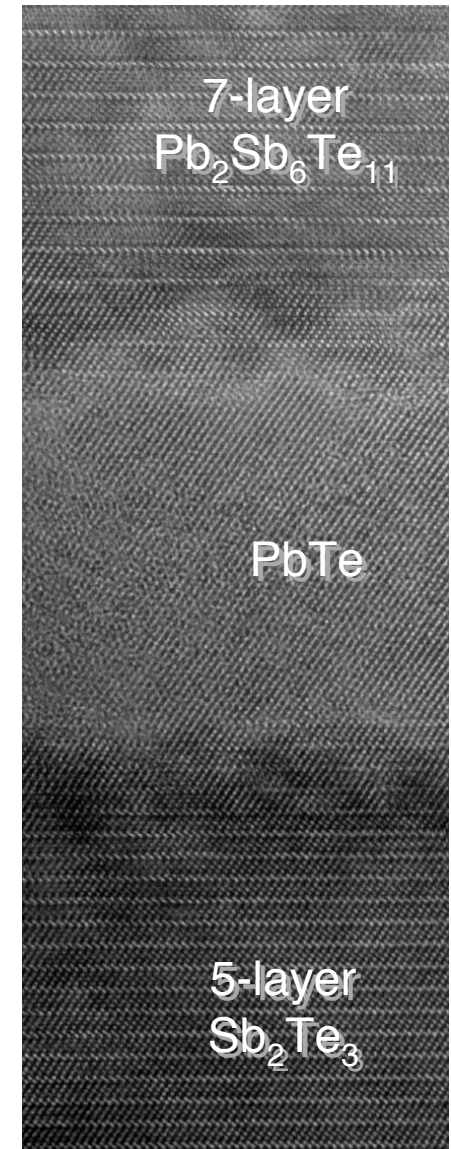
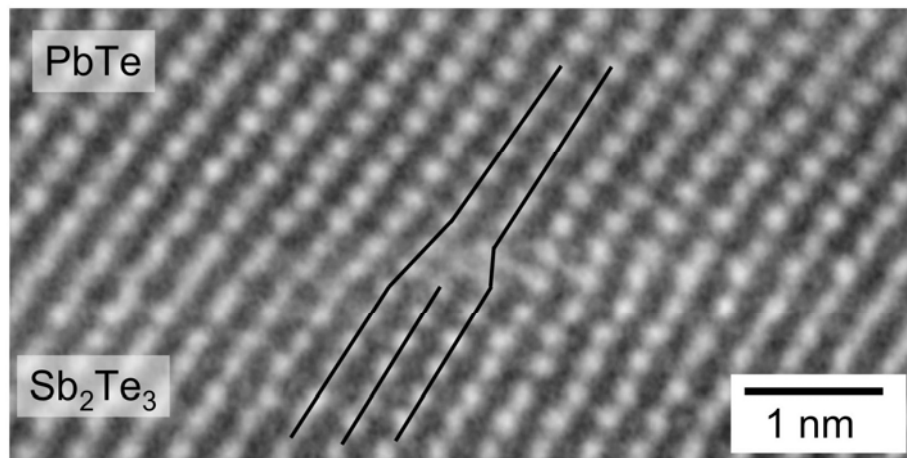
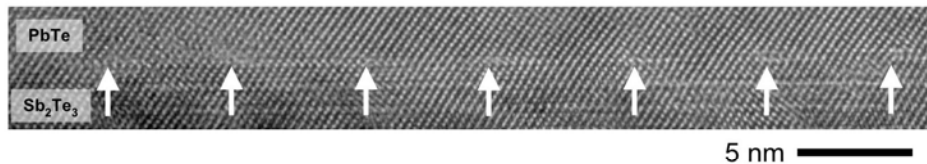
Transmission Electron Microscopy (TEM)

Reveals Crystal orientation

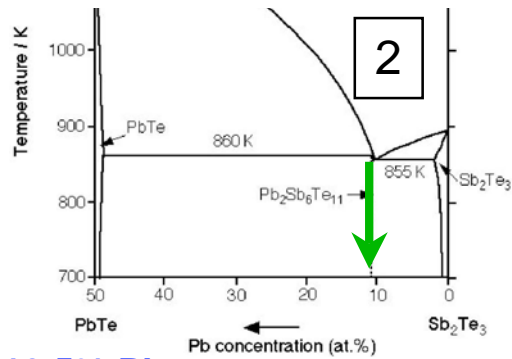
Sb_2Te_3 (001) aligned with PbTe (111)

6% lattice mismatch

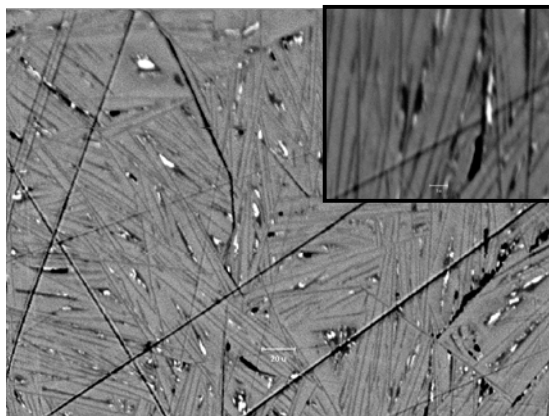
Accommodated by periodic dislocations at surface



PbTe - Sb₂Te₃ nano-lamellae

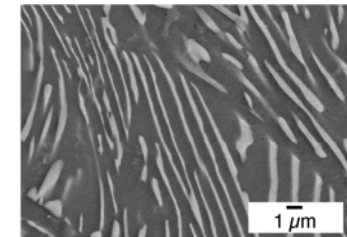
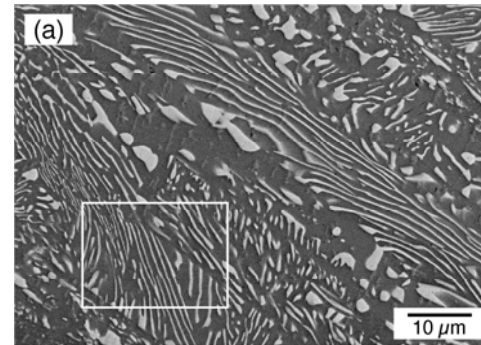


10.5% Pb

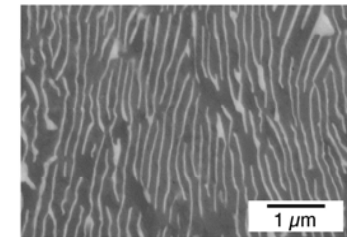
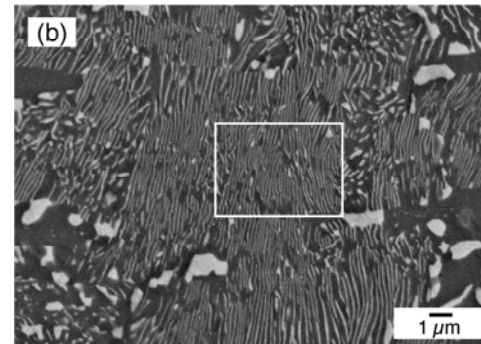


20 μm

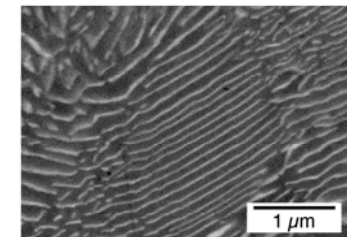
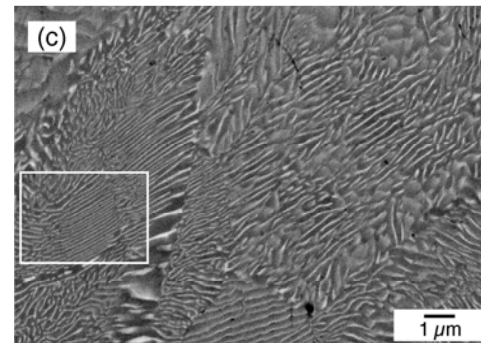
500° C



400° C



300° C



40nm PbTe
140nm Sb₂Te₃

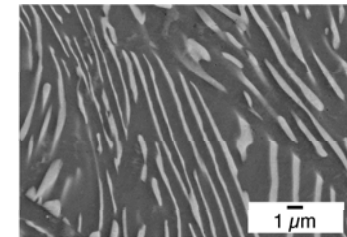
Ikeda, et al. *Chem. Materials* **19**, p 763 (2007)

Control of Lamellae size

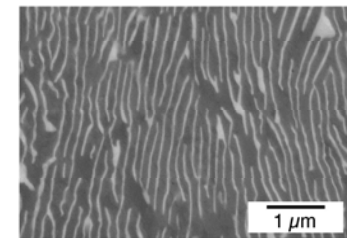
Control by time (t) and Temperature (T)
 Solid-State transformation
 Slow diffusion = Fine microstructure

$$\lambda - \lambda_0 = \frac{KD(T)t}{T}$$

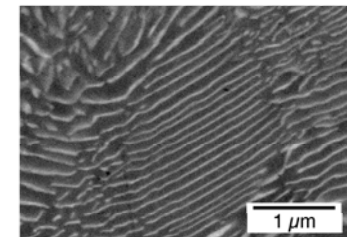
$$\lambda_0 = \frac{4\gamma T_E V_m}{\Delta H \Delta T}$$



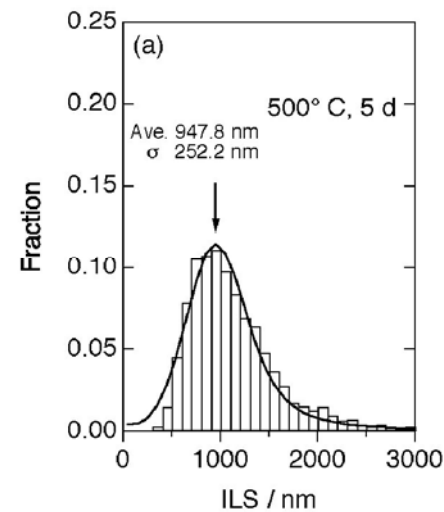
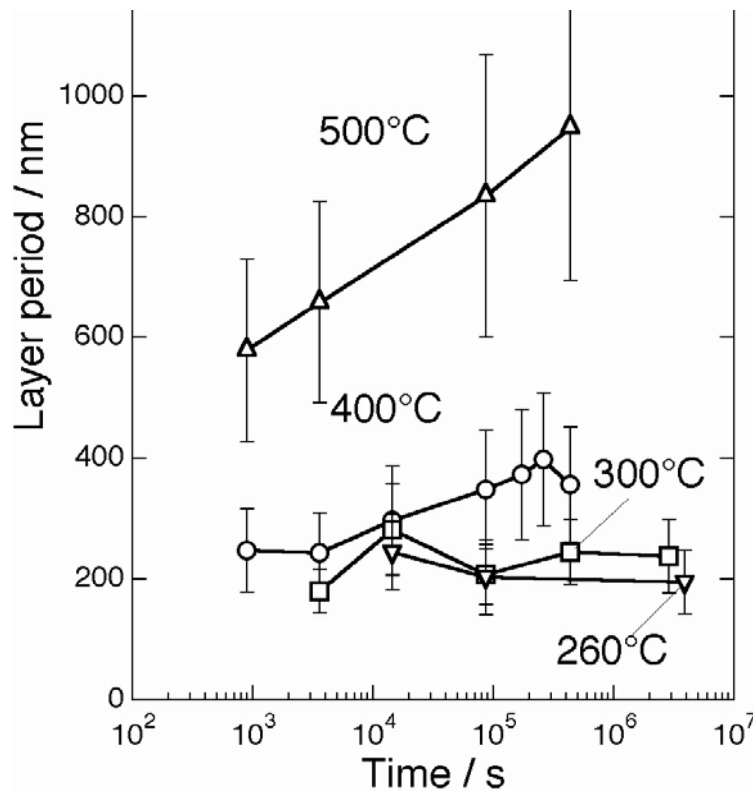
500° C



400° C



300° C



Summary



Most Thermoelectrics are Heavily Doped SC

Near Valence Balance using Zintl Rules

- needed for Band Gap

Slight Valence Imbalance

- provides metallic carriers

Electronic Transport of Complex Materials

Semi-Emperical approach successful

Ab initio is not accurate to be quantitative and difficulty including disorder

Opportunities for theory

Thermal transport

Exotic electronic states

- resonant, Kondo, correlations

Nanostructures

- composites

