A Novel Coarse-Grained Force Field for Unbiased Protein-Membrane Simulations



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Introduction



http://www.animalport.com/img/Animal-Cell.jpg

Membranes are crucial for the functioning of cells.

Proteins do the work. They also help shaping membranes.

Things often happen on large length- and time-scales.

Protein structure not always known and not always fixed.

Are there any generic principles at work?





Introduction

We would like to look into this, using tools of the following ilk:

chemically realistic but coarse-grained modeling of both membranes and proteins

no bias on secondary structure

a modicum of transferability

implicit solvent





Introduction

We think we have accomplished a little bit of that tall order. Let me thus talk about:

- 1. Our membrane model
- 2. Our peptide model
- 3. Our method of cross-parametrizing them
- 4. Some examples of what we can do





1. Our membrane model



Z.-J. Wang and M. Deserno, J. Phys. Chem. B **114**, 11207 (2010).Z.-J. Wang and M. Deserno, New J. Phys. **12**, 095004 (2010).







Construction principles:

Structure-based coarse-graining (for most non-bonded interactions)

density and pressure profiles (from experiment and simulation)

area per lipid (from experiment)



Area per lipid

The electron density at an area-per-lipid fixed to the correct value is reproduced very well!



...but that doesn't mean the stresses are right. Need to fix them independently. We can do that while modeling the implicit solvent.

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Strategy:

Use aa simulations at the correct area to infer structure

Introduce phenomenological cohesion in CG model to drive aggregation

Choose value of that cohesion to fix the zero-tension area per lipid.

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Strategy:



Range and amplitude can vary between different bead types. This is optimized to reproduce the pressure profile.



Example for structure



Example for structure aa CG



Example for structure aa CG

15

20

genuine inability to get the right structure. (Note: Henderson's theorem does *not* apply to this complicated case!)

Self-assembly

self-assembly run of 288 CG-POPC lipids in a box of ~(100 Å)³ using parallel tempering.

random 🤄

local packing

bilayer

bilayer is the thermodynamically stable phase.

Elastic properties Bending modulus *k*

4608 lipids, (~40nm)² box, several µs

from fluctuations: $\kappa = \frac{k_B T}{L^2 q^4 \left\langle \left| h_q \right|^2 \right\rangle}$ (8...10)×10⁻²⁰ J

experimentally, it's about: 8.5×10-20 mN/m

N. Kučerka, S. Tristram-Nagle & J.F. Nagle, J. Mem. Biol. 208, 193 (2005).

Phase behavior

Our bilayer shows a liquid-gel transition.

T=279 K

T=310 K

Phase behavior

Diffusion

Time mapping

intrinsic computer time: $\tau = \sigma \sqrt{m/\epsilon} = 0.062 \,\mathrm{ps}$

Using this to translate times gives $D \simeq 4.5 \times 10^{-12} \ \mu m^2 / (0.062 \text{ ps}) \simeq 73 \ \mu m^2 / \text{s}$

Experimentally, more like $D \simeq (1...10) \, \mu m^2/s$

However:

Reduced CG resolution implies reduced friction and thus an *expected* speed-up! (Here, between one and two orders of magnitude)

Transferability

Example: "tail swapping"

Transferability

Area per lipid

2. Our peptide model

T. Bereau and M. Deserno, J. Chem. Phys. 130, 235106 (2009);

- T. Bereau, M. Bachmann, and M. Deserno, J. Am. Chem. Soc. 132, 13129 (2010);
- T. Bereau, M. Deserno, and M. Bachmann, Biophys. J. 100, 2764 (2011);
- T. Bereau, C. Globisch, M. Deserno, and C. Peter, J. Chem. Theo. Comput. ASAP (2012).

Interactions

harmonic bonds

harmonic angles

improper dihedrals fix chirality at C_α

non-bonded

bonded

- excluded volume Ramachandran plot
- hydrophobic interactions

dihedrals: simple cosine

- hydrogen bonds
 - peptide dipoles

folding characteristics

Ramachandran plots 180 54.5120 4 3.5 60 3 Ψ 2.5 (2 ω C_{α} -60 1.5-120 GGG 0.5 V -180 -180-120 -60 60 120 180 () 180 54.5 120 4 3.5 \mathbf{n} 3 Ψ 2.5 2 -60 1.5 tripeptide -120 GAG 0.5 -180 -180-120 -60 60 120 180 **Carnegie Mellon**

α

Hydrogen bonds

extremely important for secondary structure formation!

 α_1

Potential *range* adjusted according to known properties of hydrogen bonds.

Potential *strength* adjusted according to folding properties of the model.

A. Irbäck, F. Sjunneson, and S. Wallin, PNAS 97, 13614 (2000)

Peptide bond dipoles

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N.Y. Chen, Z.Y. Su, and C.Y. Mou, Phys. Rev. Lett. 96, 078103 (2006)

Peptide bond dipoles

 β -sheets are electrostatically more favorable than α -helices.

Incorporate this effect in the (φ,ψ)-dihedrals!

N.Y. Chen, Z.Y. Su, and C.Y. Mou, Phys. Rev. Lett. **96**, *078103* (2006) Y. Mu and Y.Q. Gao, J. Chem. Phys. **127**, *105102* (2007)

Hydrophobic interactions

Miyazawa-Jernigan matrix

 $\mathbf{\epsilon}_{ij}^{\mathrm{MJ}}$

\mathcal{I}				
	Ser	Gly	Phe	
Ser	-1.67	-1.82	-4.02	• • •
Gly	-1.82	-2.24	-4.13	•••
Phe	-4.02	-4.13	-7.26	•••

S. Miyazawa and R.L. Jernigan, J. Mol. Biol. 256, 623 (1996) Carnegie Mellon

Miyazawa-Jernigan matrix

$$\epsilon_{ij}^{\text{MJ}} \approx \epsilon_{ij} = \sqrt{\epsilon_i \epsilon_j}$$
 (Lorentz-Berthelot)

normalize:
$$\mathcal{E}_i^{\prime} = \frac{\mathcal{E}_i - \min \mathcal{E}_k}{\max \mathcal{E}_k - \min \mathcal{E}_k}$$

correlation between ϵ_{ij}^{MJ} and ϵ_{ij} : 98% correlation between ϵ_i^{NJ} and ϵ_i : 87%

> hydrophobicity scale by Fauchere and Pliska J.L. Fauchere and V. Pliska, Eur. J. Med. Chem. **18**, 369 (1983).

S. Miyazawa and R.L. Jernigan, J. Mol. Biol. 256, 623 (1996)

Miyazawa-Jernigan matrix

$$\epsilon_{ij}^{\text{MJ}} \approx \epsilon_{ij} = \sqrt{\epsilon_i \epsilon_j}$$
 (Lorentz-Berthelot)

normalize:
$$\mathcal{E}_{i}^{*} = \frac{\mathcal{E}_{i}^{*} - \min \mathcal{E}_{k}}{\max \mathcal{E}_{k}^{*} - \min \mathcal{E}_{k}}$$

Use $\mathcal{E}_{overall} \times \mathcal{E}_{i}^{\prime}$ as the strength in a Lennard-Jones potential.

S. Miyazawa and R.L. Jernigan, J. Mol. Biol. 256, 623 (1996) Carnegie Mellon

73 aa

Folding of a 3-helix bundle

MGSWAEFKQRLAAIKTRLQALGGSEAELAAFEKEIAA… …FESELQAYKGKGNPEVEALRKEAAAIRDELQAYRHN

Parallel tempering at 8 temperatures; shown: configuration in lowest replica. Total simulation time: ~10 hours. Carnegie Mel

Folding of a 3-helix bundle

73 aa

MGSWAEFKQRLAAIKTRLQALGGSEAELAAFEKEIAA… …FESELQAYKGKGNPEVEALRKEAAAIRDELQAYRHN

> hydrophobicity: proper helix-alignment, but not a collapsed globule

hydrogen bond:

secondary structure formation, but not just one long α -helix.

dipole term:

 β -sheets maximally stable without disabling α -helices.

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2A3D

Example application

Microcanonical folding analysis of this 3-helix bundle.

T. Bereau, M. Bachmann, and M. Deserno, J. Am. Chem. Soc. **132**, 13129 (2010); T. Bereau, M. Deserno, and M. Bachmann, Biophys. J. **100**, 2764 (2011);

3. Cross-parametrization

Aim: Reproduce *free energy of insertion* of individual amino acids into the bilayer.

Target data for this:

J. L. MacCallum, W. F. D. Bennett, and D. P. Tieleman, Biophys. J. **94**, 3393 (2008).

Free energy of insertion

Ala

Gln

atomistic CG

Generic features captured semiquantitatively.

"Sharp" features difficult to match with low-resolution model.

Final interaction potentials likely not unique, but this is not a problem.

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coarse grained atomistic can observe "water defects" (without water) <u>Carnegie Mellon</u>

Folding of WALP23 in membrane environment

WALP23 in 62 lipid POPC bilayer at T=356K. CG REMD, total time ~ $10^6 \tau$ (~~8µs)

Pore formation by APs $t = 0 \tau$

8 magainin, 288 POPC, T=308K. CG REMD. Total time ~~2µs

Pore is created without even one AP actually spanning it. Helices break.

Results secondary at this point. But notice the kinds of problems you could treat in this way.

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