

Coarse-Grained Simulations of Polymer-Grafted Nanoparticle Monolayers

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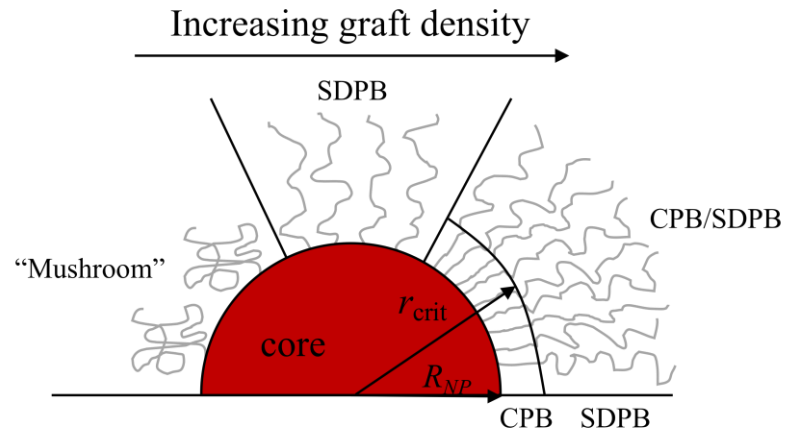
Polymer-Grafted Nanoparticles (PGNs)

Traditional Polymer Nanocomposites

- ❑ Inorganic nanoparticle (NP) fillers often added to polymer matrices to enhance optical, electrical, and mechanical properties
- ❑ Key challenges: controlling and predicting particle dispersion

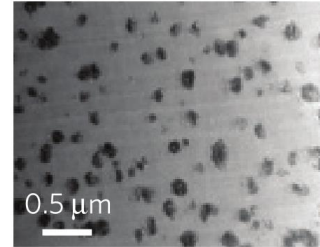
Polymer-Grafted Nanoparticles (PGNs)

- ❑ Can improve dispersion in composites
- ❑ Can be **used neat** (our focus; cannot demix macroscopically)
- ❑ Polymer conformation modified by constraint of NP surface

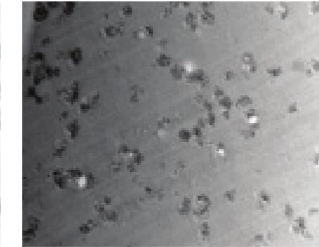


Ohno *et. al. Macromolecules* **2007**, *40*, 9143-9150

$\sigma = 0.01$ chains nm^{-2}
 $M_g = 25$ kg mol^{-1}



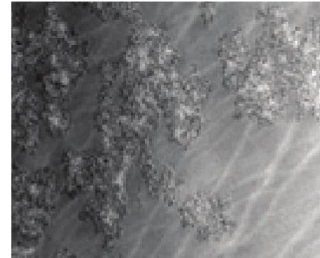
$M_g = 51$ kg mol^{-1}



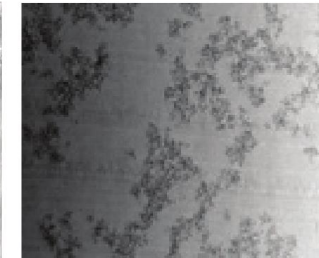
$M_g = 158$ kg mol^{-1}



$\sigma = 0.05$ chains nm^{-2}
 $M_g = 17$ kg mol^{-1}



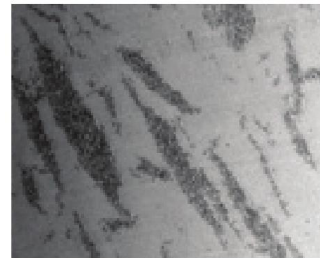
$M_g = 34$ kg mol^{-1}



$M_g = 106$ kg mol^{-1}



$\sigma = 0.1$ chains nm^{-2}
 $M_g = 24$ kg mol^{-1}



$M_g = 45$ kg mol^{-1}



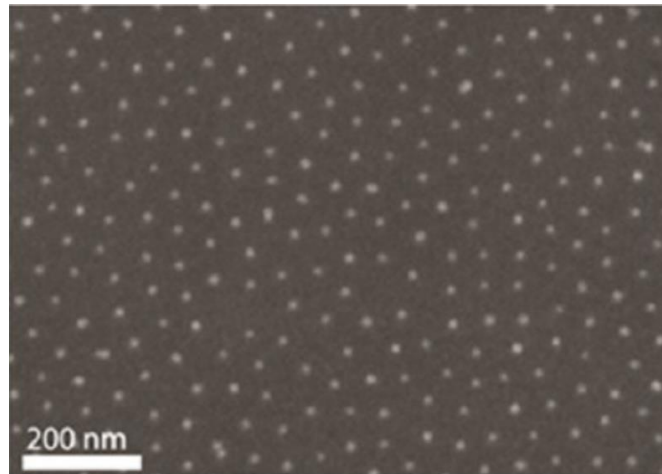
$M_g = 100$ kg mol^{-1}



Akcora, Liu, Kumar, *et al. Nature Materials*, **2009**, *8*, 354-359

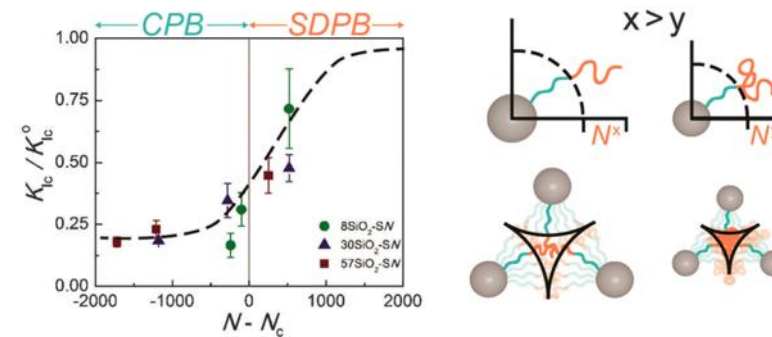
Polymer-Grafted Nanoparticle (PGN) Assemblies

- ❑ Structure depends on NP size, graft density, chain length
- ❑ On surfaces, NPs can form hexagonal arrays



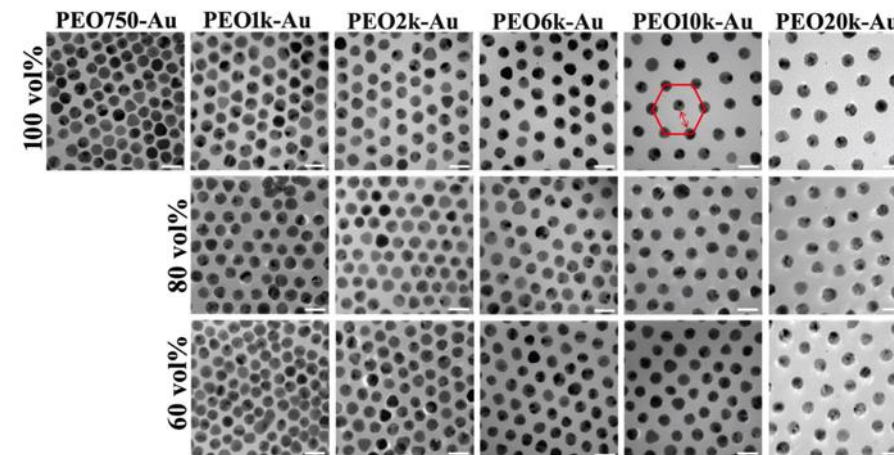
Che, Park, Grabowski, *et al.*,
Macromolecules, **2016**, *49*, 1834–1847

Toughness changes with graft length



Schmitt, Michael, *et al.* *Soft Matter* **2016**, *12*, 3527-3537

Particle spacing changes with graft length



Yang, Guang, *et al.* *Macromolecular Chemistry and Physics* **2018**, 1700417

Molecular Dynamics Simulations

- ❑ Coarse-grained bead-spring chains
- ❑ Randomly grafted to spherical nanoparticle
- ❑ LJ units (monomer diameter= 1σ)

Bonds:
Finitely Extensible Nonlinear Elastic (FENE)

$$u = -kR_0^2 \ln\left(1 - \left(r / R_0\right)^2\right)$$

Monomer-monomer interactions:
Lennard-Jones (LJ) potential

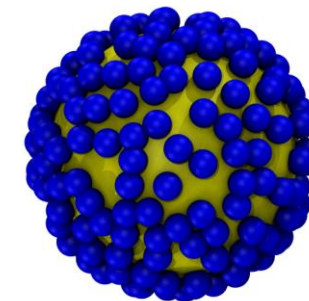
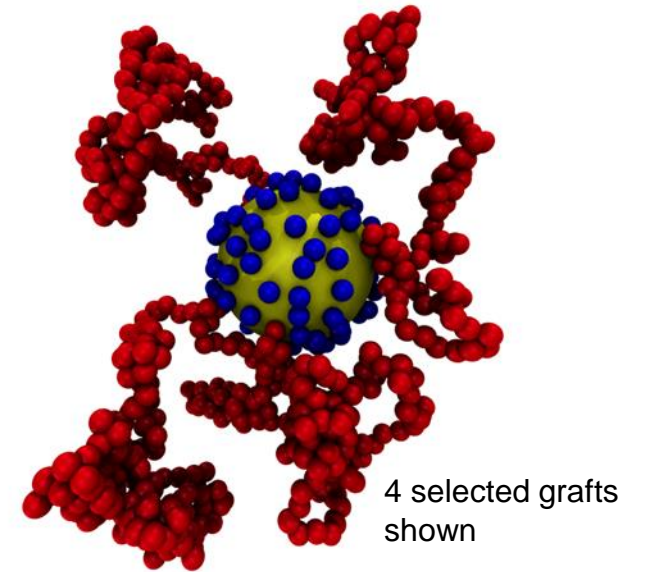
$$U_{LJ}(r) = 4e_{ij} \left[\left(\frac{S}{r}\right)^{12} - \left(\frac{S}{r}\right)^6 \right] \text{ cut off at } r_c = 2.5$$

NP-monomer interactions:

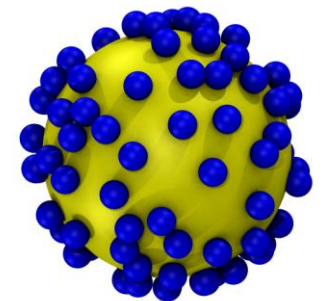
LJ with r shifted by 4.5 (particle diameter 10σ)

Wall-monomer interactions:
9-3 LJ potential

$$U_{LJ,W}(r) = e_{wp} \left[\frac{2}{15} \left(\frac{S}{r}\right)^9 - \left(\frac{S}{r}\right)^3 \right]$$



(0.6 chains/ σ^2)



(0.3 chains/ σ^2)

Systems Simulated

12 hexagonally packed polymer-grafted NPs

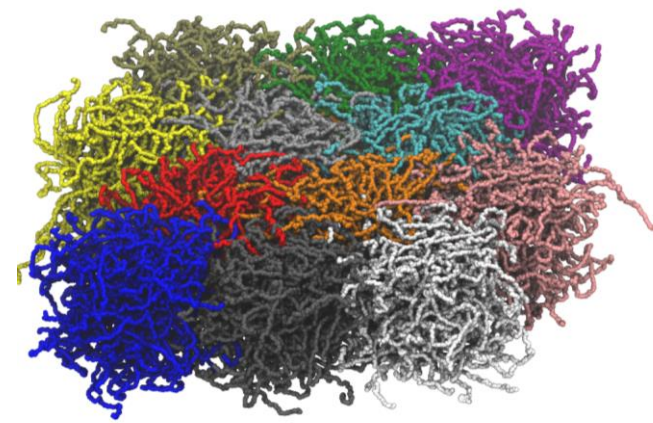
- ❑ 35 (unentangled) or 160 (entangled) beads/chain
- ❑ Graft density 0.3 – 0.6 chains/ σ^2
- ❑ System names: PGN(graft density*10)-chain length

- ❑ Periodic in x and y directions, wall at $z = 0$
- ❑ $\epsilon_{w-p} = 3.5$ (stable monolayer)
- ❑ Equilibrate in melt state on surface

- ❑ (later) Cooled (glassy) films with wall removed

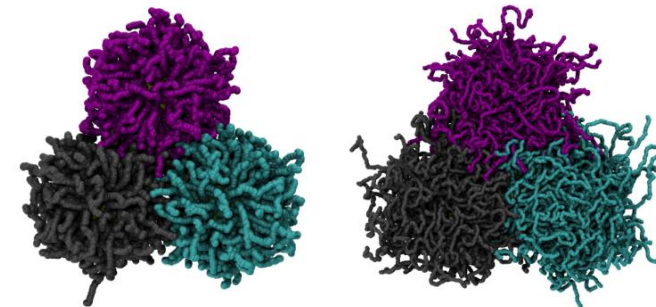
Homopolymer films (without NPs)

- ❑ Chain lengths $N = 160, 240,$ and 320
- ❑ Area set to match thickness of PGN3-160 film



Tilted view,
entire film

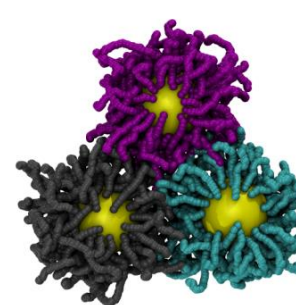
Time-averaged snapshots, unwrapped (grafted chain shown in same image as its nanoparticle)



Top view of three
selected particles
for various
systems

PGN6-35

PGN6-160



PGN3-35

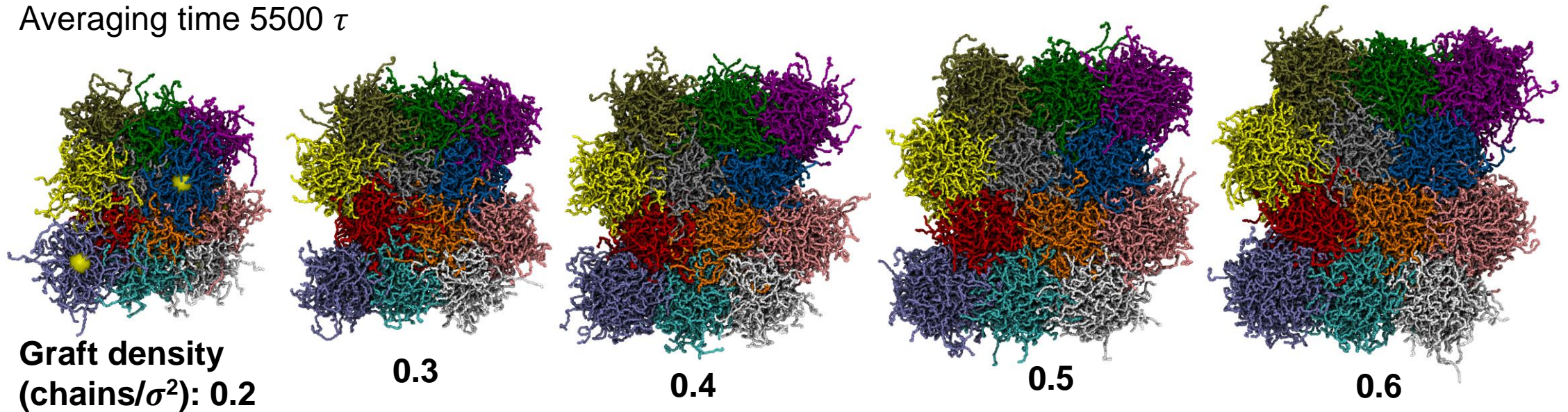
PGN3-160

Film Structure, $N=160$

- Volume fraction effect: Higher graft density means larger spacing
- Degree of hexagonal ordering increases with graft density

Time-averaged snapshots, unwrapped (grafted chain shown in same image as its nanoparticle)

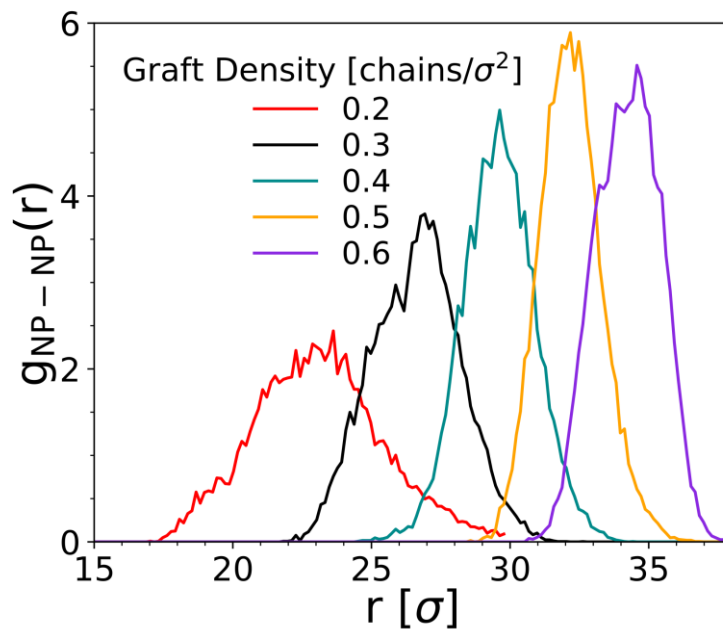
Averaging time 5500τ



Film Structure, N=160

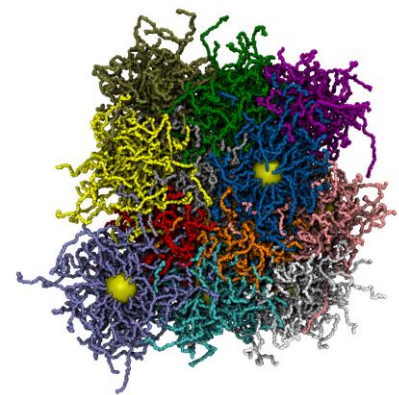
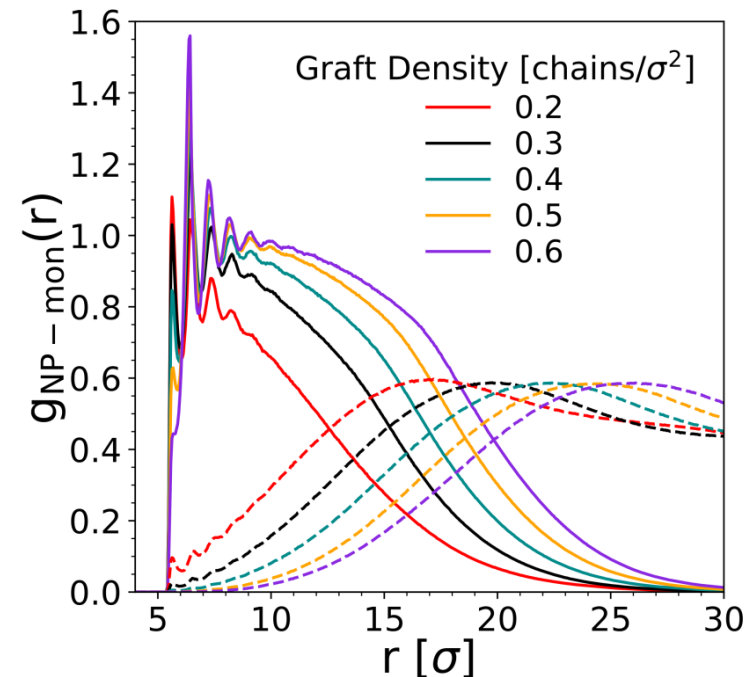
- Volume fraction effect: Higher graft density means larger spacing
- Degree of hexagonal ordering increases with graft density
- Increasing graft density decreases interpenetration

NP-NP pair correlation functions

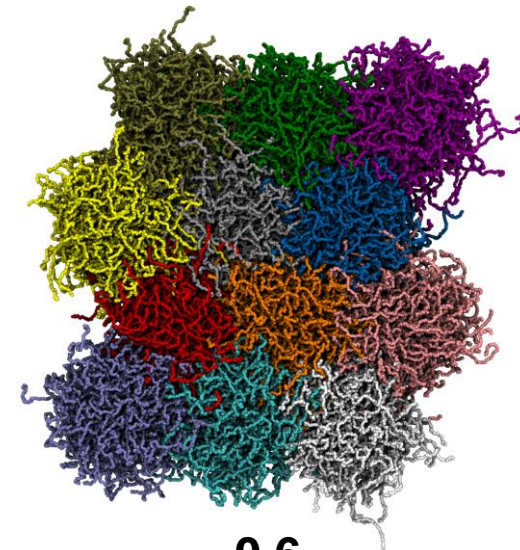


NP-monomer pair correlation functions

Solid lines: attached monomers
Dashed lines: other monomers

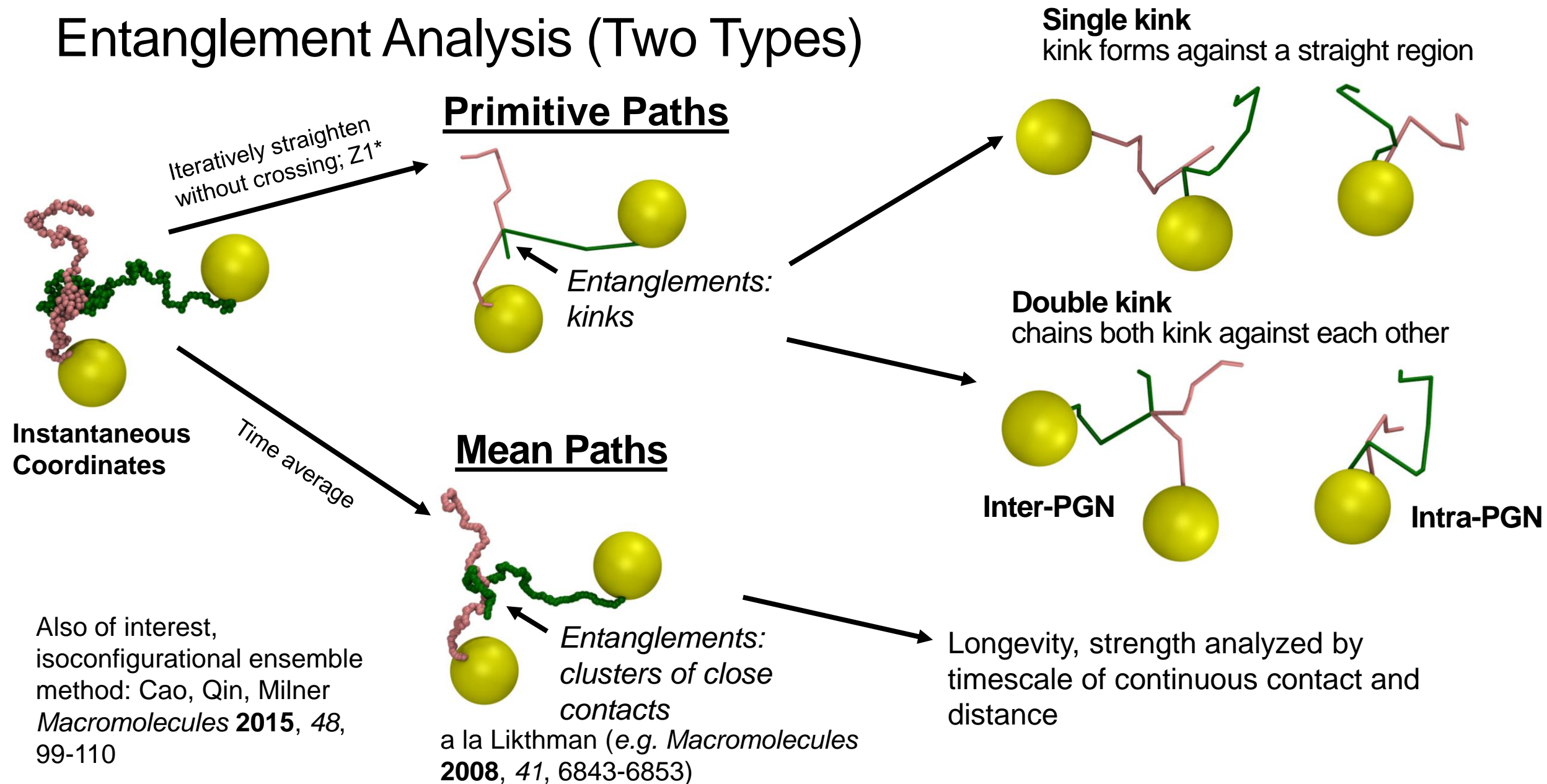


0.2



0.6

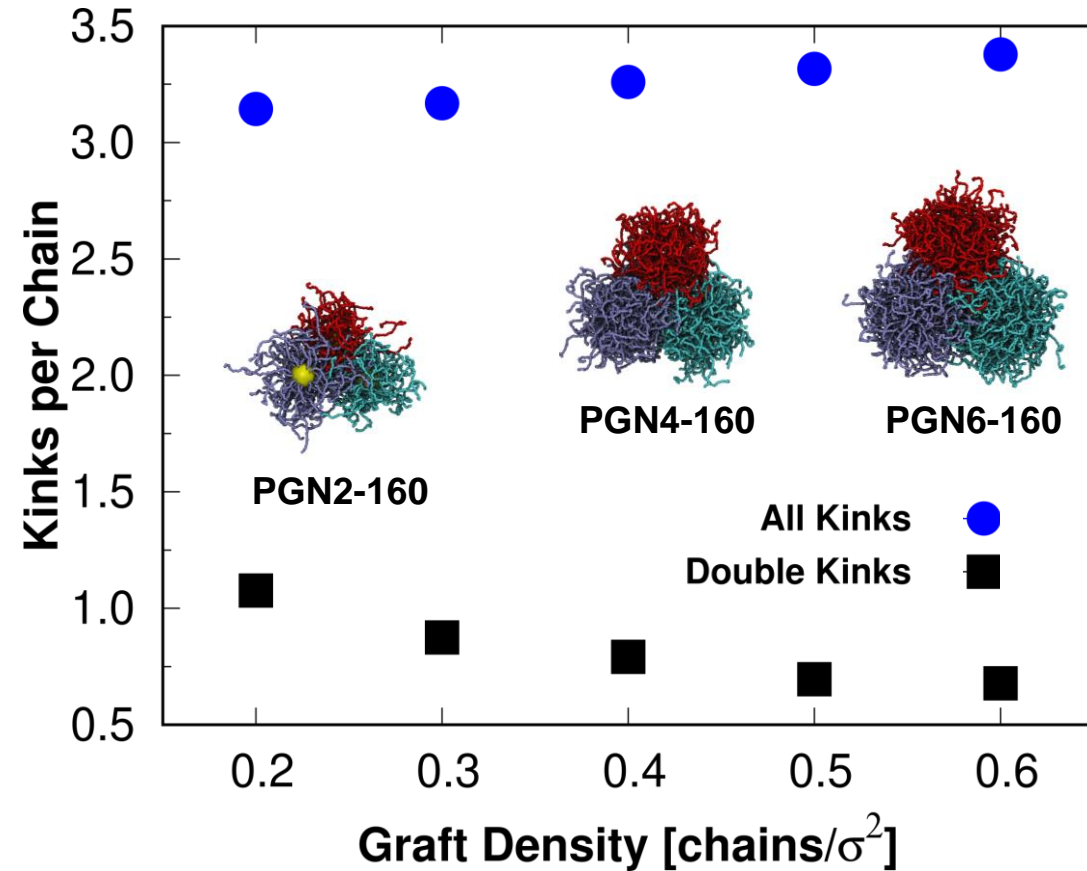
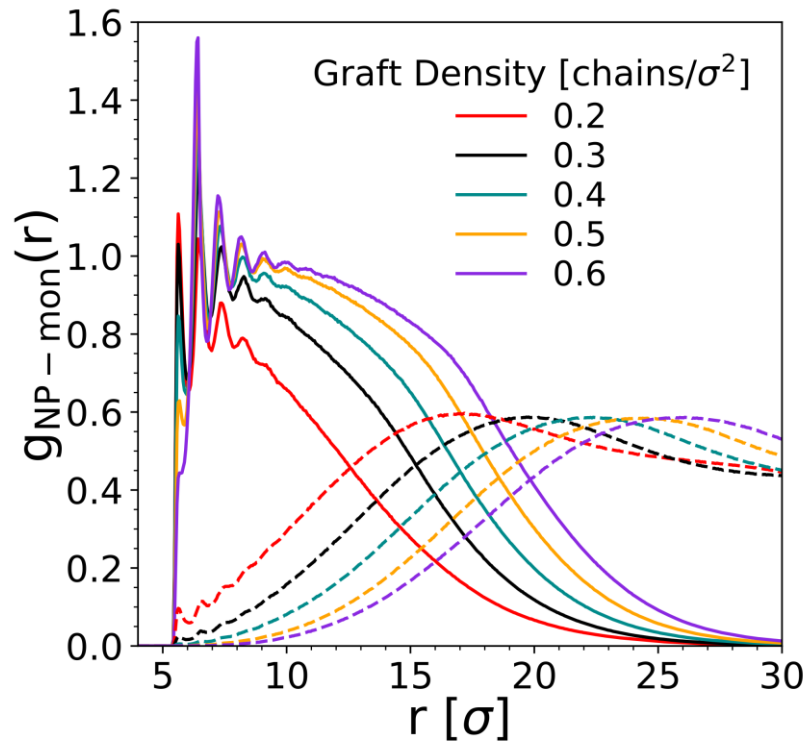
Entanglement Analysis (Two Types)



*Z1 algorithm provided by Prof. Martin Kröger

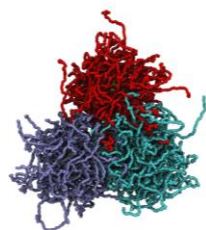
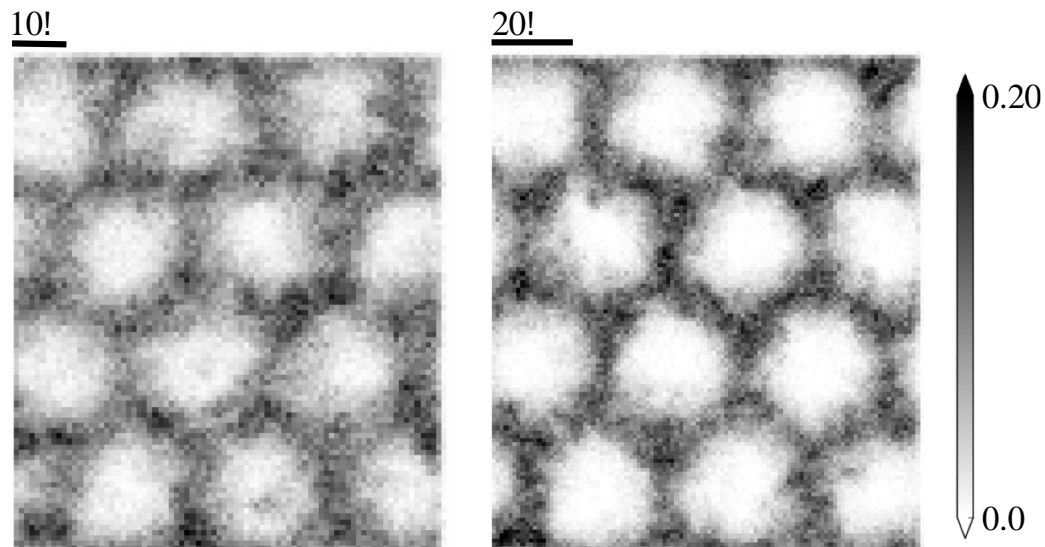
Entanglements Per Chain

- Overall (mostly single) kinks per chain increase with graft density
- Double kinks per chain decrease with graft density

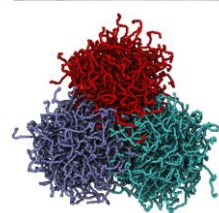


Entanglement Locations: Heat Maps

Interparticle double kinks: top view



PGN3-160
Moderate
graft density

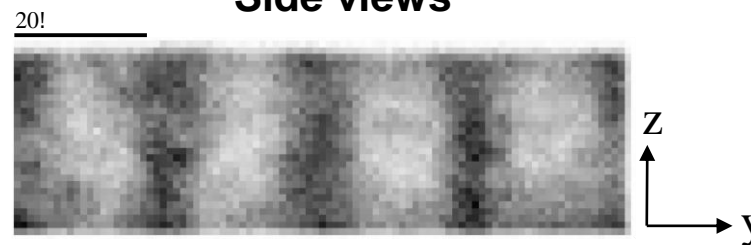


PGN6-160
High graft
density

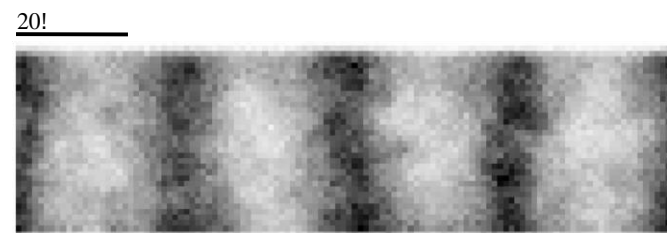
- ❑ Double kinks localized in interstitial areas at high graft density
- ❑ Lowering graft density increases entanglements above and below particles

Side views

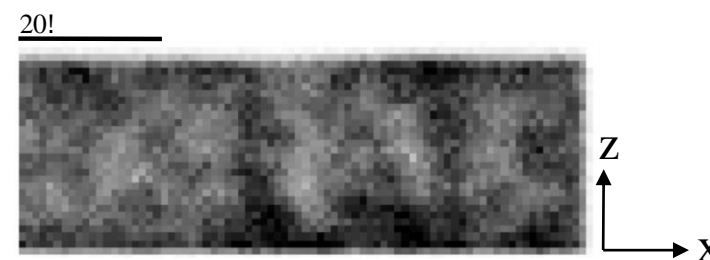
PGN3-160



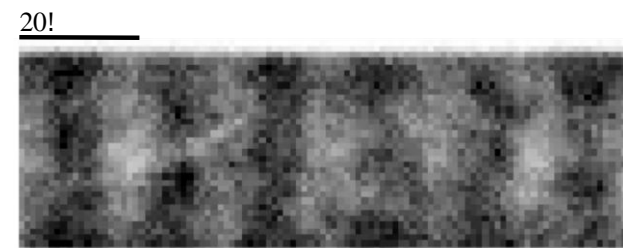
PGN6-160



PGN3-160



PGN6-160

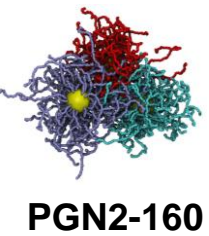
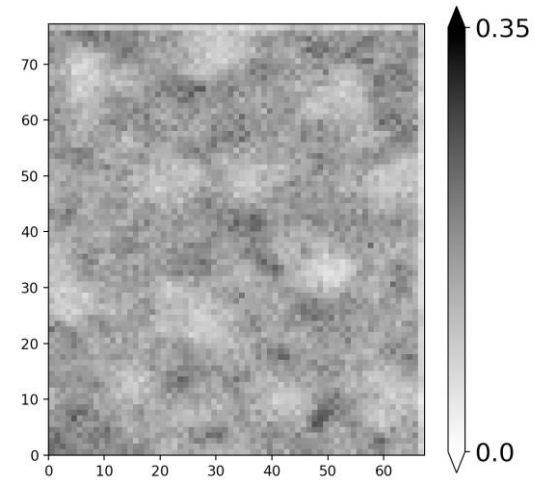
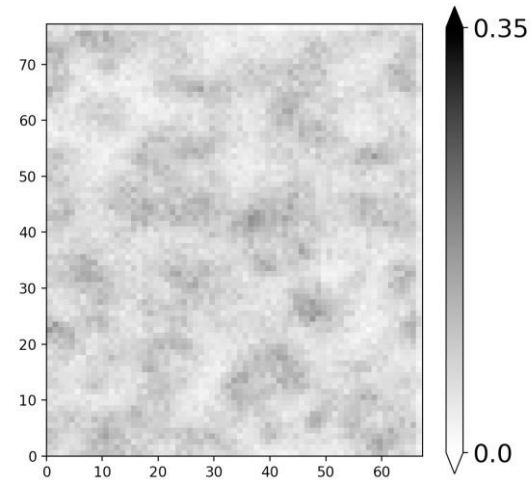
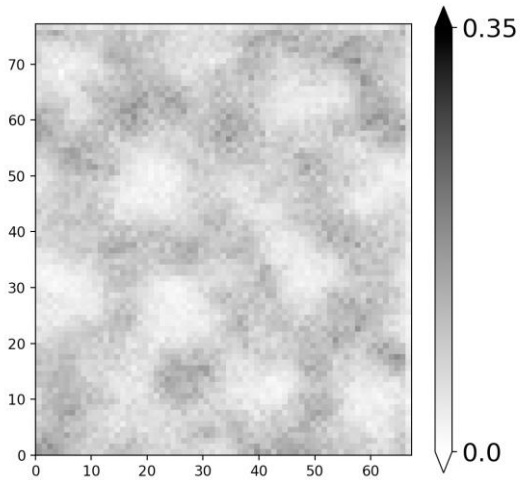


Entanglement Locations

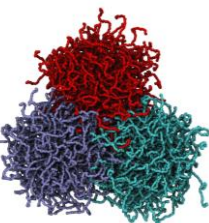
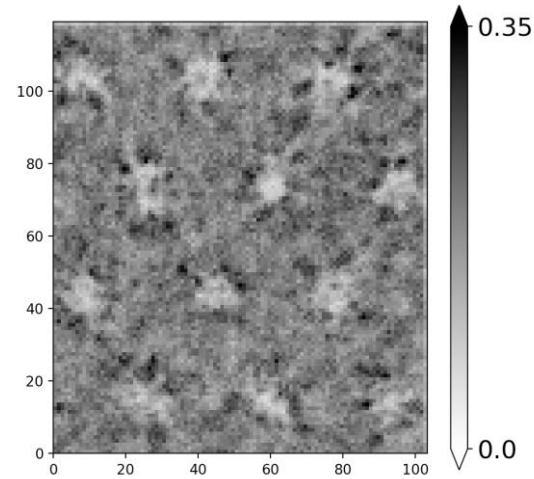
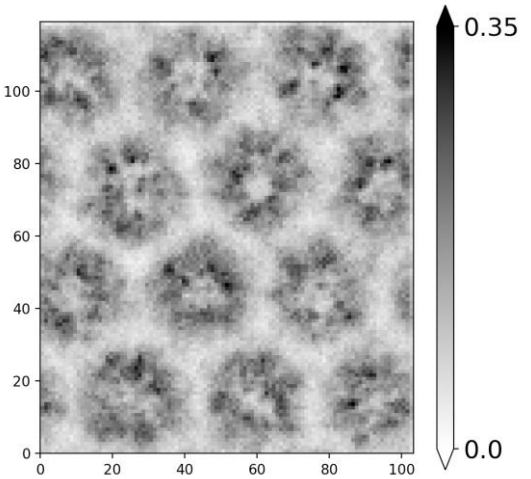
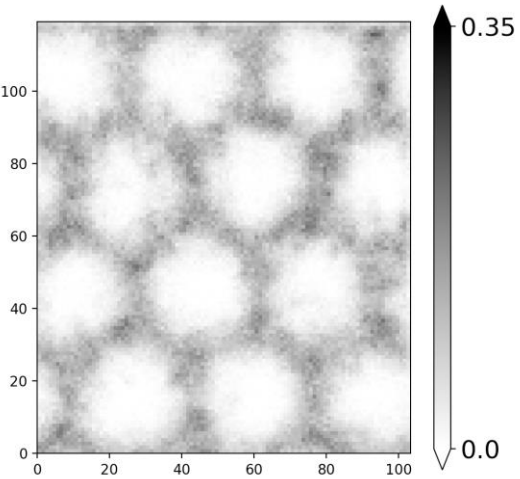
Inter-PGN

Intra-PGN

All Double Kinks



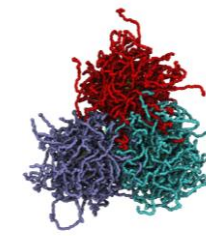
PGN2-160



PGN6-160

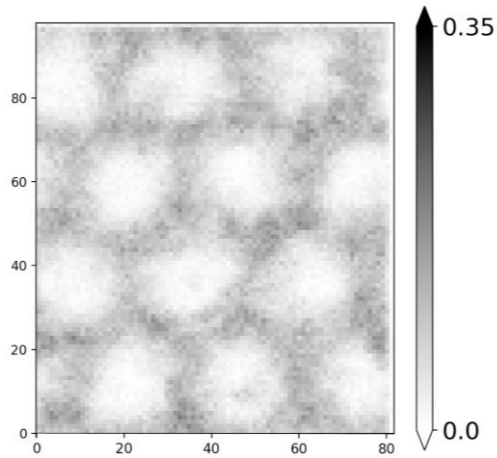
- Increased intraparticle kinks near particle surface
- Increase near surface pronounced at high graft density (local chain stretching)

Entanglement Locations

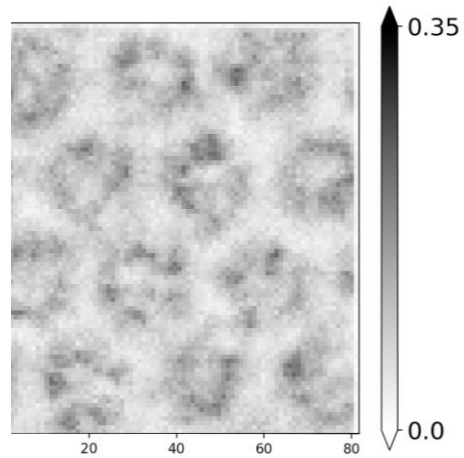


PGN3-160
Moderate
graft density

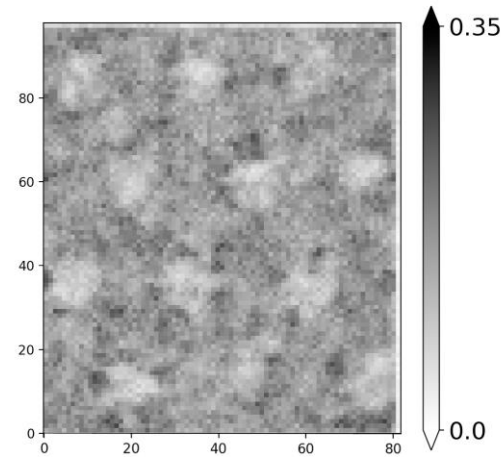
Inter-PGN



Intra-PGN

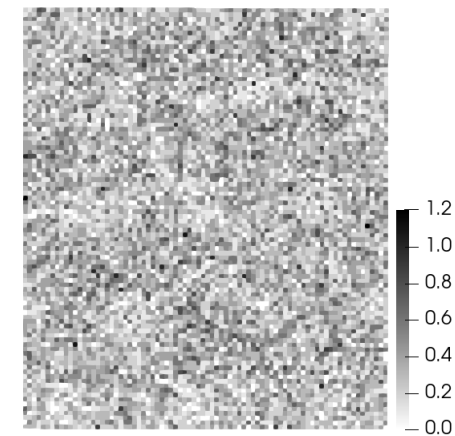
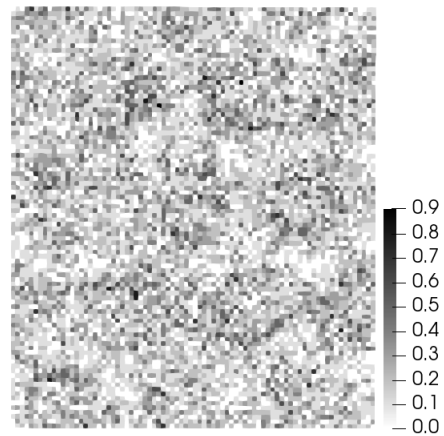
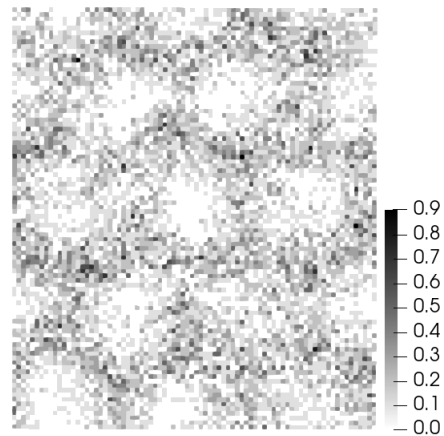


All Double Kinks

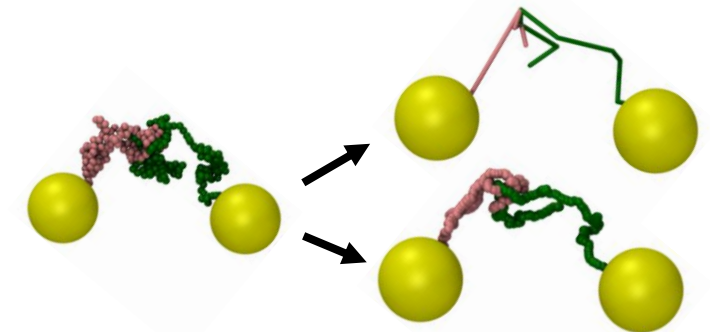


- Mean path, primitive path analyses qualitatively similar
- Mean path analysis yields tightness and timescale of contact
- Some long held mean path contact clusters correspond to multiple kinks

Double Kinks (Z1)



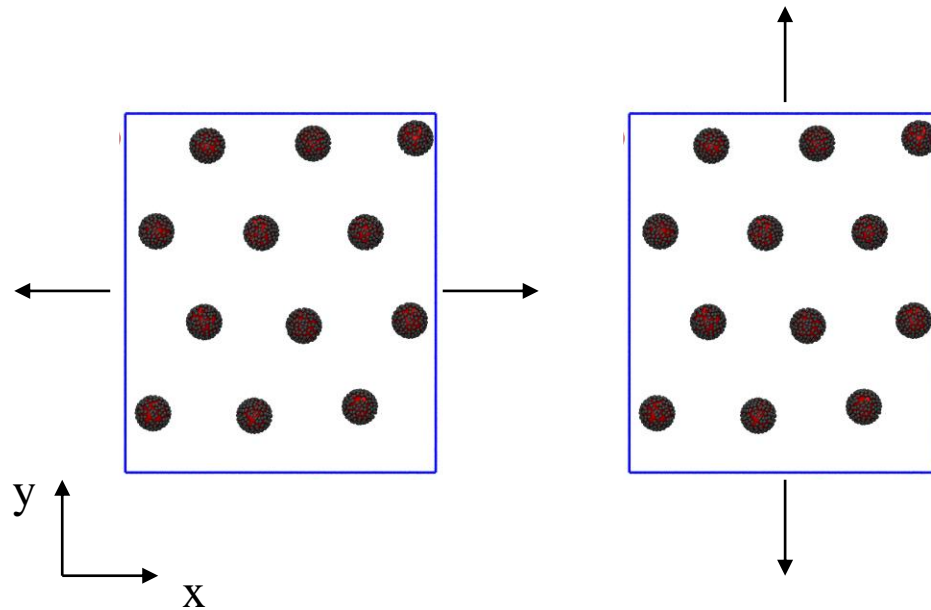
Mean Path Closest Contacts (Preliminary)



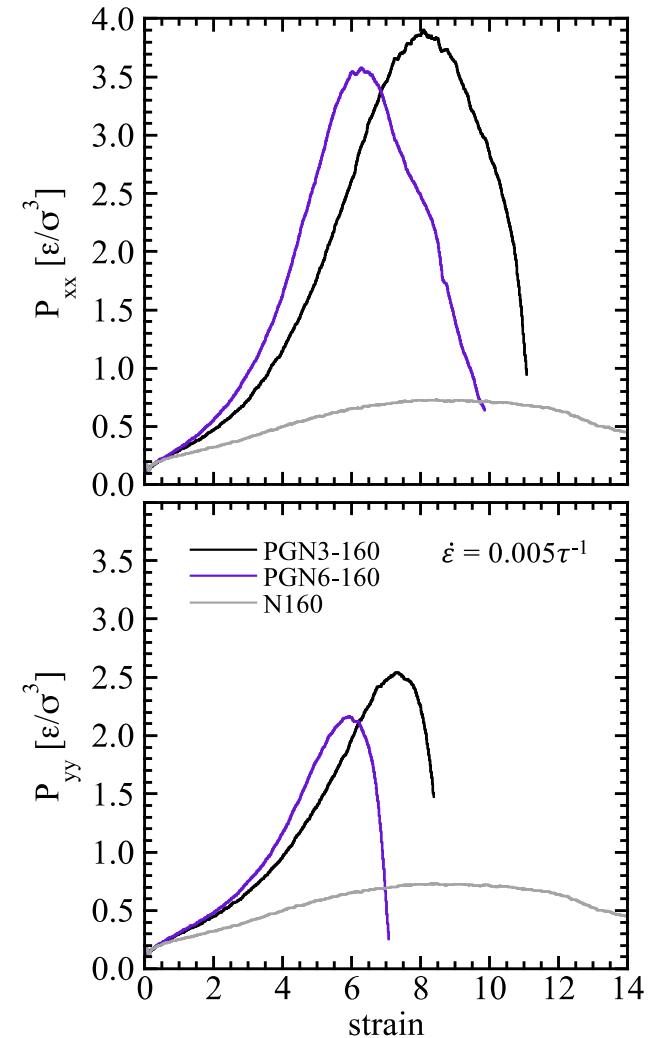
Mechanical Properties Depend on Graft Density (Melt State)

Deformation behavior: melt state, on surface

- ❑ Strain rate of $0.005 \tau^{-1}$ in x or y, constant P in other direction
- ❑ Stress recorded in direction of pull



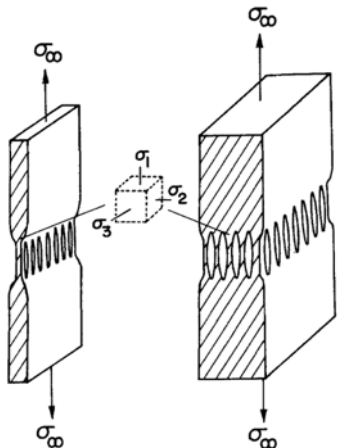
Higher graft density particles fail at lower strain due to less interparticle interaction and entanglements



Mechanical Behavior in Glassy State: Background

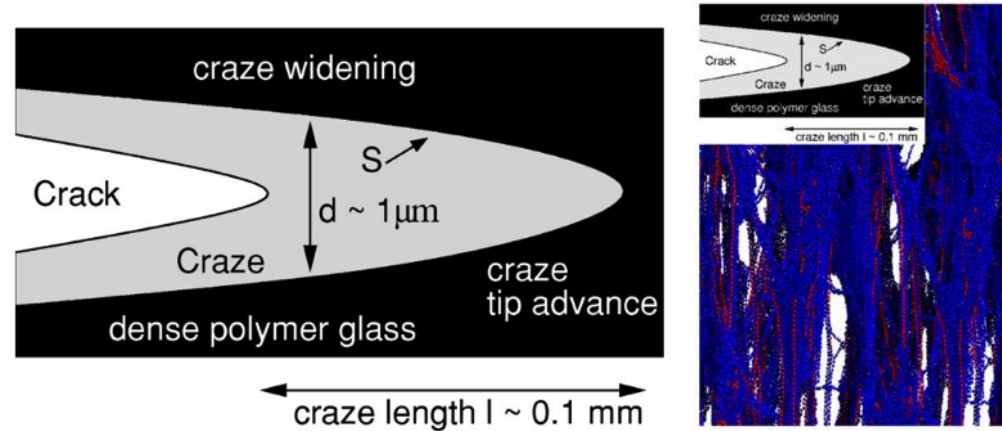
- During plastic flow, polymer is drawn into an intricate network of fibrils, or “craze”
- Significantly increases the fracture energy of the material

Thin Films vs. Bulk

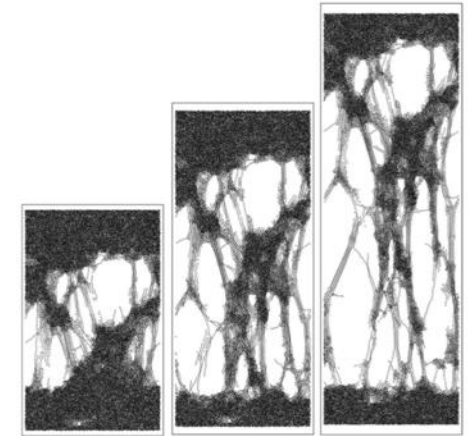


Chan, Donald, Kramer, *J. Mat. Sci.* **1981**, 16, 676

Molecular Dynamics Simulations

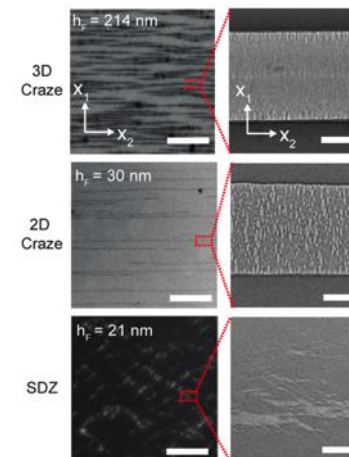


Baljon, Robbins, *Macromolecules* **2001**, 34, 4200



Rottler, Robbins, *Phys. Rev. E* **2003**, 68, 011801

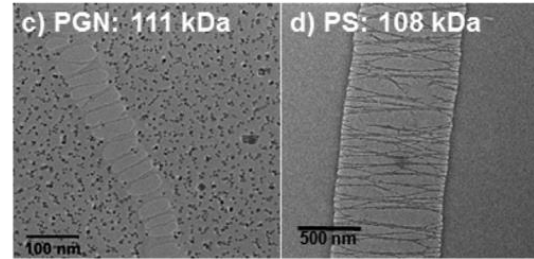
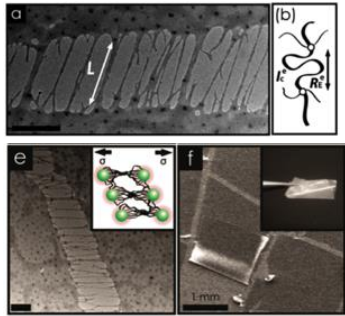
- Crazes observed in films with thickness > 30nm



Bay, Shimomura, Liu, Ilton, Crosby, *Macromolecules* **2018**, 51, 3647–3653

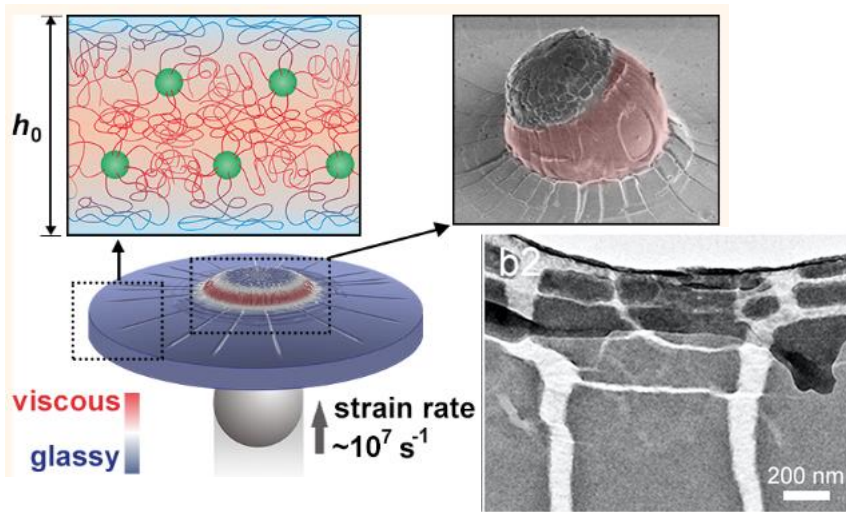
Crazing in Glassy PGN Assemblies

Neat Polymer-Grafted Nanoparticles (PGNs), Experiments

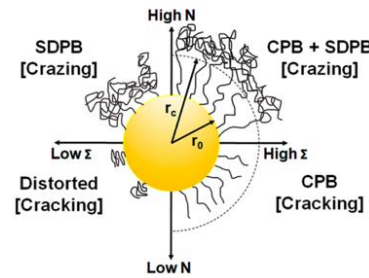


Choi, Dong, Matyjaszewski, Bockstaller, *J. Am. Chem. Soc.* **2010**, *132*, 12537–12539

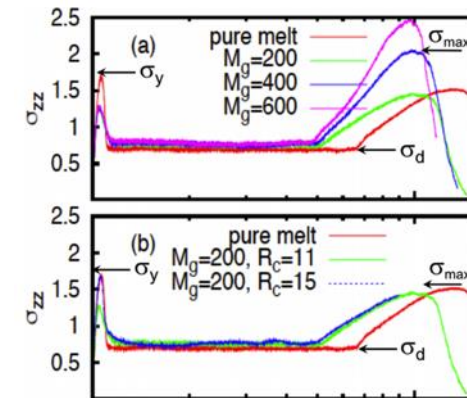
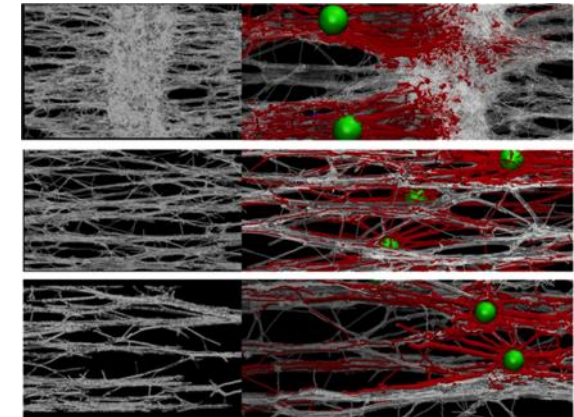
Jiao, Tibbits, Gillman, Hsiao, Buskohl, Drummy, Vaia, *Macromolecules* **2018**, *51*, 7257–7265



Hyon, Gonzales, Streit, Fried, Lawal, Jiao, Drummy, Thomas, Vaia, *ACS Nano* **2021**, *15*, 2439–2446



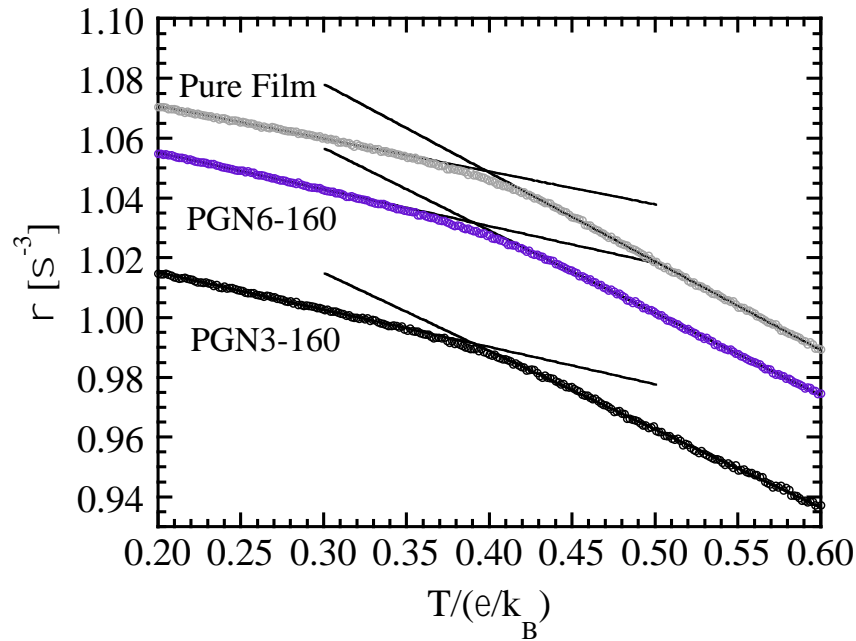
MD Simulations in Bulk PGN blends



Meng, Kumar, Ge, Robbins, Grest, *J. Chem. Phys.* **2016**, *145*, 094902

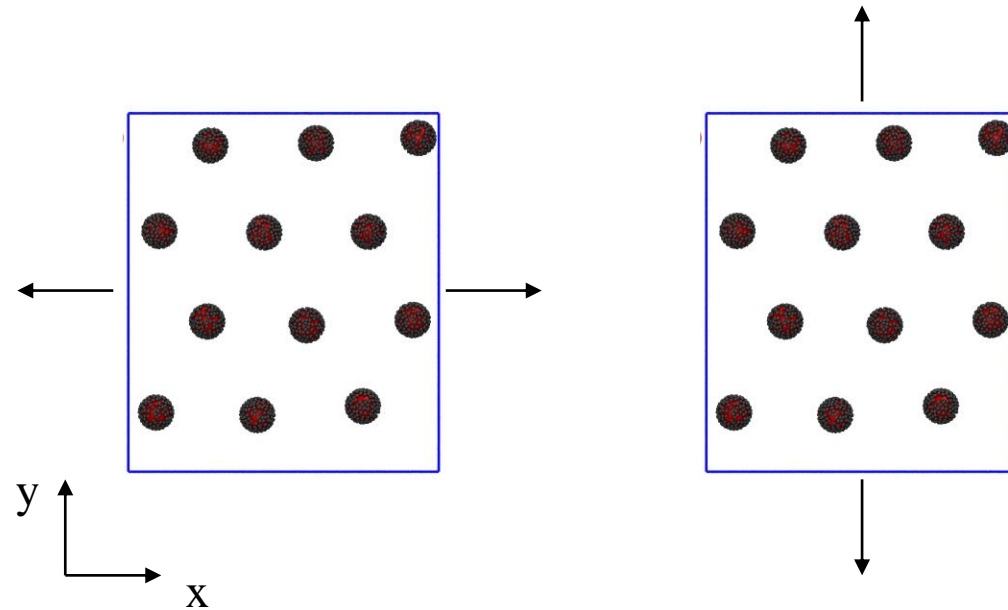
Creating Glassy Monolayer Films

- ❑ Switch to a quartic bond potential (to allow bonds to break during deformation)
- ❑ Cool monolayers to $T^* = 0.2$ ($T_g = 0.39-0.40$) and remove the surface

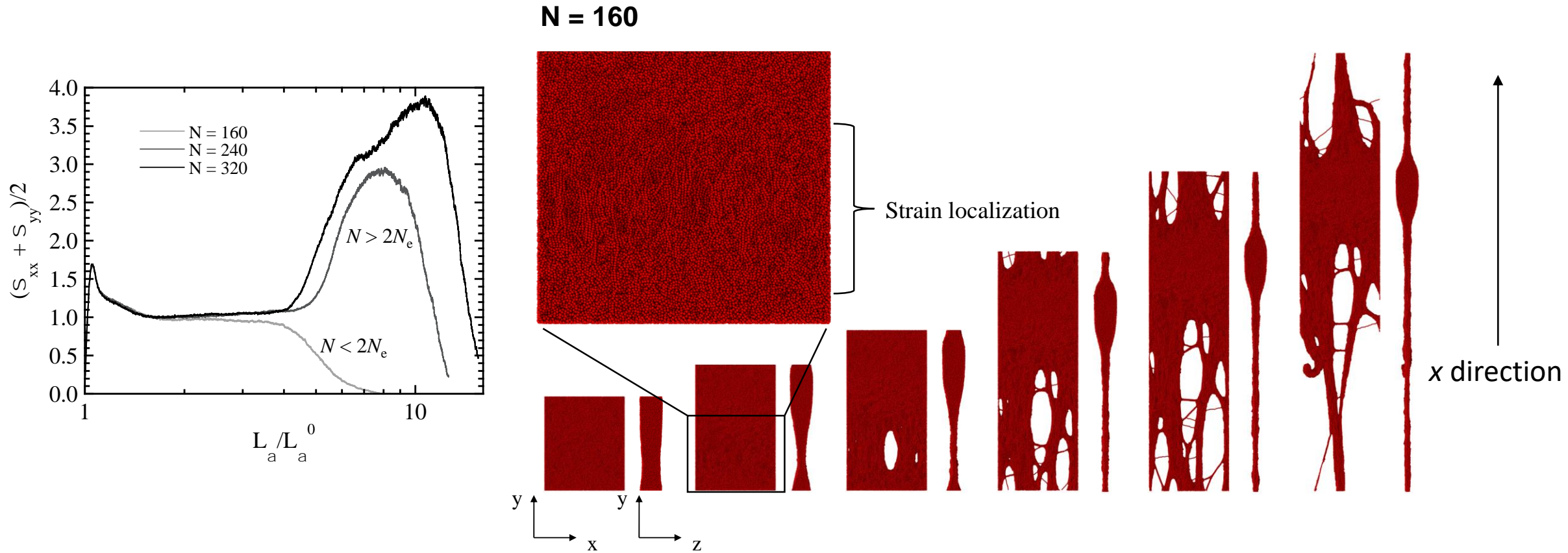


Deformation

- ❑ Deform uniaxially at constant deformation rate of $0.03 \sigma/\tau$
- ❑ Keep non-deformed side length constant
- ❑ Record stress in direction of pull

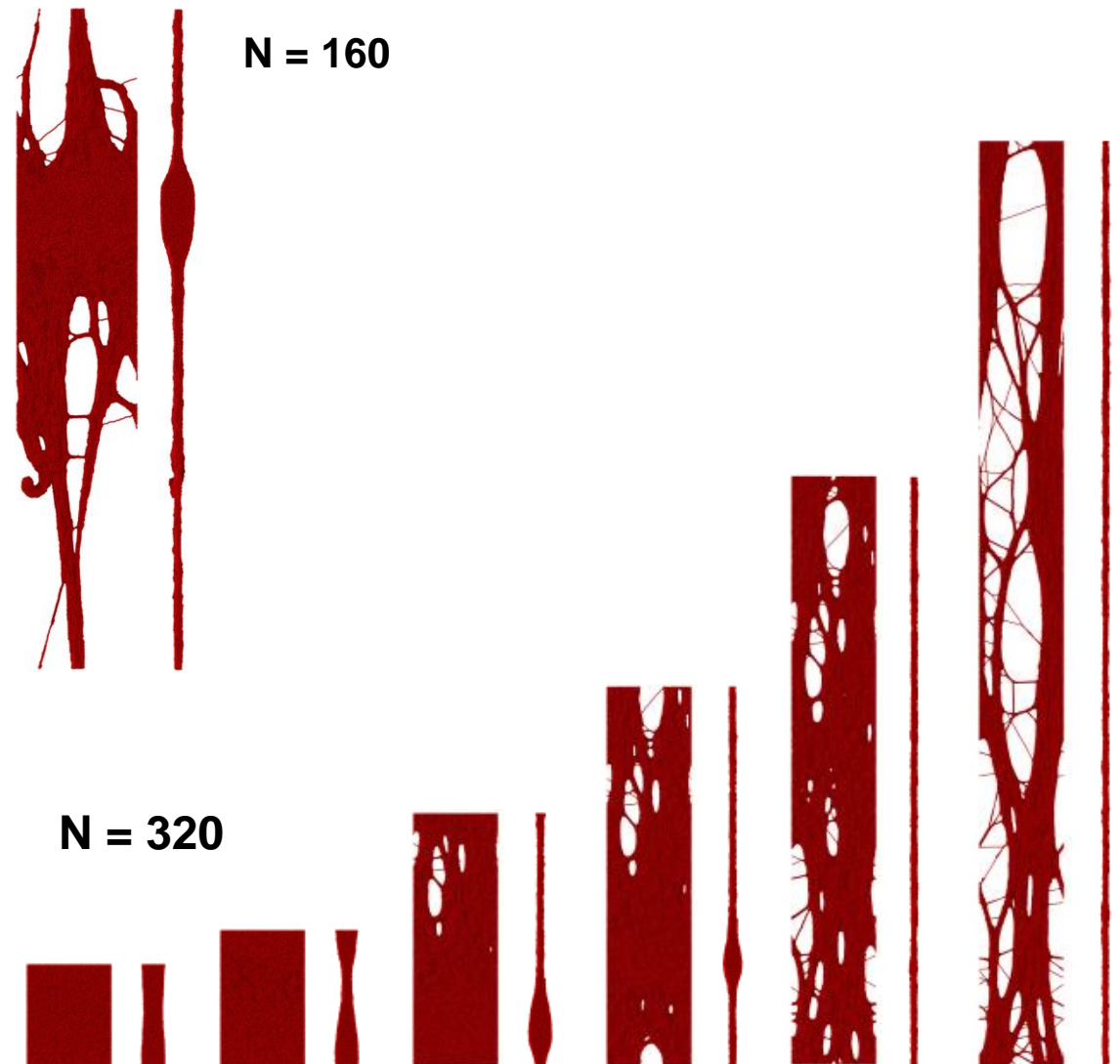
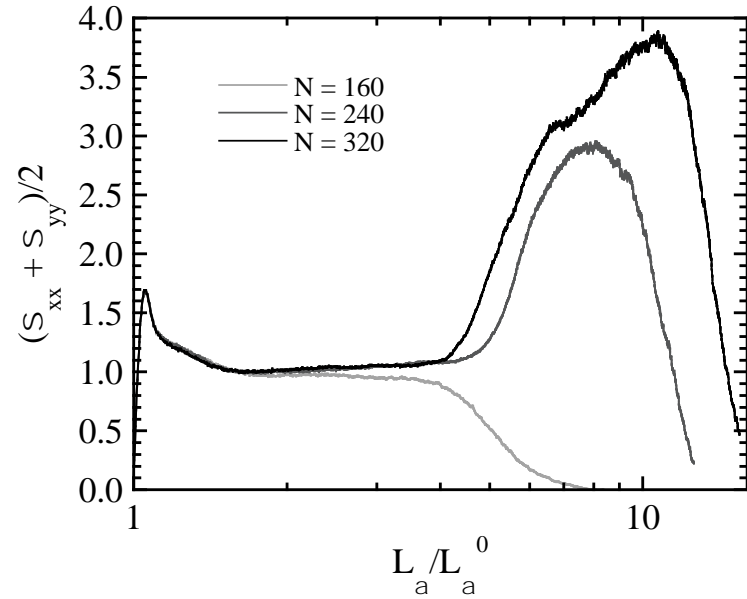


Homopolymer Films – Uniaxial Deformation



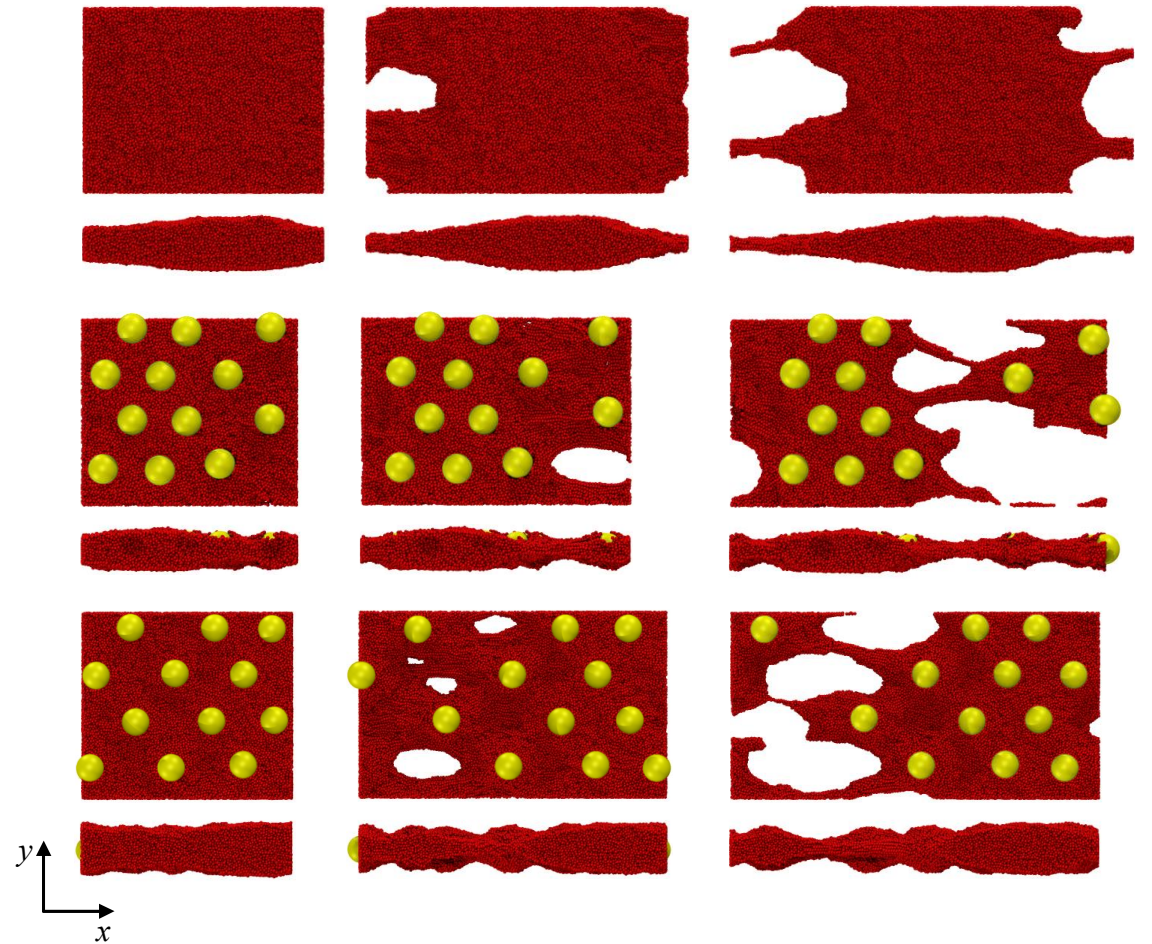
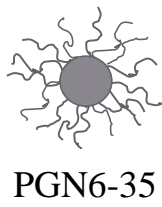
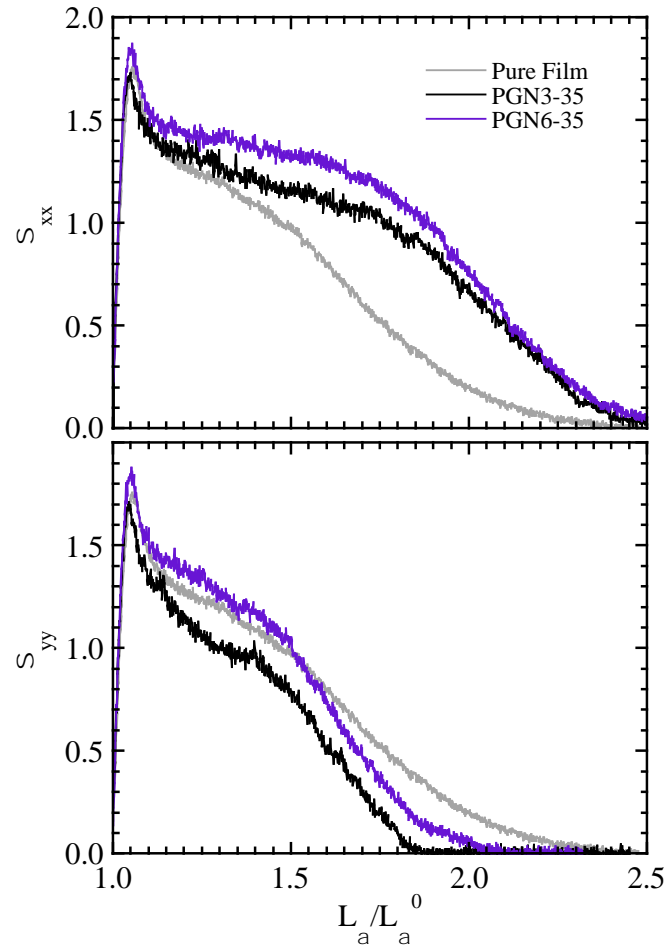
Stress-strain curves similar to expectations for bulk polymer

Homopolymer Films – Uniaxial Deformation



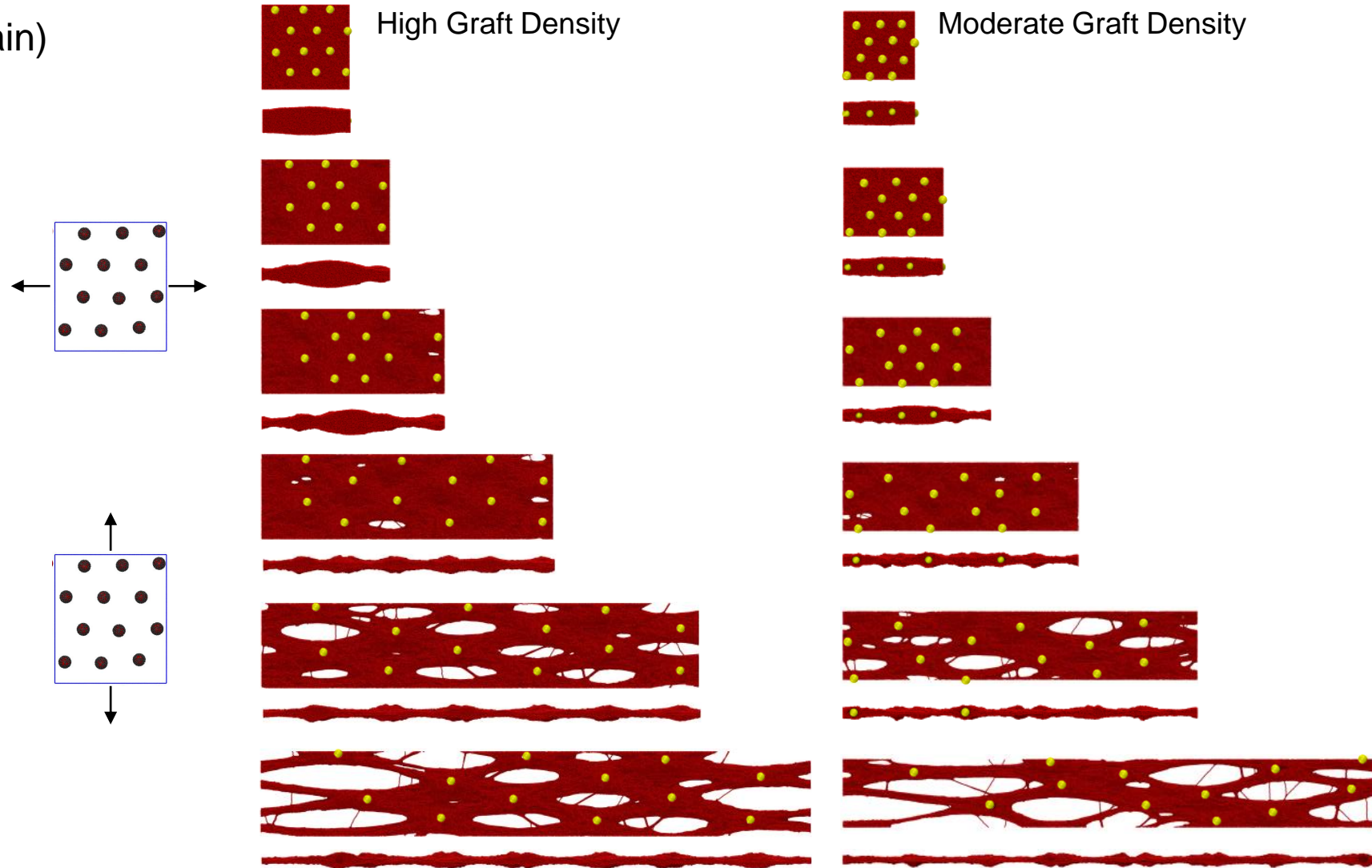
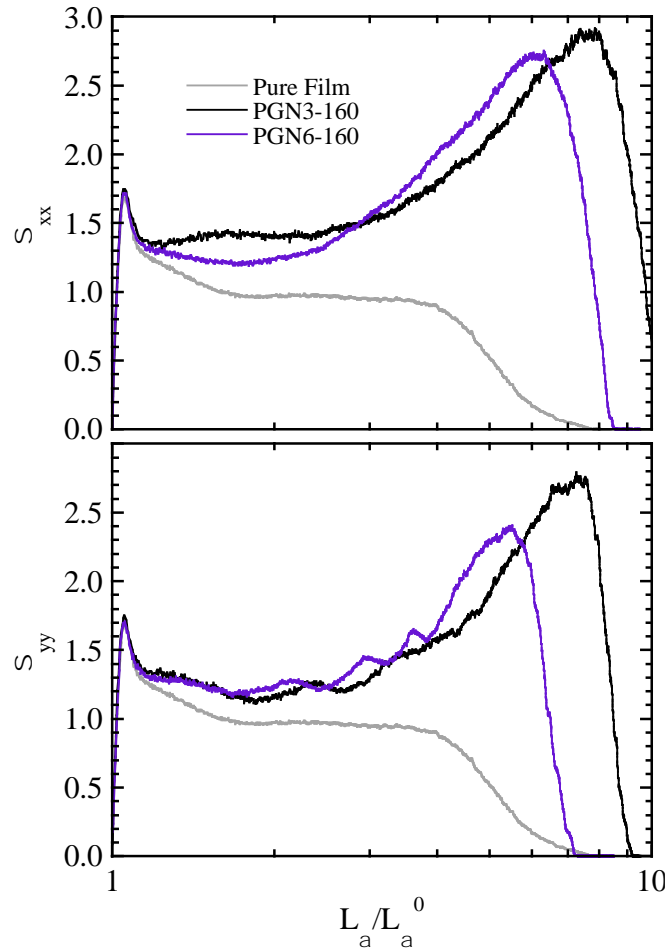
Glassy Monolayer Films – Uniaxial Deformation

Short Chains ($N = 35$ beads/chain)



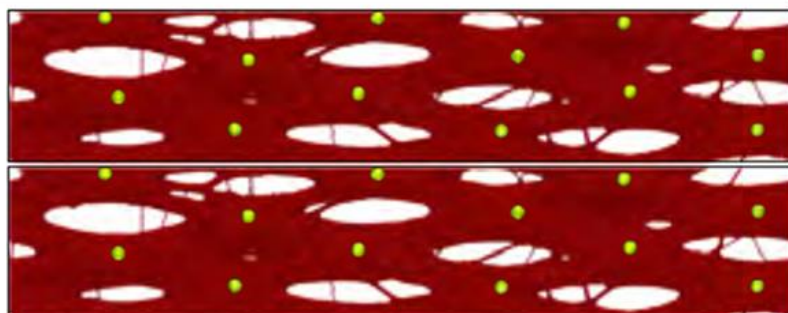
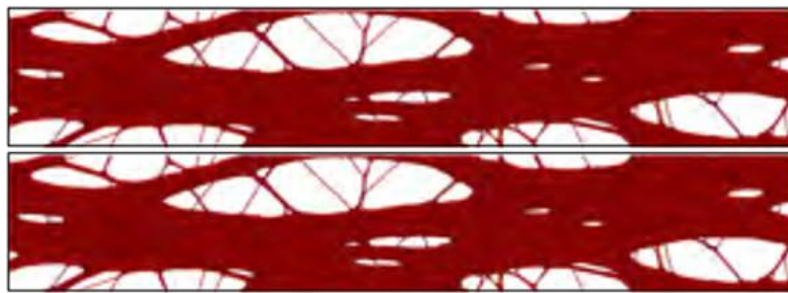
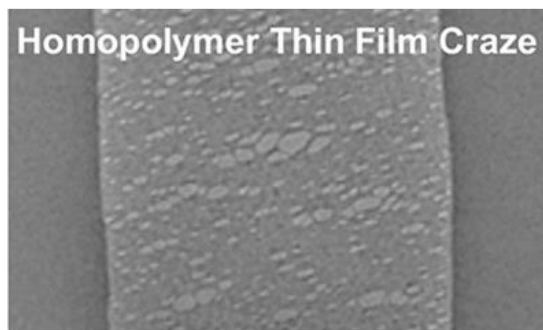
Glassy PGN Monolayer Films – Uniaxial Deformation

Long Chains ($N = 160$ beads/chain)



Comparison with Experiment

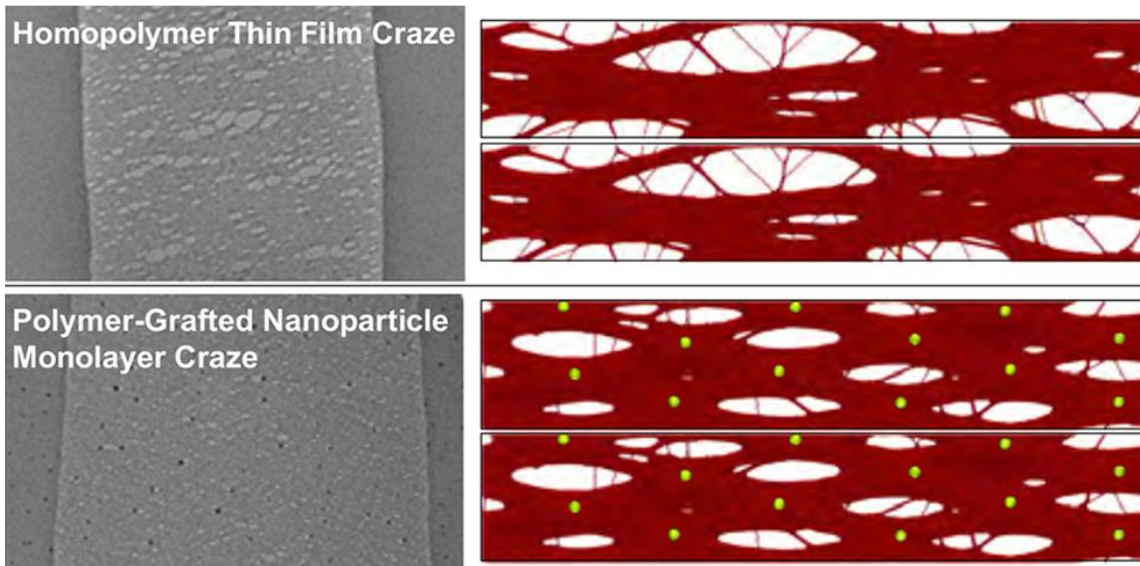
Collaboration with Drummy and Vaia at Air Force Research Lab: PGN monolayer films



- ❑ Qualitative comparison: no attempt to match chain length, particle size
- ❑ Homopolymer films have wide distribution of void sizes
- ❑ PGN films have more uniform voids
- ❑ PGN films are tougher

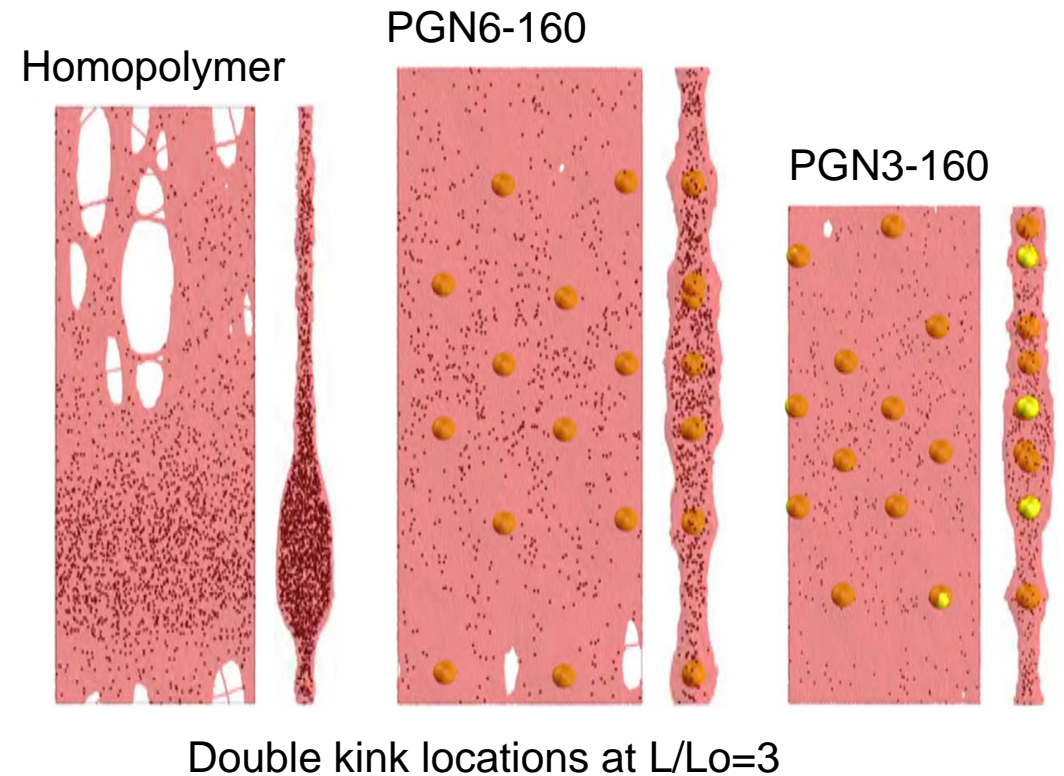
Conclusions

- ❑ Topological (Z1) and mean path entanglement analyses provide similar information; mean path method avoids fixing chain ends
- ❑ Grafting to nanoparticle adds a constraint on chain motion, as though there are additional entanglements vs. homopolymers of same length



Ongoing Work

- ❑ PGN phase behavior in solutions
- ❑ Bulk neat PGNs
- ❑ How does entanglement location change with strain?
- ❑ How can we relate entanglements to local and overall mechanical properties?

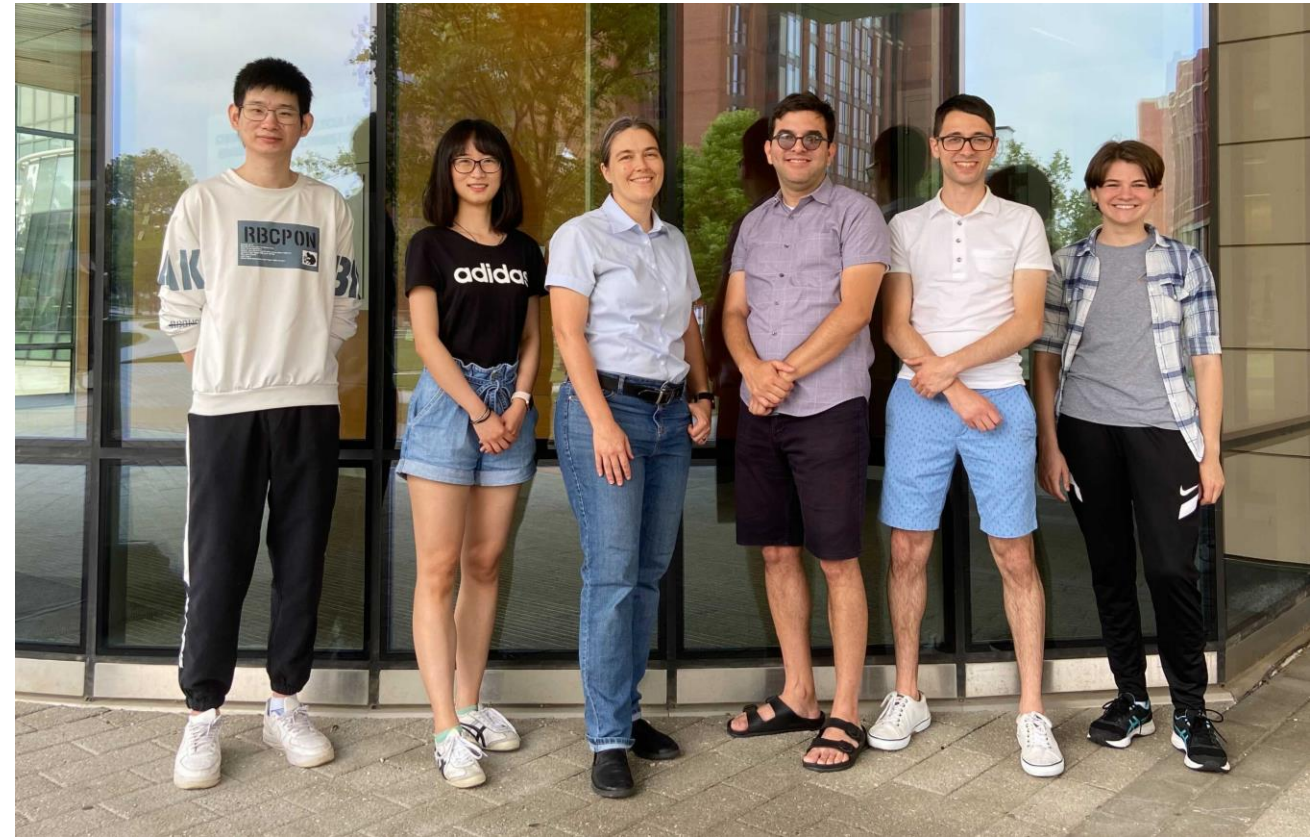


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- Jeff Ethier
- Nicholas Liesen
- Anna Schuler
- Felipe Pacci Evaristo
- Richard Vaia, AFRL (experimental)
- OSC HPC Resources
- DoD Modernization Program (additional HPC resources)

Funding

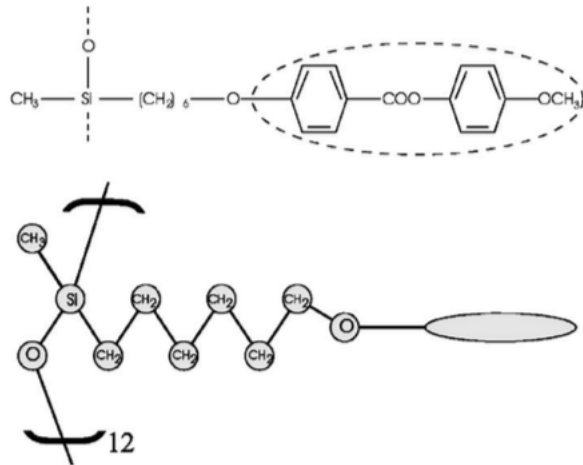
- AFRL/DAGSI Ohio Student-Faculty Research Fellowship and AFRL Minority Leaders Program
- AFOSR with Daniel Hallinan



THE OHIO STATE UNIVERSITY

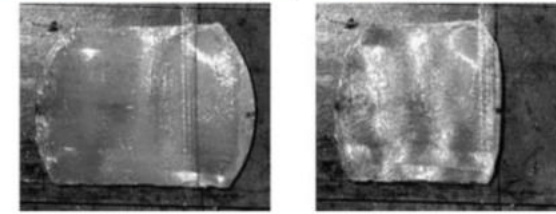
New Direction: LC Polymers

- ❑ Relatively flexible backbone with liquid crystal groups attached on side chains: either end-on or side-on
- ❑ Crosslink to make responsive materials
- ❑ Can we mix different LC attachment types to control response?

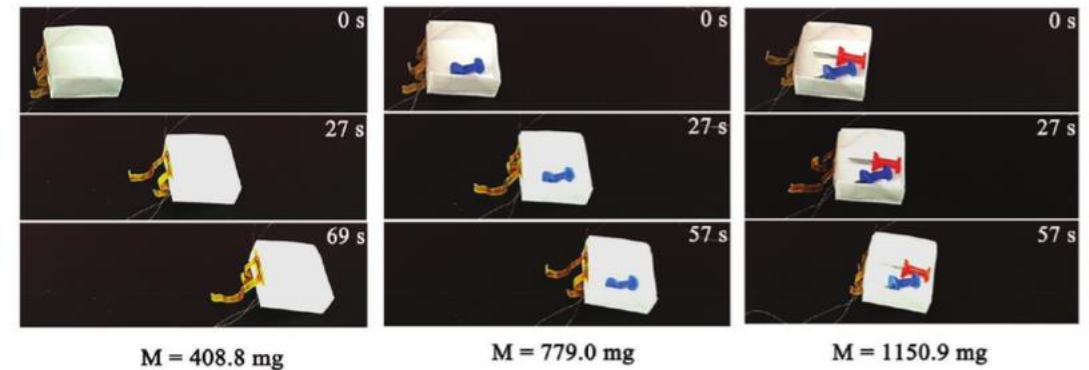
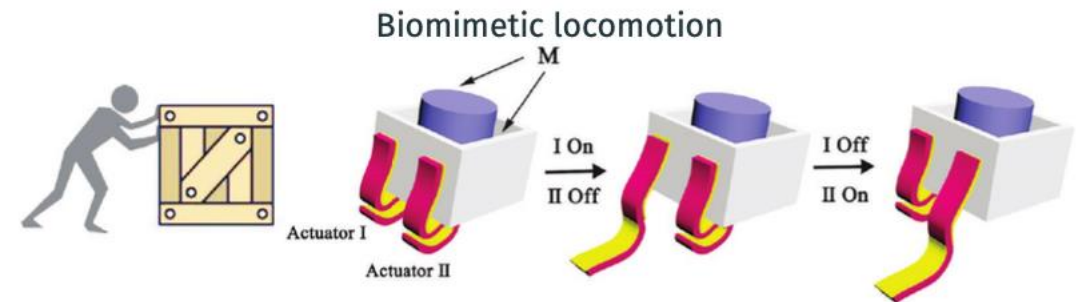


Stimson and Wilson, *J. Chem. Phys.*, 123, 2005

Artificial muscles (first proposed by de Gennes, 1997)



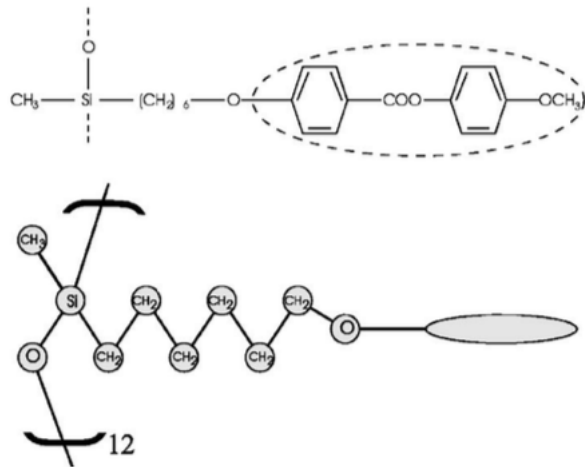
M. Li and P. Keller, *Phil. Trans. R. Soc. A*, 364, 2006



Xiao, et al., *Advanced Materials*, 31, 2019

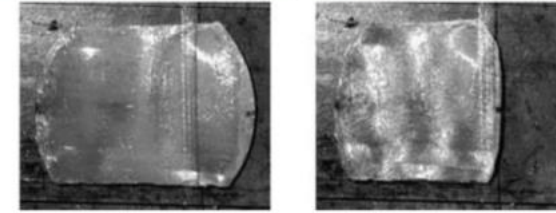
New Direction: LC Polymers

- ❑ Acknowledgements: PRF-ND award, experimental collaborator **William Wang**, OSU; postdoc **Diego Becerra**
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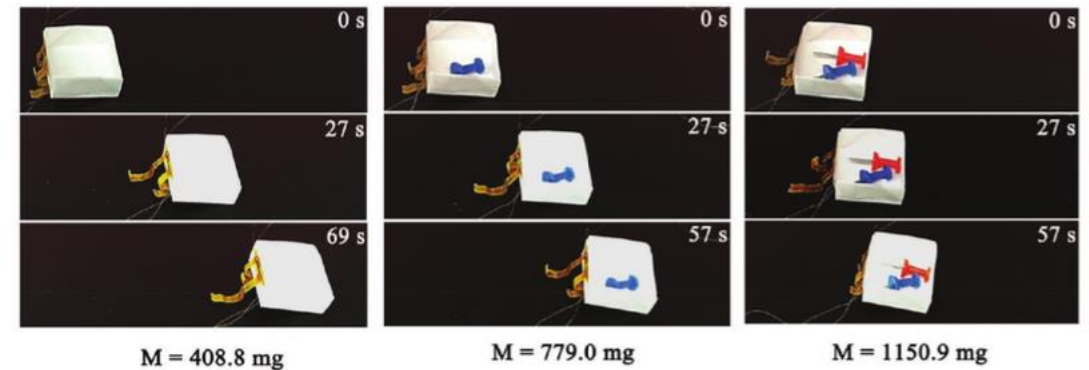
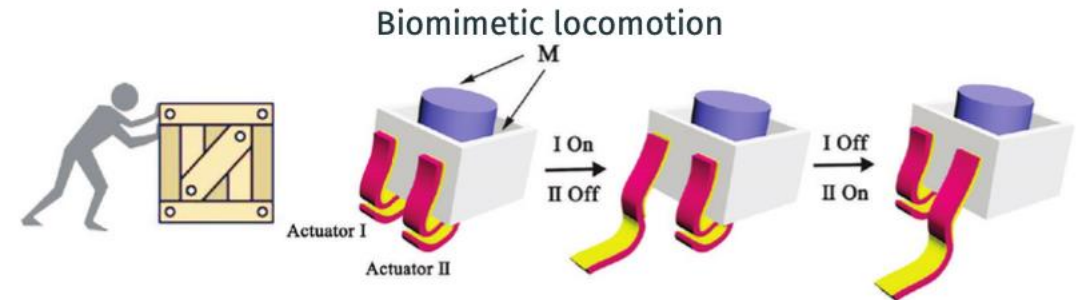


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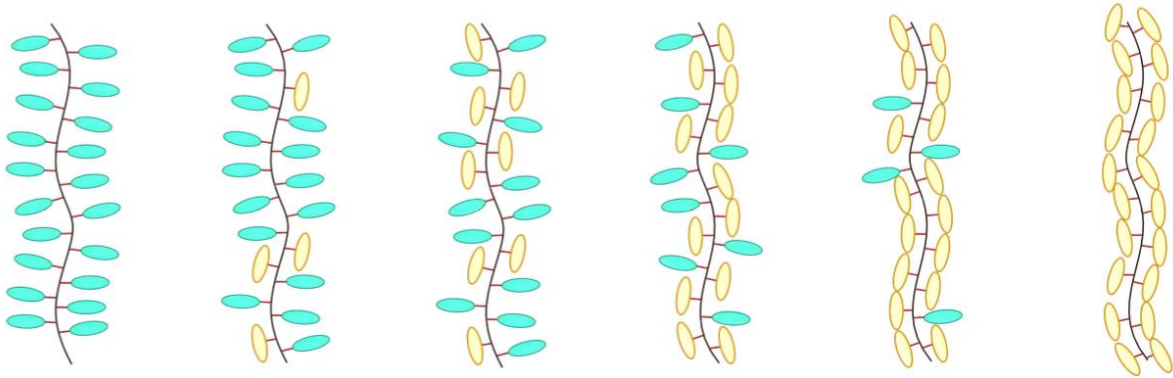
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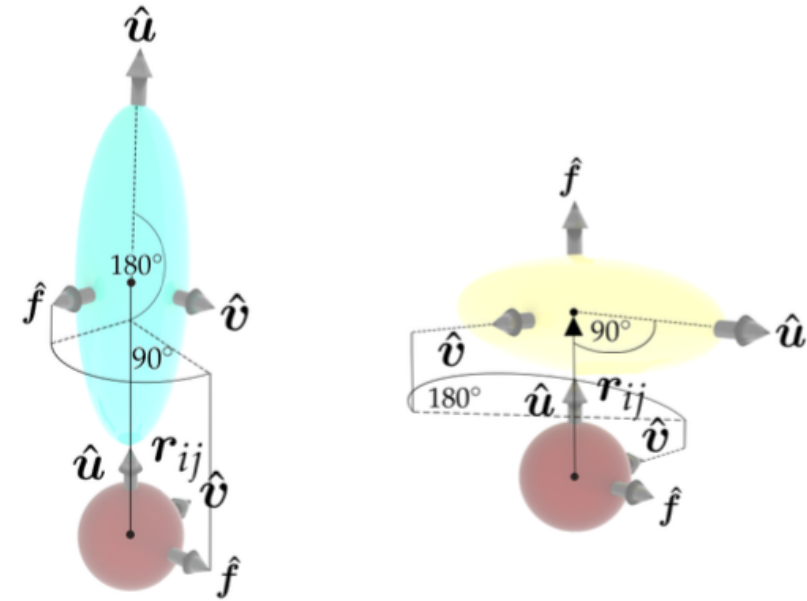
LC Polymer Model

- ❑ Include orientations/torsions on all beads
- ❑ LC groups: Gay-Berne ellipsoids
- ❑ Can we control response by mixing type of attachment?



Experimental systems: 4-(6-acryloxy-hex-1-yl-oxy) phenyl 4-(hexyloxy) benzoate end-on LC groups or 4''-acryloyloxybutyl 2,5-di(4'-butyloxybenzoyloxy) benzoate side-on groups on polydimethylsiloxane (PDMS) backbones.

Bonded interactions include torsional interactions that constrain both the relative position and orientation of bonded sites.



The **nonbonded interactions** between two sites depends on the site types i and j and distance between them, $\mathbf{r}_{ij} = \mathbf{r}_i - \mathbf{r}_j$.

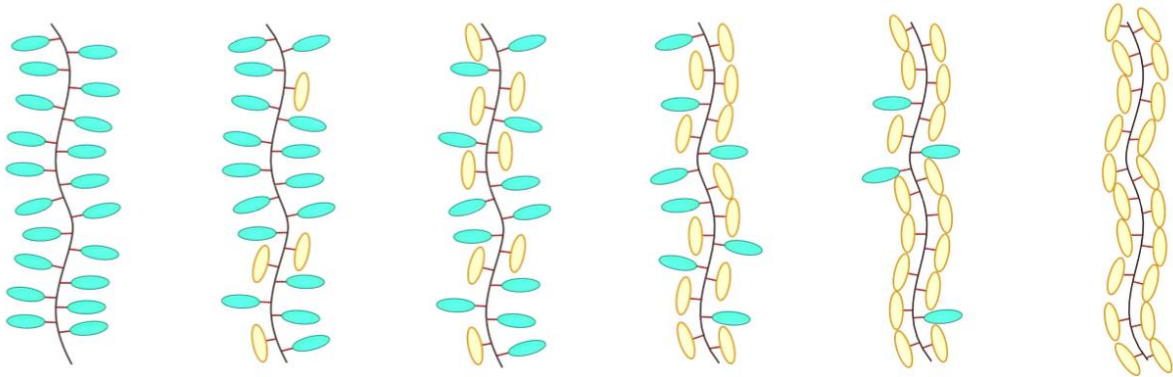
$U_{\text{LJ}}(\mathbf{r}_{ij})$	if $i \in \text{Polymer}, j \in \text{Polymer}$
$U_{\text{GB}}(\hat{\mathbf{u}}_i, \hat{\mathbf{u}}_j, \mathbf{r}_{ij})$	if $i \in \text{LC group}, j \in \text{LC group}$
$U_{\text{GB/LJ}}(\hat{\mathbf{u}}_i, \mathbf{r}_{ij})$	if $i \in \text{LC group}, j \in \text{Polymer}$
$U_{\text{GB/LJ}}(\hat{\mathbf{u}}_j, \mathbf{r}_{ij})$	if $i \in \text{Polymer}, j \in \text{LC group}$

LJ ~ Lennard-Jones potential; GB ~ Gay-Berne potential;

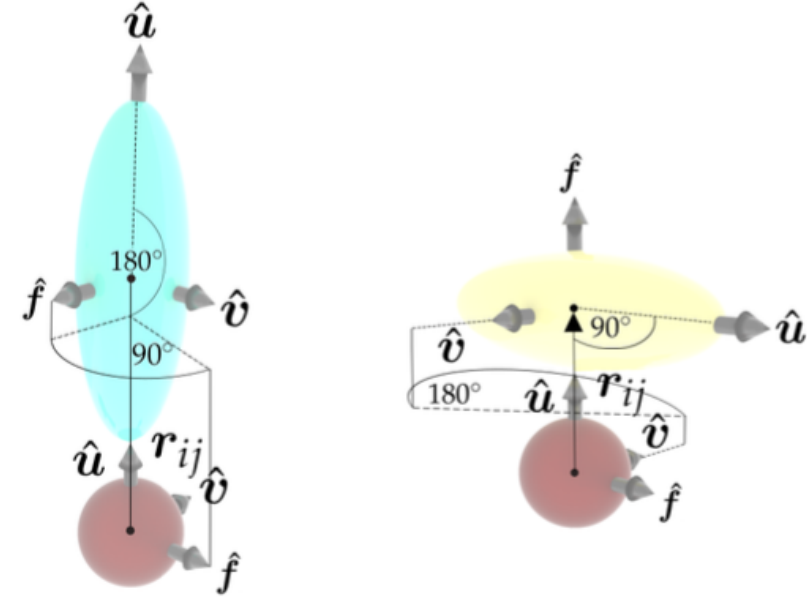
$\hat{\mathbf{u}}_i$ and $\hat{\mathbf{u}}_j$ ~ orientations of the main axis of the mesogenic groups of sites i and j .

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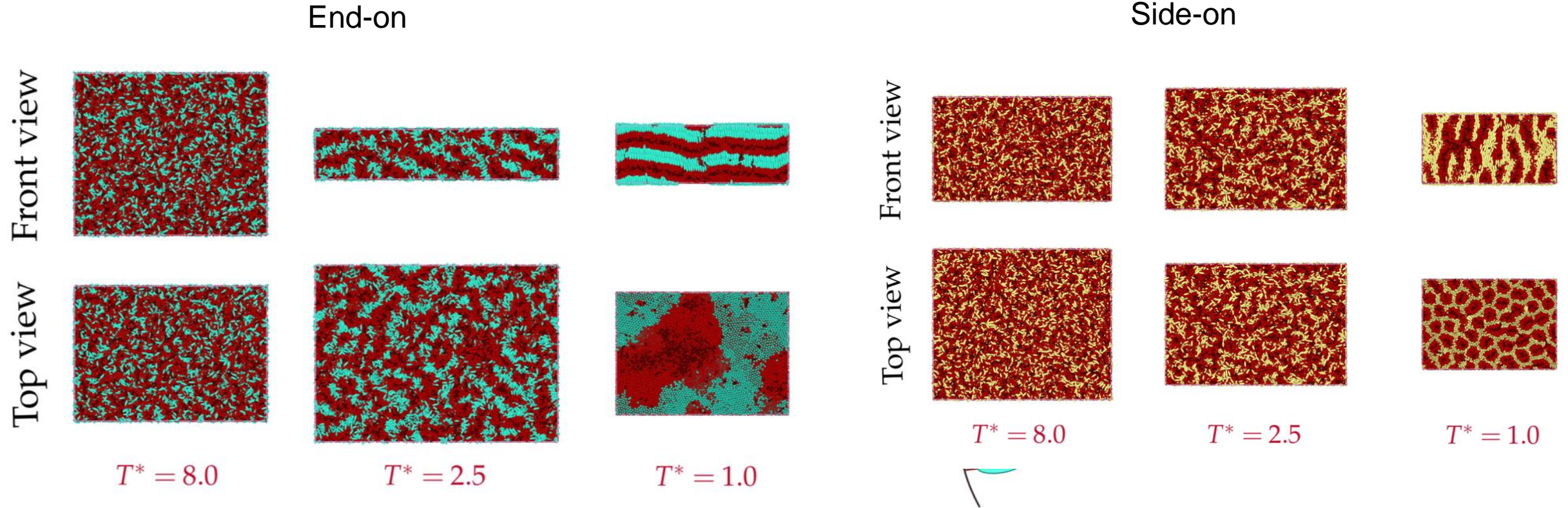
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LJ ~ Lennard-Jones potential; GB ~ Gay-Berne potential;

$\hat{\mathbf{u}}_i$ and $\hat{\mathbf{u}}_j$ ~ orientations of the main axis of the mesogenic groups of sites i and j .

Mesoscale structure depends on attachment type

Equilibrate at high T , cool while intermittently imposing alignment field to avoid multiple grains

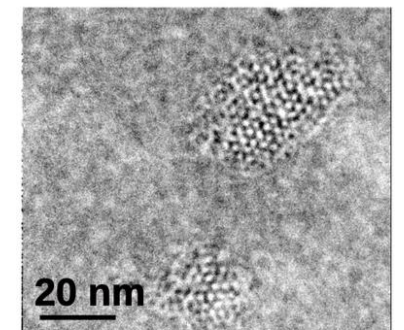
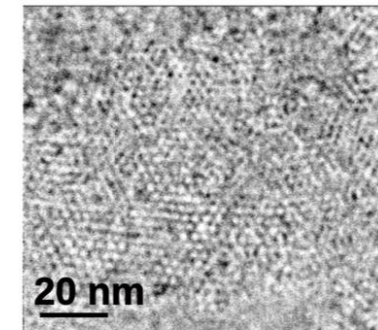
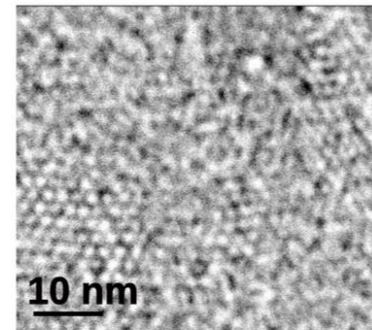
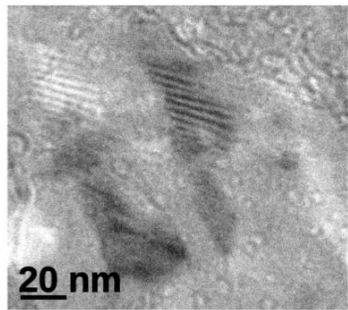
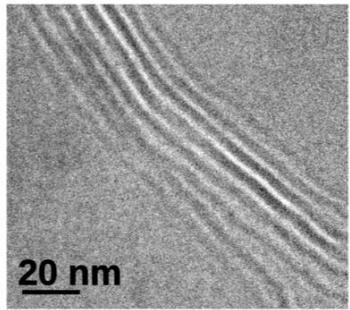


Mesoscale structure depends on attachment type

Qualitatively matches experiment

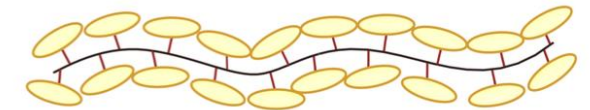
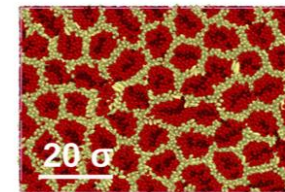
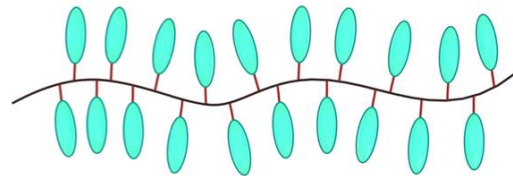
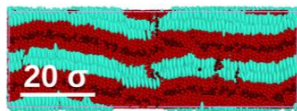
End-on

Side-on



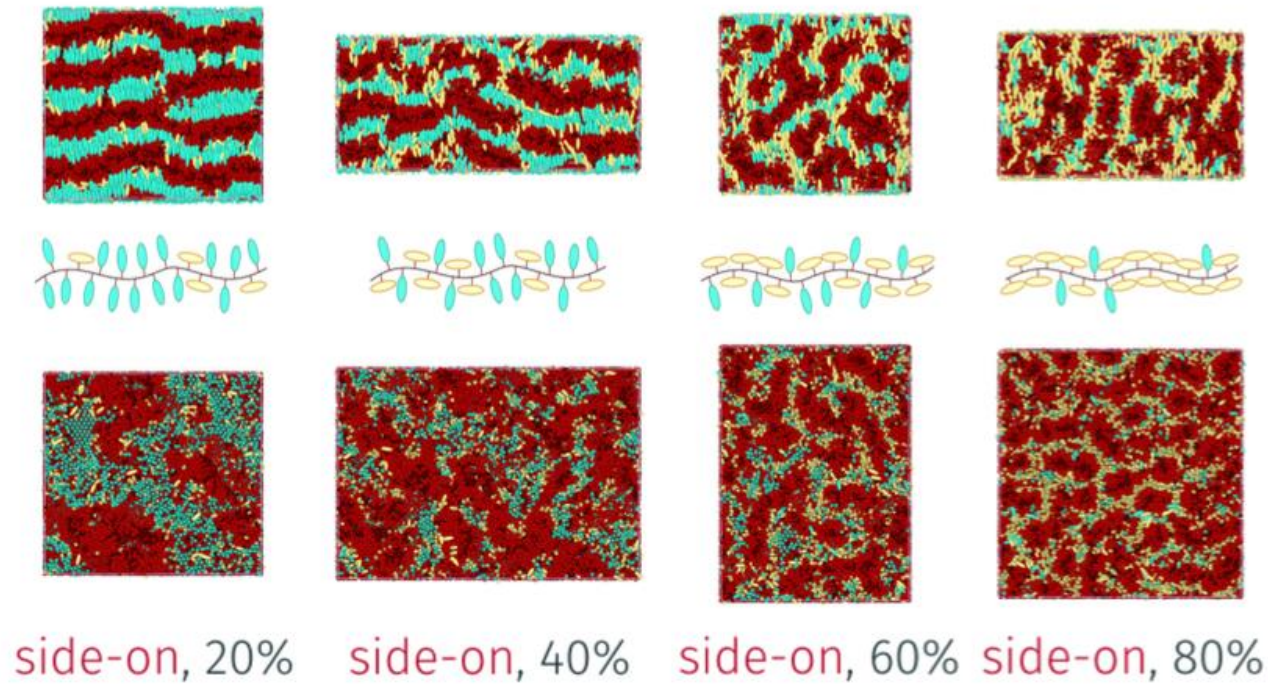
Front view

Top view

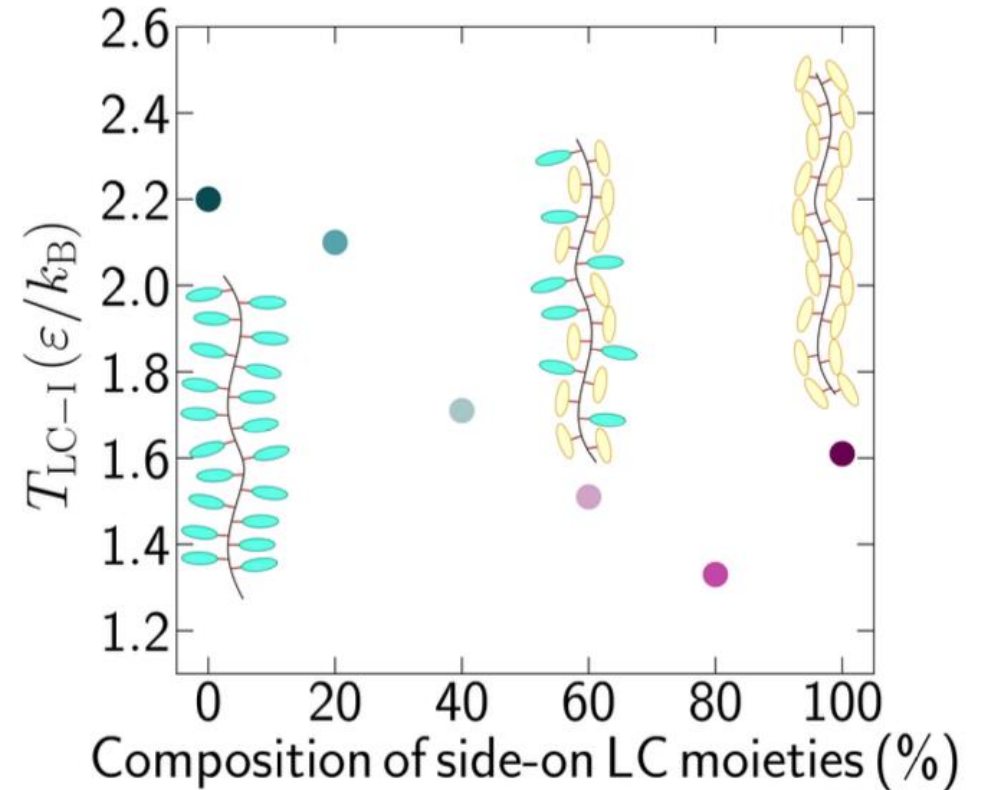


Experimental TEMs at room temperature courtesy of William Wang: 4-(6-acryloxy-hex-1-yl-oxy) phenyl 4-(hexyloxy) benzoate end-on LC groups or 4'-acryloyloxybutyl 2,5-di(4'-butyloxybenzoyloxy) benzoate side-on groups on polydimethylsiloxane (PDMS) backbones. Benzene-rich groups stained (LC groups dark in images).

Mixing attachment types disrupts ordering



Transition temperature matches experimental trend



Mixing attachment types changes conformations

