

Molecule formation in turbulent gas

Simon Glover
(AIP & KITP)

Key questions

- How quickly do molecular clouds form?
- How is the molecular gas distributed?
- What role does supersonic turbulence play?

Simulations

- ZEUS-MP v1 - MHD, self-gravity (periodic BCs)
- Detailed atomic cooling function
- Simple hydrogen chemistry
- Highly approximate treatment of shielding

Chemistry

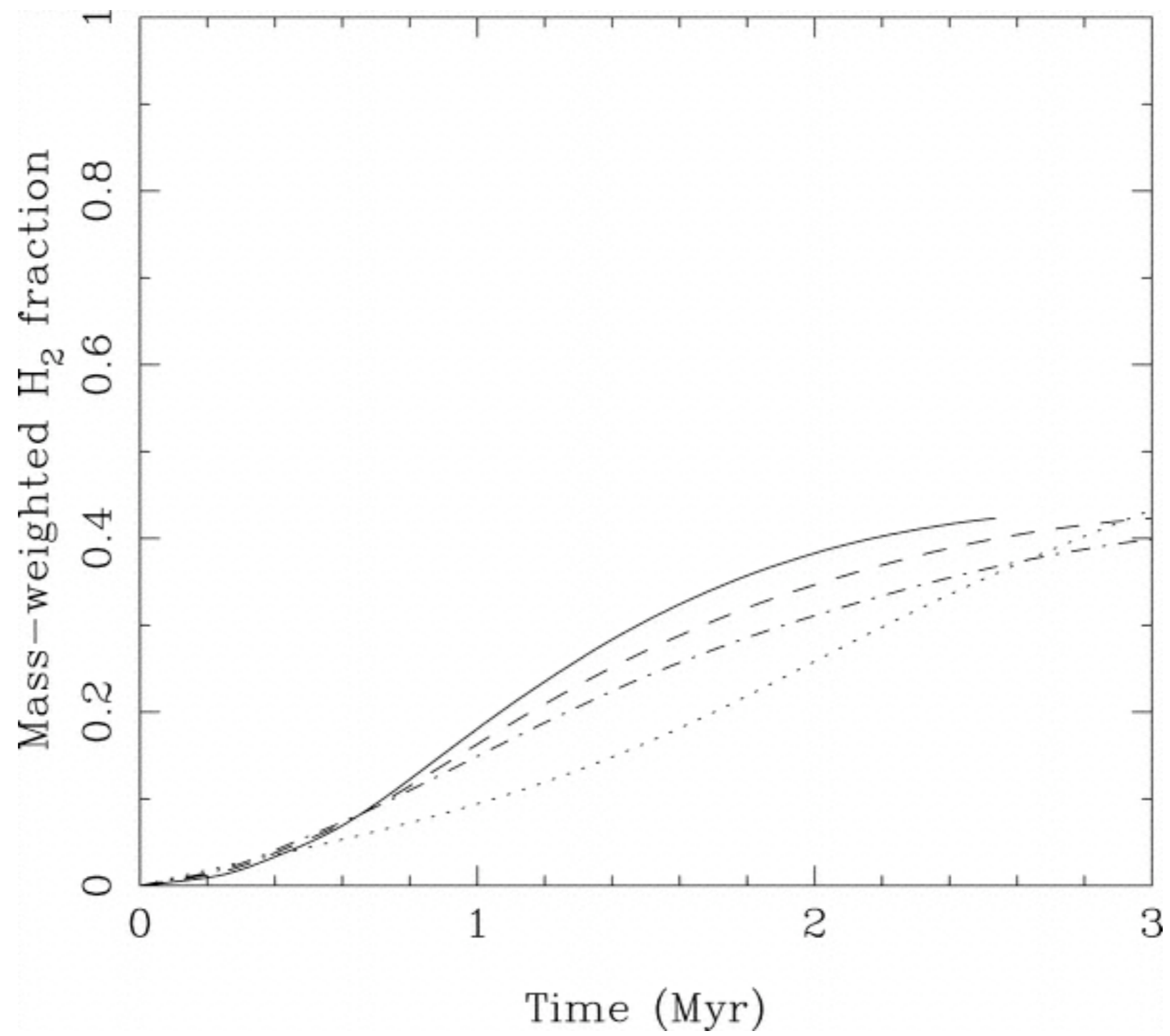
- H₂ formation on dust grains
- H₂ collisional dissociation
- H₂ photodissociation
- H ionization (collisional, cosmic rays)
- H⁺ recombination (gas-phase, grains)

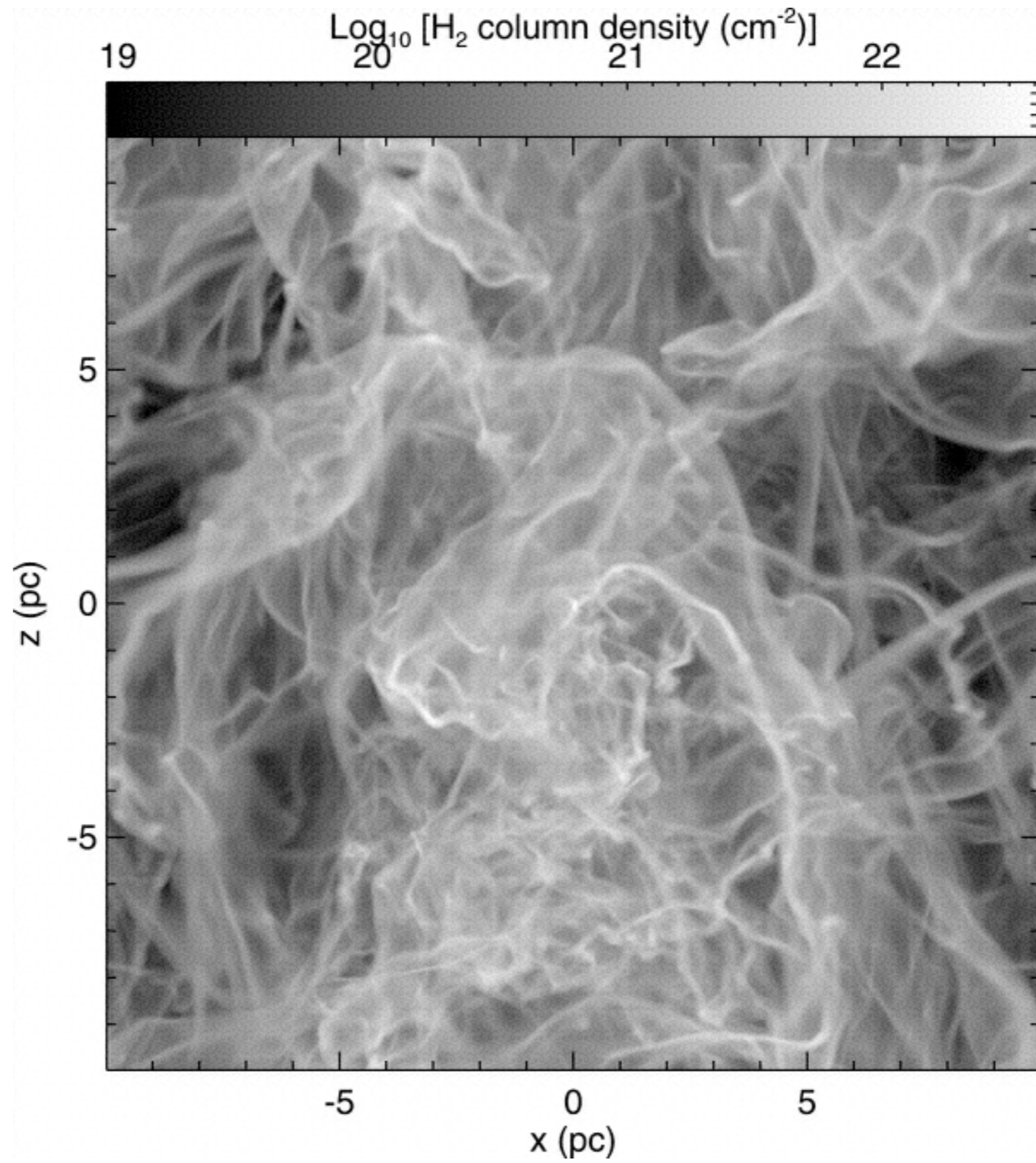
Shielding

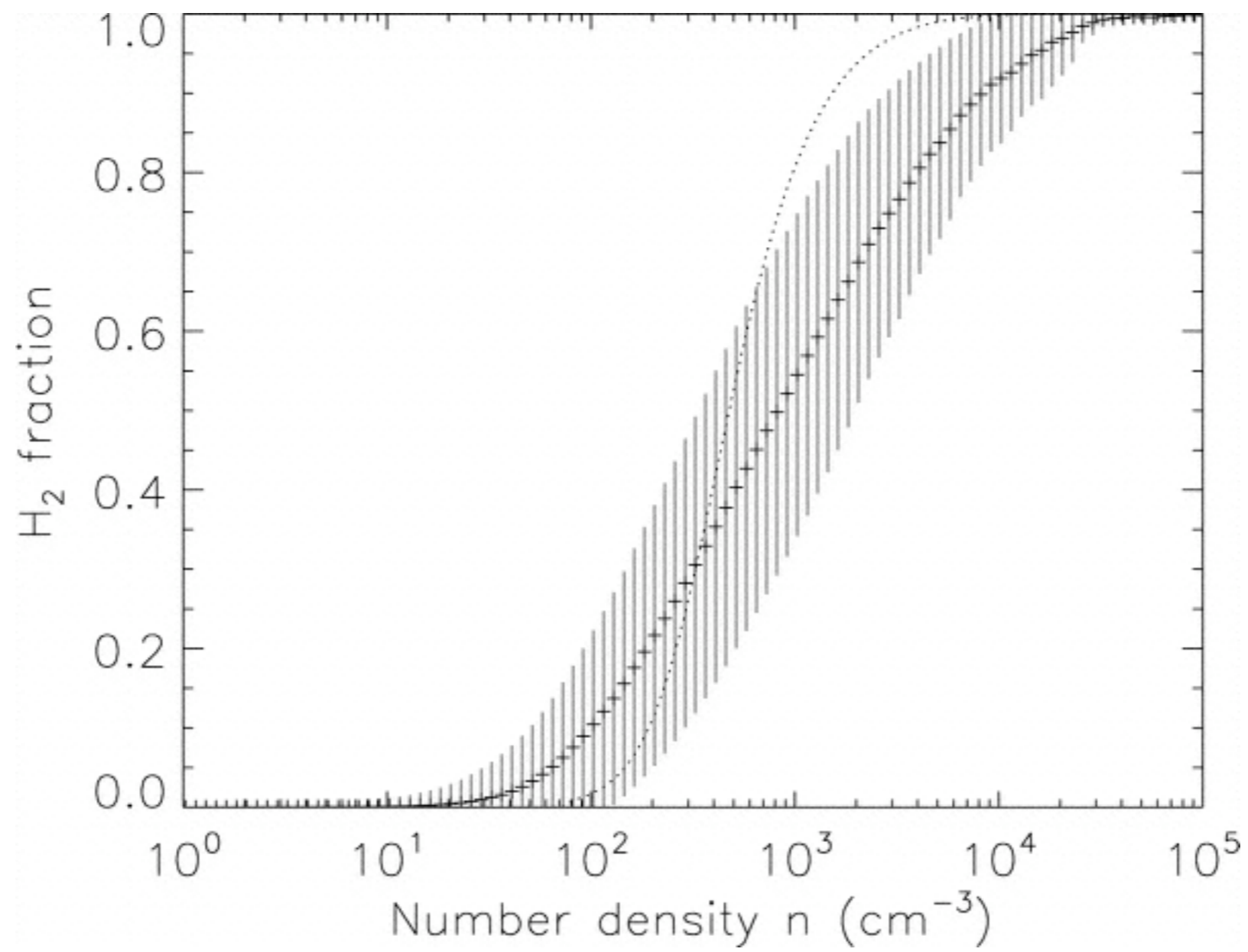
- Two forms of shielding: H₂ self-shielding, dust
- Accurate treatment computationally prohibitive, so use very simple approximations
- Local approximation: only gas within the grid cell contributes
- Six-ray approximation: ray-tracing along coordinate axes of simulation volume

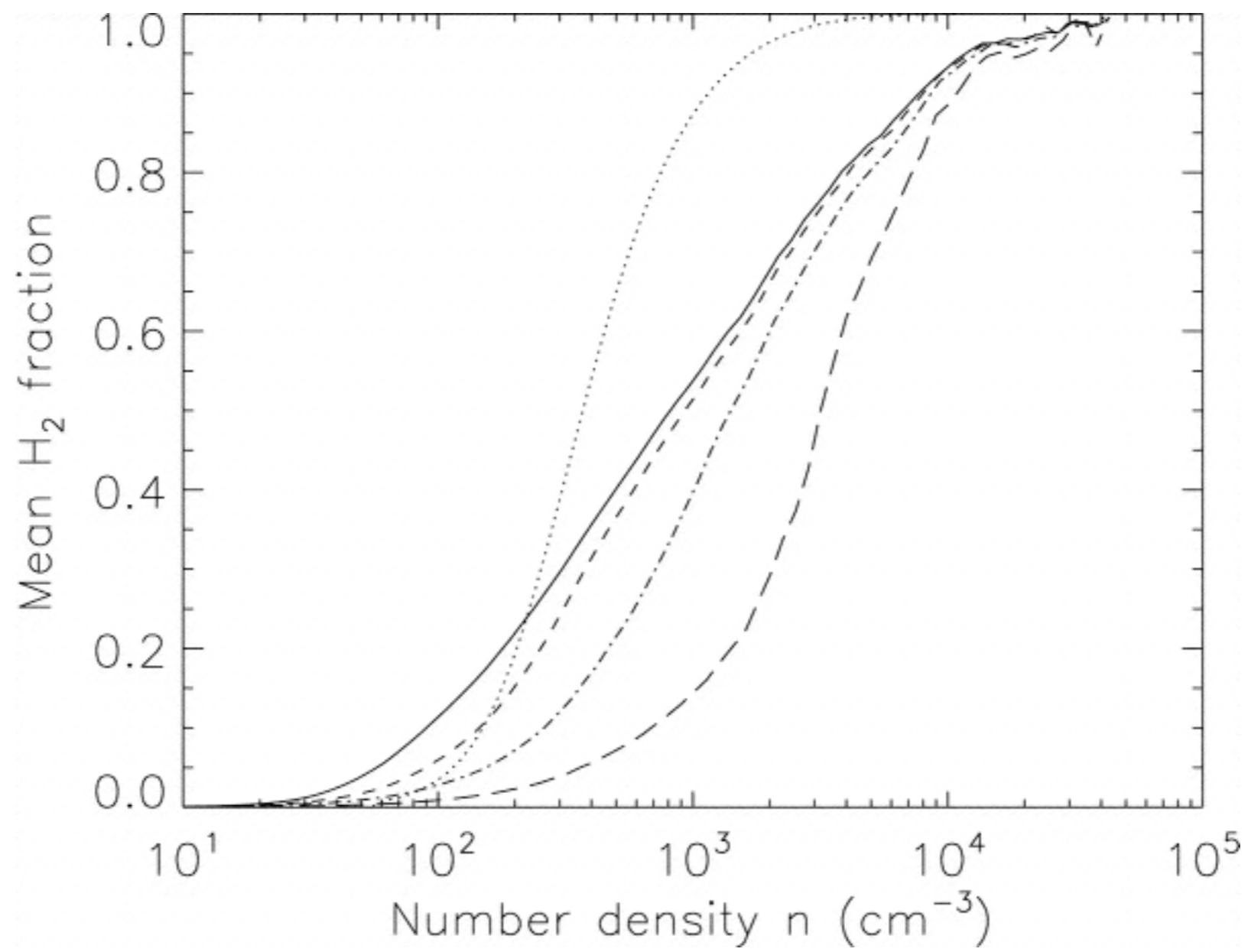
Simulation details

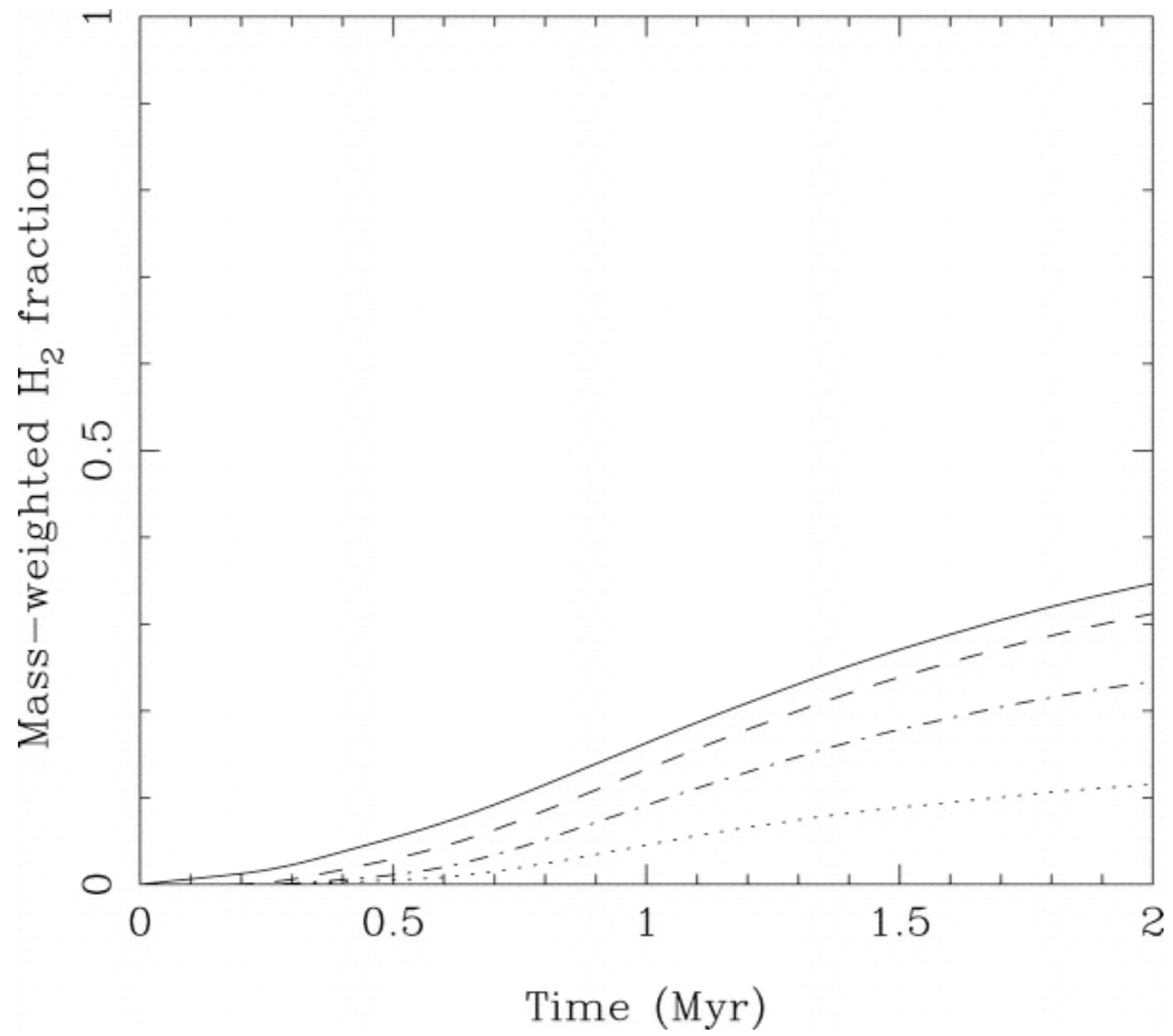
- Resolution: 64^3 , 128^3 , **256^3** , 512^3
- Box size: **20**, 40, 60 pc
- Density: 10, 30, **100** cm^{-3}
- Velocity: 1.0, 2.5, 5.0, **10.0** kms^{-1}
- Magnetic field: 0.0, **5.85**, 11.7, 23.4, 58.5 μG

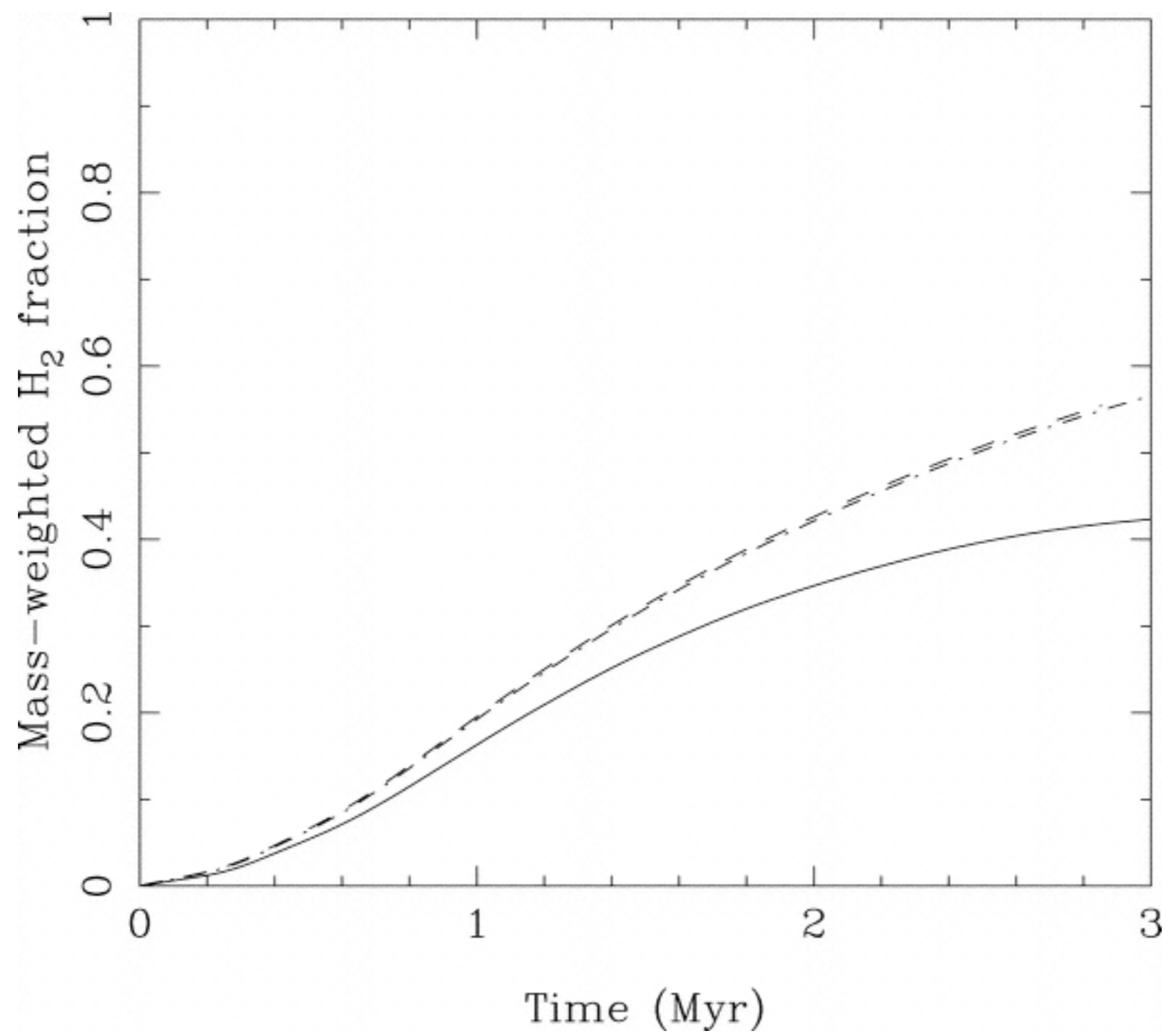


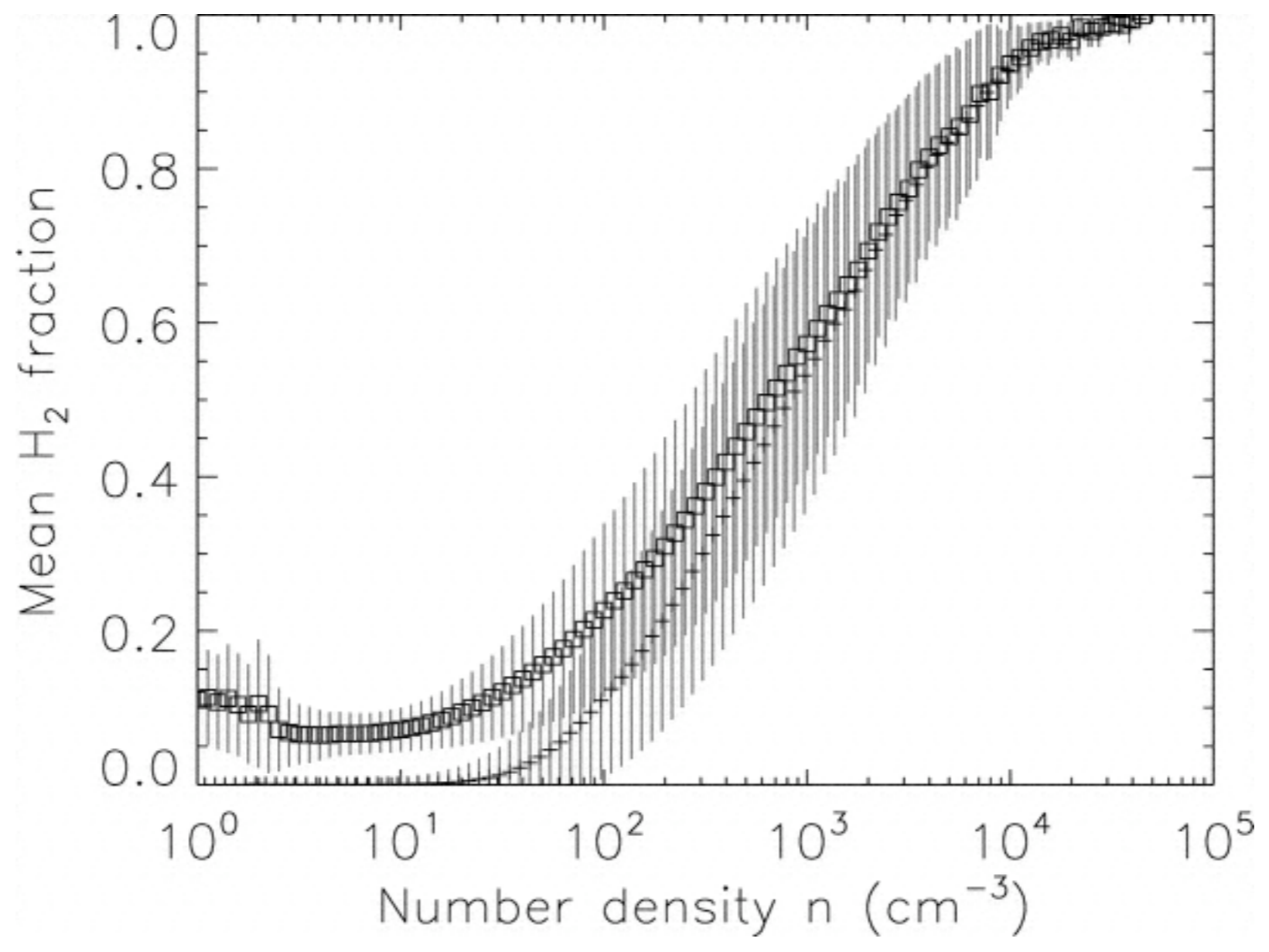


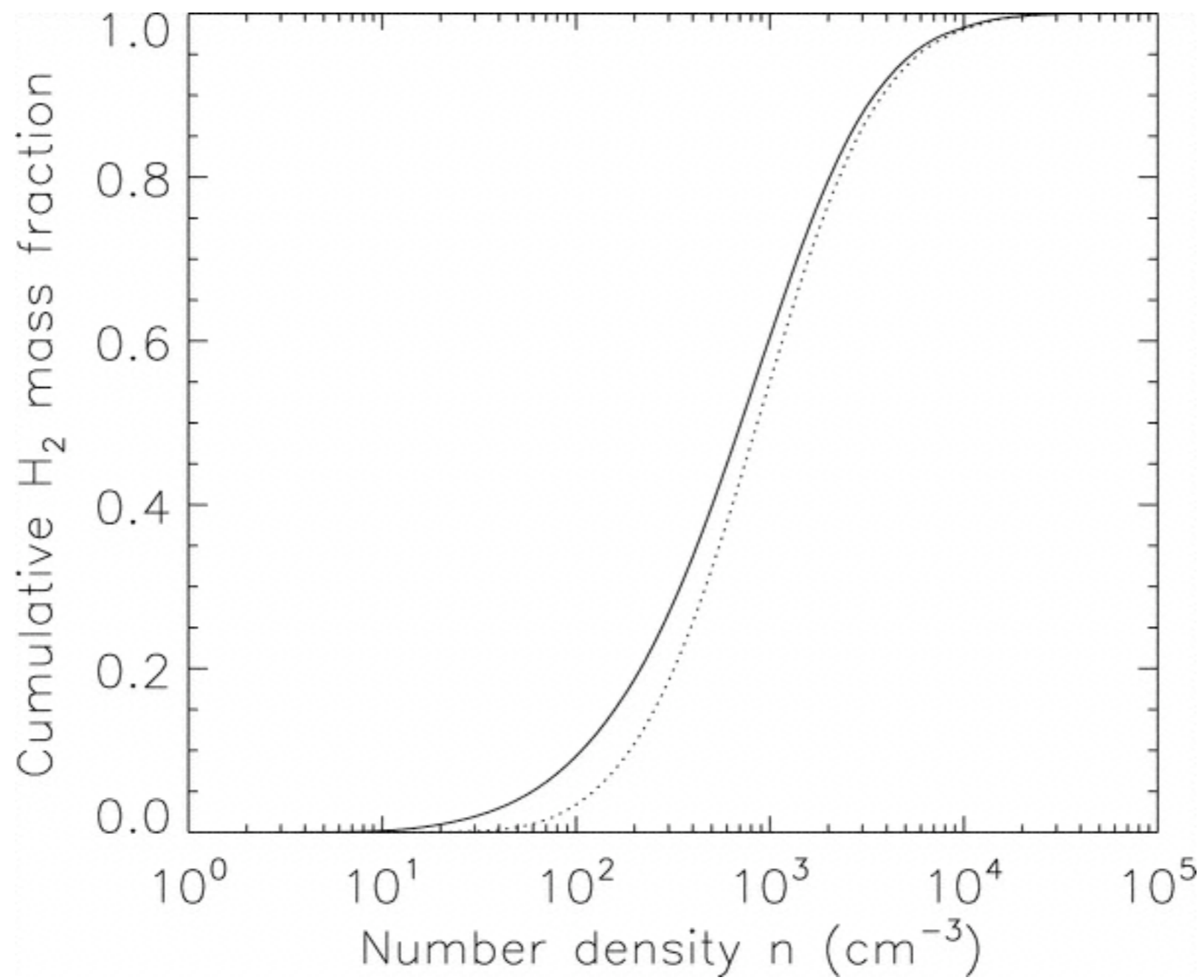


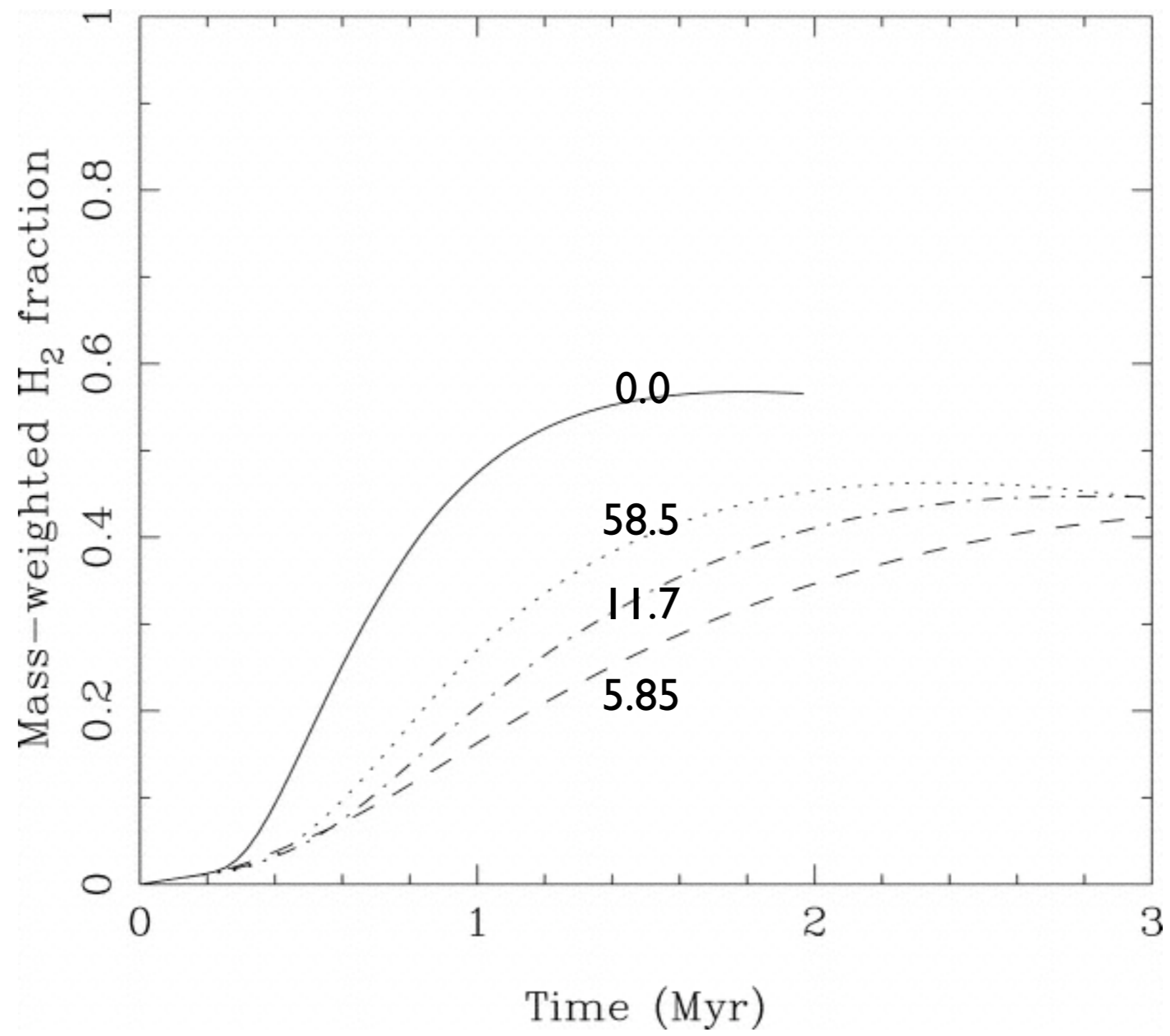


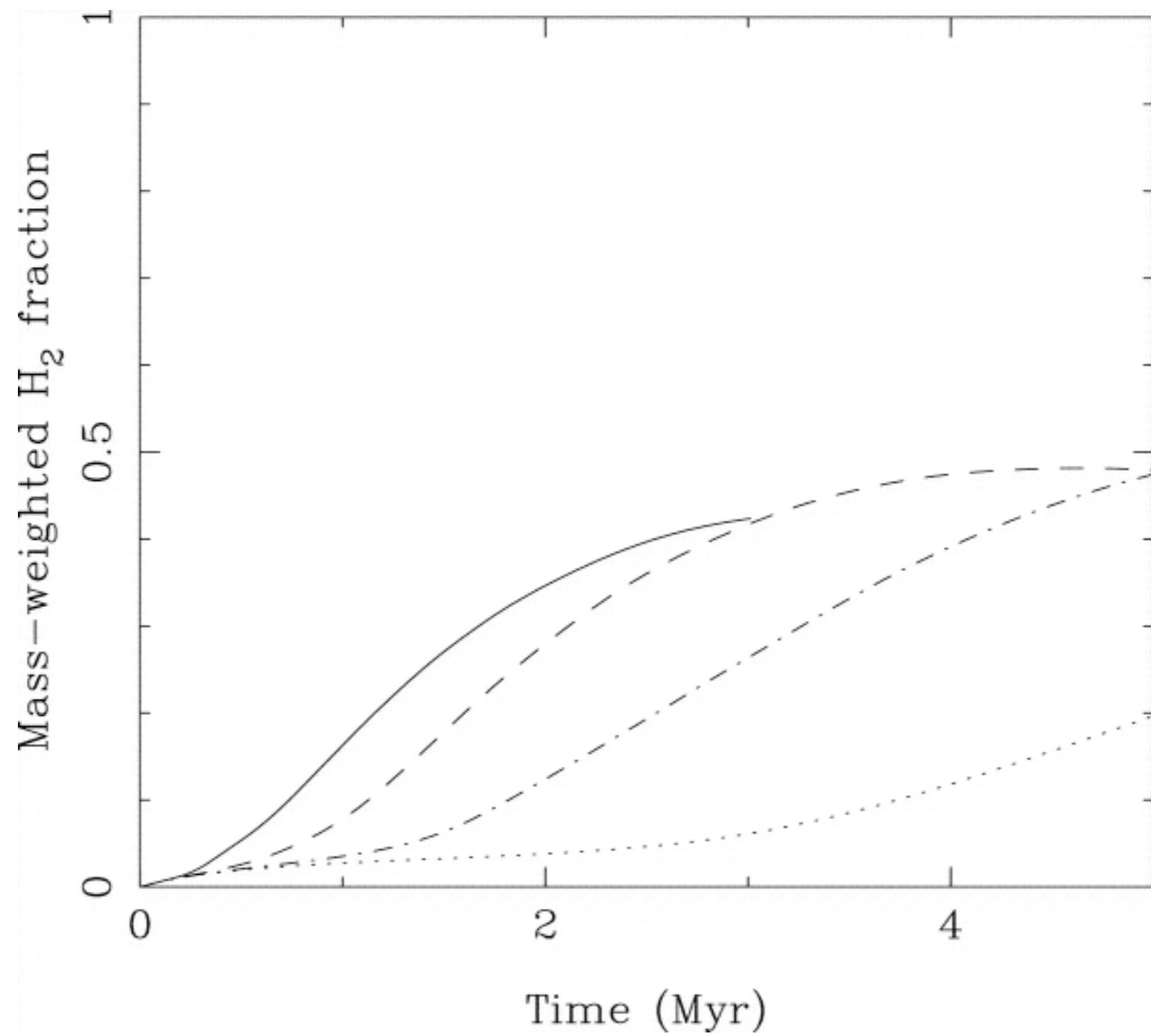


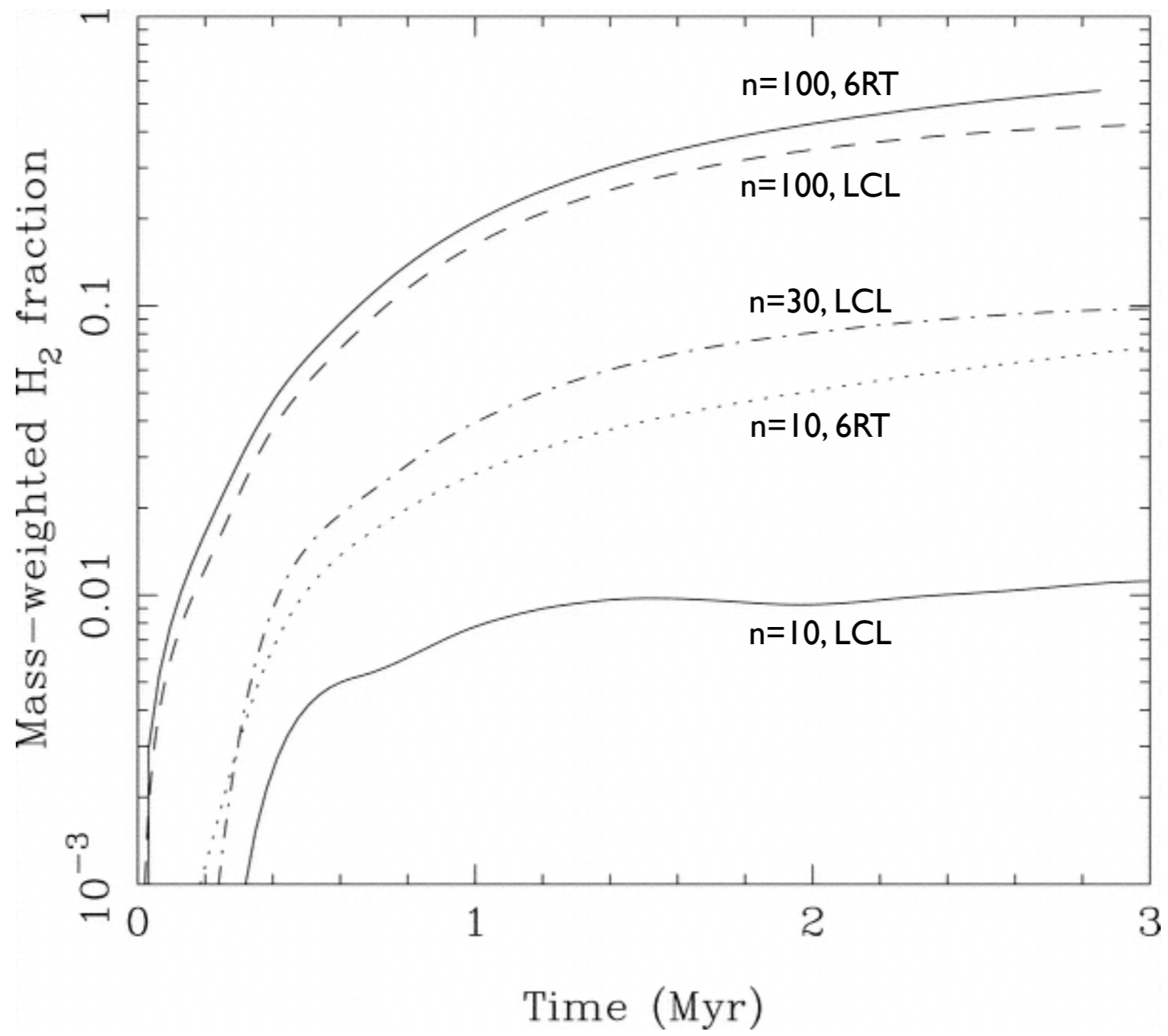












CO formation

- Same basic approach as H₂ formation models
- BUT: chemical network is far more complex...
- 18 advected species, 13 equilibrium species, ~ 300 reactions (not including grain surface chemistry)

