

# Suspensions of non-Brownian particles in a shear thickening matrix

**Sarah Hormozi**



## Ph.D. Student:



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Ohio University, Athens, Ohio

- Shear thickening: Transition from CST to DST In prep for Phys. Rev. Letter. 2018
- Enhancing shear thickening, Phys. Rev. Fluids 2, 033301.



**NSF**

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ANR-11-IDEX-0001-02

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**Elizabeth Guazzelli**

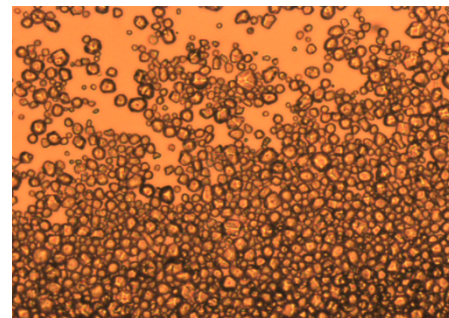
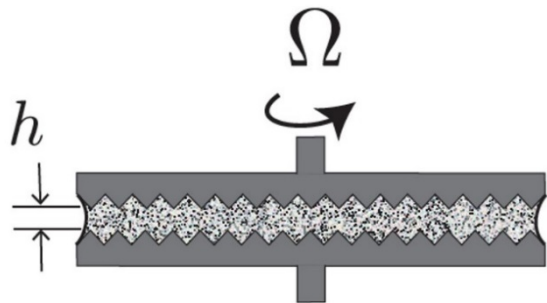
Aix- Marseille Univ, CNRS, IUSTI, Marseille, France



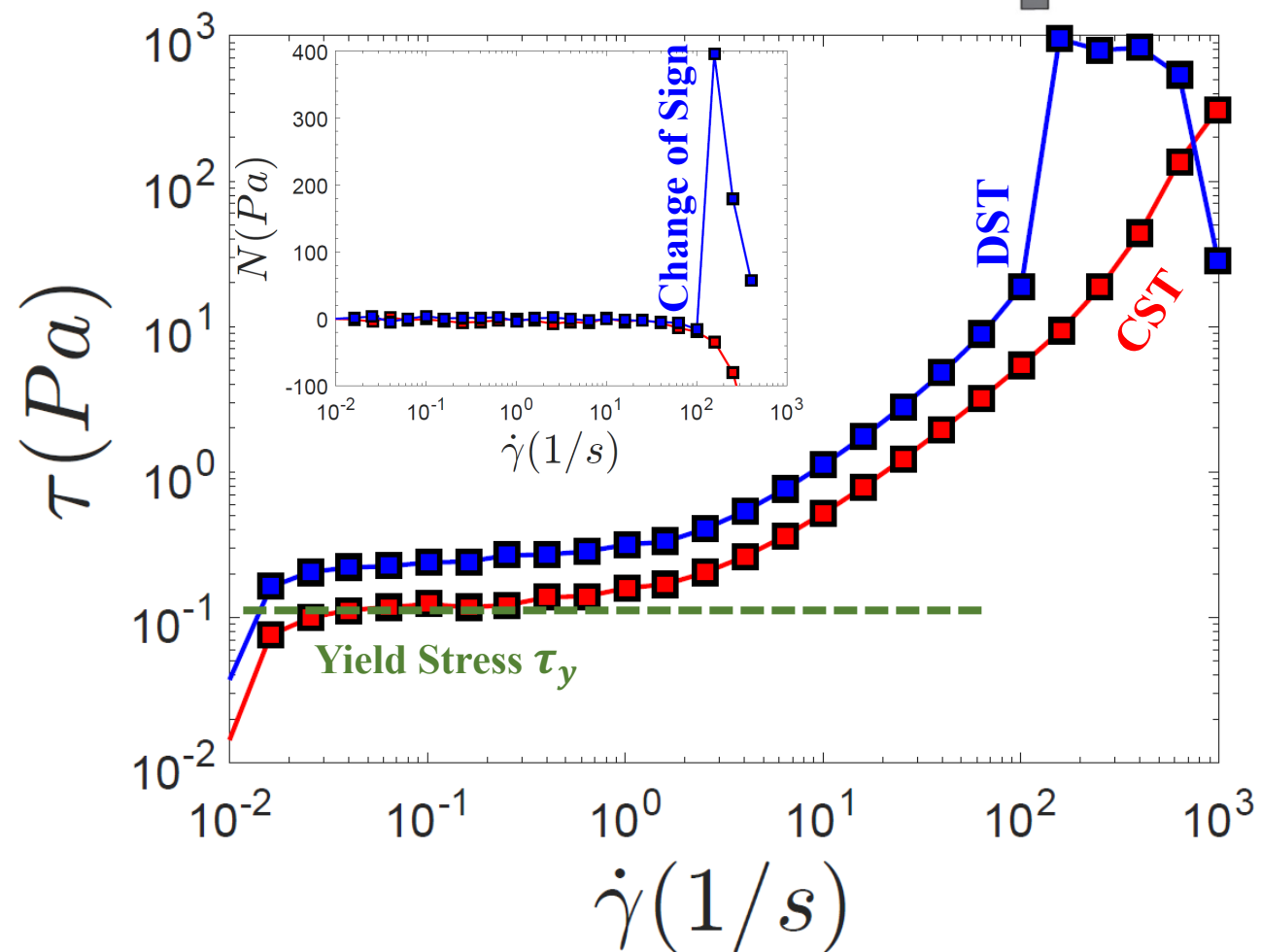
**Olivier Pouliquen**

Aix- Marseille Univ, CNRS, IUSTI, Marseille, France

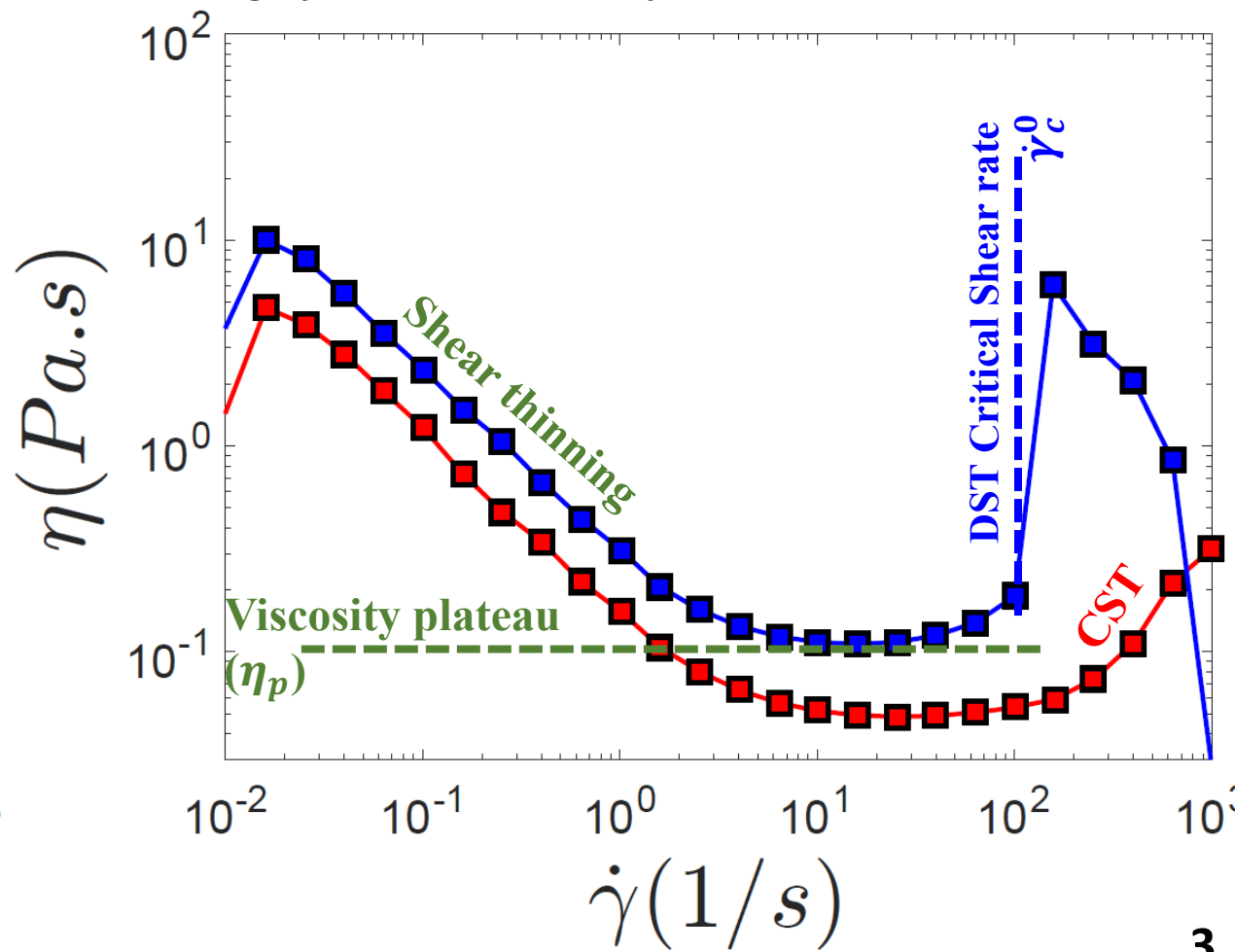
# Suspensions of cornstarch in water (Oobleck)



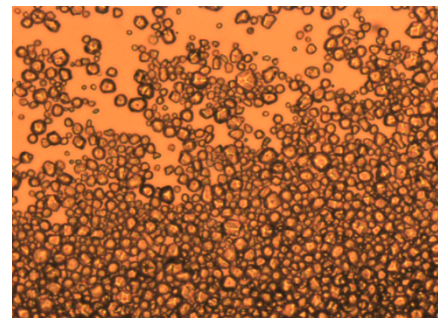
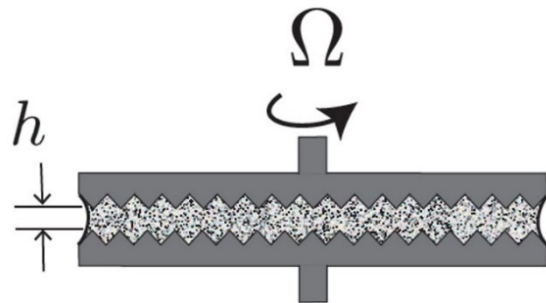
$5\mu\text{m} < d_{cs} < 20\mu\text{m}$   
 Cornstarch:  $\rho = 1.68 \frac{\text{gr}}{\text{cm}^3}$   
 water+CsCl:  $\rho \approx 1.68 \text{ gr/cm}^3$



Micrograph of cornstarch suspension



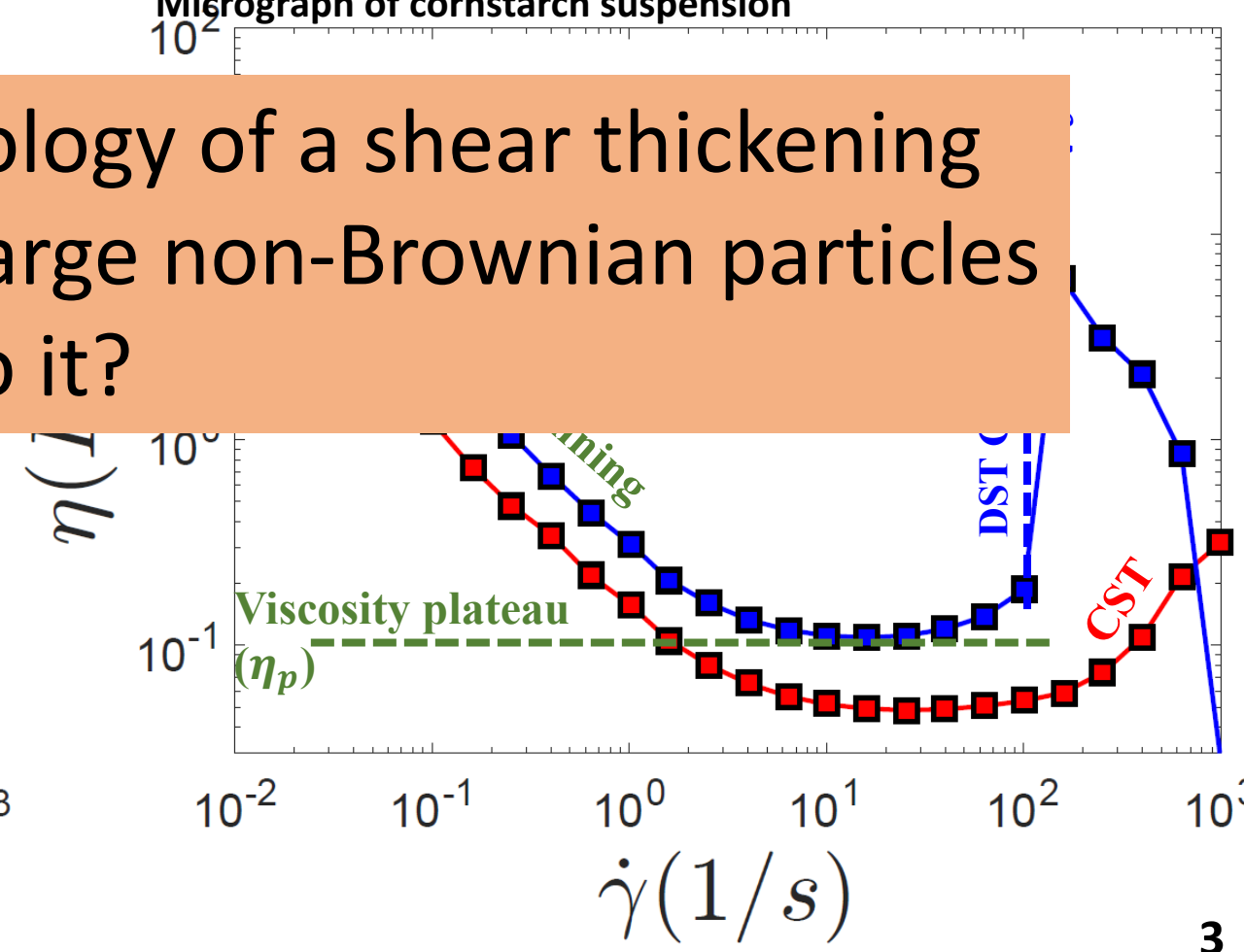
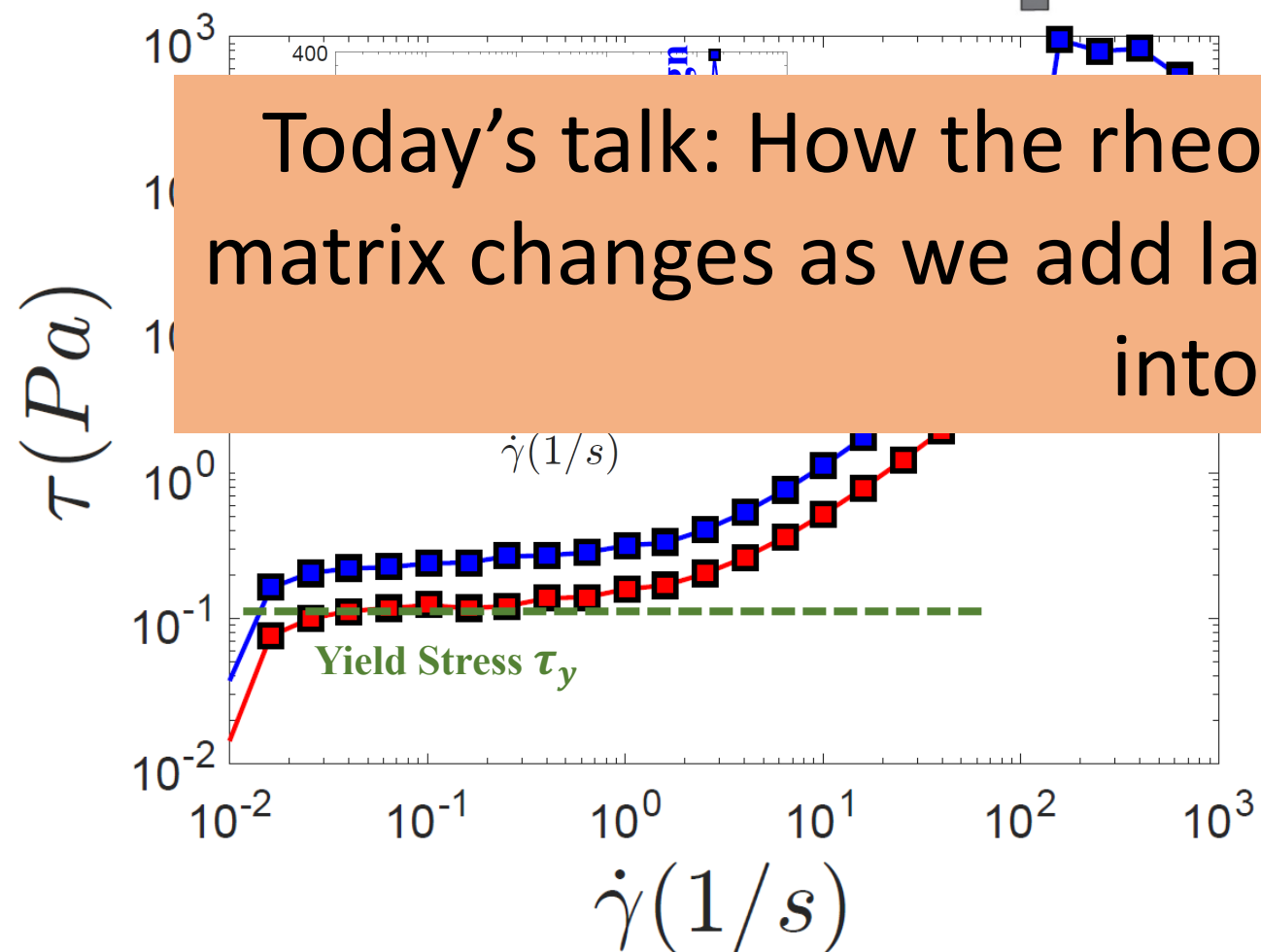
# Rheology: suspension of cornstarch in water



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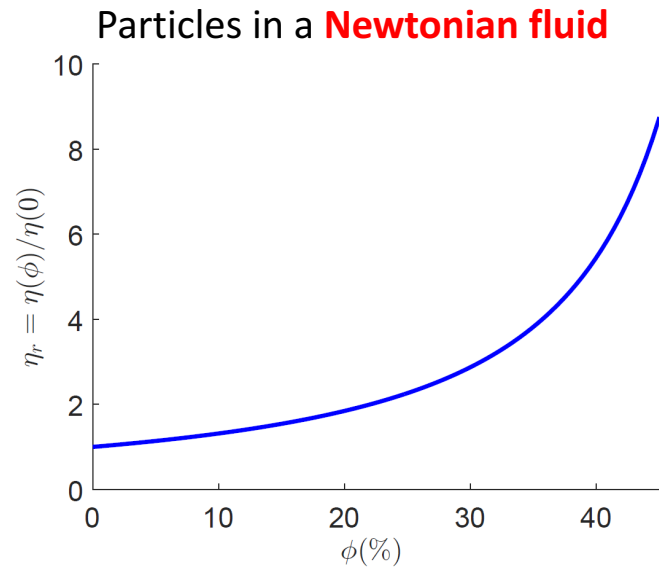
Today's talk: How the rheology of a shear thickening matrix changes as we add large non-Brownian particles into it?





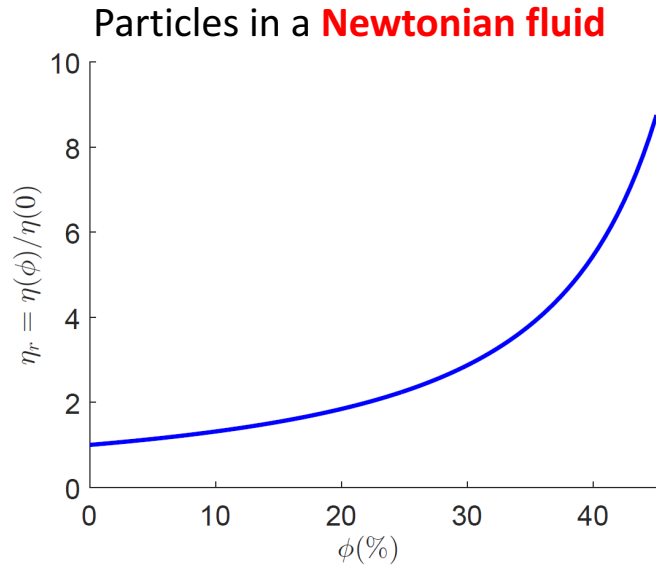
# Adding non-Brownian particles into Newtonian and generalized Newtonian fluids, Stokes limit

(Krieger & Dougherty, 1950)

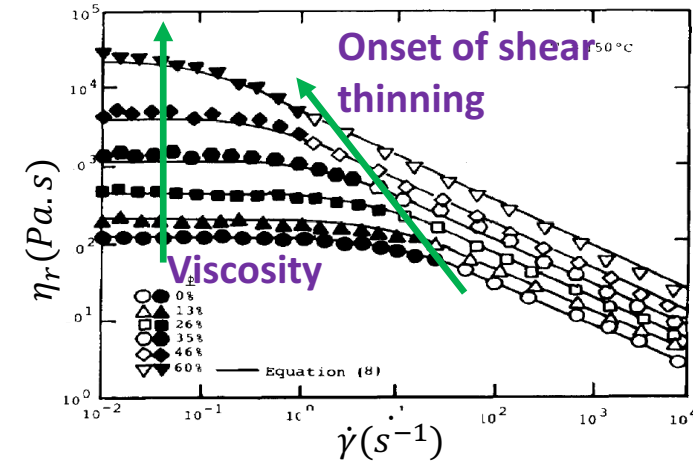


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Particles in a **shear thinning suspension**

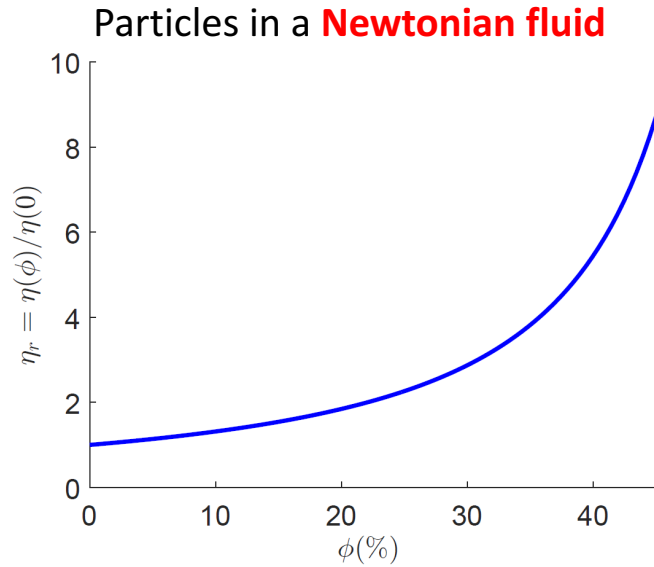


(Polinski et al., *JOR* 1988; Liard, et al., 2014)

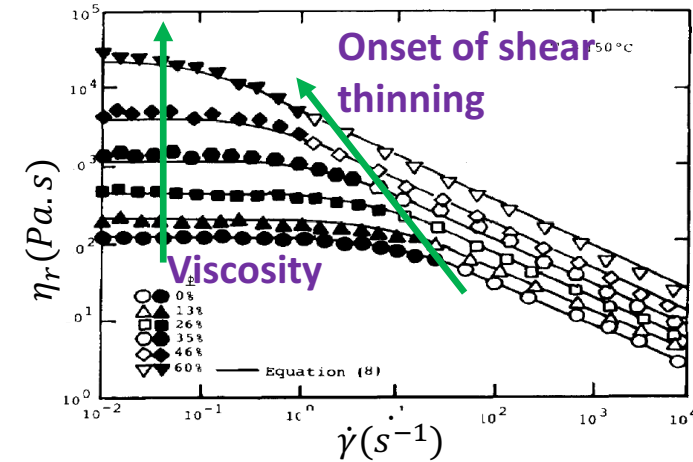
Fig. 4. Shear viscosity of glass spheres dispersed in the thermoplastic polymer at 150°C.  $\circ$  = 0%,  $\triangle$  = 13%,  $\square$  = 26%,  $\diamond$  = 35%,  $\nabla$  = 46%,  $\nabla$  = 60%, — = Eq. (8).

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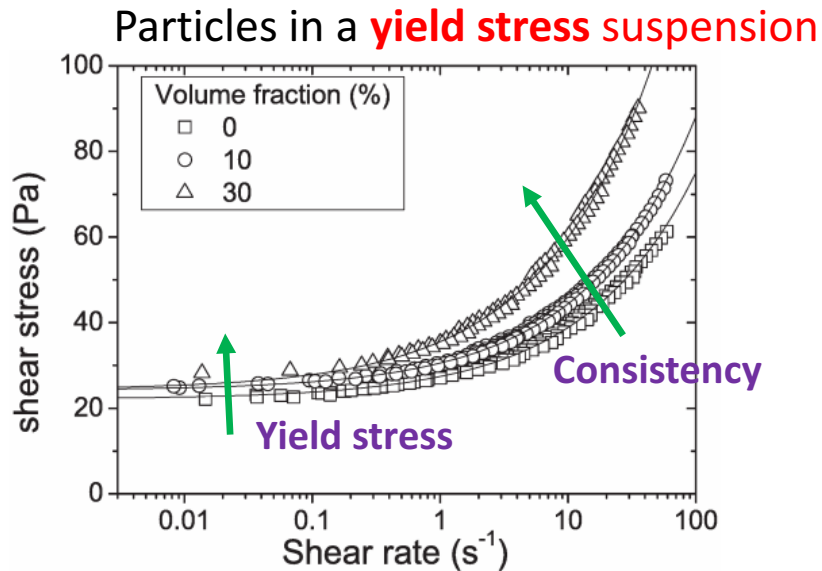
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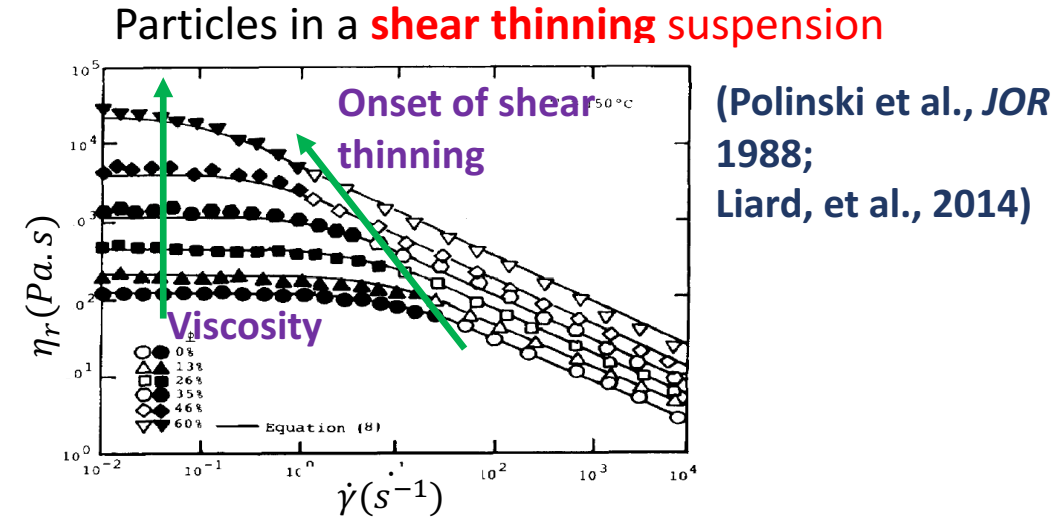
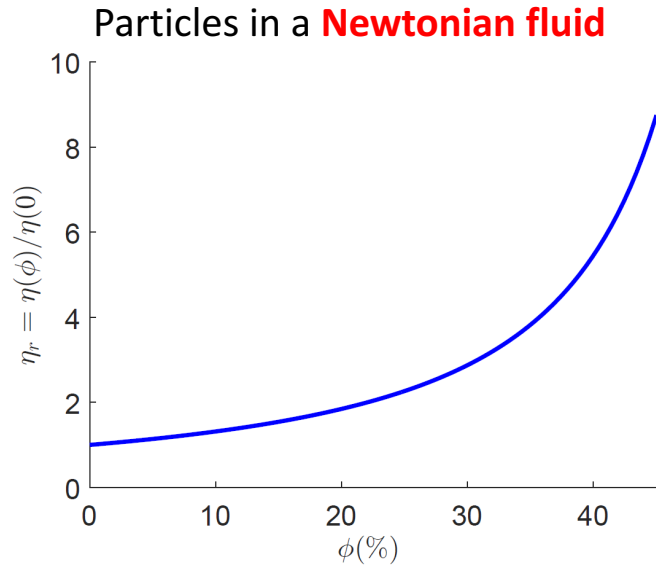
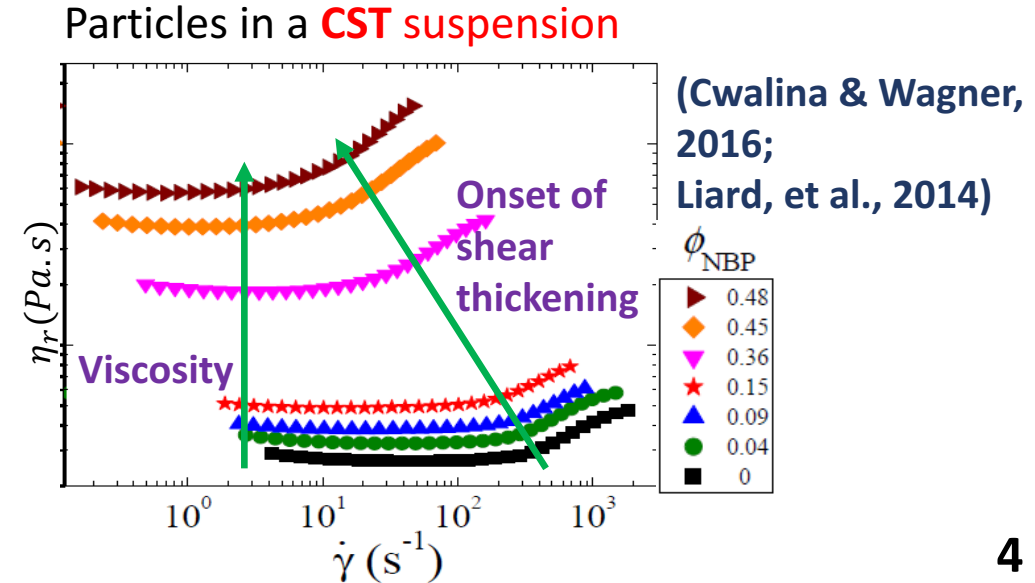
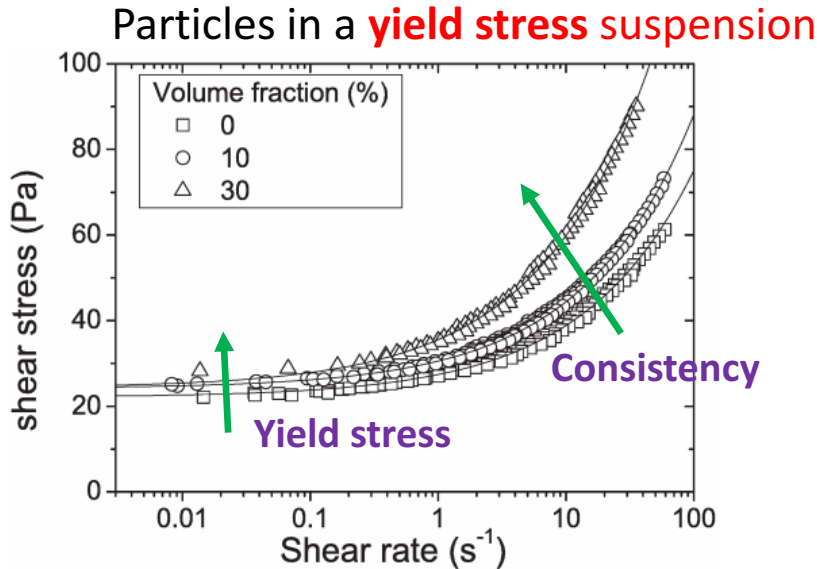


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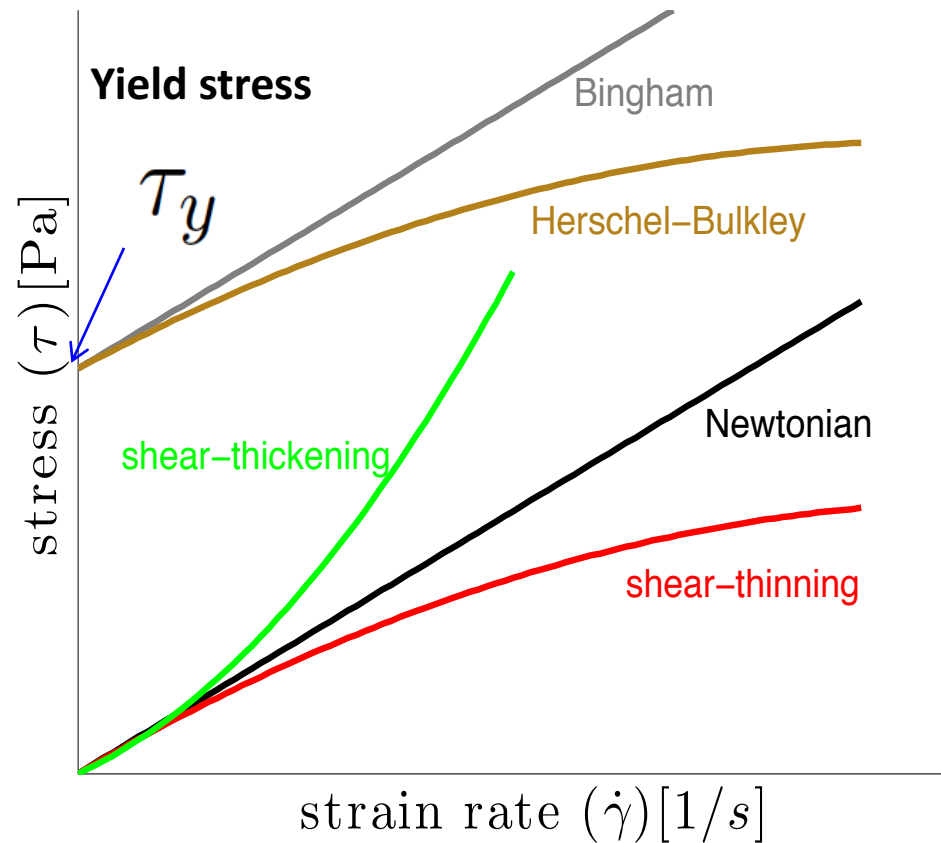
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# Explanation via a scaling argument

1. A generalized Newtonian fluid model

$$\begin{cases} \tau = k\dot{\gamma}^n & \text{Power-law model} \\ \tau = \tau_y + k\dot{\gamma}^n & \text{Herschel-Bulkley model} \end{cases}$$



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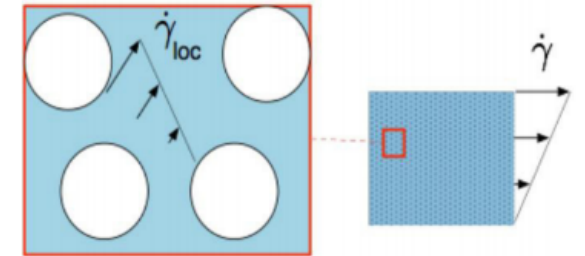
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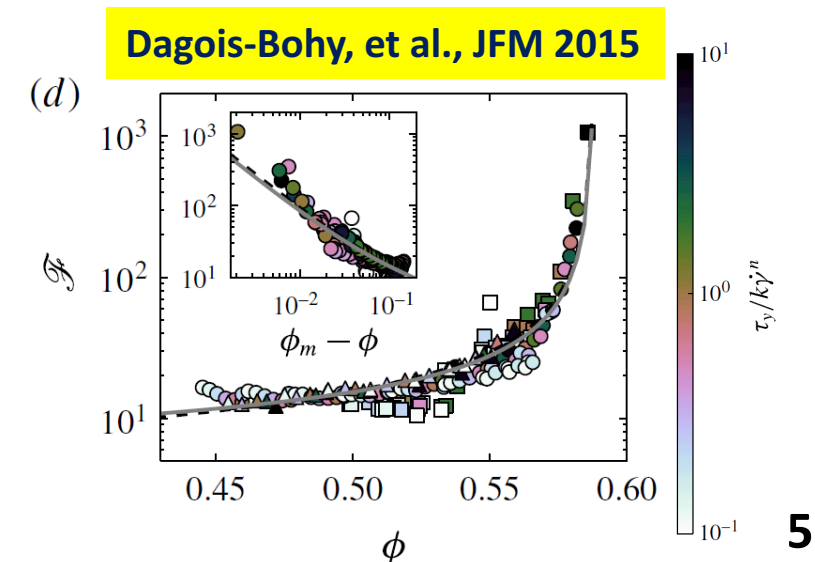
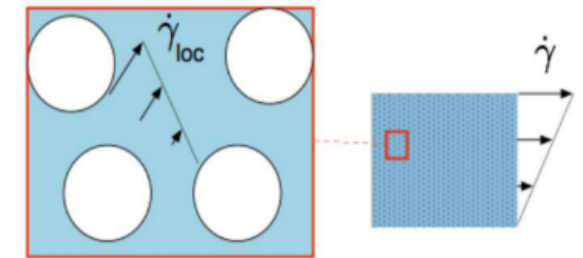
$$\eta_{app}(\dot{\gamma}) \rightarrow \eta_{app}(\dot{\gamma}_{local})$$

4. Local shear rate mainly controlled by geometrical constraints, i.e.  $\phi$

$$\dot{\gamma}_{local} = \dot{\gamma}\mathcal{F}(\phi)$$

(Chateau, et al., JOR 2008; Lerner, et al., Phys Rev E 2012;

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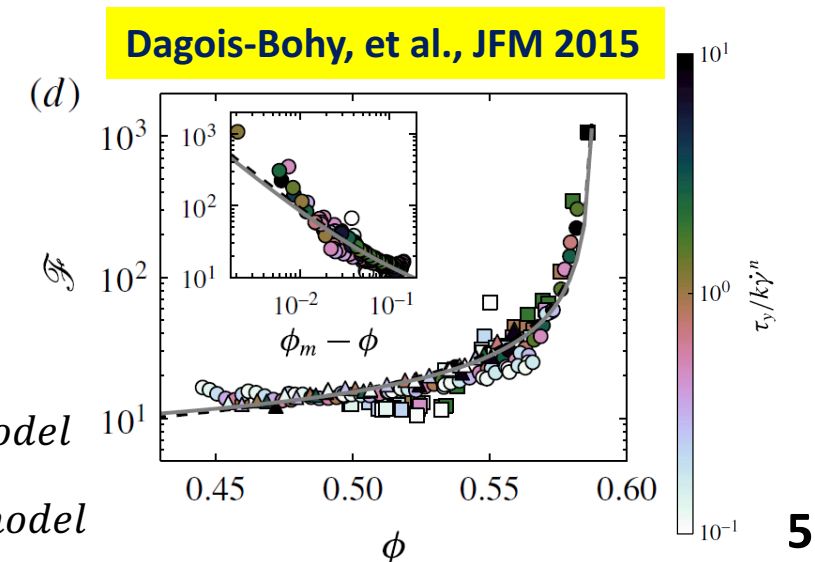
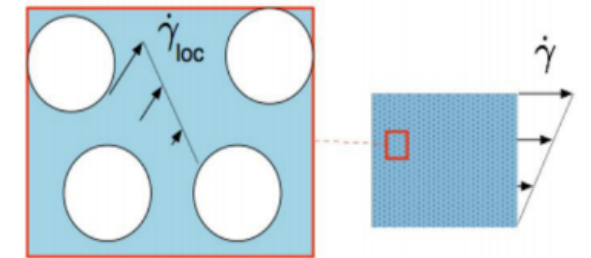
$$\dot{\gamma}_{local} = \dot{\gamma}\mathcal{F}(\phi)$$

(Chateau, et al., JOR 2008; Lerner, et al., Phys Rev E 2012;

**Dagois-Bohy, et al., JFM 2015; Madraki et al., PRF 2017)**

Constitutive laws for suspensions of particles in generalized Newtonian fluids:

$$\tau = \eta_{app}(\dot{\gamma})\dot{\gamma} \begin{cases} \tau = k(\dot{\gamma}\mathcal{F}(\phi))^{n-1}\dot{\gamma} = \kappa_s(\phi)\dot{\gamma}^n & \text{Power-law model} \\ \tau = \frac{\eta_r(\phi)\tau_y}{\mathcal{F}(\phi)} + k\mathcal{F}(\phi)^{n-1}\eta_r(\phi)\dot{\gamma}^n = \tau_{ys}(\phi) + \kappa_s(\phi)\dot{\gamma}^n & \text{Herschel-Bulkley model} \end{cases}$$

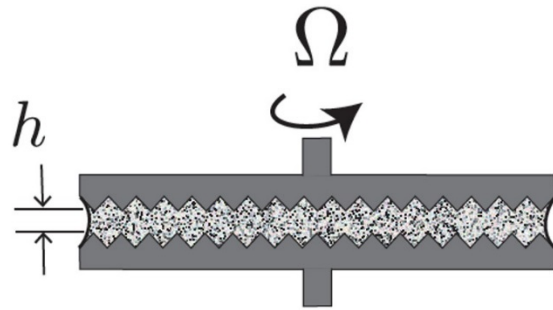


Adding large non-Brownian particles to an  
oobleck exhibiting DST!

# Methods

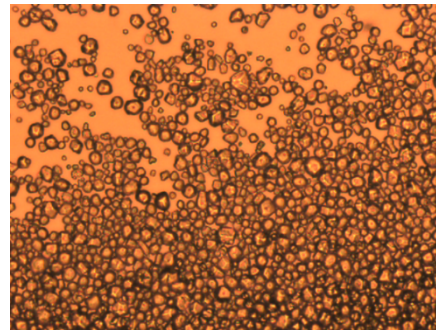
## Experiment

- Serrated parallel plate was used to eliminate wall slip effect
- Shear ramp was imposed to obtain rheological results

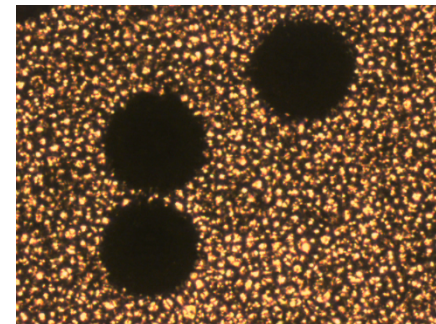


## Materials

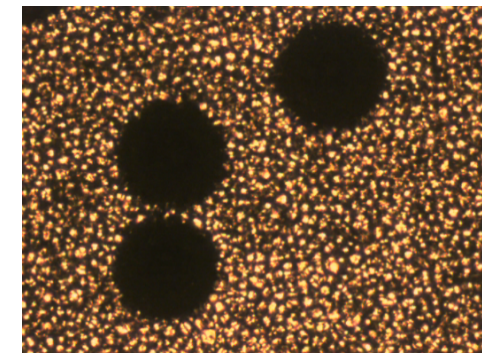
- Cornstarch:  $\rho = 1.68 \text{ gr/cm}^3$
- Distilled water+CsCl:  $\rho \simeq 1.68 \text{ gr/cm}^3$
- Silver coated PMMA\*:  $\rho = 1.34 \text{ gr/cm}^3$



Micrograph of cornstarch suspension  
 $5\mu\text{m} < d_{cs} < 20\mu\text{m}$



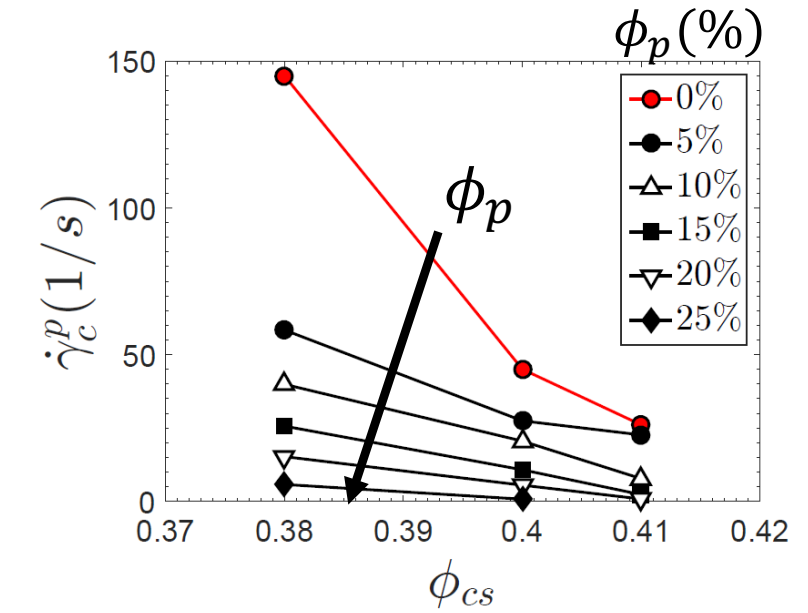
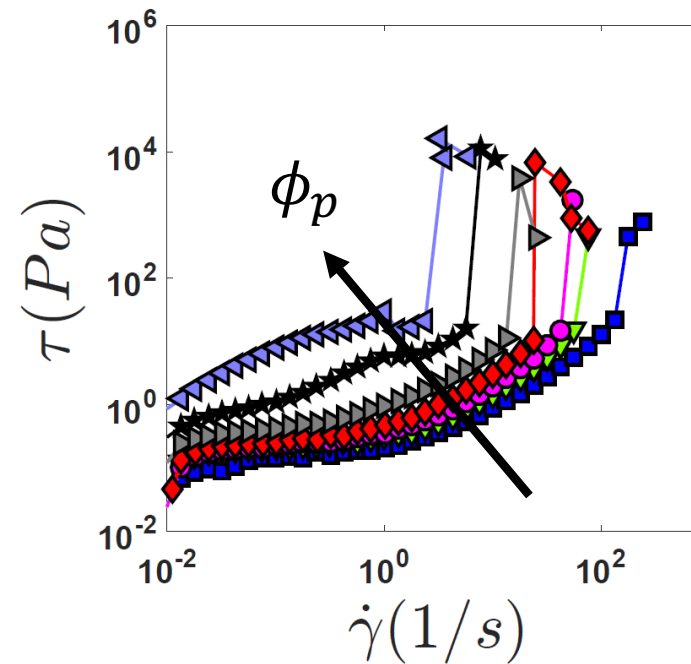
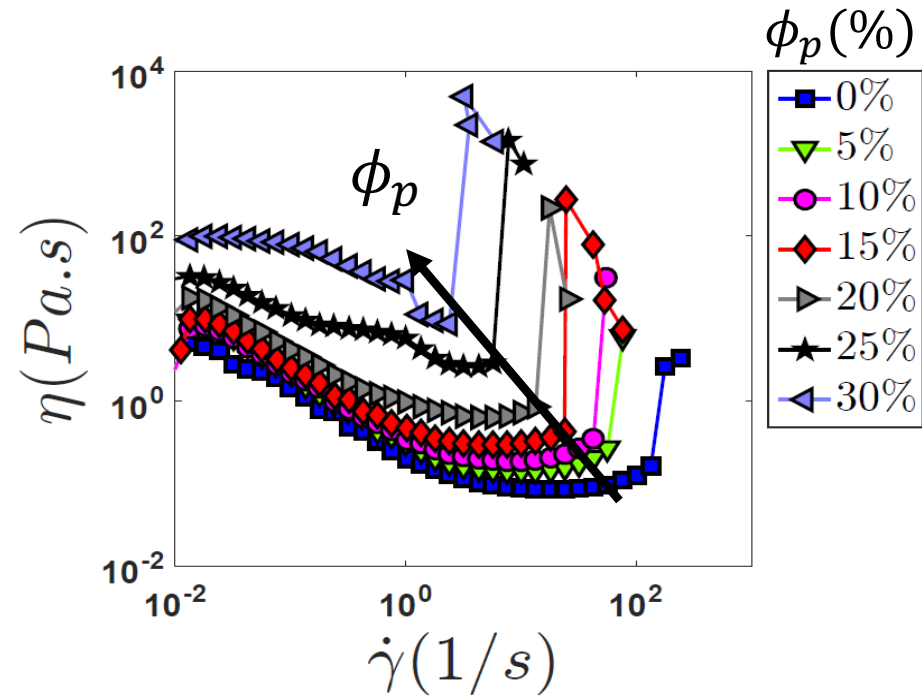
Micrograph of large particles (silver coated PMMA) in cornstarch suspension  
 $106\mu\text{m} < d_p < 125\mu\text{m}$



# Rheology of particles in cornstarch suspensions

Adding large particles to a cornstarch suspension has dramatic effect on rheological behavior:

- DST transition
- Effective viscosity of the mixture

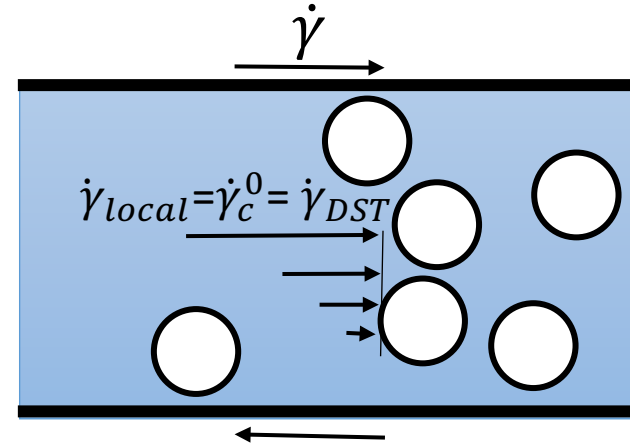
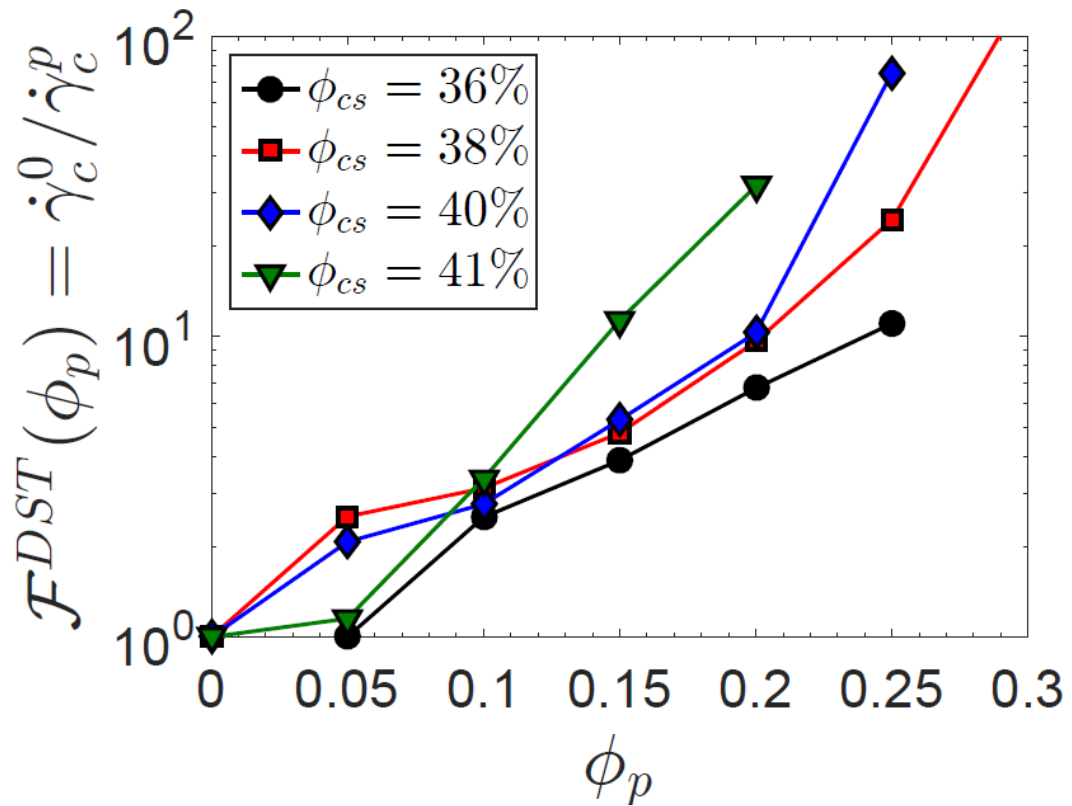


$$\phi_p = \frac{V_p}{V_p + V_{\text{Cornstarch Suspension}}}$$

$\dot{\gamma}_c^p$ : The onset of DST for the mixture of large particles and cornstarch suspensions at different  $\phi_p$



# Shift of critical shear rate at DST

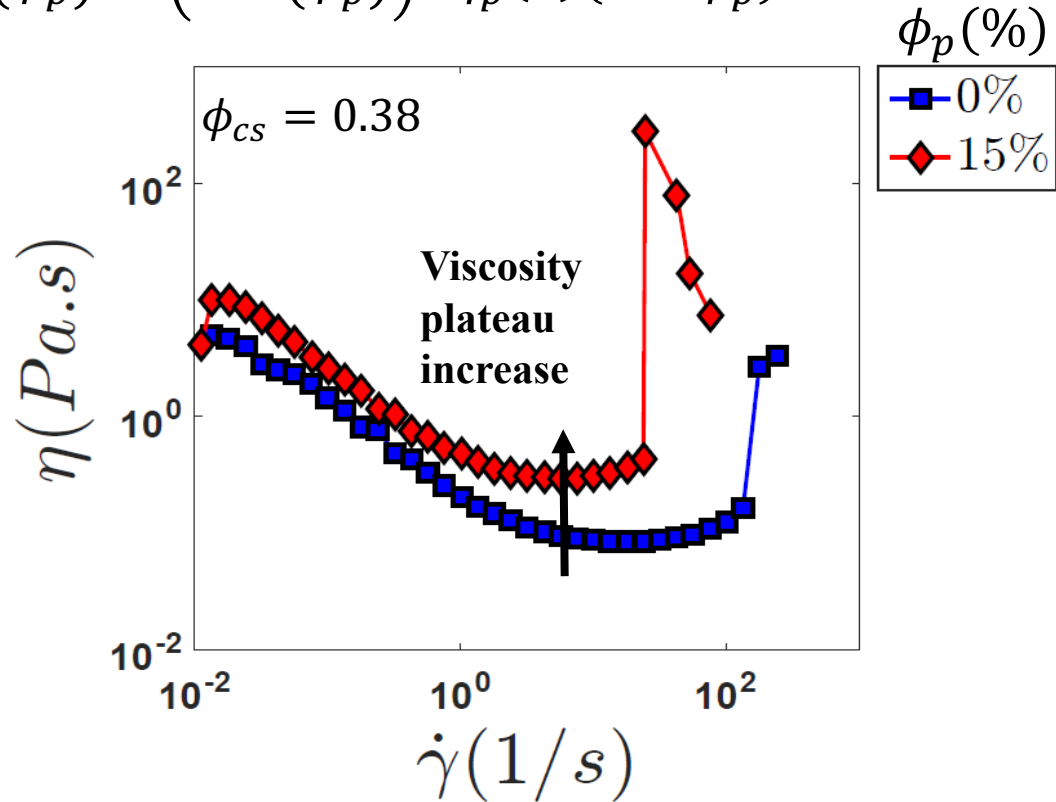


- $\dot{\gamma}_{local} = \mathcal{F}(\phi_p)\dot{\gamma}$
- DST should be obtained when  $\dot{\gamma}_{local}$  reaches the critical values for the pure cornstarch suspension,  $\dot{\gamma}_c^0$  ( $\dot{\gamma}_{local} = \dot{\gamma}_c^0$ ).
- $\mathcal{F}(\phi_p) = \frac{\dot{\gamma}_c^0}{\dot{\gamma}_c^p} = \frac{\text{critical shear rate}(\phi_p=0)}{\text{critical shear rate}(\phi_p)}$

# Shift of rheological properties prior to DST

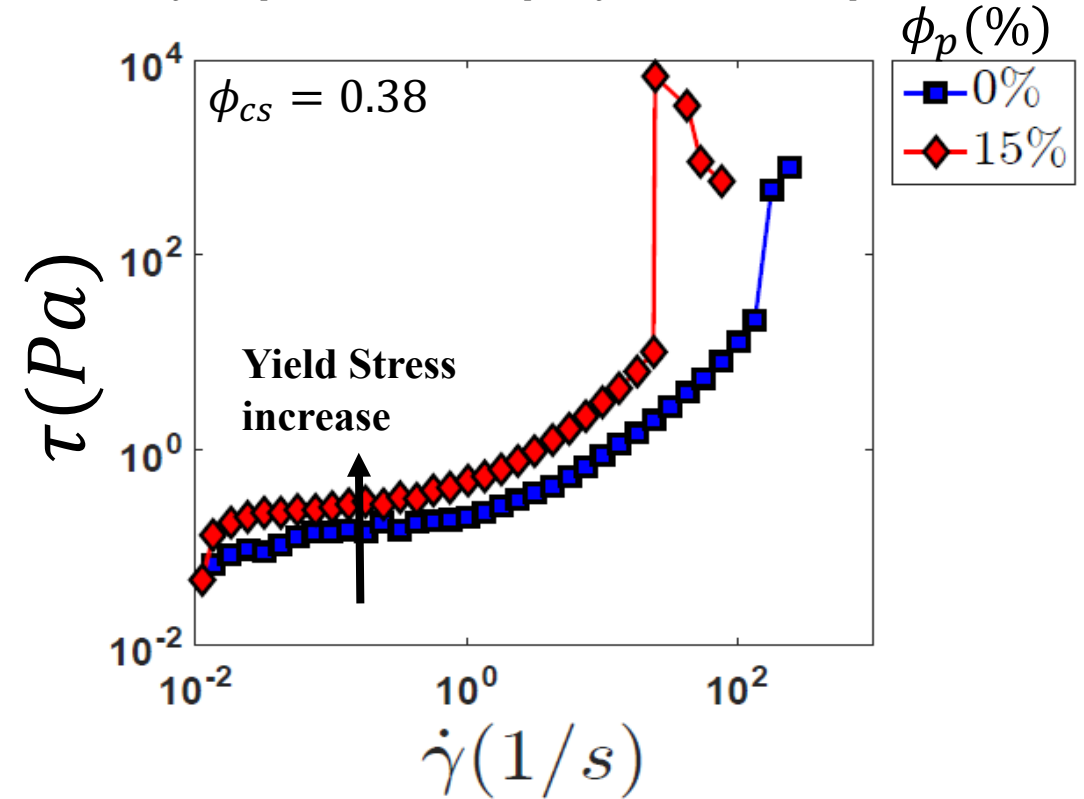
$$\eta_p(\phi_p)\dot{\gamma}^2 = (1 - \phi_p)\eta_p(0)\dot{\gamma}_{\text{local}}^2$$

$$\eta_p(\phi_p) = \left(\mathcal{F}^{\eta_p}(\phi_p)\right)^2 \eta_p(0)(1 - \phi_p)$$



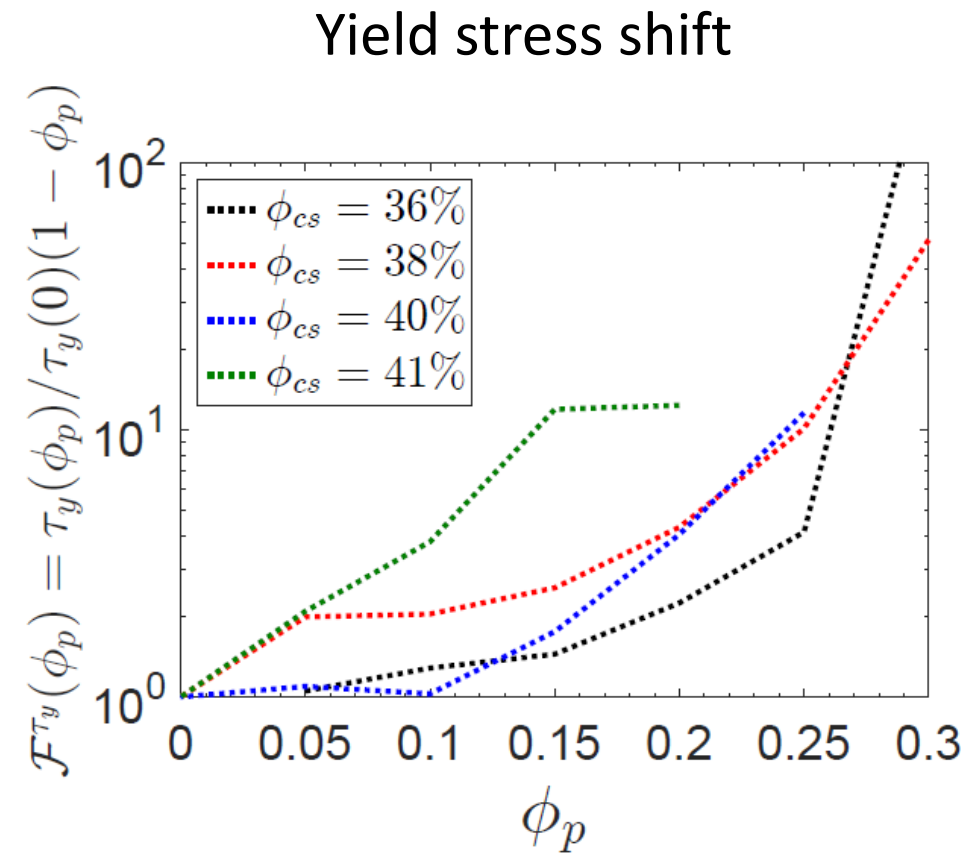
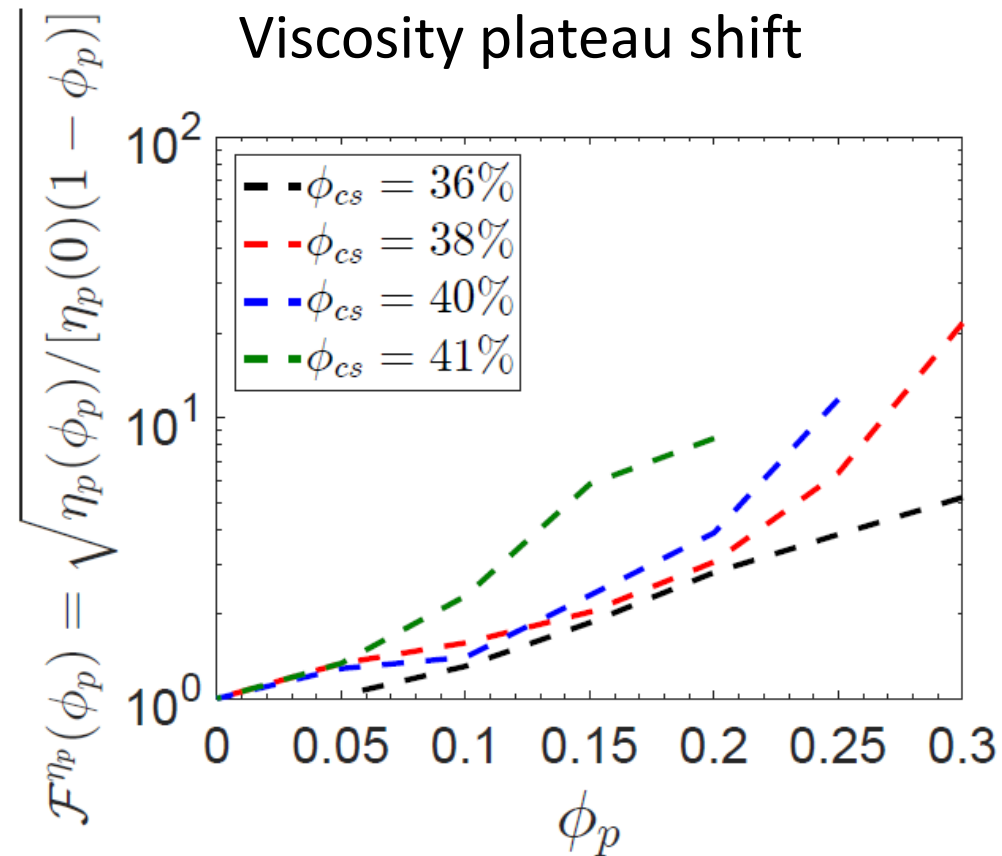
$$\tau_y(\phi_p)\dot{\gamma} = (1 - \phi_p)\tau_y(0)\dot{\gamma}_{\text{local}}$$

$$\tau_y(\phi_p) = \mathcal{F}^{\tau_y}(\phi_p)\tau_y(0)(1 - \phi_p)$$



- Not only the DST transition is shifted by adding large spheres to cornstarch suspension, but also the **viscosity plateau** and **yield stress** are shifted.

# Shift of rheological properties prior to DST

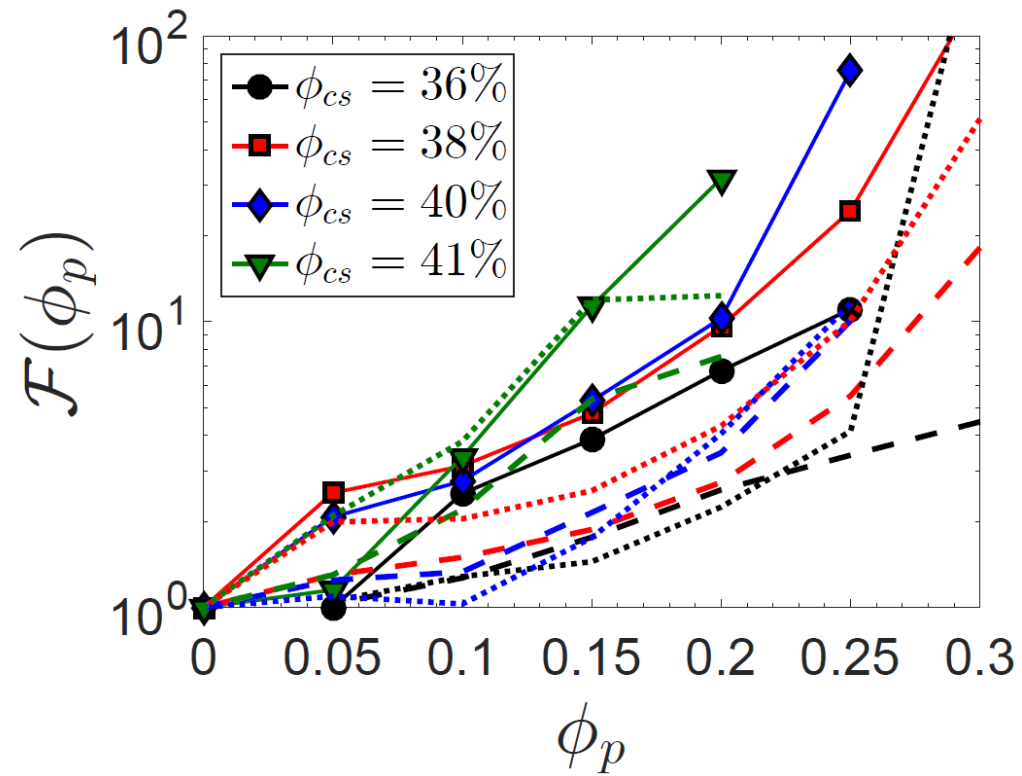


- Lever function can be alternatively estimated by measuring the shift in **viscosity plateau** and **yield stress**.

(Dagois-Bohy, et al., 2015; Chateau, et al., 2008; Ovarlez, et al., 2015)

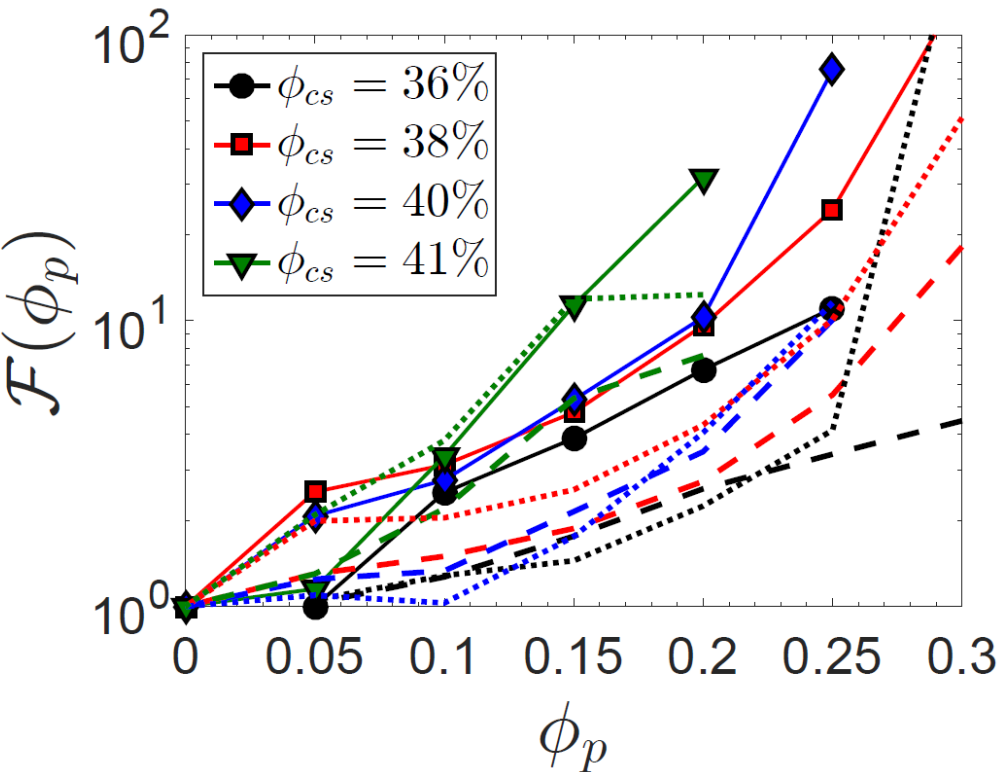
# Three values of $\mathcal{F}(\phi_p)$

- Predictions from the DST shift,  $\mathcal{F}(\phi_p)$ , are systematically larger than other two evaluations.



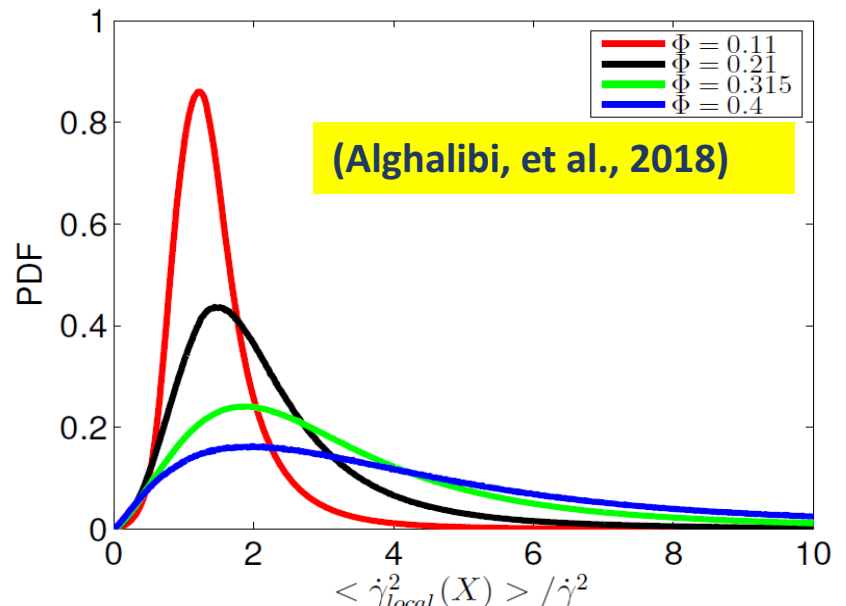
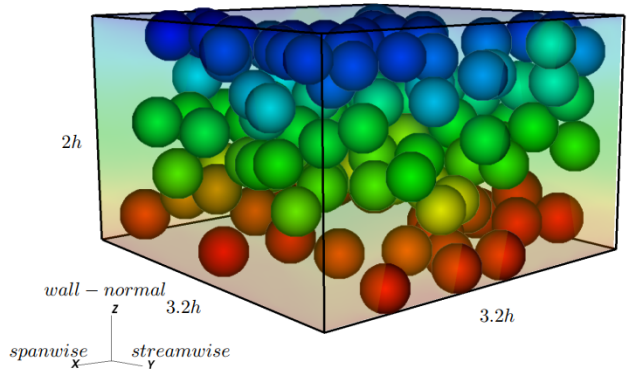
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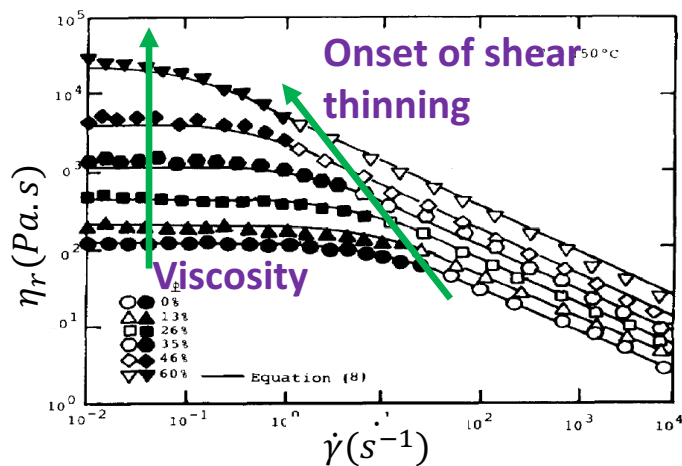
- Probability distribution function of local shear rate present increase of extreme values by adding larger particles to a background fluid. DST may be controlled by the extreme values of the local shear rate distribution.

(Liard, et al., 2014;  
 Alghalibi, et al., 2018;  
 Souzy, et al., 2017)



# Adding non-Brownian particles into Newtonian and generalized Newtonian fluids, Stokes limit

Particles in a **shear thinning** suspension

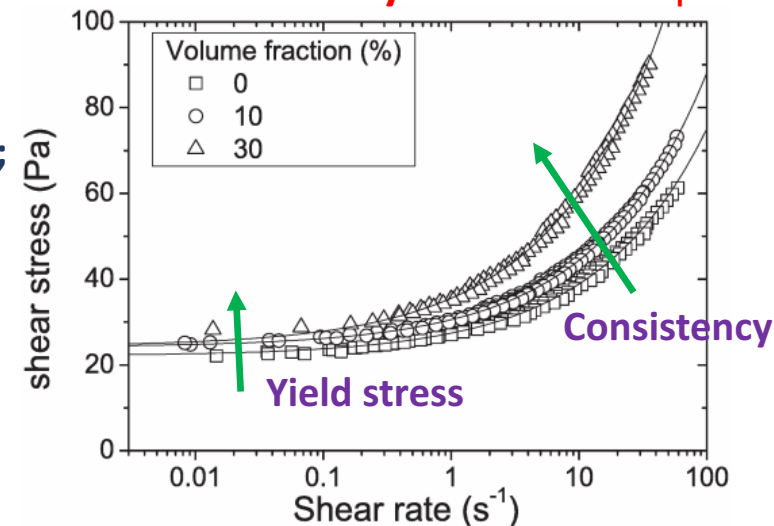


(Polinski et al., *JOR* 1988;  
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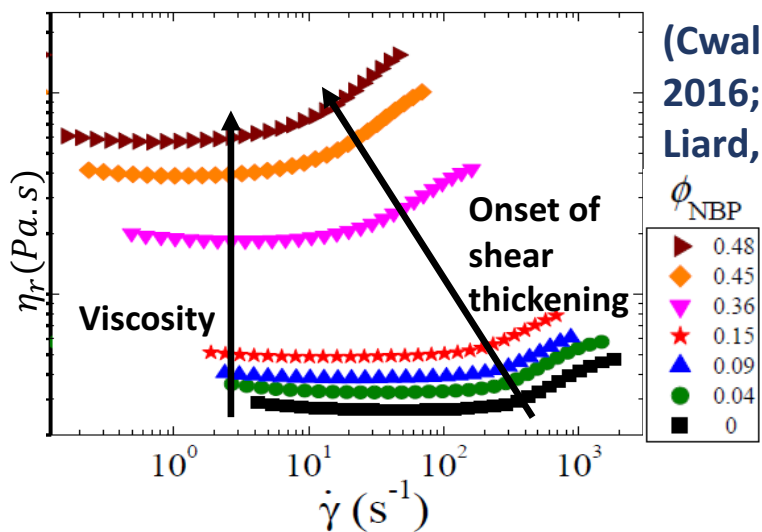
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Fig. 4. Shear viscosity of glass spheres dispersed in the thermoplastic polymer at 150°C.  $\circ$   $\bullet$  = 0%,  $\triangle$   $\blacktriangle$  = 13%,  $\square$   $\blacksquare$  = 26%,  $\diamond$   $\blacklozenge$  = 35%,  $\diamond$   $\blacklozenge$  = 46%,  $\nabla$   $\blacktriangledown$  = 60%, — = Eq. (8).

Particles in a **yield stress** suspension



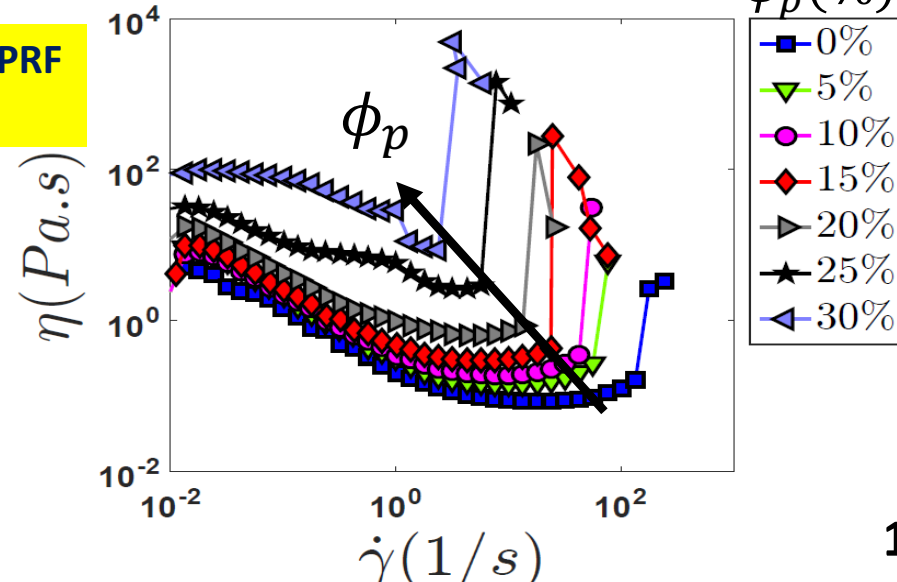
Particles in a **CST** suspension



(Cwalina & Wagner, 2016;  
Liard, et al., 2014)

(Madraki et al., *PRF* 2017.)

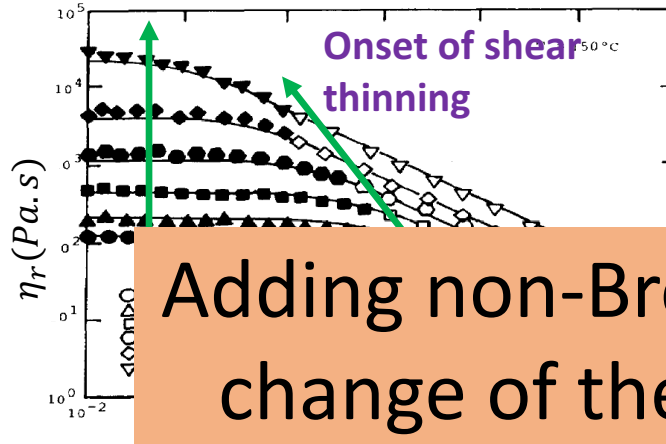
Particles in a **DST** suspension  $\phi_p$  (%)





# Adding non-Brownian particles into Newtonian and generalized Newtonian fluids, Stokes limit

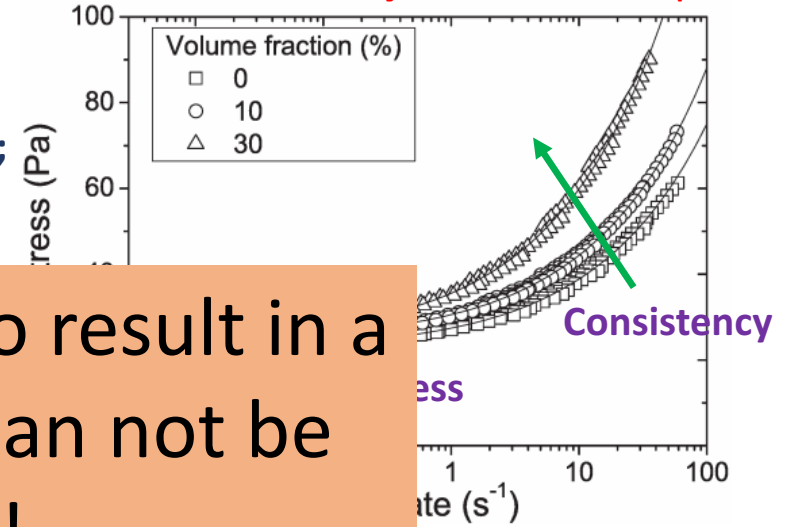
Particles in a **shear thinning** suspension



(Polinski et al., *JOR* 1988; Liard, et al., 2014)

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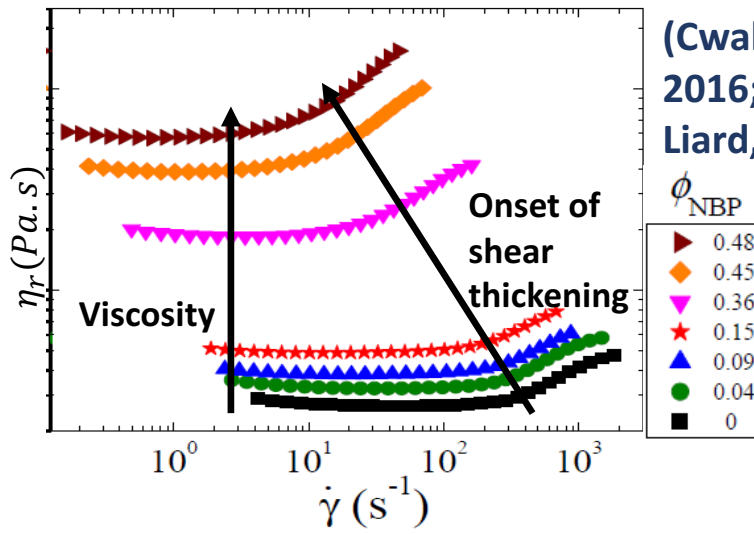
Particles in a **yield stress** suspension



Adding non-Brownian particles may also result in a change of the Rheological behavior! Can not be explained via this framework!

Fig. 4. Shear rate vs shear stress at 150°C

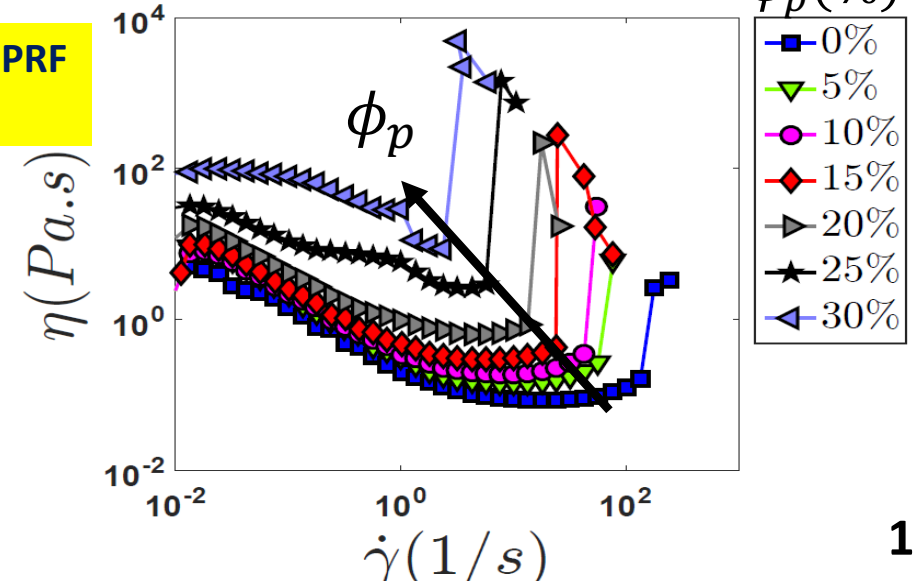
Particles in a **CST** suspension



(Cwalina & Wagner, 2016; Liard, et al., 2014)

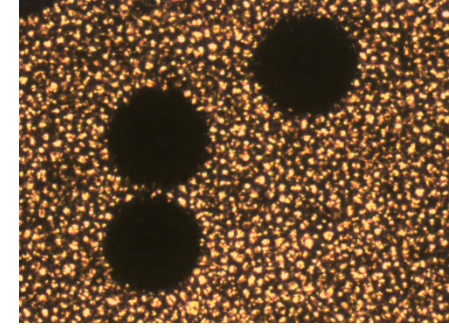
(Madraki et al., *PRF* 2017.)

Particles in a **DST** suspension  $\phi_p$  (%)

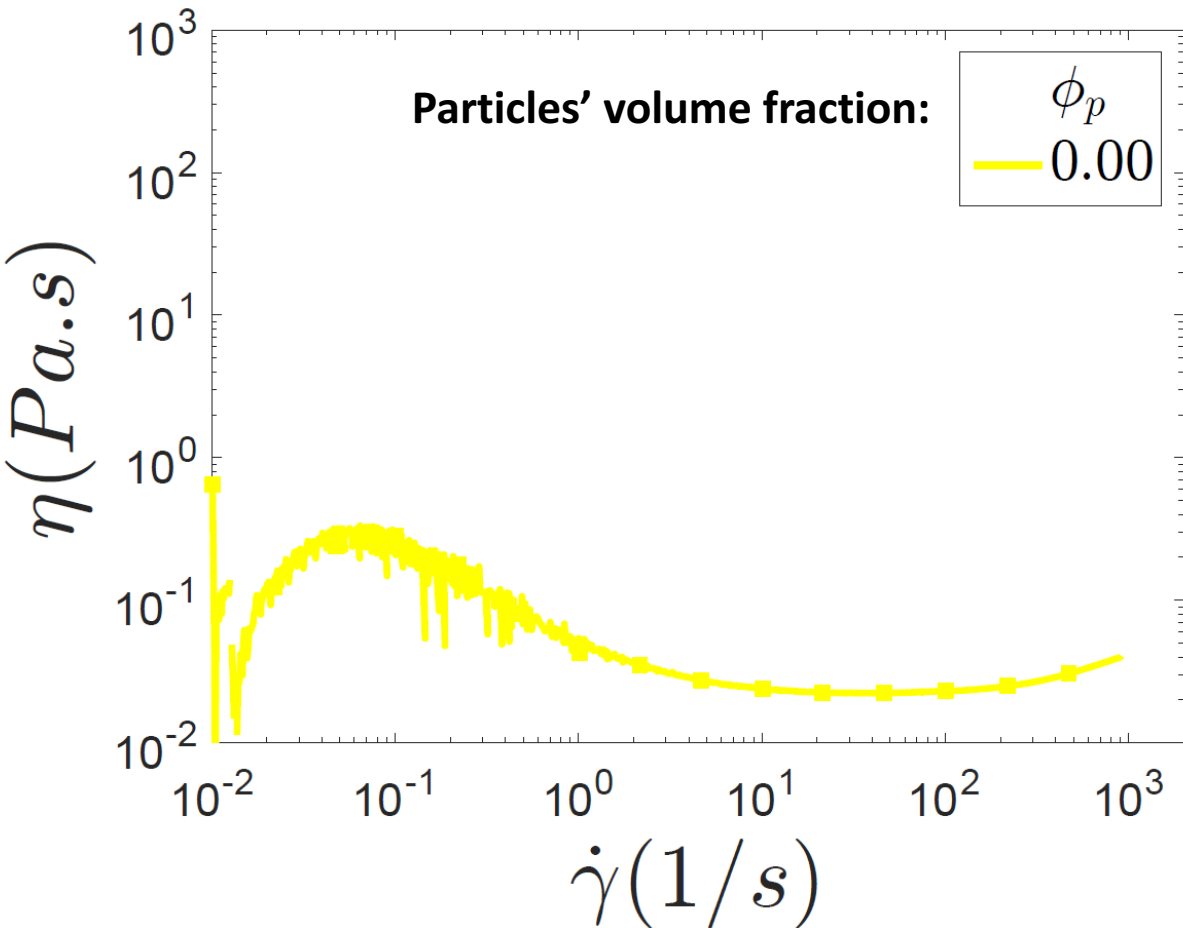


# CST to DST transition

Adding particles to a CST suspension, i.e. Cornstarch suspension  $\phi_{CS} = 0.3$ :



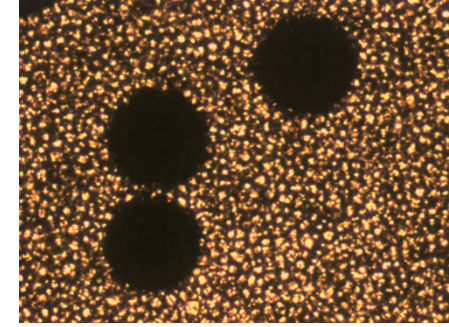
Silver coated PMMA in cornstarch suspension



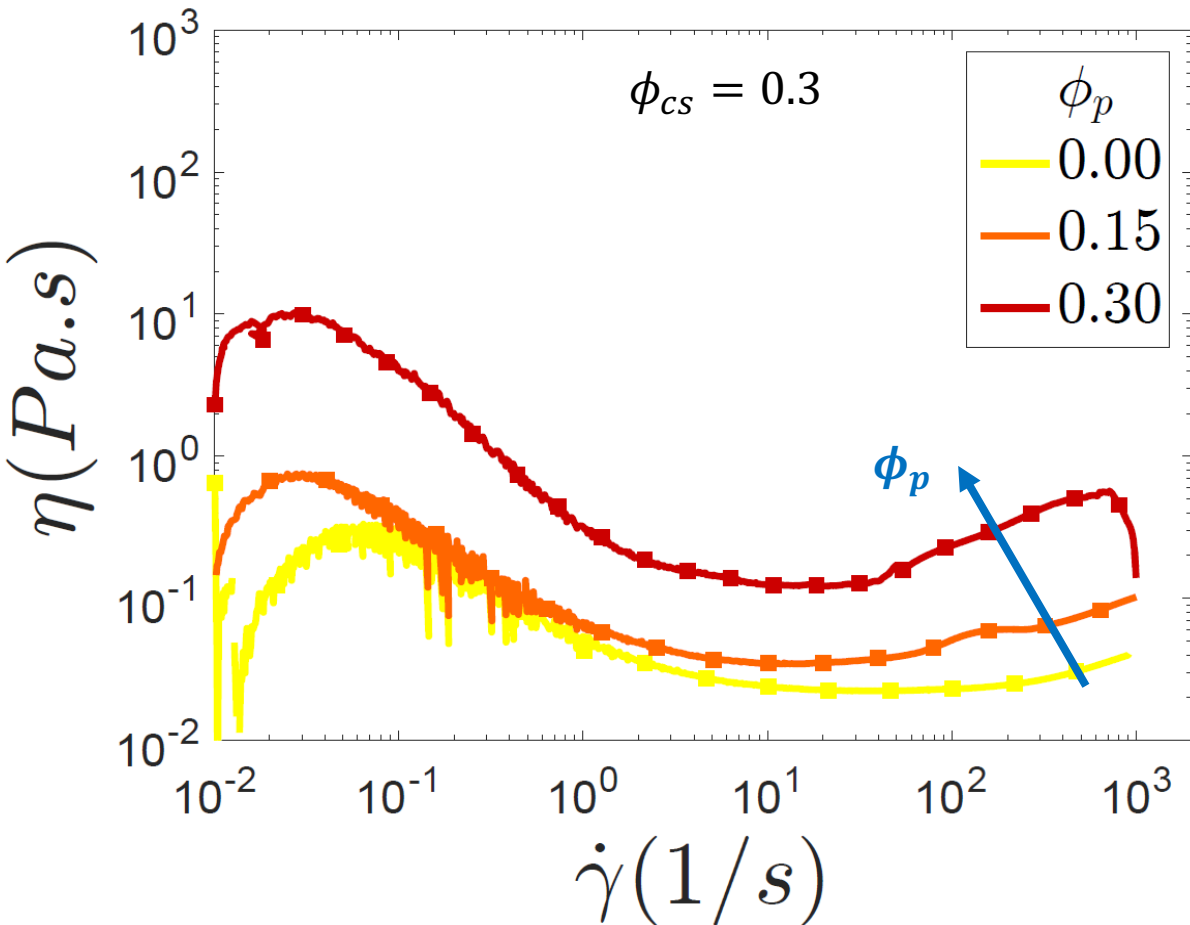
# CST to DST transition

Adding particles to CST suspension, i.e. Cornstarch suspension  $\phi_{cs} = 0.3$ :

- **Low** particles' volume fraction : **increases the viscosity** and **expedites CST** (Cwalina & Wagner, 2015; Liard, et al., 2014)



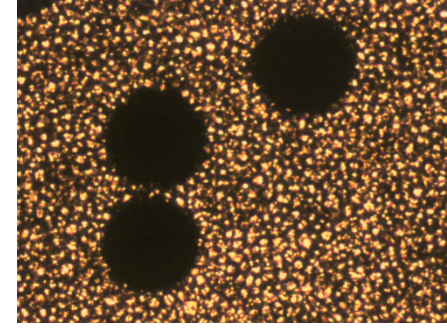
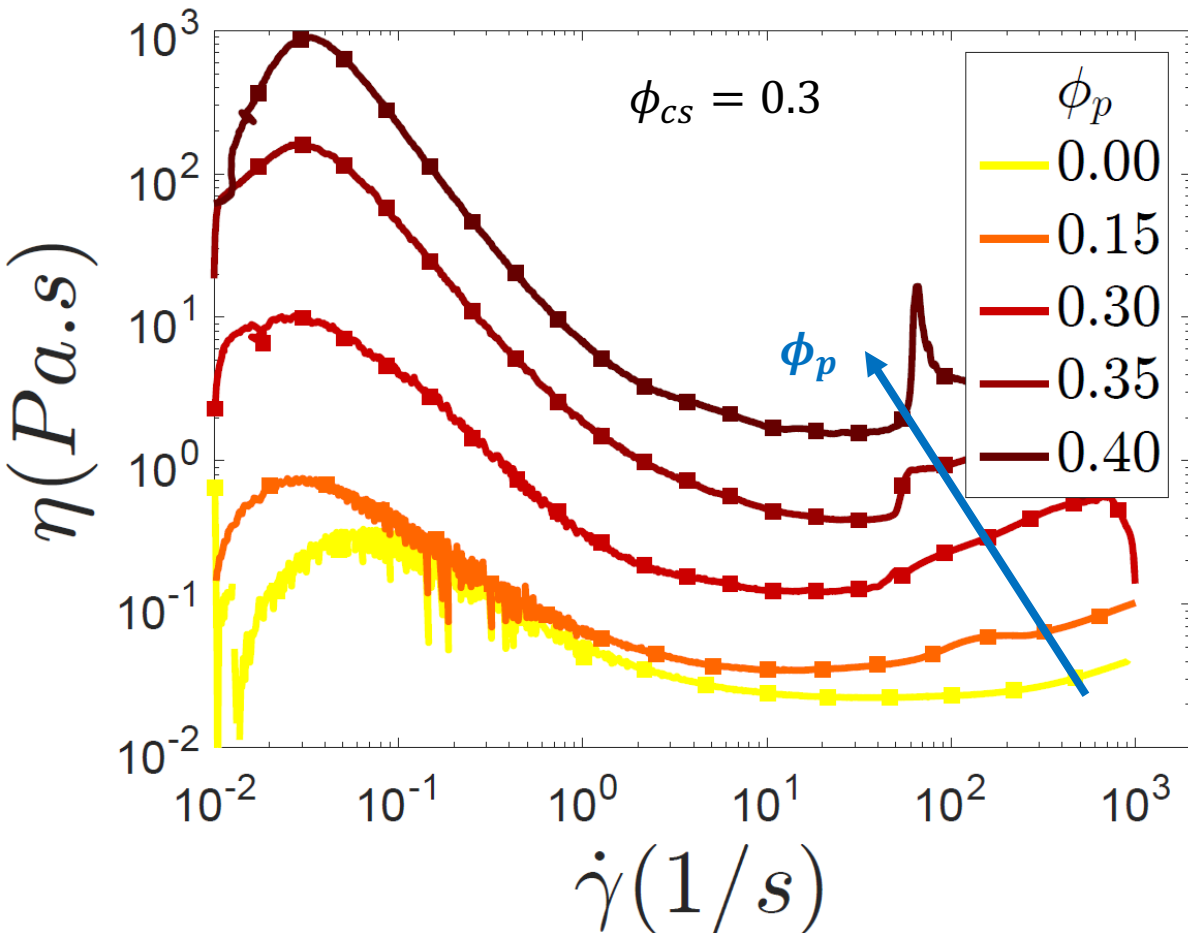
Silver coated PMMA in cornstarch suspension



# CST to DST transition

Adding particles to CST suspension, i.e. Cornstarch suspension  $\phi_{cs} = 0.3$ :

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- **High** particles' volume fraction concentrations: **transitions the CST to DST**

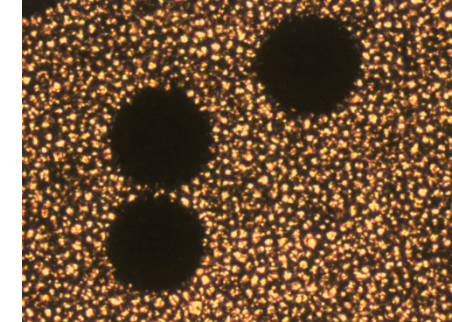


Silver coated PMMA in cornstarch suspension

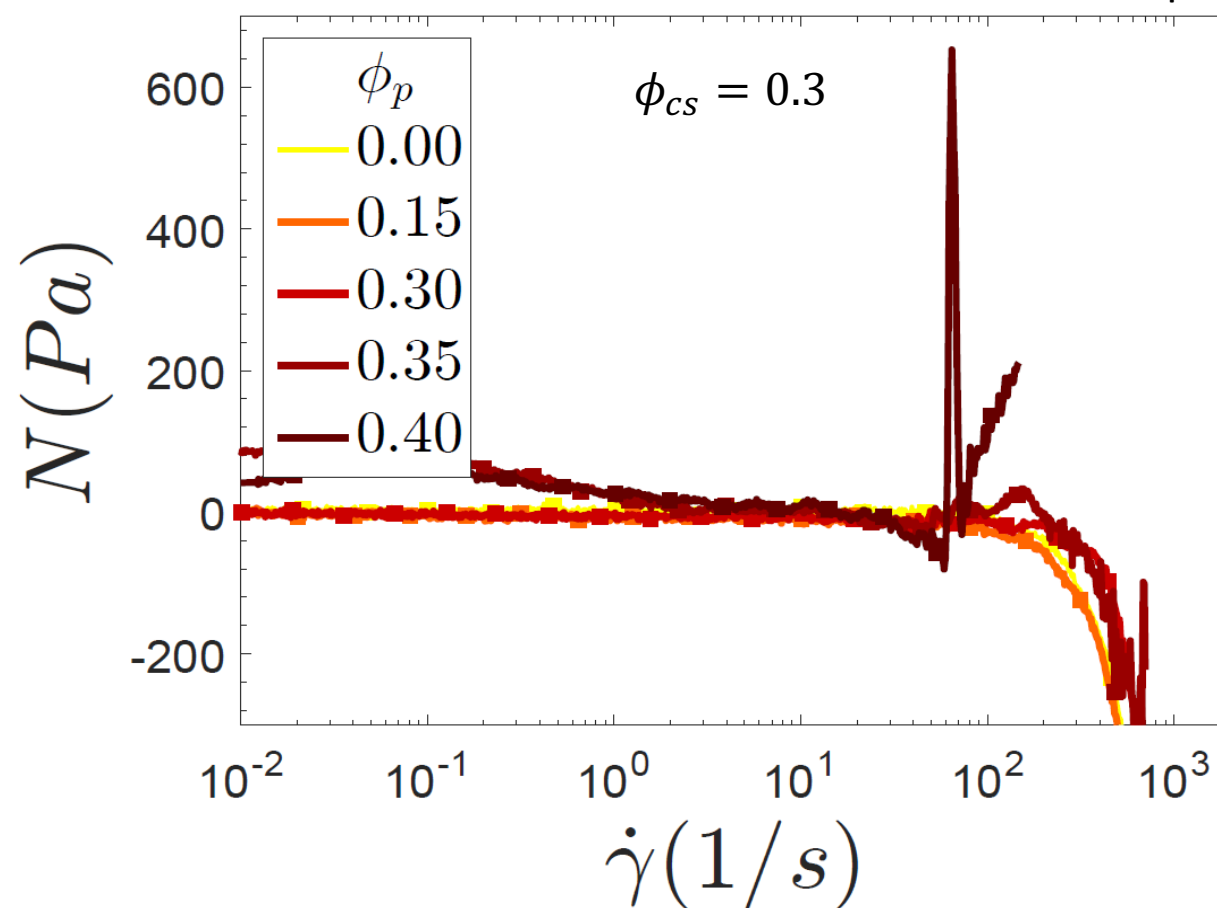
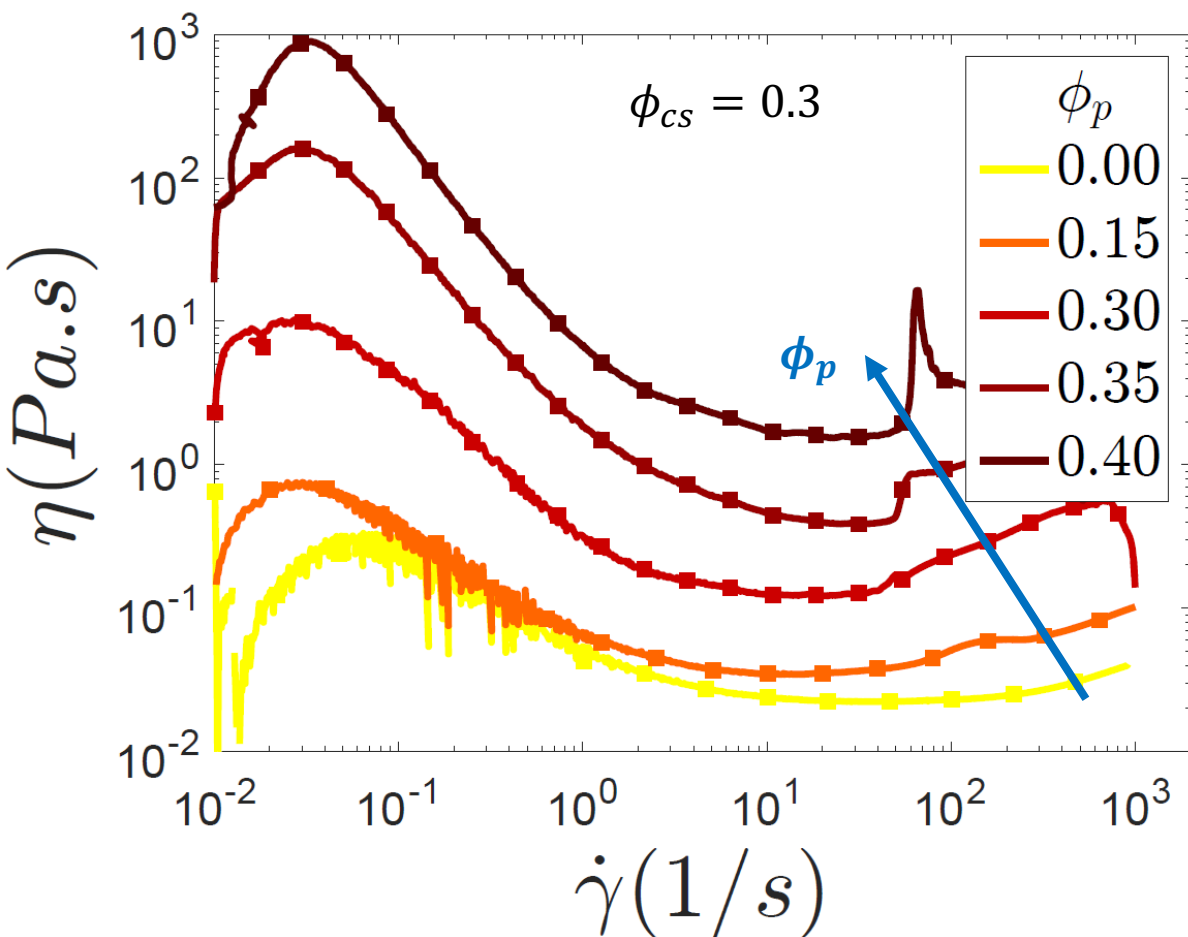
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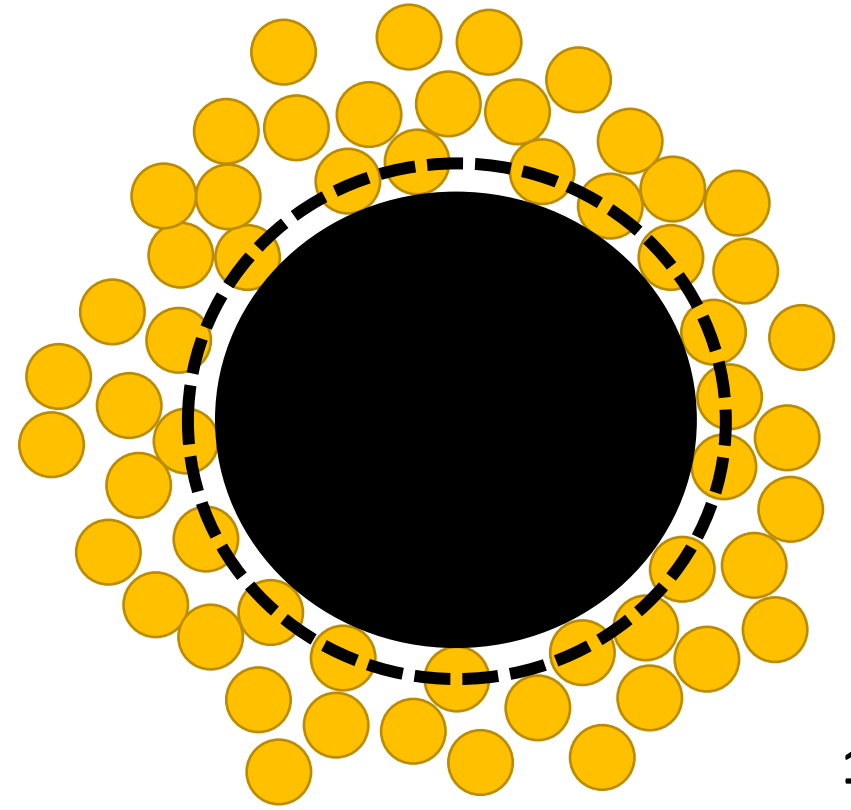
Silver coated PMMA in cornstarch suspension



## CST to DST transition: Excluded volume effect

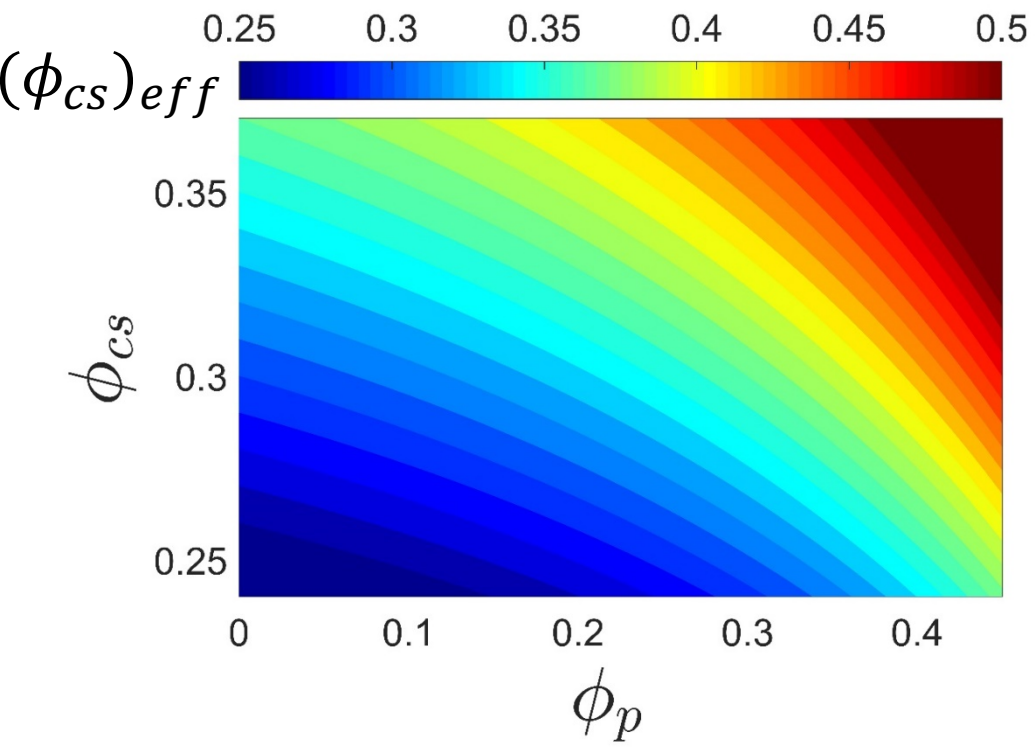
- The center of cornstarch grains cannot come closer than their radius to the surface of a large particle.
- We must subtract the volume of this shell, mostly composed of water from the volume of water in the rest of the suspension.
- There is an increase of cornstarch concentration outside of these shells.
- local cornstarch volume fraction decreases at the large particle surface, compensated by increase elsewhere

The effective cornstarch volume fraction:  $(\phi_{cs})_{eff} = f(\phi_{cs}, \phi_p, d_p)$

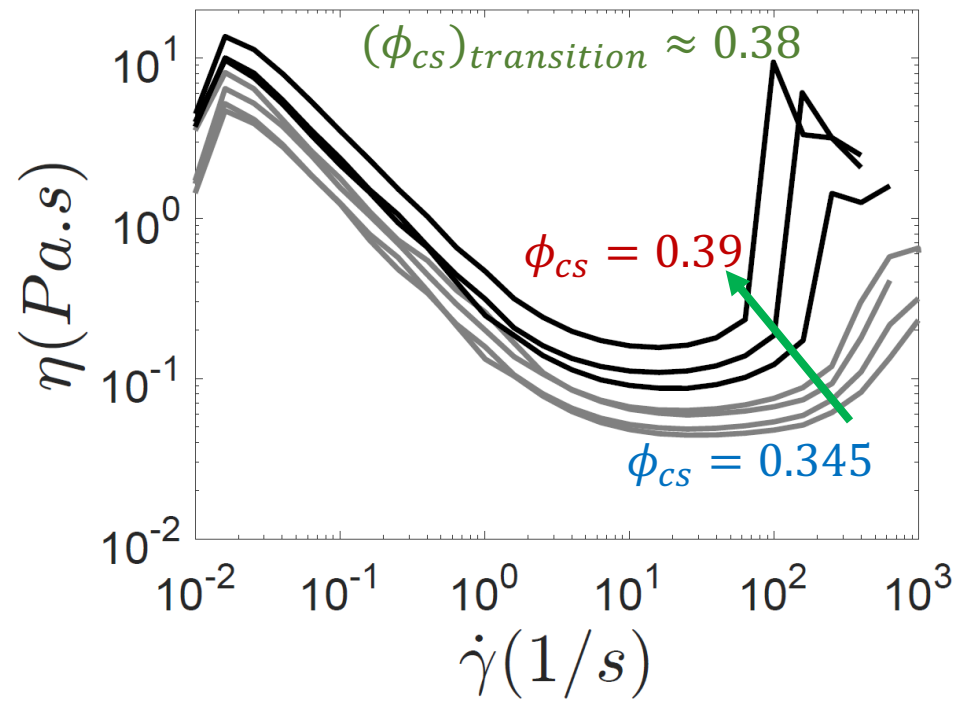
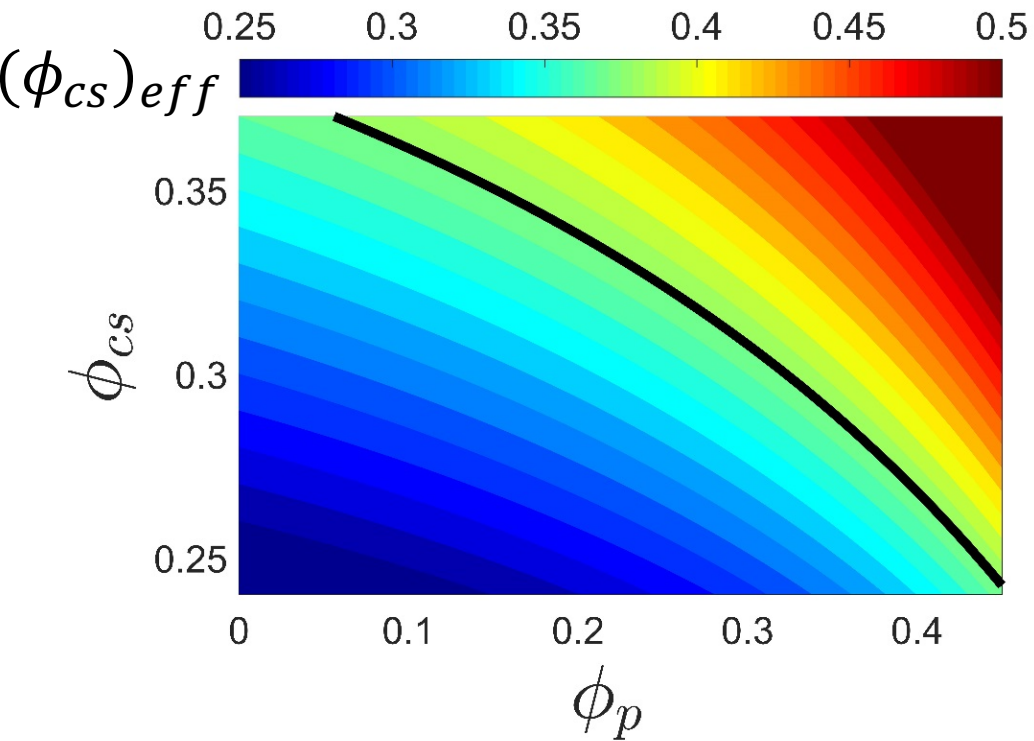




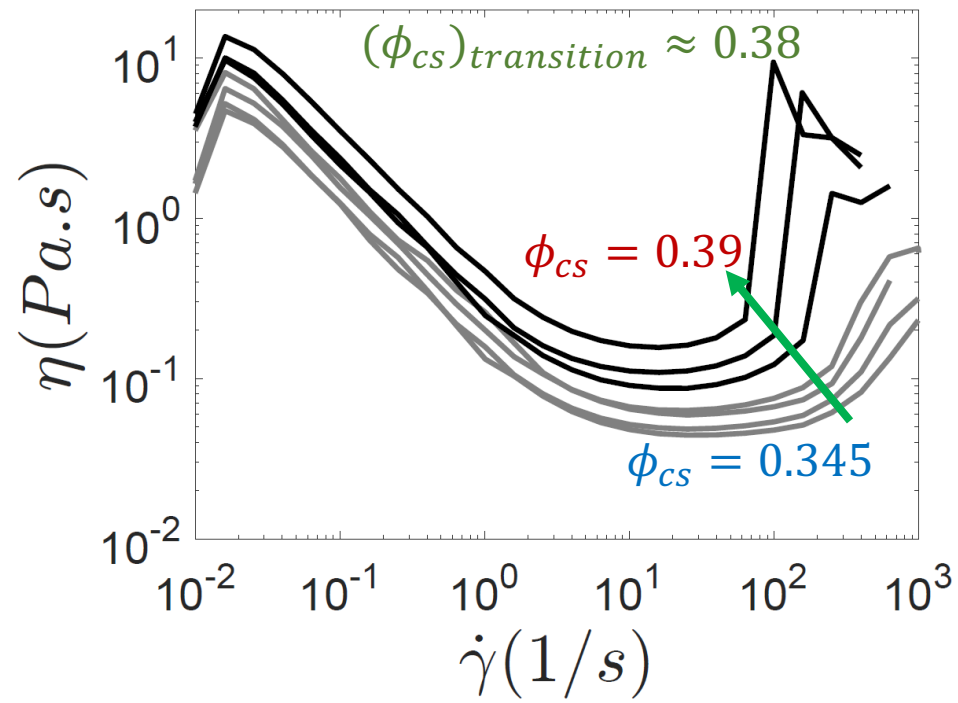
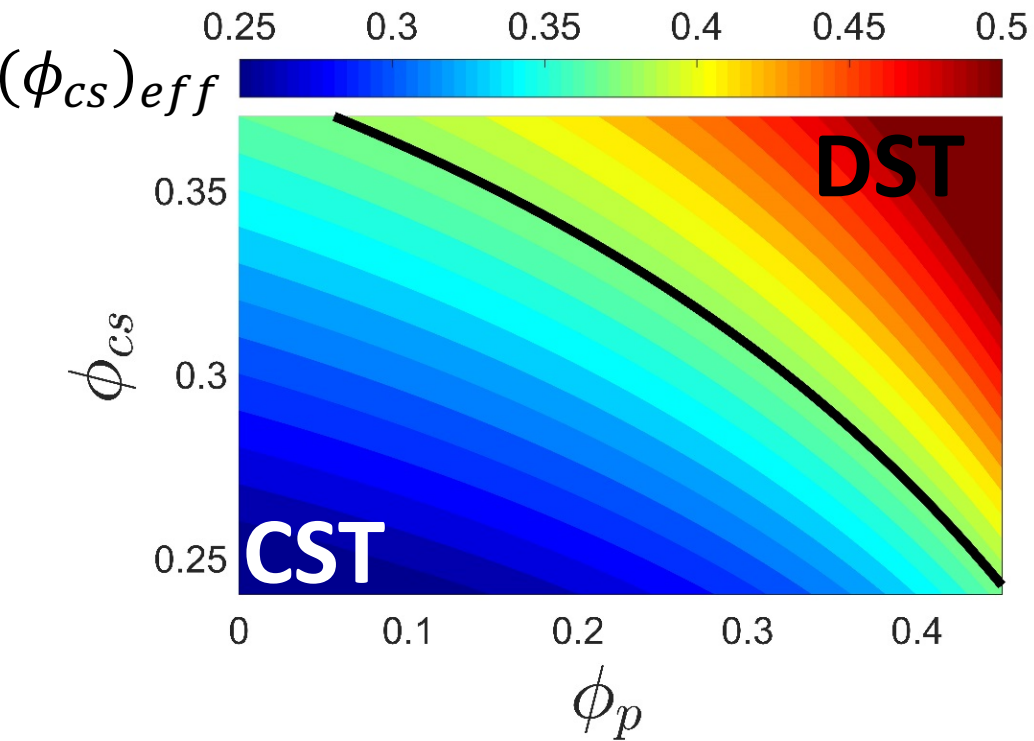
# Excluded volume effect: Plotting the $(\phi_{cs})_{eff} = f(\phi_{cs}, \phi_p, d_p)$



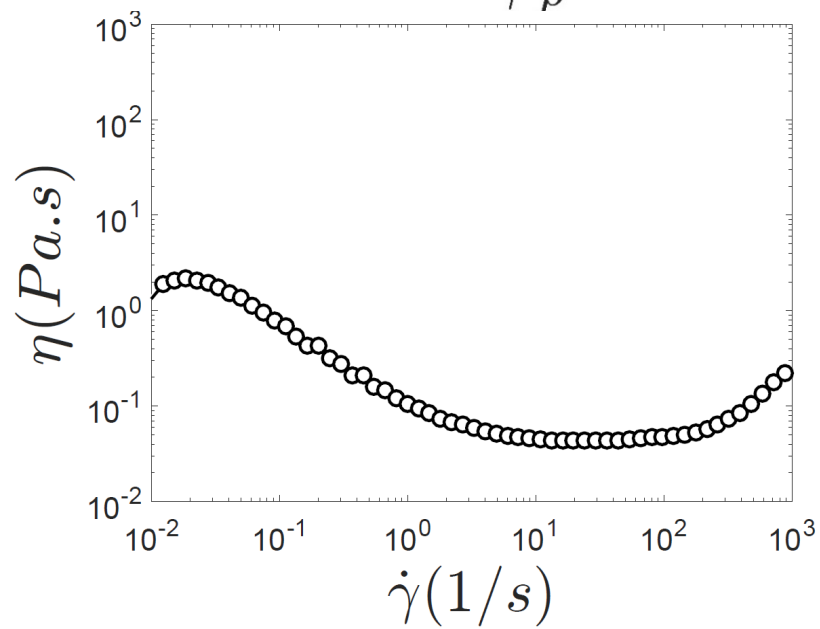
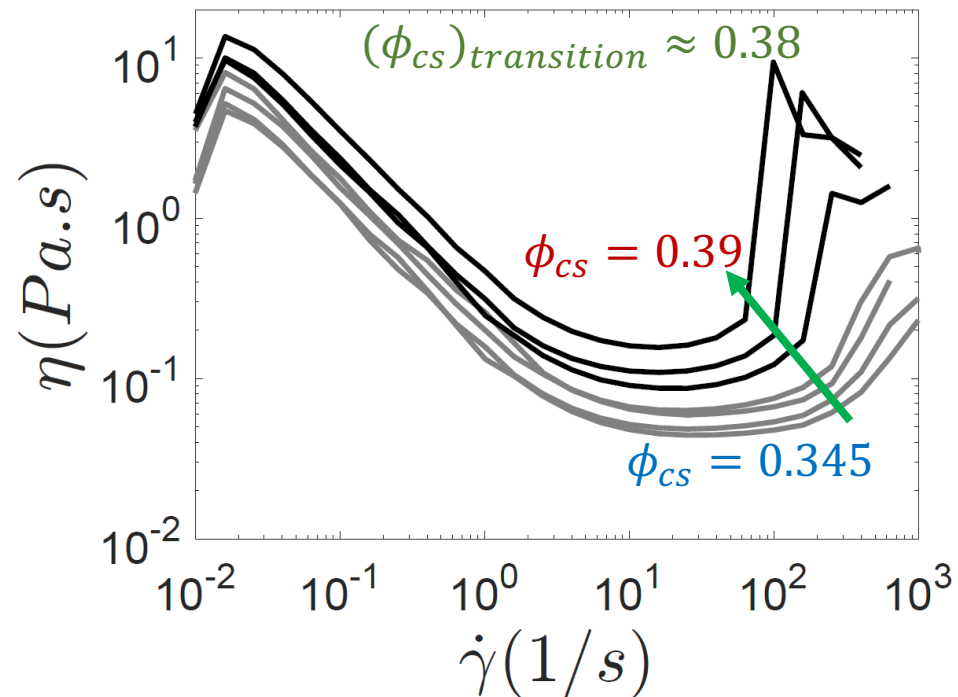
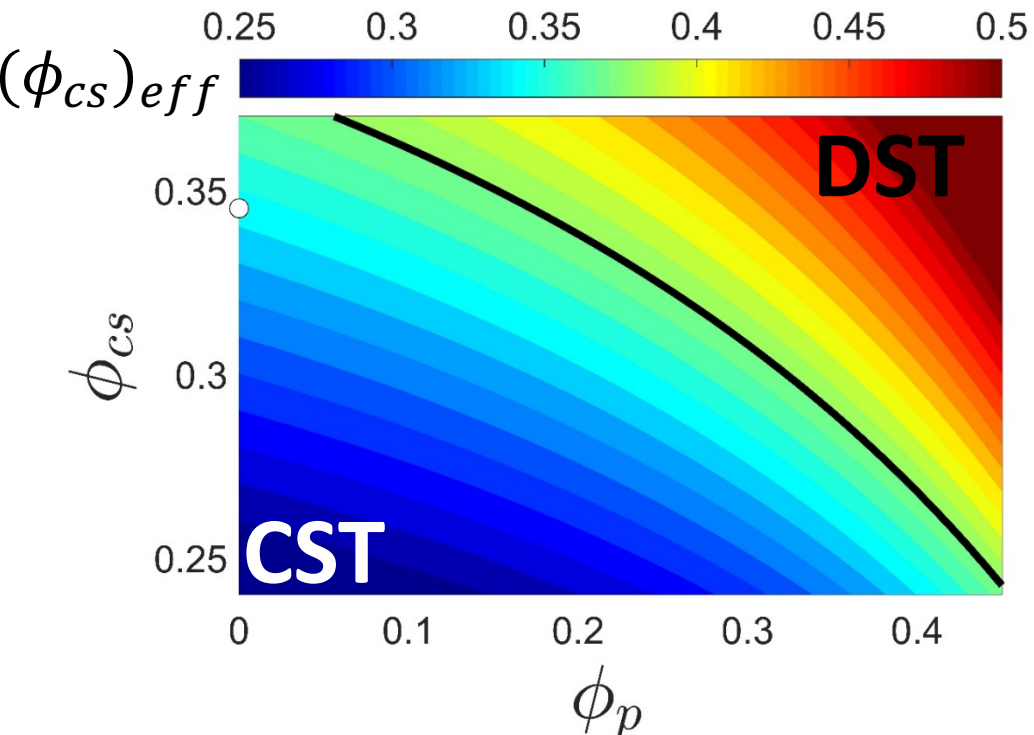
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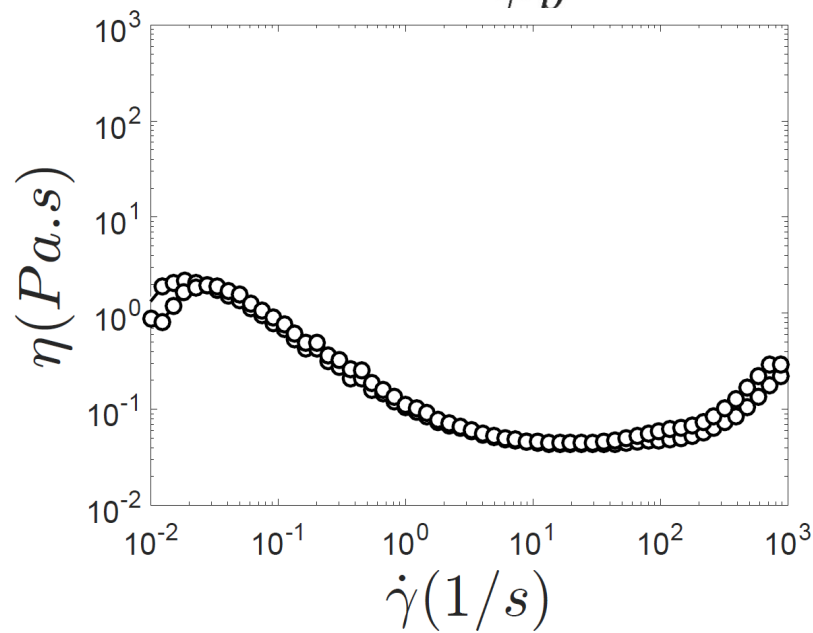
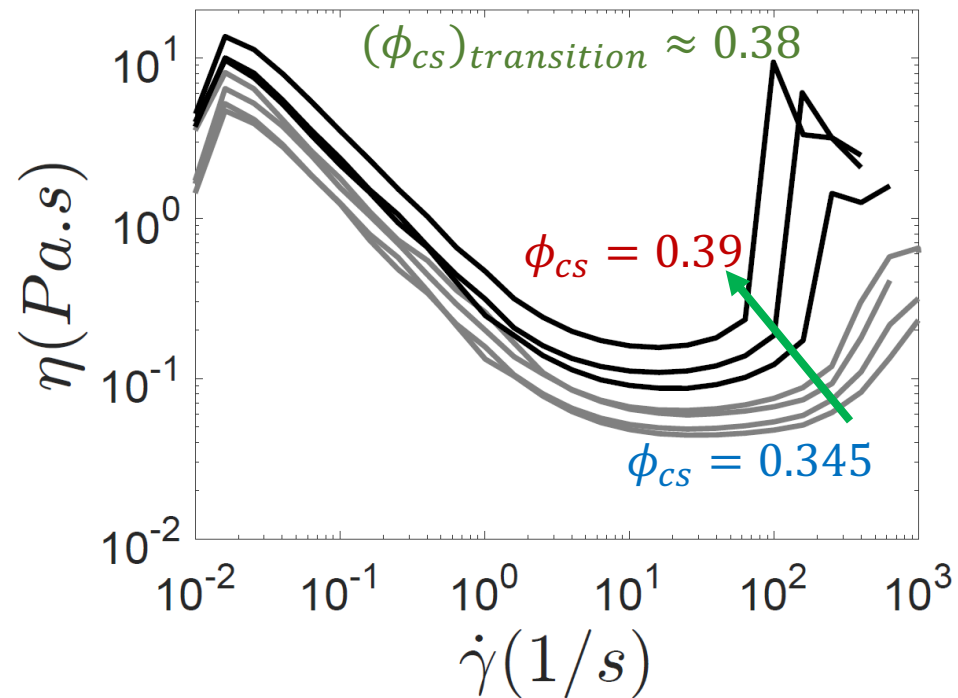
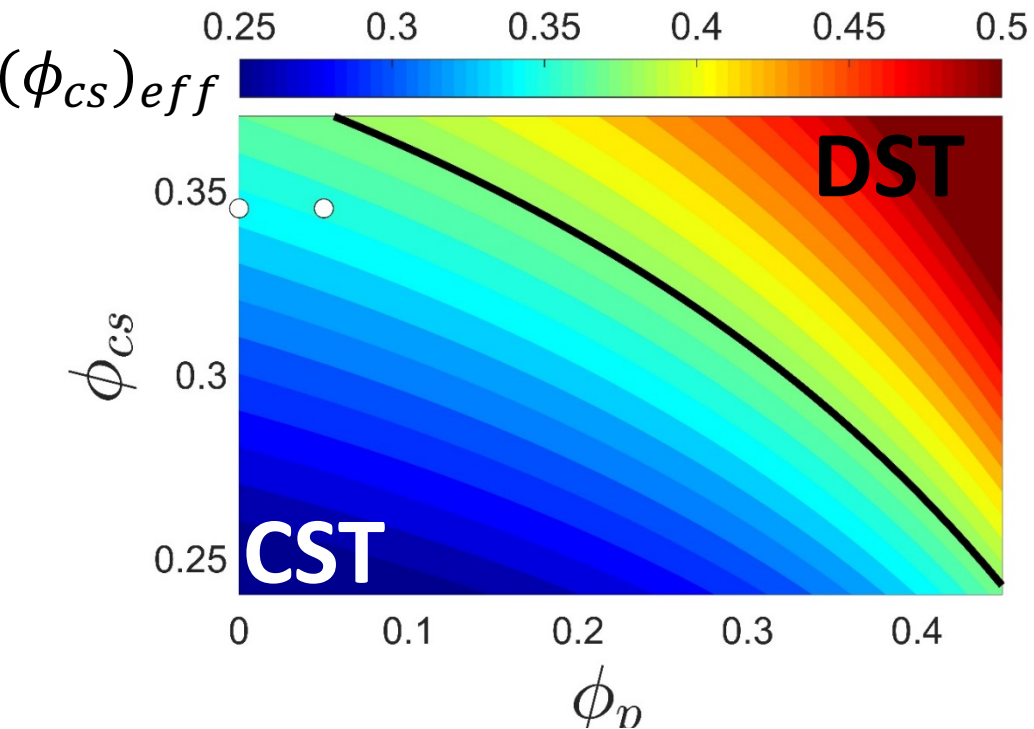


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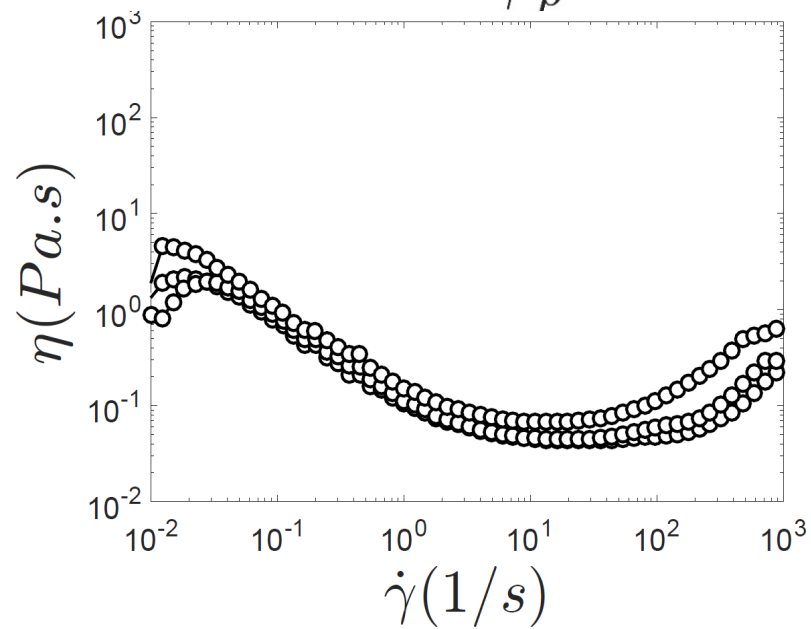
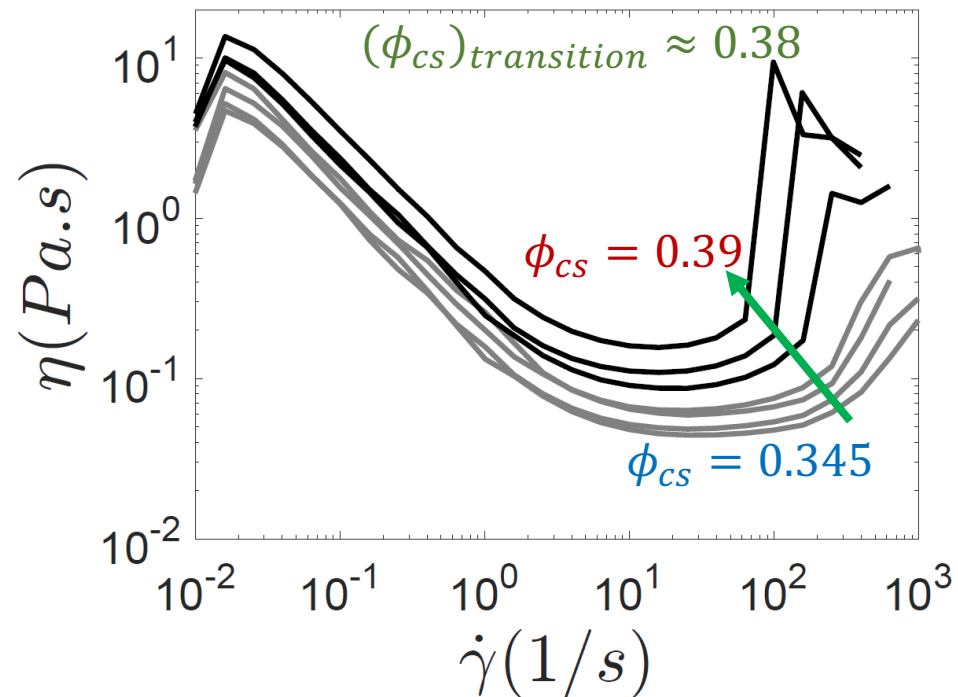
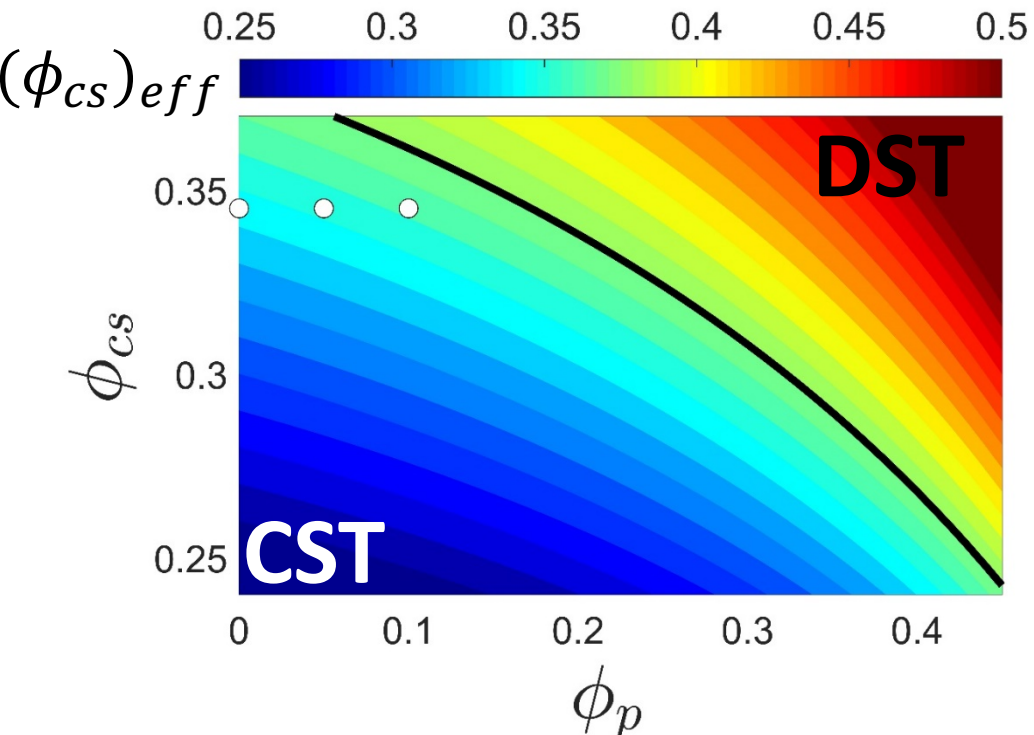
$\phi_{cs} = 34.5\%$   
 Silver Coated PMMA  
 $d_p = 106 - 125 \mu m$

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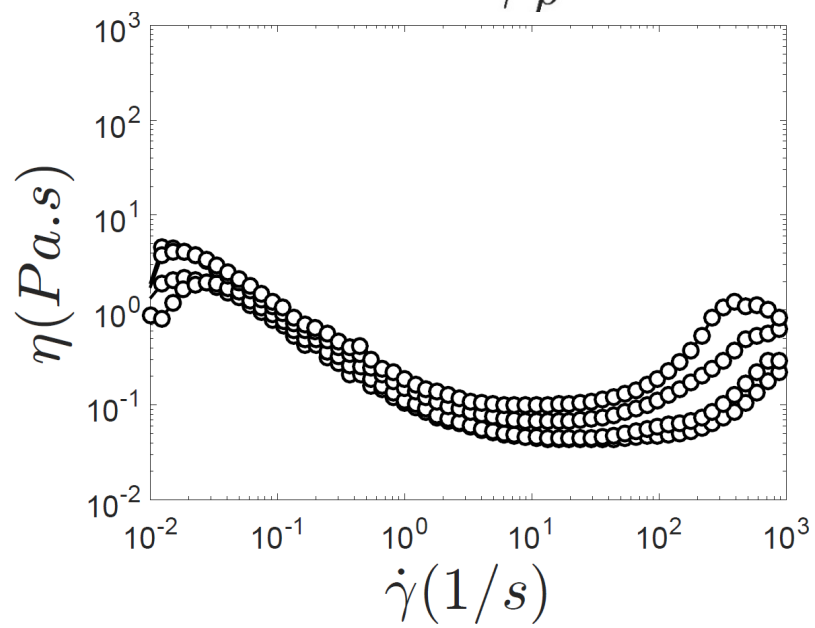
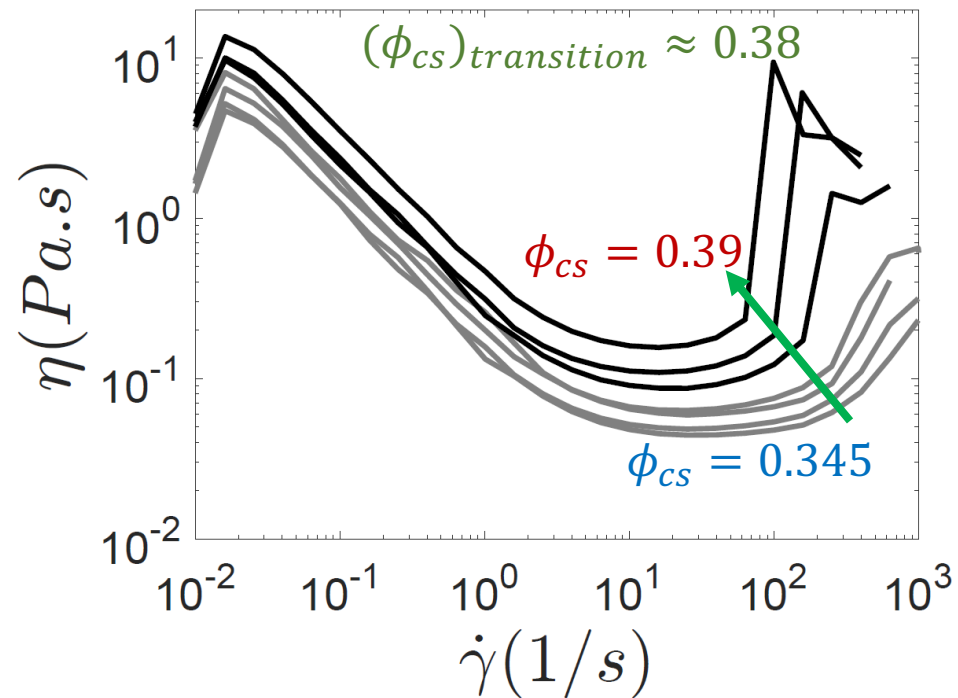
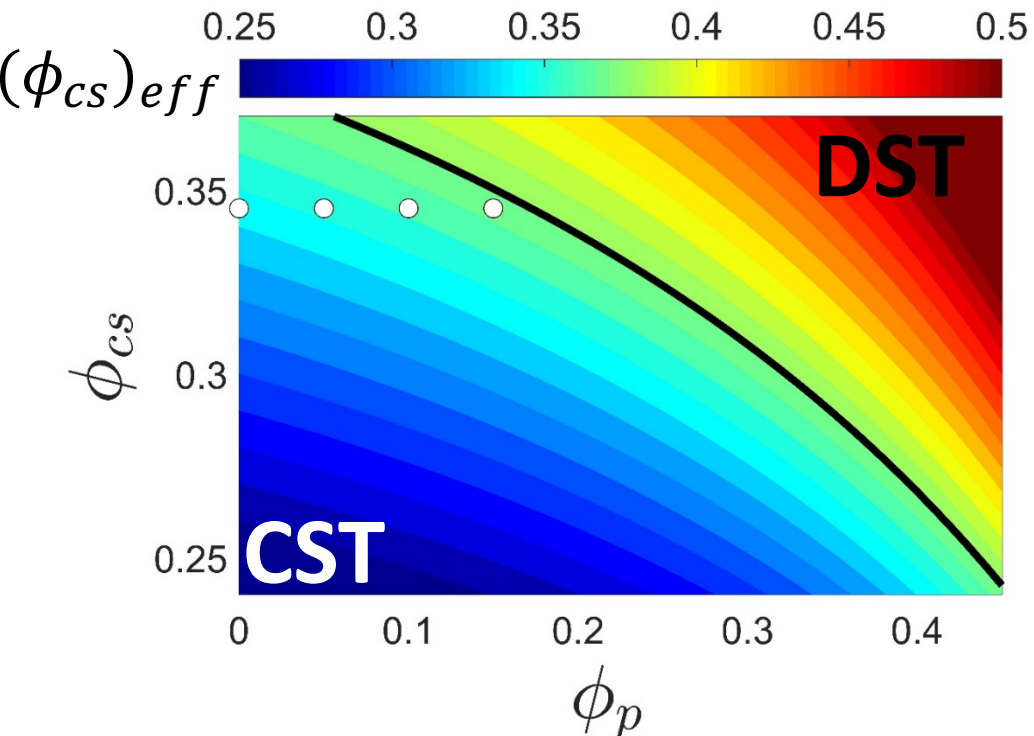
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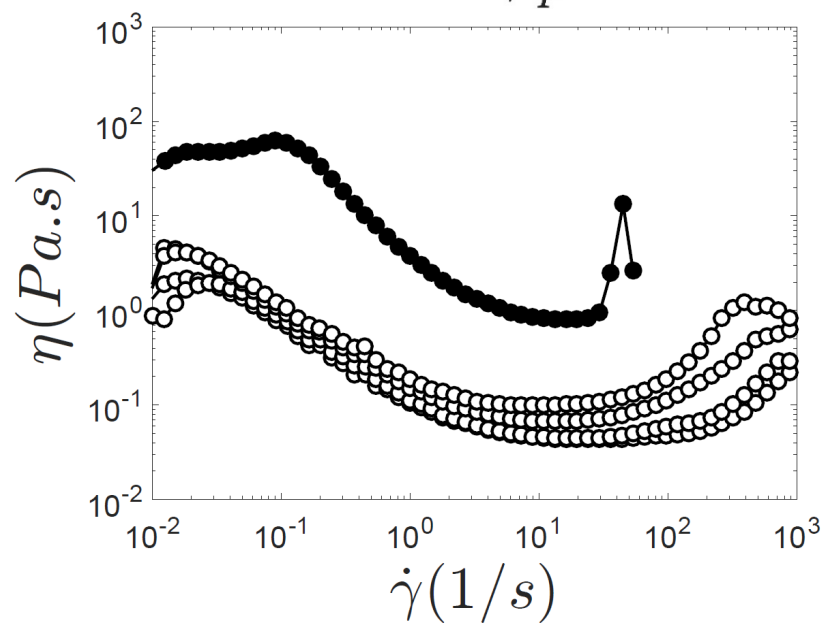
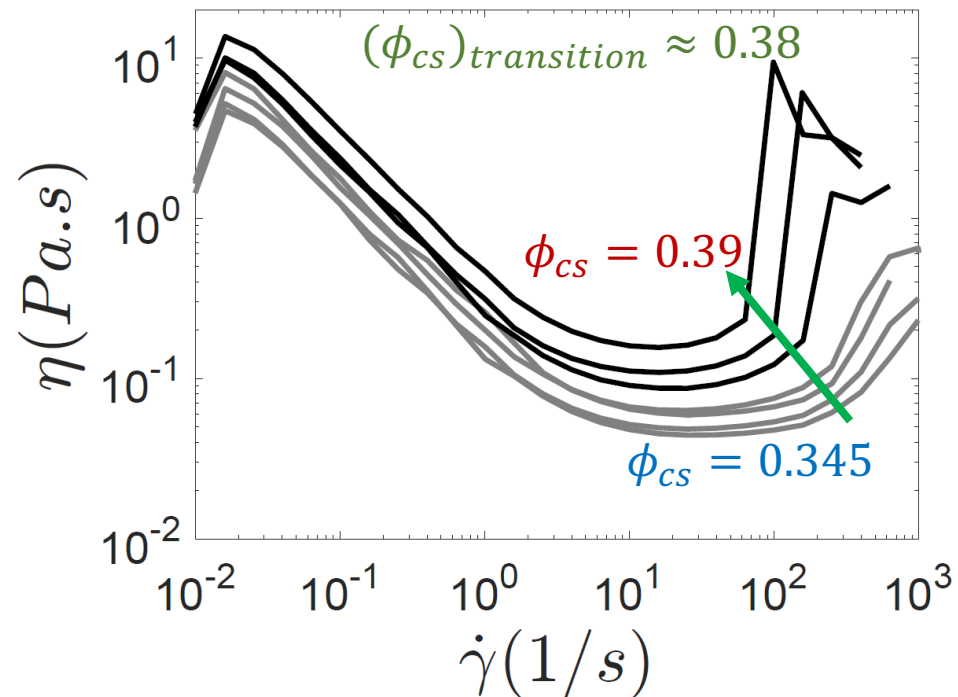
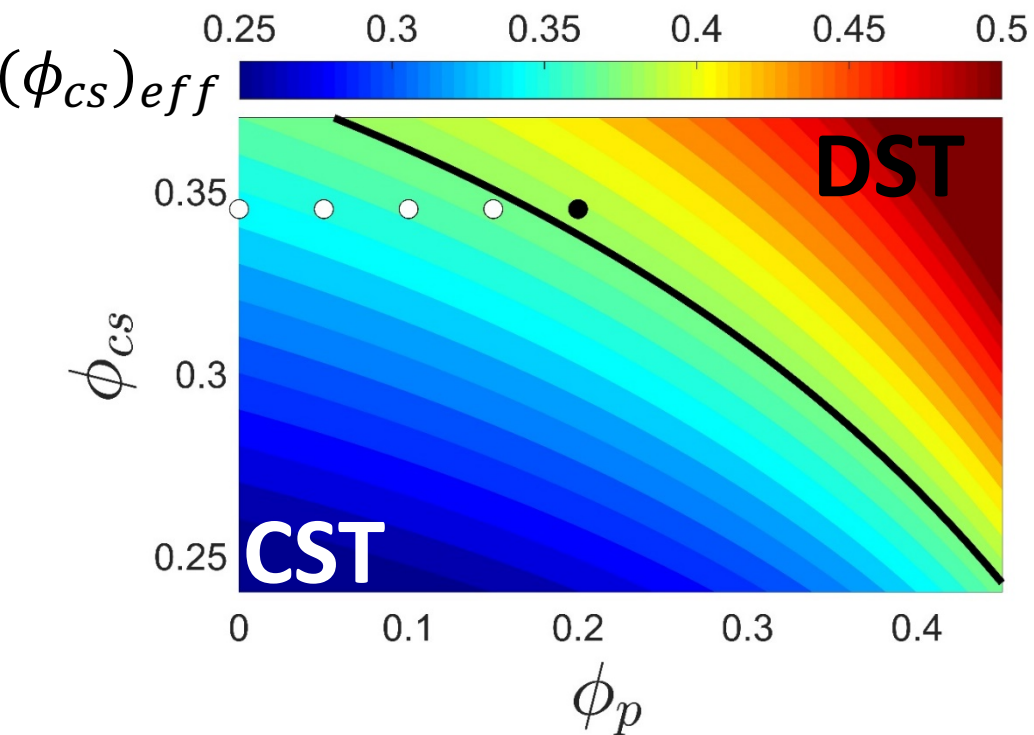


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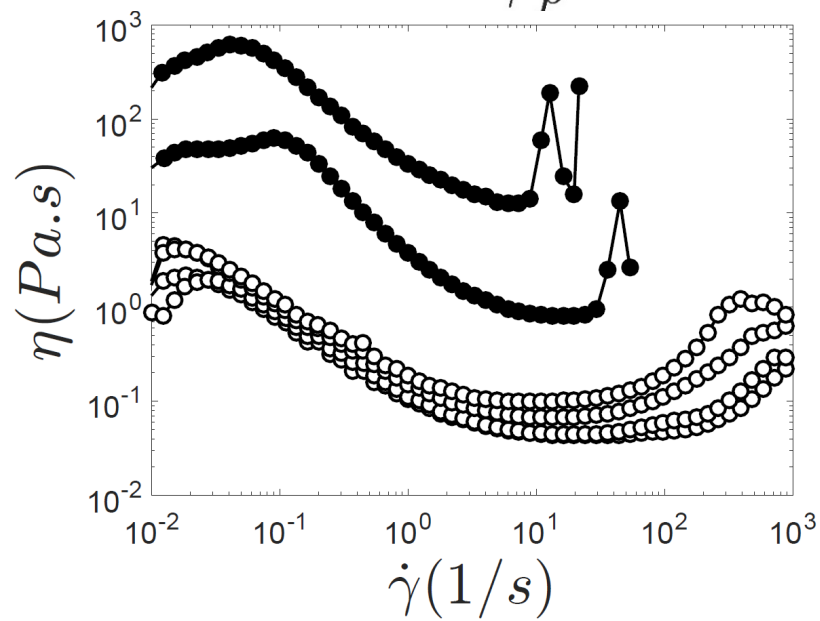
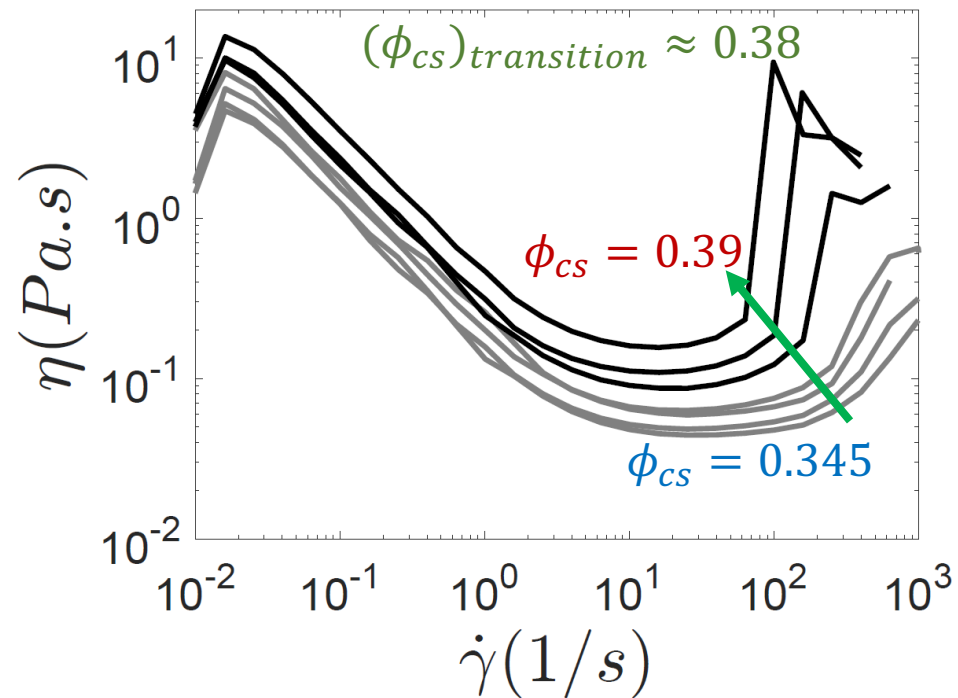
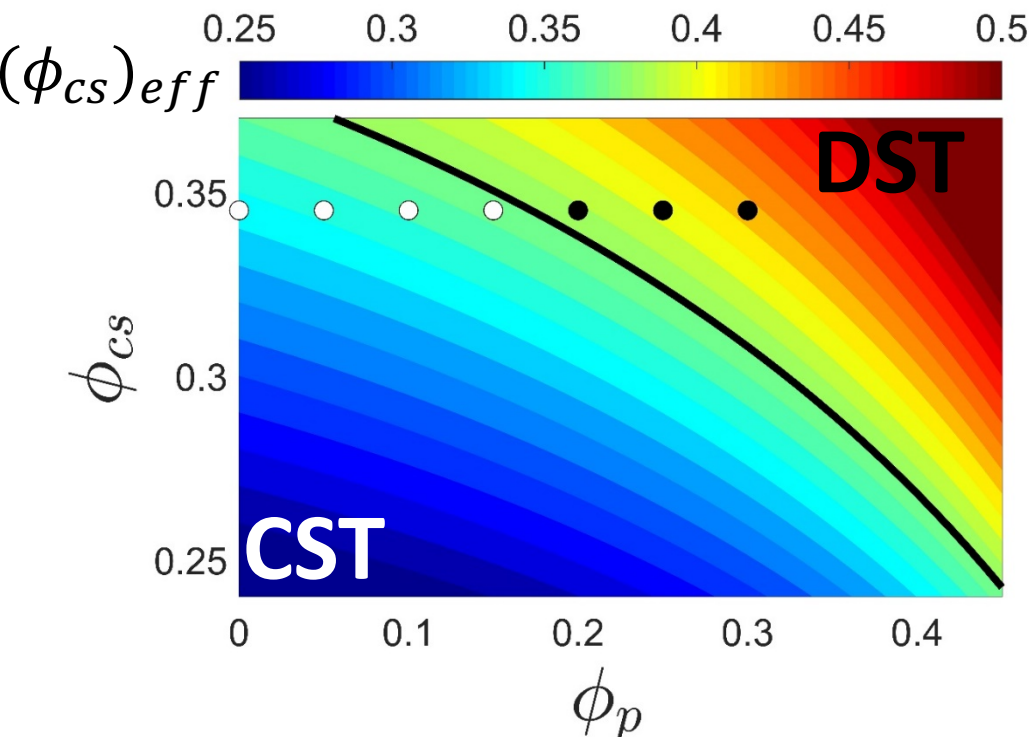
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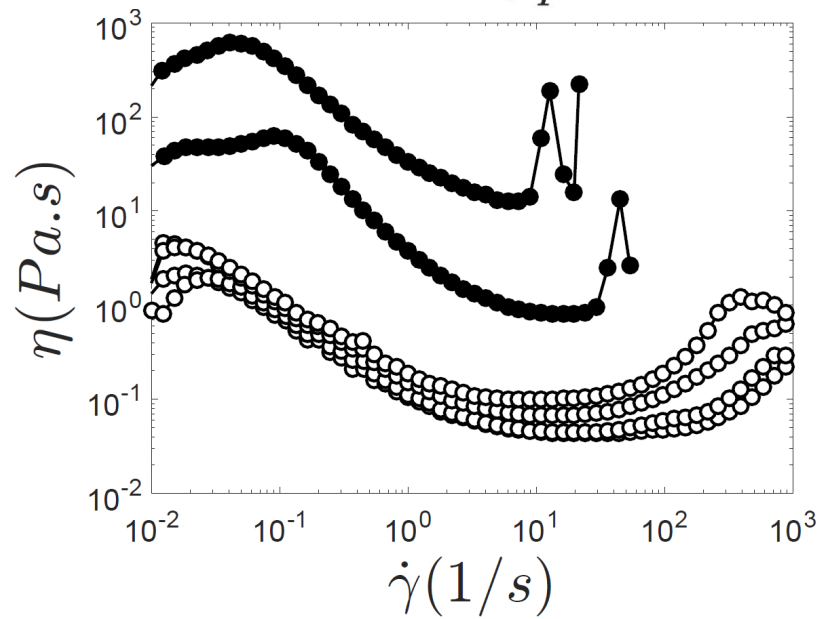
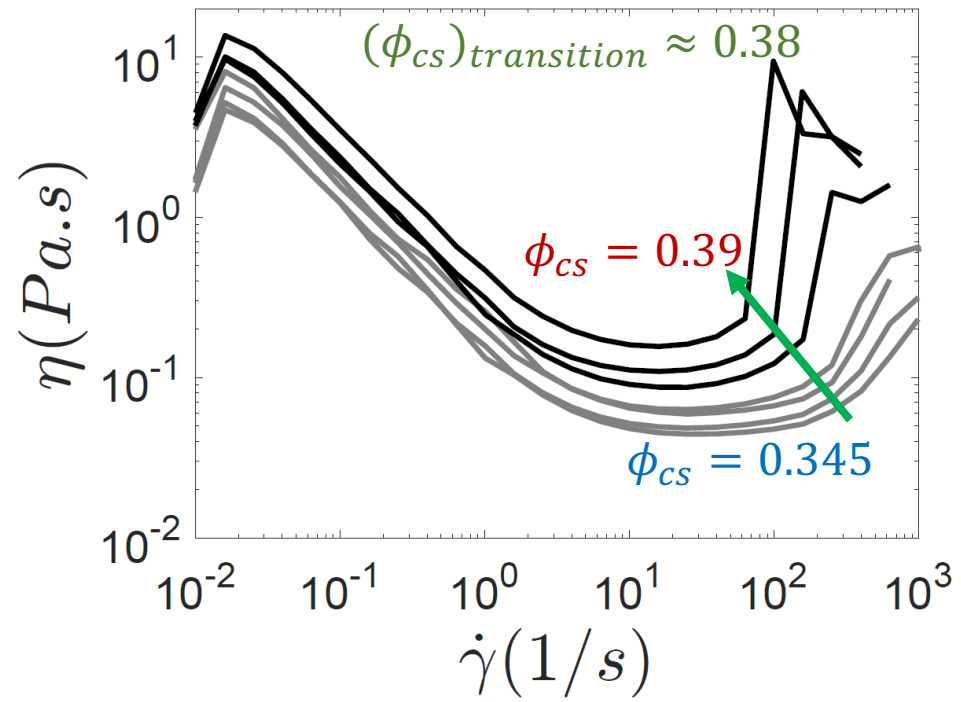
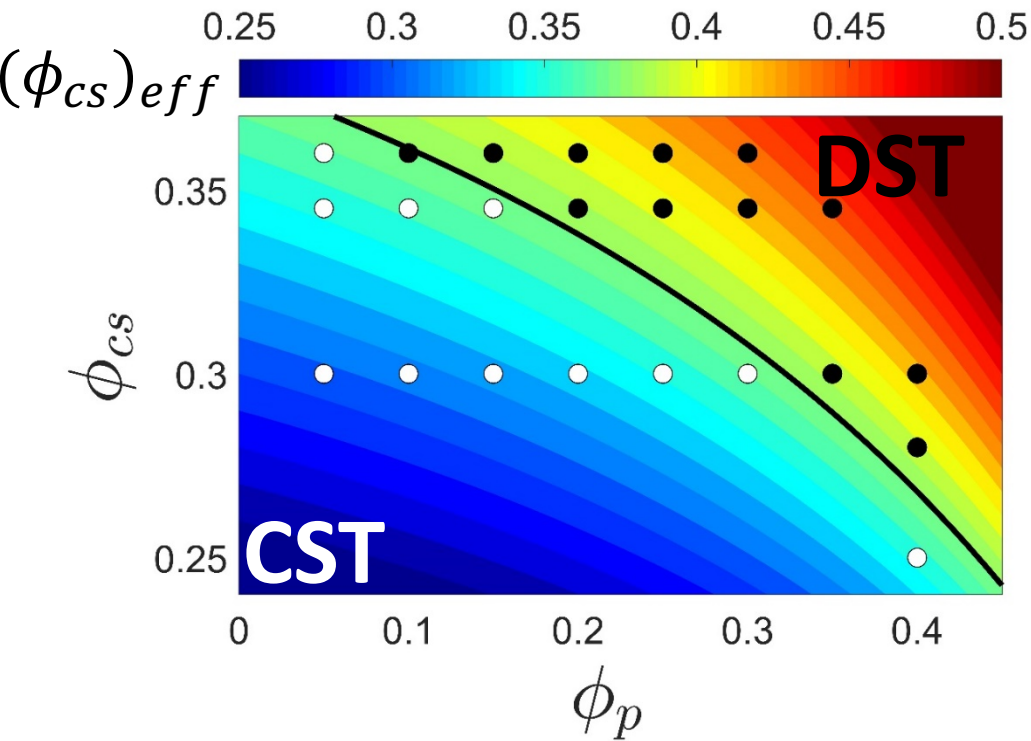


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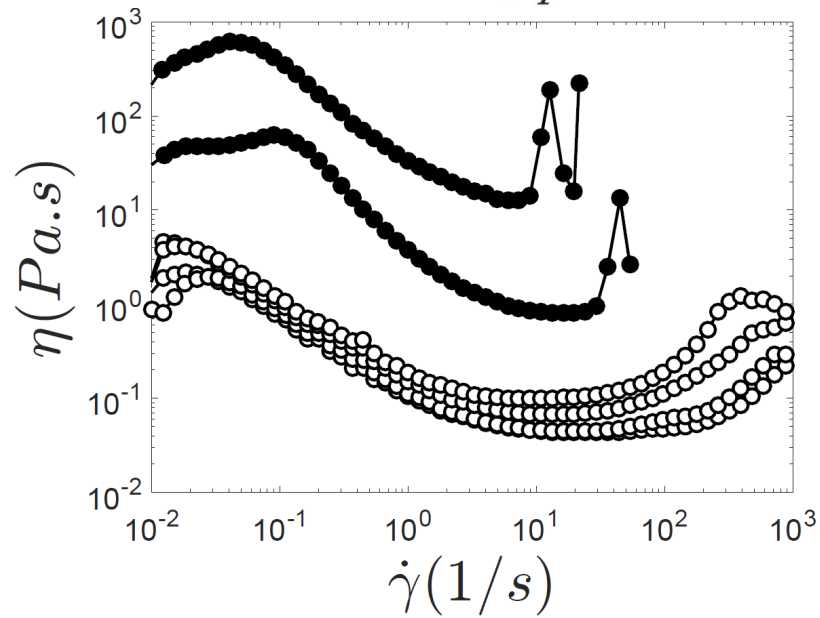
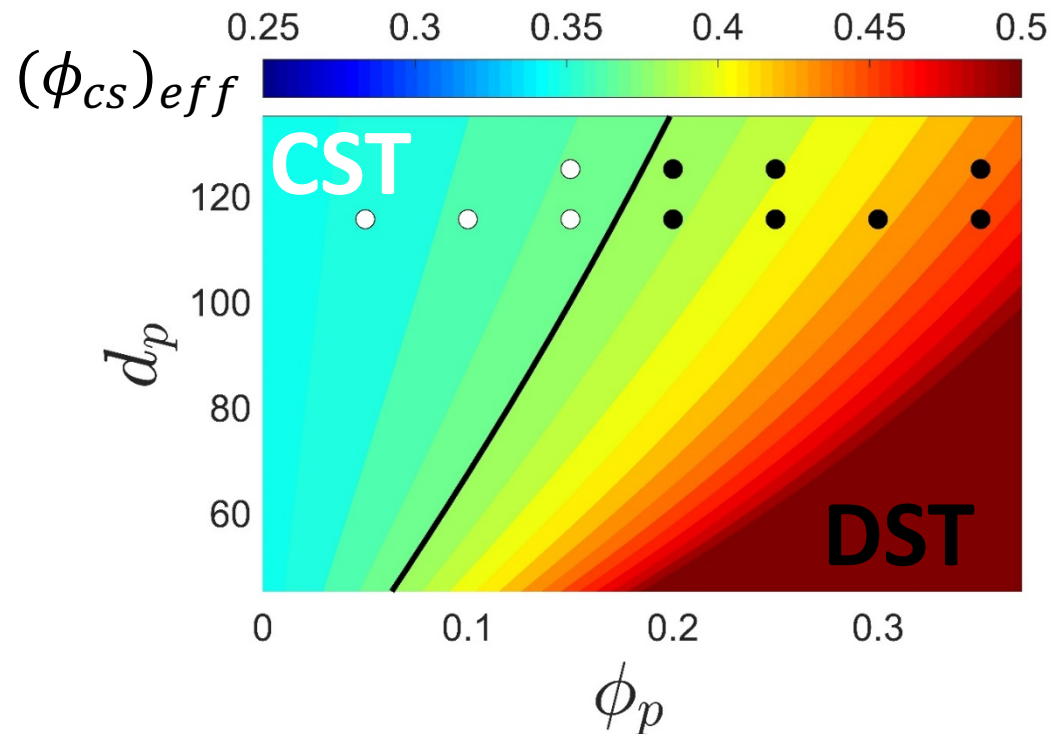
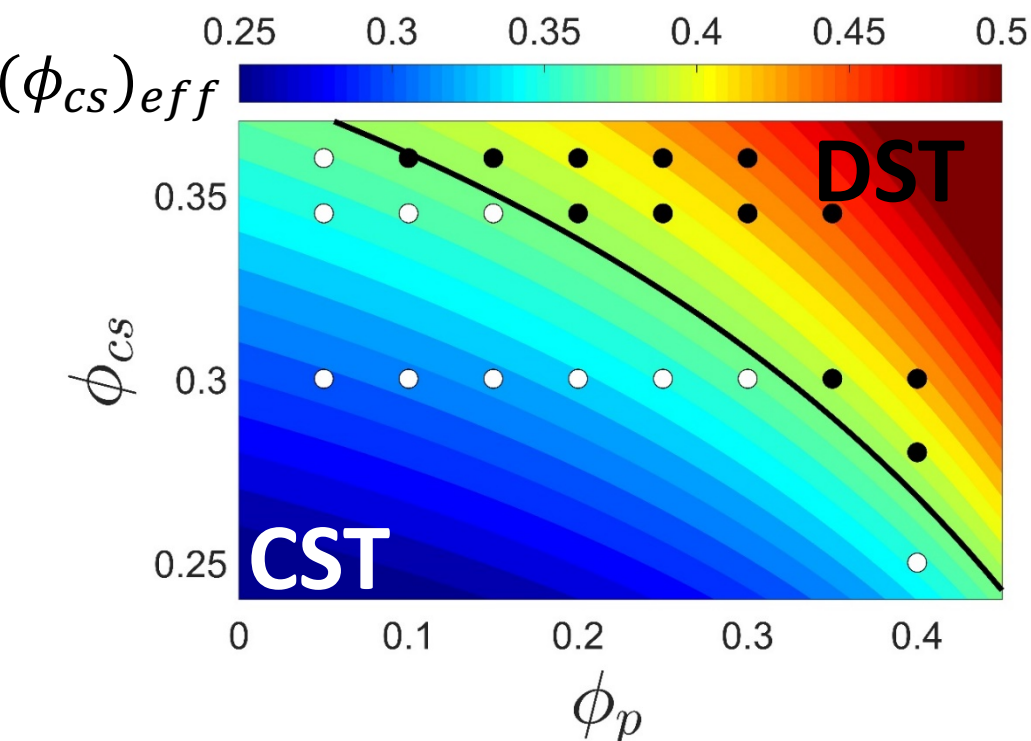
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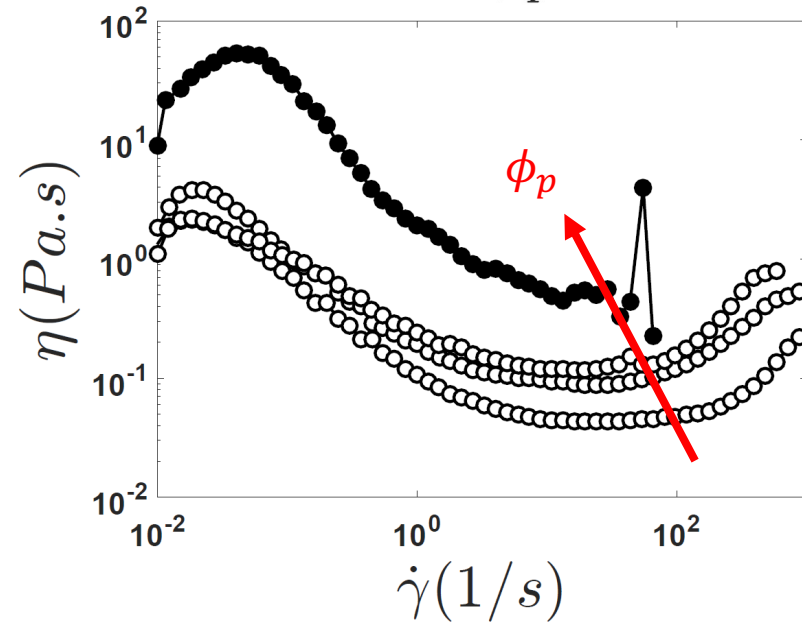


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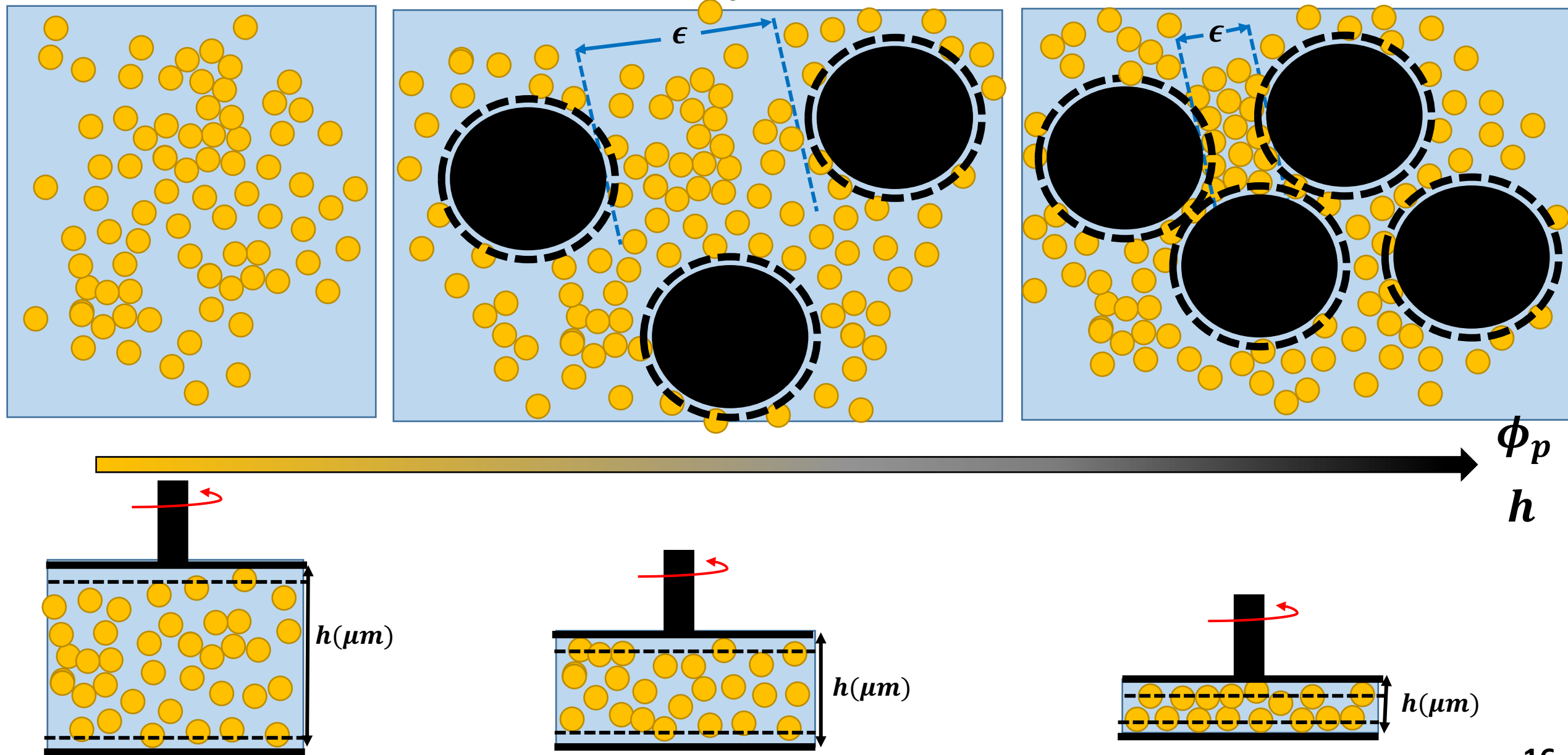


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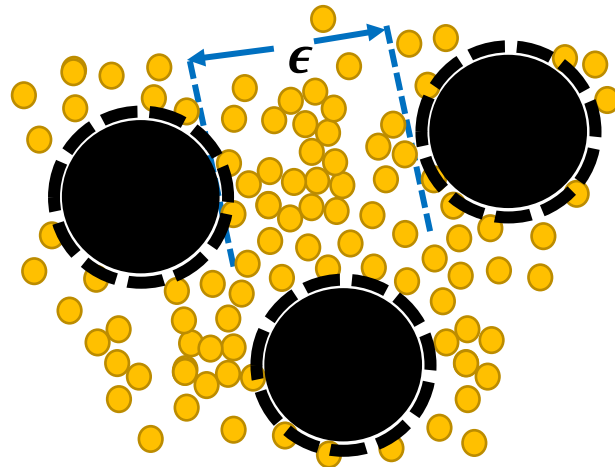
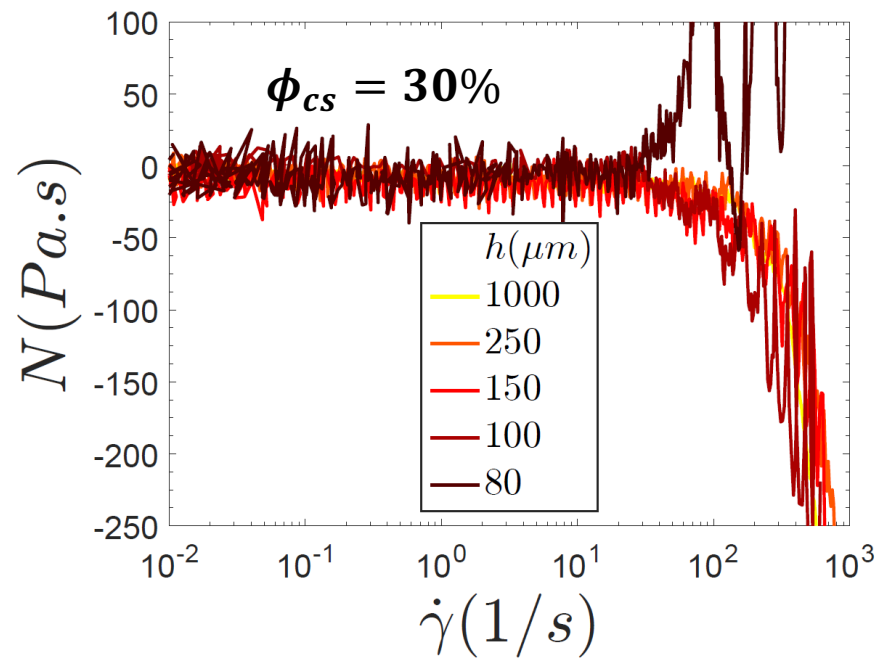
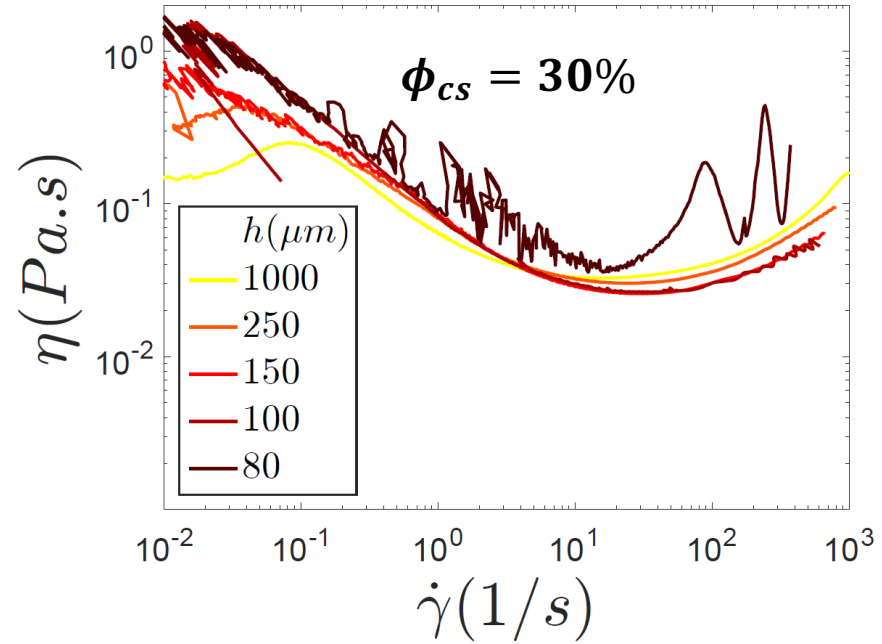


$\phi_{cs} = 34.5\%$   
 $d_p = 125 - 135 \mu\text{m}$   
Silver Coated PMMA

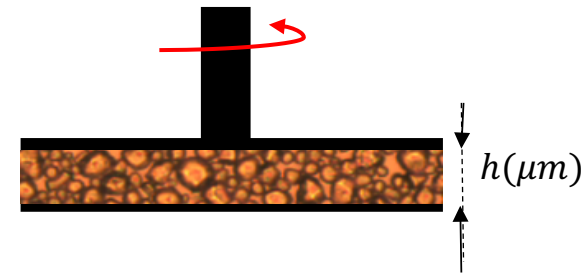
# Simulating the excluded volume effect via rheometry of the pure cornstarch suspension



# Simulating the excluded volume effect via rheometry of the pure cornstarch suspension



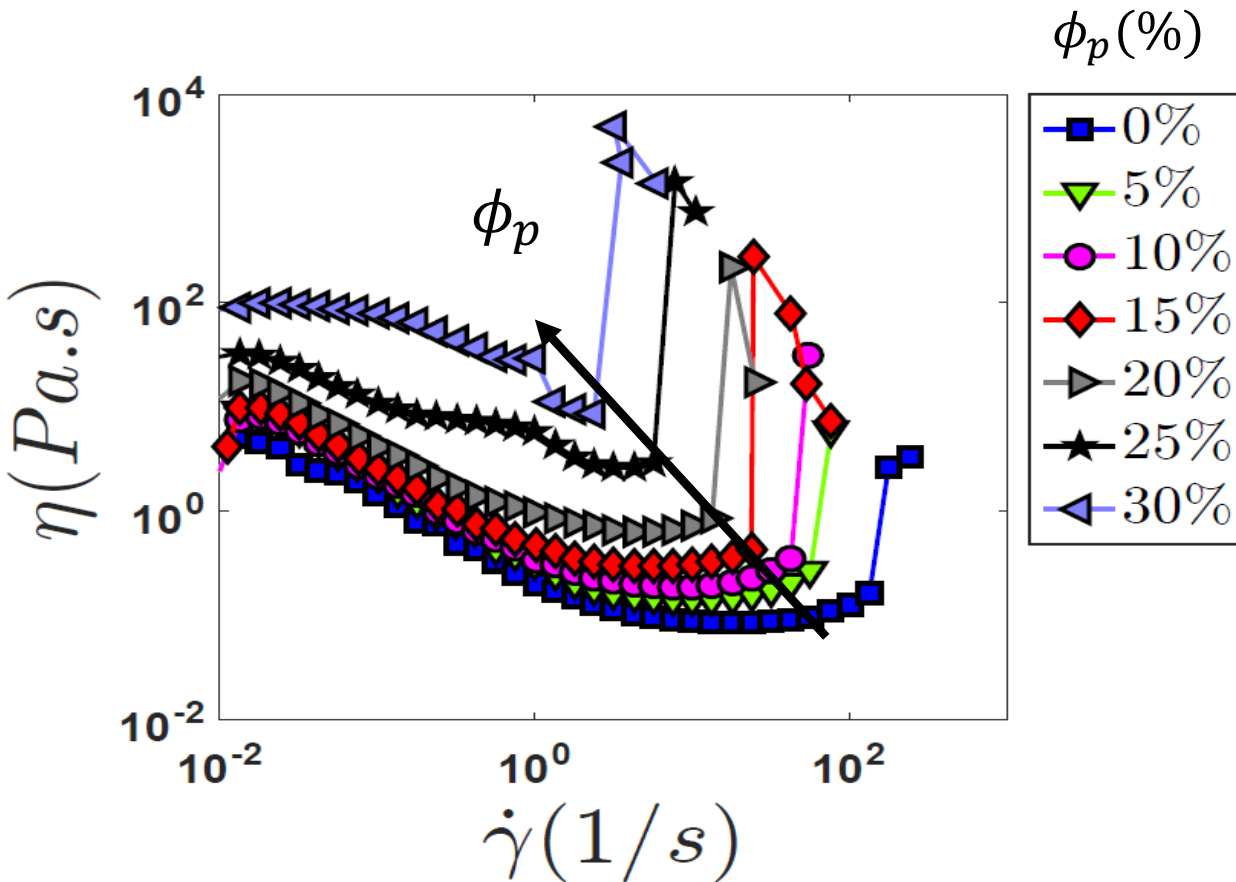
Simulated as:



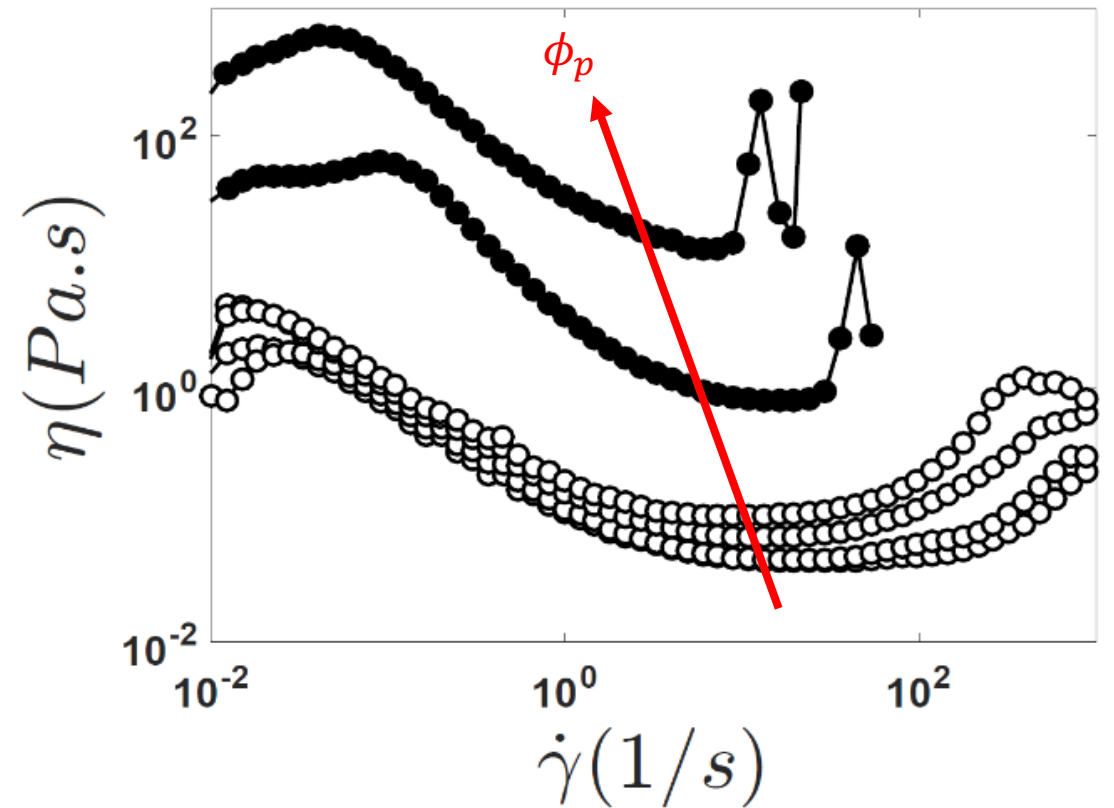


# Summary

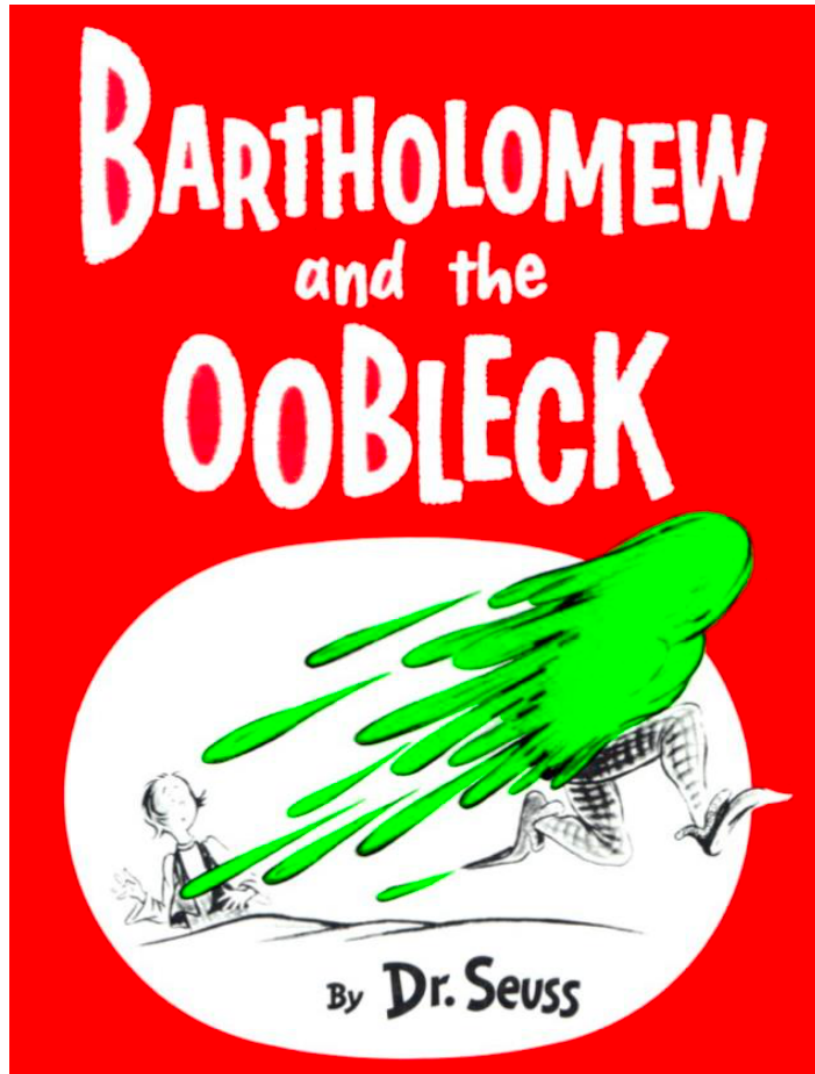
- By increasing large particles volume fraction in a DST suspension shear thickening enhances, specifically DST, can be explained via a scaling argument!



- By adding high volume fractions of large particles to a CST suspension, the suspension transitions from CST to DST, can be explained via an excluded volume effect.



# Thank You!



Bartholomew and the Oobleck is a 1949 book by Dr. Seuss. It follows the adventures of a young boy named Bartholomew, who must rescue his kingdom from a sticky substance called Oobleck!